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Thermodynamic and structural determinants of calcium-independent interactions of Calmodulin

Michael Dennis Feldkamp
University of Iowa

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THERMODYNAMIC AND STRUCTURAL DETERMINANTS OF CALCIUM-
INDEPENDENT INTERACTIONS OF CALMODULIN

by

Michael Dennis Feldkamp

An Abstract

Of a thesis submitted in partial fulfillment
of the requirements for the Doctor of
Philosophy degree in Biochemistry
in the Graduate College of
The University of Iowa

July 2010

Thesis Supervisor: Professor Madeline A. Shea

ABSTRACT

Calmodulin (CaM) is an essential protein found in all eukaryotes ranging from vertebrates to unicellular organisms such as *Paramecia*. CaM is a calcium sensor protein composed of two domains (N and C) responsible for the regulation of numerous calcium-mediated signaling pathways. Four calcium ions bind to CaM, changing its conformation and determining how it recognizes and regulates its cellular targets. Since the discovery of CaM, most studies have focused on the role of its calcium-saturated form.

However, an increasing number of target proteins have been discovered that preferentially bind apo (calcium-depleted) CaM. My study focused on understanding how apo CaM recognizes drugs and protein sequences, and how those interactions differ from those of calcium-saturated CaM. I have used spectroscopic methods to explore CaM binding the drug Trifluoperazine (TFP) and the IQ-motif of the type 2 Voltage-Dependent Sodium Channel ($\text{Na}_v1.2_{\text{IQp}}$). These studies have shown that both TFP and $\text{Na}_v1.2_{\text{IQp}}$ preferentially bind to the “semi-open” conformation of apo CaM.

TFP was shown to be an unusual allosteric effector of calcium binding to CaM. Using ^{15}N -HSQC NMR spectroscopy, I determined the stoichiometry of TFP binding to apo CaM to be 2:1 and to $(\text{Ca}^{2+})_4\text{-CaM}$ to be 4:1 TFP:CaM. That difference in stoichiometry determined whether TFP decreased or increased the affinity of CaM for calcium. Analysis of residue-specific chemical shift differences indicated that TFP binding to apo and $(\text{Ca}^{2+})_4\text{-CaM}$ perturbed the C-domain more than the N-domain, prompting high-resolution structural studies of the isolated C-domain of CaM.

Crystallographic studies of TFP bound to a calcium-saturated C-domain fragment of CaM (CaM_{76-148}) revealed that CaM adopted an “open” tertiary conformation. The unit cell contained two protein and 4 drug molecules. The orientation of TFP revealed that its trifluoromethyl group was found in two alternative positions (one in each protein in the unit cell), and that Met 144 acted as a gatekeeper to select the orientation of TFP.

In contrast to TFP binding to the “open” conformation of calcium-saturated CaM₇₆₋₁₄₈, my NMR studies showed that TFP bound the “semi-open” conformation of apo CaM₇₆₋₁₄₈. TFP interacted with CaM residues near the perimeter of the hydrophobic pocket, but did not contact residues that are solvent-accessible only in the “open” form. Allosteric effects due to TFP binding were observed in the calcium-binding loops of apo CaM₇₆₋₁₄₈. These properties suggest that TFP may antagonize interactions between apo CaM and target proteins such as ion channels that preferentially bind apo CaM.

Na_v1.2, is responsible for the passage of Na⁺ ion across cellular membranes. Apo binding of CaM to Na_v1.2 poises it for action upon calcium release in the cell. My NMR studies of CaM binding to the Na_v1.2 IQ-motif sequence (Na_v1.2_{IQp}) showed that the C-domain of apo CaM was necessary and sufficient for binding. My high-resolution structure of the isolated C-domain of CaM bound to Na_v1.2_{IQp} revealed that the domain adopted a “semi-open” conformation. At the interface between the IQ-motif and CaM, the highly conserved I and two Y residues of Na_v1.2_{IQp} interacted with hydrophobic residues of CaM, while the invariant Q residue interacted with residues in the loop between helices F and G of CaM. This is the first CaM-IQ complex to be determined by NMR; the only other available structure of apo CaM bound to an IQ-motif was determined crystallographically.

To accomplish its regulatory roles in response to cellular Ca²⁺ fluxes, CaM has evolved multiple binding interfaces that are allosterically linked to its Ca²⁺-ligation state. My studies of CaM binding to TFP and Na_v1.2 demonstrate the versatility of CaM functioning as a regulatory protein comprised of domains having separable functions.

Abstract Approved: _____
Thesis Supervisor

Title and Department

Date

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Thesis Supervisor: Professor Madeline A. Shea

Graduate College
The University of Iowa
Iowa City, Iowa

CERTIFICATE OF APPROVAL

PH.D. THESIS

This is to certify that the Ph.D. thesis of

Michael Dennis Feldkamp

has been approved by the Examining Committee
for the thesis requirement for the Doctor of Philosophy
degree in Biochemistry at the July 2010 graduation.

Thesis Committee: _____
Adrian Elcock, Thesis Chair

Ernesto Fuentes

Lei Geng

Shahram Khademi

Daniel Weeks

For those who value truth more than dogma

To laugh often and much; to win the respect of intelligent people and the affection of children; to earn the appreciation of honest critics and endure the betrayal of false friends; to appreciate beauty; to find the best in others; to leave the world a bit better, whether by a healthy child, a garden patch or a redeemed social condition; to know even one life has breathed easier because you have lived. This is to have succeeded.

-Ralph Waldo Emerson

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LIST OF ABBREVIATIONS

APBS	Adaptive Poisson-Boltzmann Solver
BAA	Basic Amphipathic Alpha-helix
CaMBD	Calmodulin binding domain
CaMKII	CaM Kinase II (CaMKII)
DPD	N-(3,3,-Diphenylpropyl-N'-[1-R-(2 3,4-Bis-Butoxyphenyl)-Ethyl]- Propylenediamine
EGTA	Ethylene Glycol bis(2 (aminoethylether)=N,N',N',N' Tetraacetic Acid
FRET	Fluorescence Resonance Energy Transfer
HPLC	High Pressure Liquid Chromotography
KAR-2	3"-(Beta-Chloroethyl)-2",4"-Dioxo-3, 5"-Spiro-Oxazolidino-4- Deacetoxy-Vinblastine
LPBE	Linearized Poisson-Boltzmann Equation
mCaM	Mammalian Calmodulin
MLCK	Myosin Light Chain Kinase
Nav1.2	Voltage-Dependent Sodium Channel type 2
Nav1.2 _{BAA}	BAA-motif of Nav1.2
Nav1.2 _{v1.2IQP}	Voltage-Dependent Sodium Channel type 2 IQ-motif
NOE	Nuclear Overhauser Effect
NOESY	Nuclear Overhauser Effect Spectroscopy
PCaM	<i>Paramecium</i> Calmodulin
RMSD	Root Mean Standard Deviation
SAXS	Small Angle x-ray Scattering
SPAN	Value determined by subtracting minimum data value from maximum data value, for use in normalization of data

T2	Transverse Relaxation
TFP	10-[3-(4-Methyl-Piperazin-1-yl)-Propyl]-2-Trifluoromethyl-10H-Phenothiazine
TFP	Trifluoperazine
TOCSY	Total Correlation Spectroscopy
W-7	N-(6-Aminohexyl)-5-Chloro-1-Naphthalenesulfonamide

CHAPTER I

INTRODUCTION

The Ca^{2+} ion plays a vital role as a secondary messenger where its spatial and temporal location can be tightly regulated to perform a myriad of regulatory functions. Upon signaling, Ca^{2+} can be released into the cytosol from both intracellular and extracellular stores where it is then sensed by a variety of Ca^{2+} -binding proteins. The most prominent of these Ca^{2+} -binding proteins is Calmodulin, which regulates over 300 target proteins in a Ca^{2+} -dependent manner (Yap et al., 2000). To accomplish its regulatory functions, CaM interacts with its protein targets using a variety of Ca^{2+} ligation states. The first CaM-dependent proteins identified were Ca^{2+} -dependent, and the conclusion was drawn (prematurely) that Ca^{2+} was required for target binding by CaM. Targets have since been identified that interact with either apo or partially Ca^{2+} -saturated CaM, representing new classes of proteins whose interaction with one or both domains of CaM is Ca^{2+} -independent. This thesis seeks to examine the structural and thermodynamic properties of apo CaM when interacting with the anti-psychotic drug Trifluoperazine (TFP) as well as a peptide that represents the CaM binding domain of the voltage dependent sodium channel 1.2 isoform ($\text{Na}_v1.2$).

CaM Background

CaM is a Ca^{2+} sensor protein that is essential to many eukaryotic signal transduction pathways. The CaM sequence (148 a.a.) is highly acidic (pI of 4) and divided between two homologous domains which are connected by a flexible 5-residue linker (**Figure 1.1**). Each domain binds two Ca^{2+} ions cooperatively in neighboring EF-Hand motifs, giving rise to a total of 4 bound Ca^{2+} ions per molecule of CaM (Chattopadhyaya et al., 1992; Crouch and Klee, 1980; Pedigo and Shea, 1995; Rao et al., 1993). Sequence comparison of EF-hand Ca^{2+} -binding sites reveals a high level of sequence conservation among residues that coordinate the Ca^{2+} ion via pentagonal

bipyramidal geometry (Strynadka and James, 1989; Wilson and Brunger, 2000; Yang et al., 2002) (**Figure 1.2**). The primary sequence of CaM is highly conserved among eukaryotes. *Paramecium* CaM (PCaM) is 88% identical to all mammalian CaM (mCaM) sequences, with only four differences in the N-domain (residues 1-80, sites I and II) and 13 differences in the C-domain (residues 76-148, sites III and IV) (Kung et al., 1992; VanScyoc et al., 2006) (**Figure 1.3**). Although the two domains are similar in sequence and structure, the N-domain binds Ca^{2+} with a 10-fold lower affinity than the C-domain (Linse and Chazin, 1995; VanScyoc et al., 2002).

CaM has been observed to adopt multiple conformations (closed, semi-open, and open) dependent upon its ligation state (**Figure 1.4**) (Ataman et al., 2007; Chattopadhyaya et al., 1992; Houdusse et al., 2006; Kuboniwa et al., 1995; Meador et al., 1992). These conformations are defined by the inter-helical angles adopted by the EF-hand motifs within each domain. The “open” domain conformation of CaM is adopted by the N- and C-domains of CaM when Ca^{2+} binds to them either in the absence or presence of a target (Ataman et al., 2007; Swindells and Ikura, 1996; Wilson and Brunger, 2000). The “semi-open” domain conformation of CaM, which has only been observed in the C-domain of CaM, is adopted when binding to targets in the absence of Ca^{2+} (Houdusse et al., 2006; Swindells and Ikura, 1996). More commonly the absence of Ca^{2+} results in a “closed” conformation for both the N- and C-domains (Kuboniwa et al., 1995).

Interaction of $(\text{Ca}^{2+})_4$ -CaM with Protein Targets

Changes in intracellular Ca^{2+} levels are linked to cellular events by the effect of Ca^{2+} on CaM: it triggers conformational changes that expose hydrophobic surfaces in both domains, altering its binding affinity for many target proteins (Bayley et al., 1996; Colbran, 1992; Klee, 1980). The Ca^{2+} binding affinity of CaM is “tuned” dependent upon which target it is bound to, allowing CaM to regulate numerous cellular processes

dependent upon [intra-cellular Ca^{2+}] (Evans and Shea, 2006; Peersen et al., 1997). The canonical mode of interaction of $(\text{Ca}^{2+})_4$ -CaM with intra-cellular protein targets such as metabolic enzymes, cyclases, kinases, phosphatases, and ion channels is a compact ellipsoidal conformation (Hoeflich and Ikura, 2002; Meador et al., 1993; Mori et al., 2000). The protein target sequence recognized by CaM is often a Basic Amphipathic Alpha-helix (BAA) motif where CaM typically binds in a Ca^{2+} -saturated manner (O'Neil and DeGrado, 1990). When bound to Ca^{2+} -dependent targets such as CaM Kinase II (CaMKII), or myosin light chain kinase (MLCK), the domains of CaM adopt an “open” conformation that exposes hydrophobic patches used for target binding resulting in Ca^{2+} -dependent regulation as shown previously in **Figure 1.4** (O'Neil et al., 1987).

Interaction of apo CaM with Protein Targets

Historically, activation of target proteins by CaM was thought to occur in a strictly Ca^{2+} -dependent manner where target binding always increased the Ca^{2+} affinity of CaM (Prozialeck and Weiss, 1982; Roberts and Harmon, 1992; Wang and Sharma, 1980; Weiss and Wallace, 1980). An emerging class of Ca^{2+} -independent CaM targets, typically found with ion channels and myosin motor proteins, interact more favorably with apo CaM instead of $(\text{Ca}^{2+})_4$ -CaM. These targets typically contain IQ-motifs, although other non IQ-motif apo CaM Binding Domains (apo CaMBD's) have been observed (Fanger et al., 1999; Gerendasy et al., 1994; Liu and Storm, 1990; Martin and Bayley, 2004; Swindells and Ikura, 1996). Compared to structural studies of $(\text{Ca}^{2+})_4$ -CaM, structural studies of apo-CaM interacting with targets are much less common. The disparity in representation of targets bound to $(\text{Ca}^{2+})_4$ -CaM compared to apo-CaM is likely due to the mobility of the N-domain relative to the C-domain in the absence of Ca^{2+} , as well as the lack of an ordered structure within the Ca^{2+} binding loops of CaM in the absence of Ca^{2+} . The two high-resolution structures available of an apo C-domain

CaM interacting with peptides derived from either Myosin V (IQ-motif) or the SK-channel (non IQ-motif), depict CaM in an extended conformation where its C-domain interacts with each target through a “semi-open” conformation (**Figure 1.5**) (Houdusse et al., 2006; Schumacher et al., 2001).

The IQ-motif was first discovered and characterized from neuromodulin, a neurospecific apo CaM binding protein (Chapman et al., 1991; Liu and Storm, 1990). The IQ-motif is approximately 11 amino acids in length and distributed across multiple protein families ranging from myosins and ion channels, to Ras exchange and neuronal growth proteins (Bähler and Rhoads, 2002). The consensus sequence of this motif is (IQxxxBGxxxB, B=Lys or Arg), which forms an amphiphilic α -helix characterized as capable of binding to calmodulin in a Ca^{2+} -independent manner (**Figure 1.6**) (Swindells and Ikura, 1996). Analysis of genomic sequencing results indicate that there are at least 208 IQ-motifs in 108 proteins within the human genome (Venter et al., 2001). Examination of IQ-motifs across all eukaryotes reveals that of the IQ-motif defining residue Gln at position 1 is invariant, while IQ-motif position 6 (Gly) is the most variable (**Figure 1.6**). As indicated by the higher number of IQ-motifs than number of proteins which contain them (208 IQ-motifs per 108 proteins), several proteins contain multiple IQ-motifs (Houdusse et al., 2006; Martin and Bayley, 2004; Trybus et al., 2007). These proteins are mainly found in the myosin family, which depending upon the myosin variant, possess between 1 and 7 IQ-motifs typically separated by 9-16 residues (Koide et al., 2006).

The Voltage-Dependent Sodium Channel_v1.2

The voltage-dependent sodium channel_v1.2 (Na_v1.2) is an integral membrane protein comprised of one pore-forming α -subunit (2005 aa, 260 kDa), and one or more β -subunits (215 aa, 33–36 kDa each) which control the kinetics and gating of the channel as shown in (**Figure 1.7**) (Yu and Catterall, 2003). The physiological role of this channel

is to selectively regulate the flow of Na^+ ions across the cell membrane of central and peripheral neurons, allowing for the creation of action potentials (Catterall, 2000b). The IQ-motif containing CaM binding region of $\text{Na}_v1.2$ is located near its C-terminus and has a high degree of sequence identity to corresponding regions of all 10 known human sodium channel isoforms (**Figure 1.8**) (Mori et al., 2003; Theoharis et al., 2008; Yu and Catterall, 2003).

The α -subunit is comprised of four domains formed from six transmembrane helices, where the fourth helix contains a voltage sensor, responsible for activation of the $\text{Na}_v1.2$ upon depolarization of the cell membrane (Cormier et al., 2002; Mantegazza et al., 2001). Inactivation of $\text{Na}_v1.2$ is achieved via an intracellular loop between domains III and IV which contains the inactivation gate and the C-terminal tail of the α -subunit (**Figure 1.7**) (Chin and Means, 2000; Herzog et al., 2003; Mantegazza et al., 2001; Yu and Catterall, 2003). The mechanism of inactivation is hypothesized to occur when the inactivation gate physically blocks the channel pore via a CaM mediated interaction with C-terminal tail of $\text{Na}_v1.2$ (Chin and Means, 2000; Herzog et al., 2003; Mantegazza et al., 2001; Yu and Catterall, 2003). Among the ten known sodium channel isoforms the C-terminal tail is responsible for the different rates of inactivation of the α -subunit (Deschênes et al., 2001; Mantegazza et al., 2001). The membrane proximal half of the intracellular C-terminal tail has been modeled to contain six α -helices, where deletion of the putative sixth α -helix region slows recovery from inactivation, maintaining $\text{Na}_v1.2$ in a closed inactivated state (Cormier et al., 2002). In this proposed sixth α -helix of the C-terminal region, between residues 1901–1927 of the $\text{Na}_v1.2$, is located a CaM binding IQ-motif (Mori et al., 2000; Theoharis et al., 2008). Studies of $\text{Na}_v1.2$, as well as other sodium channel isoforms have shown that CaM binding is necessary for functional sodium currents, indicating a regulatory role for CaM on the sodium channel (Mori et al., 2003; Yu and Catterall, 2003).

The two CaM domains (N and C) may have distinct regulatory roles in sodium channel modulation. Separable roles for the CaM domains were first shown physiologically with *in vivo* studies by Kung and associates (Gustin et al., 1986). A genetic screen of mutagenized but viable *Paramecium* identified two classes of mutants that under- or over-reacted to chemical stimuli (Preston et al., 1992). These mutants were found to have defective regulation of their Ca²⁺-dependent sodium and potassium channel currents, and the mutations were located in CaM. Under-reacting mutations that occurred between sites I and II and in site II of the N-domain of CaM were shown to affect only sodium conductance. In contrast, over-reacting mutations occurred within sites III and IV and the fourth helix of the C-domain and only affected potassium conductance.

Interaction of CaM with Drugs

In addition to interacting with naturally occurring protein targets, (Ca²⁺)₄-CaM has also been shown to interact with small molecule compounds such as N-(6-Aminoethyl)-5-Chloro-1-Naphthalenesulfonamide (W-7), N-(3,3-Diphenylpropyl-N'-[1-R-(2,3,4-Bis-Butoxyphenyl)-Ethyl]-Propylenediamine (DPD), 3''-(Beta-Chloroethyl)-2'',4''-Dioxo-3,5''-Spiro-Oxazolidino-4-Deacetoxy-Vinblastine (KAR-2), and 10-[3-(4-Methyl-Piperazin-1-yl)-Propyl]-2-Trifluoromethyl-10H-Phenothiazine (Trifluoperazine, TFP) (**Figure 1.9**) (Cook et al., 1994; Harmat et al., 2000a; Hennessey and Kung, 1984; Horvath et al., 2005; Matsushima et al., 2007; Osawa et al., 1998; Tang et al., 2006; Vandonselaar et al., 1994a; Vertessy et al., 1998b). The compounds W-7 and DPD have primarily been used *in vitro* as CaM antagonists, while KAR-2 (a potent anti-microtubular agent), and TFP (antipsychotic agent) have been used in the clinical setting (Cook et al., 1994; Harmat et al., 2000a; Hennessey and Kung, 1984; Horvath et al., 2005; Matsushima et al., 2007; Osawa et al., 1998; Tang et al., 2006; Vandonselaar et al., 1994a; Vertessy et al., 1998b). In structures of these drugs bound to (Ca²⁺)₄-CaM, their

aromatic moieties insert into the hydrophobic pockets of CaM, mimicking CaM-peptide interactions which employ similar interactions to insert aromatic groups into CaM. In all of these structures the drug bound to the “open” conformation and like peptide targets all crystallographically determined structures of these drugs bound to $(Ca^{2+})_4$ -CaM (with the exception of W-7 determined via NMR) depict CaM in a collapsed conformation as seen with peptides (**Figure 1.10**) (Cook et al., 1994; Harmat et al., 2000b; Horvath et al., 2005; Osawa et al., 1998; Vandonselaar et al., 1994a; Vertessy et al., 1998a). Although CaM has been observed to bind drugs *in vitro*, it is not the intended primary *in vivo* target of the previously mentioned compounds making *in vivo* CaM interactions with these drugs a secondary or off target effect (Sheets et al., 2006).

TFP

Trifluoperazine (TFP) is a phenothiazine class antipsychotic drug primarily used in the treatment of schizophrenia and related mental disorders (Abuzzahab, 1977; Oybir, 1962). Its first clinical trial for use in human patients with mental disorders was in 1958; more recently TFP has been indicated to reduce levels of opioid addiction (Tang et al., 2006; Wallis, 1958). Its primary function is that of a dopamine antagonist where it binds to, but does not activate, the dopamine receptor, thus blocking the action of dopamine or exogenous agonists (Clow et al., 1980; Kerwin et al., 1984; Roudebush et al., 1991). Clinically, TFP can be administered orally in solid pill form, as a liquid, or as an intramuscular injection where the daily amount administered is typically 15-20 mg per day (Carscallen et al., 1968; Gauron and Rowley, 1970; Hodes, 1960). The clinical use of TFP as an anti-psychotic has been discontinued in favor of newer formulations that do not carry the often irreversible side-effect of tardive dyskinesia typical of TFP and other first generation anti-psychotics like it (Lahti et al., 1993).

TFP has been shown *in vitro* to be a potent CaM antagonist, where it is often added to cell cultures to disrupt CaM interactions with its protein targets (Lydan and

O'Day, 1988; Pelech et al., 1983). The interaction of TFP with CaM is the most studied of small molecule CaM antagonists. However, the results of many of these studies which focused on the stoichiometry of TFP binding, its effect on Ca^{2+} -binding affinity, as well as how it alters the structure of CaM have been inconclusive (Cook et al., 1994; Massom et al., 1990b; Matsushima et al., 2000; Matsushima et al., 2007; Vogel et al., 1984; Yamaotsu et al., 2001). An example of this can be found in the stoichiometry of TFP bound to $(\text{Ca}^{2+})_4$ -CaM in the three structures of that have been determined, where depending upon the structure examined, either 1, 2, or 4 TFP are bound to $(\text{Ca}^{2+})_4$ -CaM as shown previously in (Figure 1.10).

Electrostatic Interactions of CaM with Targets

In addition to hydrophobic interactions, electrostatic interactions between CaM and its protein targets have a significant role in recognition and binding (Linse et al., 1991; Noguchi et al., 2004; Ogawa and Tanokura, 1984). At pH 7.4, CaM is highly acidic ($\text{pI} = 4$) and carries a net charge of -24 (apo) or -16 (Ca^{2+} -saturated), while the CaMBD's of target proteins contain basic residues arginine or lysine, resulting in electrostatic attraction between the two molecules. The opening and closing of ion channels necessary for cell signaling results in a constant flux of Ca^{2+} , K^+ , and Na^+ ions, altering the strength of electrostatic interactions between CaM and its targets. K^+ and Na^+ also reduce the Ca^{2+} -binding affinity of CaM, making it intriguing to learn the functional consequences of how they affect CaM-protein interactions.

Description of Thesis Content

Chapter II describes studies conducted to explore how the Ca^{2+} -binding properties of CaM are altered upon binding TFP, as well as the binding stoichiometry of TFP to apo and $(\text{Ca}^{2+})_4$ -CaM. In this chapter fluorescence-monitored Ca^{2+} titrations demonstrate that dependent upon the [TFP] examined that the Ca^{2+} -binding affinity of CaM can either increase or decrease relative to values observed in the absence of TFP.

Additionally using ^{15}N -HSQC NMR spectroscopy the stoichiometry of TFP binding was dependent upon the Ca^{2+} ligation state of CaM where 2 and 4 TFP were found to bind to apo and $(\text{Ca}^{2+})_4$ -CaM respectively. The major findings of experiments conducted in **Chapter II** were reported in a manuscript titled “*Allosteric Effects of the Anti-Psychotic Drug Trifluoperazine on the Energetics of Calcium Binding by Calmodulin*” that has been published in *Proteins: Structure, Function, and Bioinformatics*.

Chapter III addresses how TFP alters the structures of apo and $(\text{Ca}^{2+})_4$ -CaM at the atomic level. In this chapter I show, using x-ray crystallography, that TFP binds $(\text{Ca}^{2+})_2$ -CaM₇₆₋₁₄₈ in two different orientations, unifying what were otherwise conflicting observations within the field. I will also show, using NMR spectroscopy, that the chemical environment of specific atoms of apo CaM₇₆₋₁₄₈ are perturbed upon TFP addition. The major findings of experiments conducted in **Chapter III** will be reported in a manuscript that is in preparation for submission

Chapter IV discusses how binding of Na_v1.2_{IQP} alters the structure of apo CaM₇₆₋₁₄₈. In this chapter I will show using NMR spectroscopy how apo CaM₇₆₋₁₄₈ adopts a “semi-open” conformation upon binding Na_v1.2_{IQP}. I will also demonstrate, using NMR spectroscopy, that when Na_v1.2_{IQP} binds to apo CaM₁₋₁₄₈, the N- and C-domains adopt “closed” and “semi-open” domain conformations respectively. The major findings of experiments conducted in **Chapter IV** will be reported in a manuscript that is in preparation for submission.

Chapter V presents preliminary studies of how changing the solution ionic strength via KCl or NaCl alters the strength of electrostatic interaction between CaM and peptides (melittin, CaMKII, and Na_v1.2_{IQP}). In this chapter I show using fluorescence anisotropy the effect that varied [NaCl] and [KCl] salts have on the binding affinity of CaM for the peptides Melittin, CaMKII, and Na_v1.2_{IQP}. Poisson-Boltzman calculations were performed to theoretically examine how varying [NaCl] or [KCl] altered the electrostatic attraction between $(\text{Ca}^{2+})_4$ -CaM and CaMKII. These studies indicated that

increasing [salt] lowered the electrostatic attraction of CaM for all peptides tested. Significant differences were observed though between experimental and theoretical studies involving $(Ca^{2+})_4$ -CaM and CaMKII suggesting that another factor, possibly conformational change by CaM and/or CaMKII influence the electrostatic interaction between these two molecules.

Chapter VI discusses the results of the prior chapters and their contribution to our understanding of the role of target interactions with CaM under apo conditions. While the work described in this thesis presents a high-resolution structure of apo CaM bound to $Na_v1.2_{IQP}$, the structural basis of how $(Ca^{2+})_4$ -CaM interacts with $Na_v1.2_{IQP}$ or another proposed Ca^{2+} -dependent CaM binding site ($Na_v1.2_{BAA}$) is not clear. Future experiments will determine how $(Ca^{2+})_4$ -CaM interacts with its binding domains ($Na_v1.2_{IQP}$ and $Na_v1.2_{BAA}$) of $Na_v1.2$. Experiments are proposed to resolve the role of the N-domain in $Na_v1.2$ regulation under Ca^{2+} -saturating conditions where it has been implicated in interacting with the $Na_v1.2_{BAA}$ region of $Na_v1.2$.

Appendices provide Fortran functions used in nonlinear squares analysis, and chemical shift assignments of apo CaM_{76} , apo CaM_{76-148} when bound to TFP, and apo CaM_{76-148} when bound to $Na_v1.2_{IQP}$.

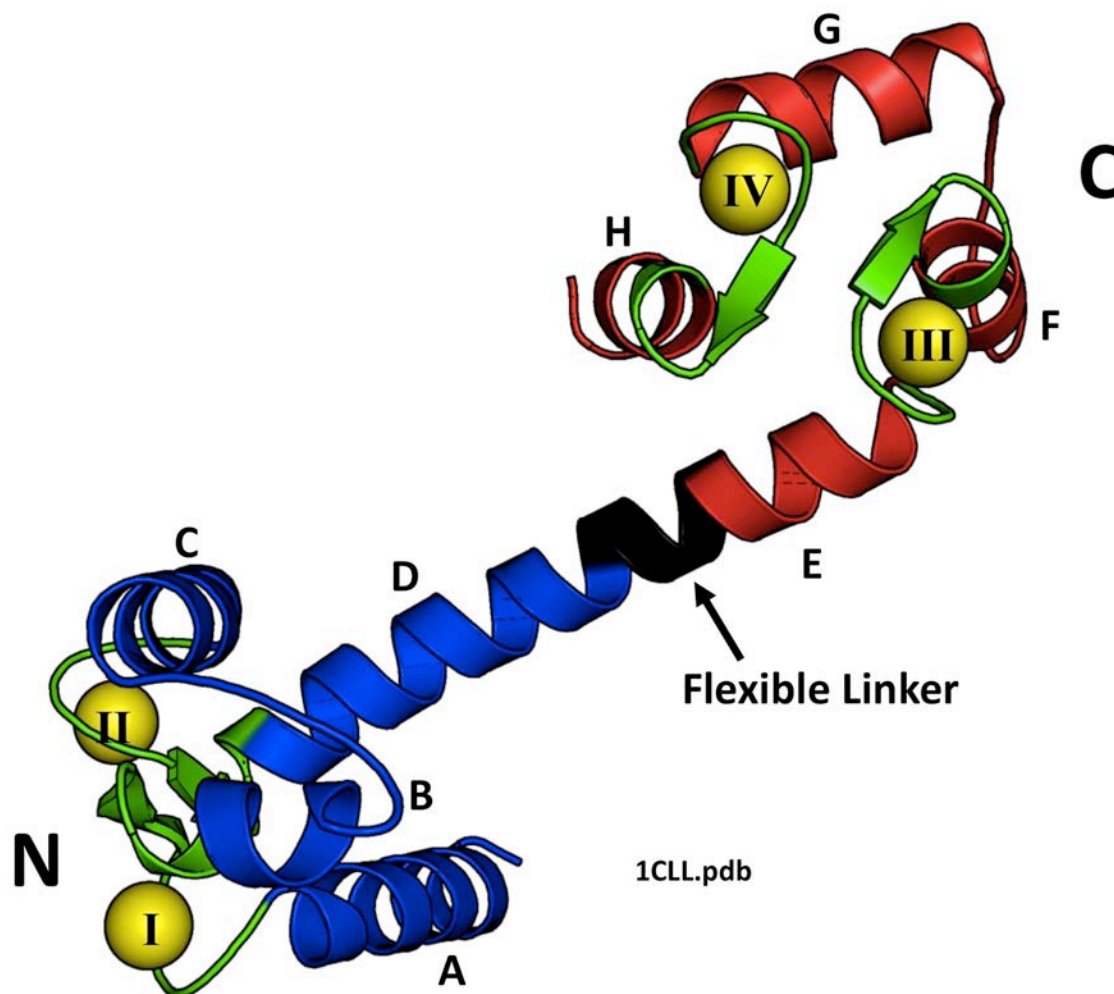
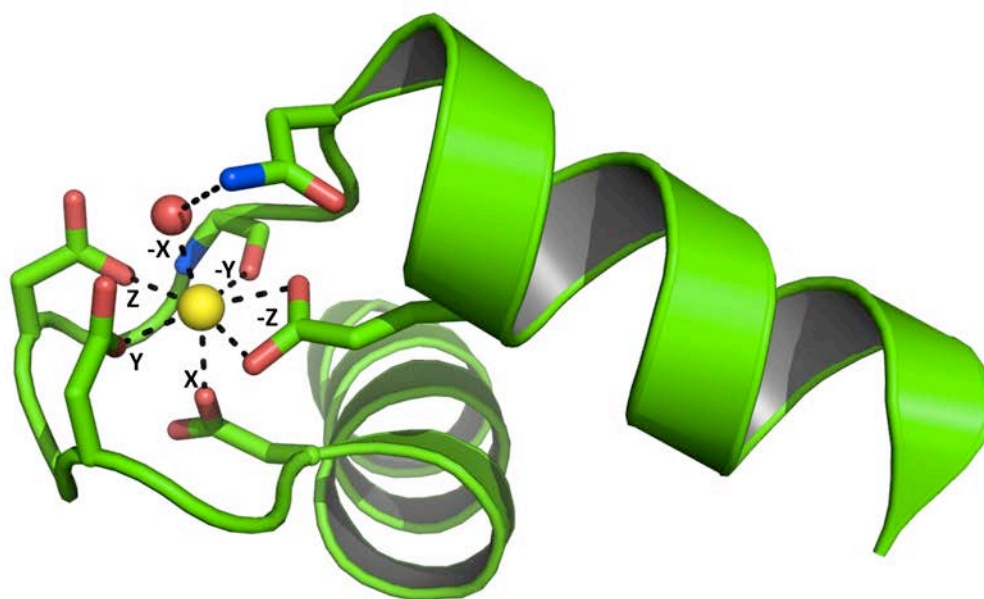


Figure 1.1: Structure of $(Ca^{2+})_4$ -CaM. Ca^{2+} -binding sites (green) I and II are located in the N-domain (blue, helices A-D), while Ca^{2+} -binding sites III and IV are located in the C-domain (red, helices E-H), allowing CaM to bind 4 Ca^{2+} ions (yellow spheres). The N- and C- domains of CaM are connected by a 5 residue linker (black).
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	X	Y	Z	-Y	-X		-Z					
PCaM	D	K	D	G	D	G	T	I	T	T	K	E
mCaM	D	K	D	G	D	G	T	I	T	T	K	E
	20	21	22	23	24	25	26	27	28	29	30	31
PCaM	D	A	D	G	N	G	T	I	D	F	P	E
mCaM	D	A	D	G	N	G	T	I	D	F	P	E
	56	57	58	59	60	61	62	63	64	65	66	67
PCaM	D	R	D	G	N	G	L	I	S	A	A	E
mCaM	D	K	D	G	N	G	Y	I	S	A	A	E
	93	94	95	96	97	98	99	100	101	102	103	104
PCaM	D	I	D	G	D	G	H	I	N	Y	E	E
mCaM	D	I	D	G	D	G	Q	V	N	Y	E	E
	129	130	131	132	133	134	135	136	137	138	139	140

Figure 1.2: Ribbon diagram of an EF-hand motif.

The EF-hand consists of two α -helices connected by a 12-residue loop. Residues 1, 3, 5, 7, 9 and 12 (sticks) of the loop contribute a side chain or backbone oxygen (red) atom necessary for Ca^{2+} (yellow sphere) binding. Residue 12, which is a Glu, contributes both oxygens from its side chain carboxylic acid group. Residue 9 of the Ca^{2+} -binding loop does not directly bind the Ca^{2+} ion but instead coordinates a water (red sphere) at position -X. Below are aligned *Paramecium* and mammalian CaM sequences where the identity of X, Y, Z, -X, -Y, and -Z are in each of the 4 Ca^{2+} -binding loops are boxed. Differences in primary sequence of the Ca^{2+} -binding loops are highlighted in gray.

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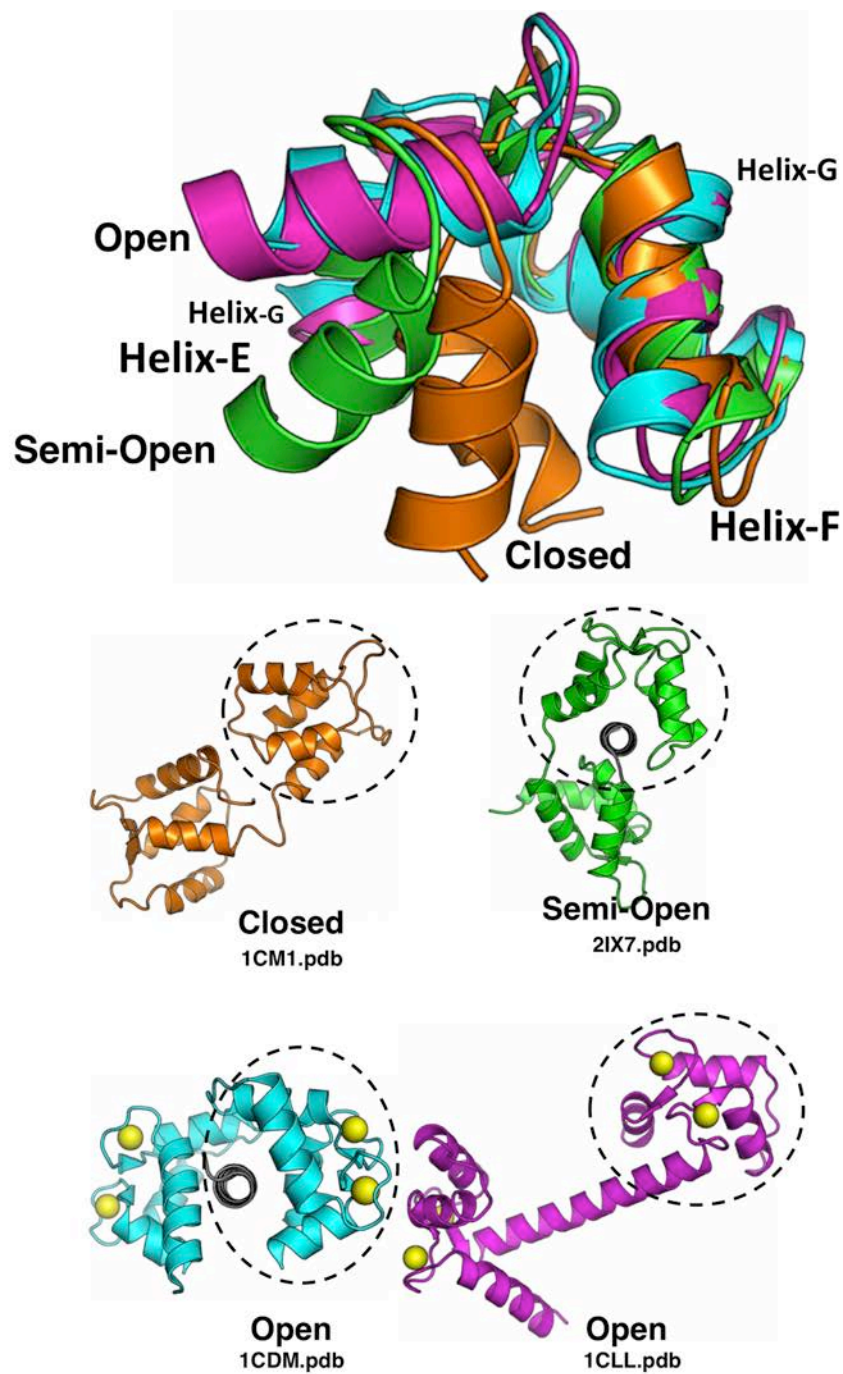


Figure 1.4: Superposition of experimentally observed conformations of CaM. Examples of the “closed” (1CFC.pdb orange), “semi-open” (2IX7.pdb green) and “open” (1CDM.pdb aqua and 1CLL.pdb magenta) conformations of the C-domain are aligned according to the positions of the F and G (first and fourth) helices of the domain. Users/nmr_mike/Thesis/Chapter_I/Figure1_4.jpg

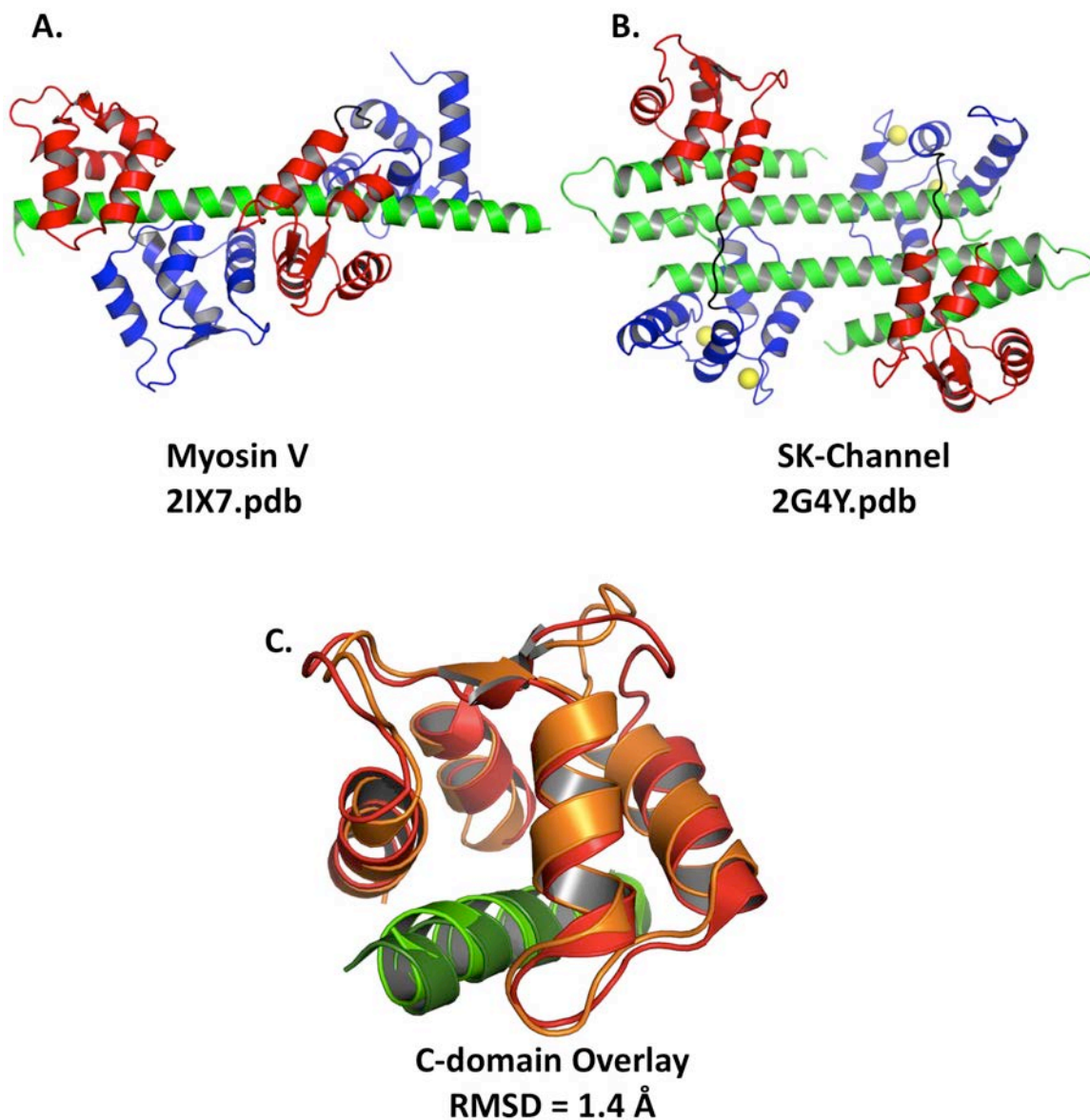


Figure 1.5: Structures of apo CaM or partially Ca^{2+} -ligated CaM interacting with a target.

A. Apo CaM (“closed” N-domain blue, “semi-open” C-domain red) bound to a peptide containing tandem IQ-motifs derived from Myosin V (green). B. Partially Ca^{2+} -ligated CaM (“open” N-domain blue, “semi-open” C-domain red) where its N-domain is Ca^{2+} -saturated (Ca^{2+} yellow sphere), and apo C-domain are bound to peptide derived from the small-conductance Ca^{2+} -activated K^+ channels (SK-channel) in green. C. Superposition of apo C-domains of CaM from myosin V (red and green) and SK-channel (orange and dark green).

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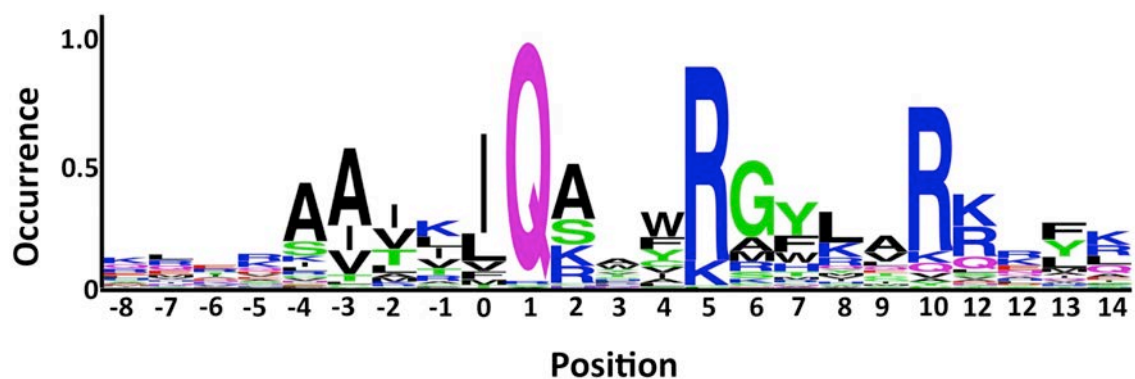


Figure 1.6: IQ-motif amino acid sequence conservation from 208 sequences derived from 108 human proteins, where Q at position 1 is almost invariant.
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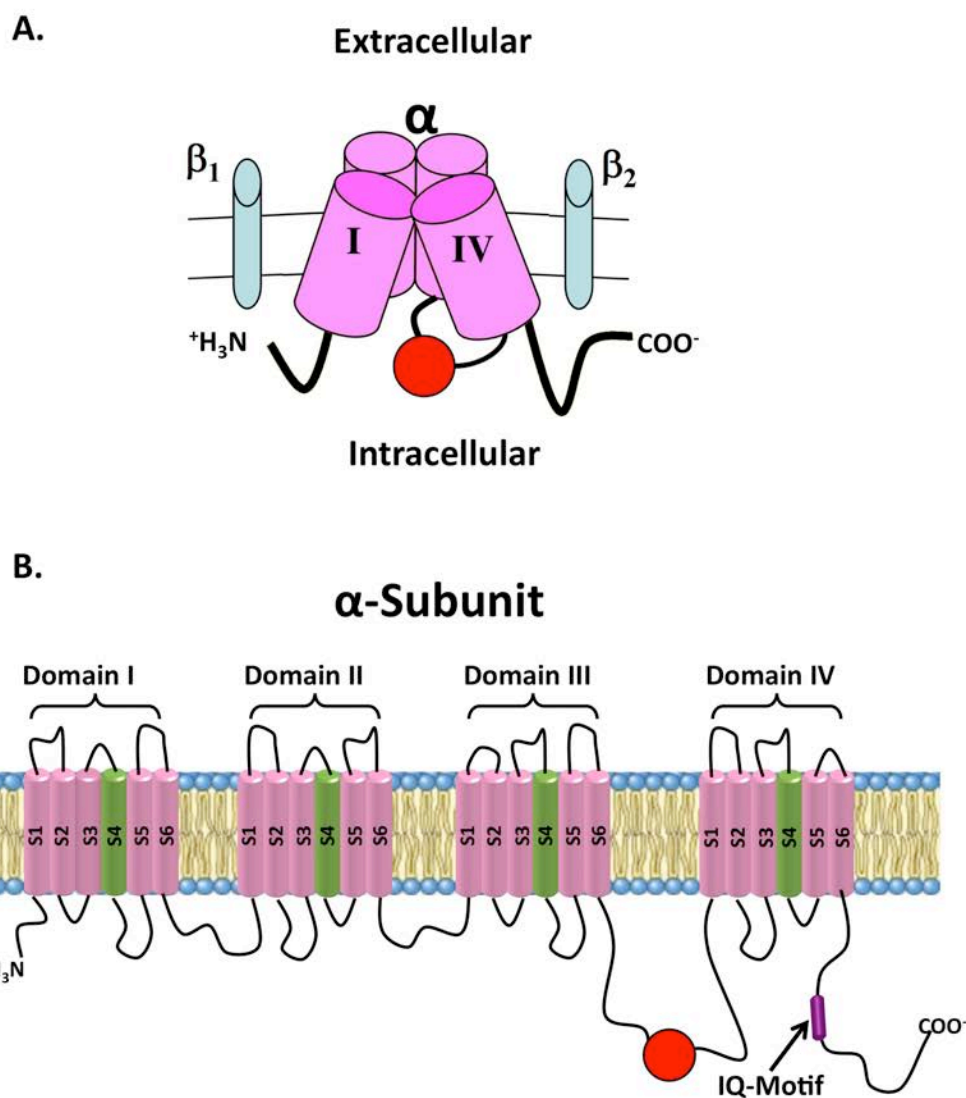


Figure 1.7: Domain structure of $\text{Na}_v1.2$.

A. $\text{Na}_v1.2$ is composed of a single 4-domain α -subunit, and multiple noncovalently attached β -subunits.

B. The α -subunit of $\text{Na}_v1.2$ is composed of 4 domains, each of which is made up of 6 helices, where the 4th helix (green) is part of the Na^+ channel pore. The loop between domains III and IV, contains the inactivation gate (red sphere) which is responsible for blocking the inside of the channel shortly after it has been activated. The CaM-binding IQ-motif (purple cylinder) is located near the C-terminus of the channel.

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		01	5	10	
		IQxxxBGxxxB			
Na _v 1.1	VSYPITTTTLKRRQEEVSAVLIQRAYRRHLIKRTVKQASFTYNKNKIKGGANLLIKEDMI	I	R	K	1960
Na _v 1.2	VSYPEITTTTLKRRQEEVSAIIQRAYRRYLKQKVKVSSSIYKDKGKCEDGTPIKEDTL	I	R	K	1950
Na _v 1.3	VSYPEITTTTLKRRQEEVSAAIQRNYRCYLKQRLKNISSKYDKETIKGRIDLPKIGDMV	I	R	K	1896
Na _v 1.4	VSYPEITTTTLKRRKHEEVCAIKIQRAYRRHLICRSMKQASYMYRSHSHDGSDDAPEKEGLL	I	R	K	1772
Na _v 1.5	ISYPEITTTTLRRKHEEVSAMVIQRARRRHLICRSLKHASFLFRQQAGSGLSEEDAPEREG	I	R	K	1945
Na _v 1.6	VSYPEITTTTLRRKQEEVSAVVLQRAYRGHLARGFICKKTTSNKLE-----	I	R	K	1926
Na _v 1.7	VSYPEITTTTLKRRQEDVSATVIQRAYRRYRIQNVKNISSIYIKDGDRD-DDLLNKKDMA	I	R	K	1922
Na _v 1.8	SSYPEIATTLRWKQEDISATVIQKAYRSYVHIRSMALSNTPCVPRA---EEEEASLPDEG	I	R	K	1893
Na _v 1.9	KLYEPIVTTTLKRKEEERGAAIQRKAFRRKYMVKVTKGQGD-----QNDLENGPHSP	I	R	K	1769
Na _v x	ITCEPIVTTTLKRRQEAVSATVIQRAYKNYRIQRNDKNTSDIHMDG-----	I	R	K	1656

Figure 1.8: Sequence alignment of all ten human Na⁺ channel isoforms. The locations of IQ-motif defining residues have been highlighted in red. Users/nmr_mike/Thesis/Chapter_I/Figure1_8.jpg

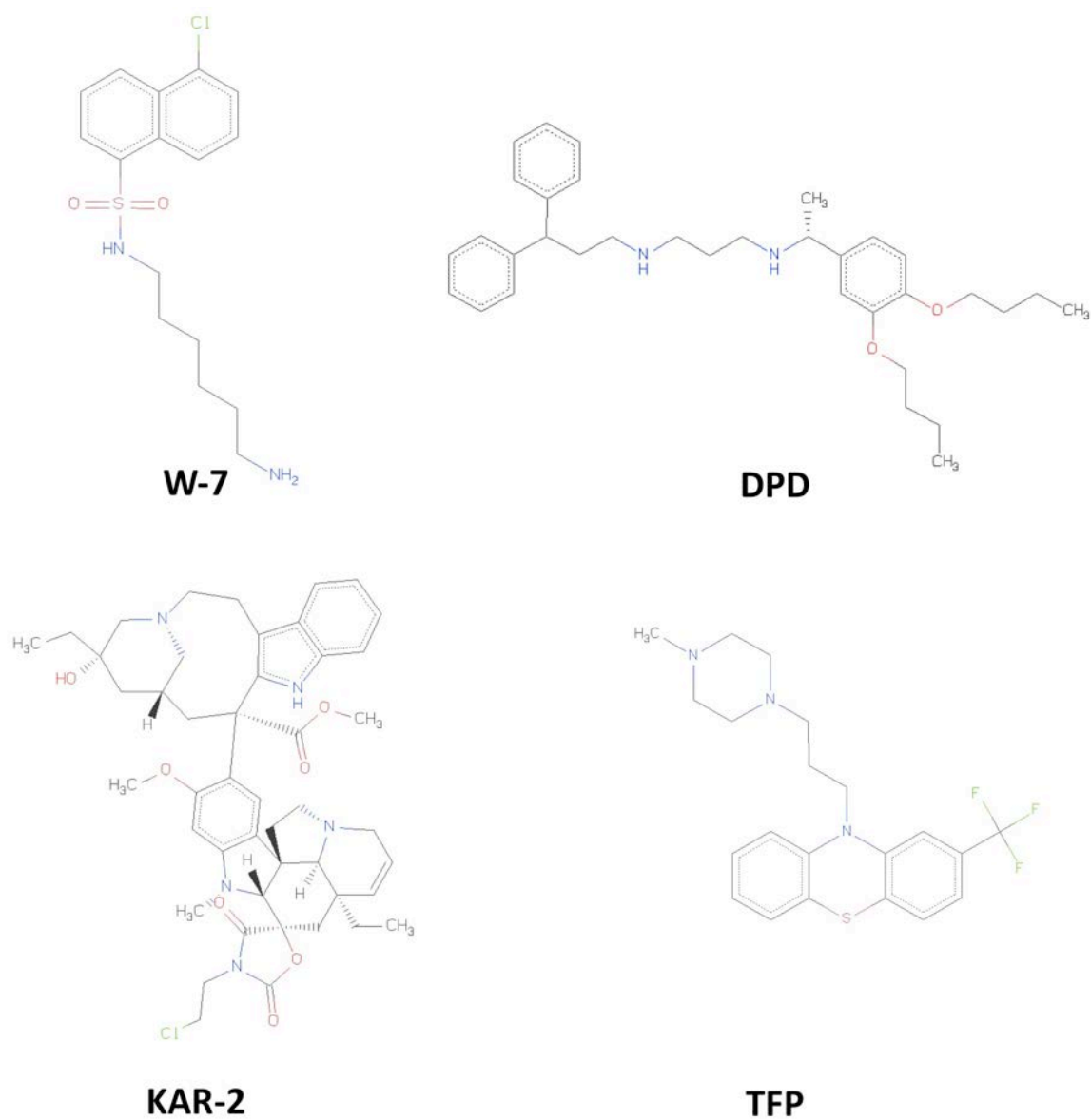


Figure 1.9: Chemical structures of CaM-binding drugs whose binding interface with CaM has been determined structurally.
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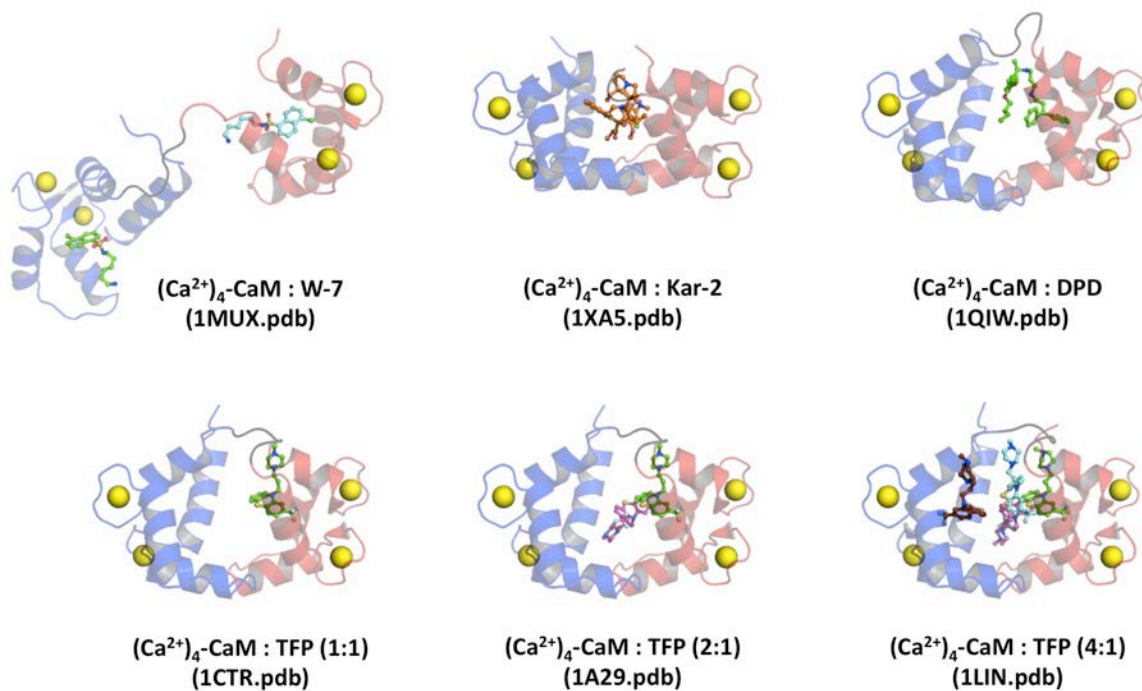


Figure 1.10: Structures of (Ca²⁺)₄-CaM bound to drugs. X-ray (Kar-2, DPD, TFP) or NMR (W-7) determined structures of (Ca²⁺)₄-CaM bound to drugs where the N- and C-domain of (Ca²⁺)₄-CaM are colored blue and red respectively, drug molecules are shown in sticks, while Ca²⁺ ions are represented by yellow spheres. Users/nmr_mike/Thesis/Chapter_I/Figure1_10.jpg

CHAPTER II
**ALLOSTERIC EFFECTS OF THE ANTI-
PSYCHOTIC DRUG TRIFLUOPERAZINE ON THE
ENERGETICS OF CA²⁺ BINDING BY CAM**

Introduction

Calmodulin (CaM) is a small (148 a.a.), essential, and highly conserved eukaryotic protein that is required for many calcium-sensitive signal transduction pathways (Hidaka and Ishikawa, 1992; Newman, 2008). It is composed of two homologous domains (N and C). Each domain consists of a pair of EF-hands (a helix-loop-helix motif) that forms a 4-helix bundle, and binds two calcium ions cooperatively. The domains are connected by a linker that plays a regulatory role in determining calcium affinity, and permits the domains to adopt multiple relative orientations to optimize interactions with target proteins (**Figure 2.1a**) (Sorensen et al., 2002b; Sorensen and Shea, 1998; Zhang et al., 1995).

Although the two domains are similar in sequence and structure, the affinity of the N-domain for calcium is an order of magnitude lower than that of the C-domain (Seamon, 1980; VanScyoc et al., 2002). As the concentration of intracellular calcium increases, calcium binding to 12-residue sites in CaM triggers conformational changes, causing the pairs of helices in each 4-helix bundle to separate. This structural change associated with the transition from apo (calcium-depleted) to (Ca²⁺)₄-CaM exposes hydrophobic residues that alter the affinity of CaM for target proteins (Chattopadhyaya et al., 1992; Kuboniwa et al., 1995; Meador et al., 1992; Wilson and Brunger, 2000). (Ca²⁺)₄-CaM (4 bound calcium ions) is shown in **Figure 2.1b**.

In a CaM-target complex, the protein-protein interface is determined by the number and location of occupied calcium-binding sites of CaM, conformational change propagated from those binding sites, and the surface of the target protein. Based

on the interhelical angles adopted by the paired helices of each domain, the CaM C-domain has been categorized as adopting three distinct conformations: “closed”, “semi-open”, and “open” (Chattopadhyaya et al., 1992; Kuboniwa et al., 1995; Meador et al., 1992; Wilson and Brunger, 2000). Any of these may be adopted by free CaM (**Figure 2.2**), consistent with the hypothesis that changes in the distribution of pre-existing conformational states occur upon binding to Ca^{2+} or a target protein (Yap et al., 1999). In high-resolution structures, only apo CaM has been observed to adopt the “closed” (**Figure 2.2B**) and “semi-open” forms (**Figure 2.2C**), while only $(\text{Ca}^{2+})_4$ -CaM has been observed in the “open” form (**Figure 2.2D-E**). Binding of a target protein stabilizes the “semi-open” or “open” conformation of CaM by burying hydrophobic surface area that would otherwise be exposed to solvent. For example, a CaMKII peptide bound to calcium-saturated “open” CaM (1CM1.pdb) buries 1226 \AA^2 of the surface area of CaM; of that, 990 \AA^2 (81%) is hydrophobic (Wall et al., 1997).

For many years, activation of CaM-regulated target proteins such as metabolic enzymes, kinases, and phosphatases was thought to occur in a strictly calcium-dependent manner, such that the extent of binding of a target protein to apo CaM was negligible, and binding to $(\text{Ca}^{2+})_4$ -CaM always increased its calcium affinity (Cox, 1988). However, there are subclasses of CaM-regulated target proteins, including some ion channels, that contain IQ-motifs that reduce the calcium affinity of CaM (Martin and Bayley, 2004; Theoharis et al., 2008). These targets interact preferentially with apo CaM. The only high resolution structure of apo CaM interacting with an IQ-motif (2IX7.pdb) is the crystallographic observation of two apo CaM molecules bound to a peptide containing two adjacent IQ-motifs derived from Myosin V (Houdusse et al., 2006) (see **Figure 2.2c**). In this complex, both N-domains were “closed” whereas the C-domains were “semi-open”. Analysis of the CaM-peptide interface showed that hydrophobic residues of CaM accounted for most (71% and 75%) of the buried surface of the two CaM C-domains.

Drugs with aromatic moieties bind to CaM in a manner similar to protein targets that bury a phenylalanine or tryptophan residue in the hydrophobic pockets of CaM (Cook et al., 1994; Vandonselaar et al., 1994b; Vertessy et al., 1998b). Trifluoperazine (TFP) (**Figure 2.1c**) is a CaM antagonist historically used in the treatment of mental illness because of its interaction with the dopamine receptor; recently, it was implicated in the disruption of opioid tolerance (Tang et al., 2006). TFP is membrane-permeable and is commonly added to cell culture media to disrupt CaM-mediated processes (Barrington and Majewski, 1994; Chen et al., 2008; Frankfurt et al., 1995). Structures of three TFP-CaM complexes (Cook et al., 1994; Vandonselaar et al., 1994b; Vertessy et al., 1998b) have been determined crystallographically. In these, 1, 2, or 4 molecules of TFP are bound to $(Ca^{2+})_4$ -CaM; all of them share a TFP-binding site in the C-domain but only one structure (1LIN.pdb) has a TFP-binding site in the N-domain. Superposition of these structures showed that the backbone of CaM adopts indistinguishable conformations in all of them, despite differences in the number and location of TFP molecules bound (**Figure 2.3D**). The relative abundance of these three ligation states of TFP bound to CaM in solution is not known. In all three complexes, the tertiary structure of CaM mimics that of $(Ca^{2+})_4$ -CaM when bound to CaMKII, and other kinases that increase its calcium affinity (Clapperton et al., 2002; Heidorn et al., 1989; Ikura et al., 1992). The CaM-binding domain in those enzymes is a BAA (basic amphipathic alpha helical) motif.

The conflicting structural evidence regarding a preferred binding stoichiometry for TFP binding to CaM, as well as disagreement on the effects of TFP on the calcium-binding properties of CaM (Cook et al., 1994; Craven et al., 1996; Massom et al., 1990c; Tanokura and Yamada, 1985; Vandonselaar et al., 1994b; Vertessy et al., 1998b), motivated this study of the thermodynamic and structural properties of intermediate ligation states. These are critical for understanding how the highly homologous N- and C-domains of CaM exert different physiological effects on target proteins, and exploring

whether exogenous, pharmaceutical applications of TFP and related drugs truly target only the calcium-saturated form of CaM, as has been assumed.

TFP titrations of CaM monitored by ^{15}N -HSQC spectroscopy showed that TFP saturated apo CaM at a ratio of 2:1, but saturated $(\text{Ca}^{2+})_4\text{-CaM}$ at a ratio of 4:1. Equilibrium calcium titrations monitored by steady-state fluorescence spectroscopy demonstrated that, unlike the majority of effectors (e.g., peptides and proteins) whose binding to CaM has been examined in detail, TFP reduced the calcium affinity of CaM at low stoichiometries. Thus, thermodynamic linkage requires that TFP have a higher affinity for apo CaM than for $(\text{Ca}^{2+})_4\text{-CaM}$; this is similar to the preferential binding of most IQ-motifs to apo CaM.

However, the multiplicity of binding stoichiometries allows for reversal of this effect. At higher ratios (8:1) of TFP:CaM, the calcium affinity of CaM reversed, and was more favorable than that in the absence of TFP. These effects were found to be similar, but not identical, in each domain (N and C) of CaM and were compared to the effects of TFP on isolated domain fragments. On this basis, a model is proposed in which the “semi-open” conformation offers a “pocket” for binding of aromatic moieties that is unlike the FLMM pocket (Ataman et al., 2007) of the “open” conformation where aromatic side chains of many peptides are known to bind to CaM. TFP recognizes this site on the “semi-open” conformation of an apo domain that is not available when CaM adopts the “open” conformation.

The binding preference of a single TFP molecule for apo CaM is significant because, to the best of our knowledge, it is the only drug identified to reduce calcium affinity. This motivates a review of studies in which TFP antagonizes CaM-dependent cellular phenomena. The interference observed in those studies has been interpreted as arising from the effects of the drug on $(\text{Ca}^{2+})_4\text{-CaM}$ and its regulation of enzymes or channels. However, the findings presented here show that TFP can act as an antagonist

of apo CaM, which is critical for regulating pathways that are distinct from those modulated by $(\text{Ca}^{2+})_4\text{-CaM}$.

Materials and Methods

Protein over-expression and purification

IPTG-induced over-expression of CaM was performed using transformed in the *E. coli* strains BL21 DE3 or BL21 DE3-pLysS cells containing the recombinant pT7-7 vector of interest: full-length mammalian CaM₁₋₁₄₈, CaM₁₋₈₀, and CaM₇₆₋₁₄₈ (Pedigo et al., 1992). For ¹⁵N-labeled proteins used in the NMR studies, *Paramecium* CaM₁₋₁₄₈ (PCaM) was over-expressed in minimal medium with ¹⁵N-NH₄Cl (Cambridge Isotopes) as the sole nitrogen source. The proteins were then purified as described by Putkey (Putkey et al., 1985). The recombinant proteins were 97-99% pure as judged by silver-stained SDS-PAGE and reversed-phase HPLC. Protein concentrations were calculated from UV absorbance in 0.1 N NaOH, using the extinction coefficients for Phe and Tyr reported by Beaven and Holiday (Beaven and Holiday, 1952)

Equilibrium calcium titrations monitored by intrinsic protein fluorescence

Calcium titrations were monitored at 22 °C with a PTI-QM4 Fluorimeter (Photon Technology International, Birmingham, NJ) using bandpasses of 4 nm (for excitation) and 6 nm (for emission). CaM (CaM₁₋₁₄₈, CaM₁₋₈₀, or CaM₇₆₋₁₄₈ at 6 μM) solutions, containing 0, 6, 12, 18, 24, or 48 μM TFP (Sigma-Aldrich, St. Louis, MO), were prepared in 50 mM HEPES, 100 mM KCl, 5 mM NTA, 0.05 mM EGTA, 1 mM MgCl₂, and 5.75 nM Oregon Green (pH 7.4). The concentration of TFP was determined using an $\epsilon_{305.5}$ of 3540 M⁻¹ cm⁻¹ (Hart et al., 1983). CaM was titrated using a microburet (Micro-Metric Instrument Co., Cleveland, OH) fitted with a 250 μL Hamilton syringe (Hamilton Co., Reno, NV) containing a concentrated CaCl₂ solution prepared in a matching buffer.

Binding of calcium to sites I and II in the N-domain was monitored with λ_{ex} of 250 nm, and λ_{em} of 280 nm, based on the intrinsic phenylalanine fluorescence. Binding to sites III and IV in the C-domain was monitored with λ_{ex} of 277 nm, and λ_{em} of 320 nm, based on the intrinsic tyrosine fluorescence as previously described (VanScyoc et al., 2002). The fluorescent calcium indicator dye Oregon Green 488 BAPTA-5N (Oregon Green, 5.75 nM) (Molecular Probes, Eugene, OR) whose increase in fluorescence intensity was linearly proportional the $[\text{Calcium}]_{\text{free}}$ was used to determine the free calcium concentration at each point in the titration according to **Equation 2.1**, described previously by (VanScyoc et al., 2002)

$$[\text{calcium}]_{\text{free}} = K_d \frac{f_{[x]} - f_{\text{low}}}{f_{\text{high}} - f_{[x]}} \quad (2.1)$$

where f_{high} and f_{low} are the highest and lowest observed fluorescence intensity signals, respectively, observed for Oregon Green during the titration. The K_d of calcium binding to Oregon Green was determined previously to be 34.24 μM in 50 mM HEPES, 100 mM KCl, and 1 mM MgCl_2 (pH 7.4) at 22 C (λ_{ex} of 494 nm, λ_{em} of 521 nm) (VanScyoc et al., 2002). Each titration was repeated at least three times.

Free energies of calcium binding to the pair of sites in each domain were determined by fitting the titrations to a model-independent two-site (Adair) function (**Equation 2.2**), as described previously (VanScyoc et al., 2002),

$$\bar{Y} = \frac{K_1[X] + 2K_2[X]^2}{2(1 + K_1[X] + K_2[X]^2)} \quad (2.2)$$

where $[X]$ is free calcium, and the macroscopic association constant K_1 is the sum of intrinsic microscopic equilibrium constants ($k_1 + k_2$) for two sites: either sites I and II in the N-domain, or sites III and IV in the C-domain. This formulation allows the microscopic binding constants (k_1 and k_2) to be nonequivalent. The second macroscopic equilibrium constant K_2 ($k_1k_2k_c$) is the product of the intrinsic microscopic equilibrium

constants (k_1, k_2) and the cooperativity constant (k_c). The parameters ΔG_1 and ΔG_2 are macroscopic binding free energies, with $\Delta G_i = -RT \ln K_i$. The parameter ΔG_2 is thus the total free energy of saturating both calcium-binding sites in a domain.

Changes in fluorescence intensity for the calcium titrations were normalized to the highest and lowest experimentally determined signals. To account for experimental variations in the asymptotes of replicate titrations, we performed nonlinear least-squares analysis of the fluorescence intensity signal using the function $f(X)$, as given by **Equation 2.3**, as described previously (VanScyoc et al., 2002).

$$f(X) = Y_{[X]low} + \bar{Y} \cdot Span \quad (2.3)$$

where \bar{Y} refers to the average fractional saturation as described by **Equation 2.2**, and $Y_{[X]low}$ corresponds to the value of the fluorescence intensity in the absence of calcium. $Span$ refers the normalized range (0-1, or -1-0) of the data signal. The parameter $Span$ was negative in the case of a decreasing signal and positive in the case of an increasing signal. For monotonic titrations with well defined asymptotes, values for all parameters ($\Delta G_1, \Delta G_2, Y_{[X]low}$ and $Span$) were fit simultaneously using NONLIN (Johnson et al., 1981; Johnson and Frasier, 1985).

Note that a ratio of N:1 TFP:CaM does not indicate that all CaM molecules in the solution have N TFP molecules bound. Because it was not possible to determine the population distribution of TFP:CaM species in solution from an independent, experimentally observable property, the values of ΔG_2 determined in the presence of TFP are apparent values (i.e., ΔG_2^{app}).

NONLIN provides several measures of the goodness-of-fit for the parameters that minimize the variance of each fit. These error statistics include (a) the value of the square root of variance, (b) the values of asymmetric 65% confidence intervals, (c) the systematic trends in the distribution of residuals, (d) the magnitude of the span of residuals, and (e) the absolute value of elements of the correlation matrix. From these,

best-fit values were selected after testing multiple sets of initial guesses for parameters to probe for the presence of local minima. Free energies determined from at least three replicate titrations were averaged; those values and standard deviations are reported in **Table 2.1**.

In some titrations, the value determined for ΔG_1 was sensitive to starting guesses; in those cases, a manual grid search was conducted to obtain the lowest square root of variance. For titrations (**Figure 2.9**) that exhibited alternating increasing and decreasing calcium-dependent changes in fluorescence intensity, it was necessary to fix the values of both ΔG_1 and Span, as described in the *Results* section. Estimates of the apparent total free energy corresponding to each transition are reported separately in **Table 2.1**.

NMR Spectra

^{15}N -HSQC spectra were acquired at 25 °C on a Bruker Avance II 800 MHz US² spectrometer with a 5 mm TXI ^1H ($^{15}\text{N}/^{13}\text{C}/\text{D}$) probe featuring XYZ gradients. All spectra were processed in NMRPipe/NMRDraw (Delaglio et al., 1995b), while peak-picking and analysis were performed using SPARKY (Goddard and Kneller). TFP titrations of ^{15}N -PCaM₁₋₁₄₈ under apo conditions were carried out in 10% D₂O, 10 mM imidazole, 100 mM KCl, 50 μM EDTA, pH 6.5 at 22°C; in the case of $(\text{Ca}^{2+})_4$ -PCaM₁₋₁₄₈ TFP titration studies, 10 mM CaCl₂ was included. Starting volumes were 500 μL .

TFP titration of CaM Monitored by NMR

^{15}N -HSQC spectra of apo ^{15}N -PCaM₁₋₁₄₈ were acquired at incrementally increasing concentrations of TFP. The initial concentration of both apo- and $(\text{Ca}^{2+})_4$ - ^{15}N -PCaM₁₋₁₄₈ was 617 μM . In the TFP titration series performed under apo conditions, the $[\text{TFP}]_{\text{total}}$ was 0, 0.15, 0.29, 0.44, 0.61, 0.75, 0.90, 1.04, 1.21, 1.35, 1.49, 1.63, 1.79, 1.93, 2.07, and 2.37 mM (16 spectra). In the $(\text{Ca}^{2+})_4$ - ^{15}N -PCaM₁₋₁₄₈ TFP titration series, the $[\text{TFP}]_{\text{total}}$ was 0, 0.14, 0.28, 0.41, 0.58, 0.71, 0.85, 0.99, 1.15, 1.28, 1.41, 1.55, 1.70, 1.83, 1.96, 2.25, and 2.75 mM (17 spectra). The amide assignments for apo and $(\text{Ca}^{2+})_4$ - ^{15}N -

PCaM₁₋₁₄₈ in the absence of TFP were reported previously (Jaren et al., 2002). To determine the change in chemical shift upon TFP binding to apo and (Ca²⁺)₄-PCaM₁₋₁₄₈, chemical-shift changes in both the ¹H and ¹⁵N dimensions were quantified using the modified Pythagorean theorem shown in **Equation 2.4**, described previously (Jaren et al., 2002).

$$\Delta ppm = \sqrt{(\Delta^1 H ppm)^2 + (0.10134 \cdot \Delta^{15} N ppm)^2} \quad (2.4)$$

In this equation, Δppm refers to the linear change of a specific resonance peak from its initial starting position as TFP is titrated into solution, as done previously (Jaren et al., 2002).

Computational Modeling of TFP Binding

AutoDock Vina 1.0.3 (Trott and Olson, 2009) was used to simulate the binding of a single molecule of TFP to a fragment of CaM corresponding to the apo C-domain in three different tertiary conformations: “closed”, “semi-open” and “open”. Coordinates for residues 82-146 were extracted from these structures: 1DMO.pdb (apo, “closed”), 2IX7.pdb (apo, “semi-open”), 2HQW.pdb (calcium-saturated, “open”, bound to NR1C1 peptide), and 1LIN.pdb (calcium-saturated, “open”, 4 TFP molecules bound). To approximate an apo “open” structure (which has not been observed experimentally), calcium ions were removed from 2HQW and 1LIN. Each of the four protein fragments were placed in a cubic (45 Å³) search space with implicit water. The exhaustiveness parameter (number of times calculation was re-run) was 128. In **Figure 2.11A**, the 20 models that had the most favorable (lowest) free energies are depicted by PyMol™ v.1.2r2 (DeLano Scientific), using a gradient of green (most favorable) to white (least favorable) for the position of the unique sulfur atom in TFP. The remainder of each drug molecule is shown in light gray sticks. For each of the four CaM structures, CaM-TFP complexes calculated to have identical free energies are shown in the same color.

Results

The major aims of this study were to understand the allosteric effects of TFP on the domains of CaM by comparing the stoichiometry of TFP binding to apo and calcium-saturated domains of CaM, and determining thermodynamic effects of TFP on calcium-binding affinity.

TFP Titration of apo ^{15}N -PCaM

The stoichiometry of TFP binding to apo PCaM was determined using ^{15}N -HSQC spectra to examine changes in the local chemical environment of individual amide resonances as TFP was titrated into a solution of uniformly ^{15}N -labeled PCaM. The sample had been depleted of calcium via extensive dialysis against metal chelators.

In the absence of TFP, 126 resonances were identified for PCaM₁₋₁₄₈. TFP addition resulted in residue-specific perturbations of almost all of these resonances. Individual peaks found to be in fast exchange were tracked over the course of the TFP titration (a subset are shown in **Figure 2.4a-c**). This analysis revealed that TFP saturated apo PCaM at a stoichiometry of 2:1. At saturation by TFP, resonances corresponding to the C-domain of PCaM showed a greater average degree of chemical shift perturbation (Δppm of 0.044) than those of the N-domain (Δppm of 0.030) (**Figure 2.4c**). As shown in **Figure 2.4d**, residues having significant backbone amide chemical shifts ($\Delta\text{ppm} \geq 0.05$ a value used previously by Jaren et al., 2002 (Jaren et al., 2002)) are mapped onto corresponding residues of a high resolution solution structure of apo CaM in its extended form (Kuboniwa et al., 1995; Zhang et al., 1995).

The drug was observed to bind sequentially to the two domains of apo PCaM. The largest change in most resonances of the C-domain occurred in the range of 0 to 1 molar equivalents of TFP, whereas the largest change in most resonances of the N-domain occurred in the range between 1 and 2 molar equivalents of TFP. This indicated that TFP bound preferentially to the C-domain, despite the extensive similarity of the N-

and C-domains in sequence and structure. A distinct subset of residues (~30%) responded continuously over the range of zero to 2 molar equivalents of TFP. That group included Phe16, Ile86, Thr110, and Gly113; their response is shown in **Figure 2.4b**.

TFP Titration of $(\text{Ca}^{2+})_4$ - ^{15}N -PCaM

For calcium-saturated PCaM, 133 resonances were resolved. Saturation with TFP was reached at a ratio of 4:1 TFP:PCaM (**Figure 2.5**). Over the course of the titration, 17 resonances experienced slow or intermediate exchange, with the majority (13) of these located in the C-domain of $(\text{Ca}^{2+})_4$ -PCaM. Final chemical shift values due to TFP addition for these residues were unable to be determined, because only ^{15}N -HSQC spectra were collected for this study. Therefore, it was not possible to determine Δppm . Their positions are represented by the absence of a bar in **Figure 2.5c**.

Of the 116 resonances that were observed to be in fast exchange upon TFP addition, 98 were classified as being perturbed significantly ($\Delta\text{ppm} \geq 0.05$). They correlated closely with the location of TFP-binding sites observed in the crystallographic structure (1LIN.pdb) that showed 4 TFP bound to $(\text{Ca}^{2+})_4$ -CaM (**Figure 2.5d**). Although the calcium-binding sites of CaM are distant from the TFP-binding sites observed in all three of the crystal structures shown in **Figure 2.1**, some of their resonances were perturbed also.

Part of each ^{15}N -HSQC spectrum collected for the first and last point of the titration is shown overlapped in **Figure 2.5a**; representative titrations of individual residues are shown in **Figure 2.5b**. Of the 98 peaks that could be tracked throughout the titration, there were 74 that shifted monotonically; approximately half of those were in each domain of CaM (35 in the N-domain vs. 39 in the C-domain). The remaining 24 resonances exhibited a biphasic response to TFP addition. Some examples are shown in **Figure 2.6a-c**. Many of these residues were located at the interface between the N- and C-domains of CaM (**Figure 2.6d**).

This analysis of TFP-induced chemical shifts in $(\text{Ca}^{2+})_4\text{-}^{15}\text{N-PCaM}$ indicated that the TFP-binding sites were non-equivalent, and that some residues responded to TFP binding at more than one of its sites. Notably, some residues in the N-domain (e.g., Glu11 and Glu14) were among this group, even though the N-domain of calcium-saturated CaM has only been observed to have a single TFP-binding site.

Equilibrium Calcium Titration of CaM₁₋₁₄₈

To determine the effect of TFP on the affinity of calcium for CaM, equilibrium calcium titrations of CaM₁₋₁₄₈ were conducted in the presence of discrete molar ratios of TFP:CaM ranging from zero to eight. In the absence of TFP, calcium binding to sites I and II in the N-domain of CaM₁₋₁₄₈ was monitored by observing a decrease in intrinsic phenylalanine fluorescence intensity (**Figure 2.7a**, blue) as described in *Materials and Methods*. Nonlinear least squares analysis according to a model-independent two-site (Adair) function (Eq 2) established a reference total free energy (ΔG_2) of -13.05 ± 0.06 kcal/mol (**Table 2.1**). An increase in intrinsic tyrosine fluorescence intensity was used to monitor calcium binding to sites III and IV in the C-domain of CaM₁₋₁₄₈ (**Figure 2. b**, blue). In the absence of TFP, the total free energy was -15.00 ± 0.06 kcal/mol (**Table 2.1**).

Effect of TFP on Calcium Binding to CaM₁₋₁₄₈

Calcium titrations of CaM₁₋₁₄₈ were conducted at molar ratios of 1:1 (green), 2:1 (red), 3:1 (black), 4:1 (cyan), and 8:1 (purple) TFP:CaM₁₋₁₄₈ (**Figure 2.7, 2.8**). In these titrations, there is no experimental signal that reports directly on the number of TFP molecules bound to CaM or the fractional population of the possible ligation states of TFP bound to apo and calcium-saturated CaM. Therefore, each set of titrations will be referred to by the known independent variable: the ratio of the final mols of TFP to mols of CaM.

The calcium affinity of sites I and II of CaM₁₋₁₄₈ decreased or increased depending upon the concentration of TFP (**Figure 2.7a**). Of the TFP:CaM₁₋₁₄₈ ratios examined, a 1:1 ratio caused the largest decrease (2.08 kcal/mol) in calcium affinity at sites I and II (apparent free energy of -10.97 kcal/mol; green bar in inset). Calcium affinity was diminished at ratios of 2:1 (red) and 3:1 (black), but the effects were less severe than the ratio of 1:1. The smallest decrease (0.58 kcal/mol) in apparent free energy of binding at sites I and II occurred at a ratio of 4:1 TFP:CaM₁₋₁₄₈ (-12.47 kcal/mol, turquoise). In contrast, an 8:1 ratio reversed the effect and made calcium binding to sites I and II more favorable by -0.54 kcal/mol (relative to the binding affinity observed in the absence of TFP). This effect is represented by the bar graph inset in **Figure 2.7a** showing values of $\Delta\Delta G_2$. Although small in absolute magnitude, this reversal is considered significant because the standard deviation of replicate measurements for all of these titrations ranged from 0.04 to 0.16 kcal/mol, and was much smaller than 0.54 kcal/mol.

The effects of TFP on calcium sites III and IV of CaM₁₋₁₄₈ shared several features of its effects on sites in the N-domain. At all levels tested, TFP made calcium binding to sites III and IV of CaM₁₋₁₄₈ less favorable. The pattern of effects (**Figure 2.7b**) in response to an increasing ratio of TFP:CaM₁₋₁₄₈ was similar to that observed for sites I and II (**Figure 2.7a**). The bar graph inset shows that the free energy of -12.39 kcal/mol at a 1:1 TFP:CaM₁₋₁₄₈ ratio represented the maximum change in ΔG_2 of 2.61 kcal/mol. At a ratio of 2:1, the effect was slightly smaller; ratios of 3:1 and 4:1 both caused a decrease of ~1.8 kcal/mol in the calcium affinity of sites III and IV. A TFP:CaM₁₋₁₄₈ ratio of 8:1 had the smallest effect; the apparent ΔG_2 was -13.99 kcal/mol, representing a change of only 1 kcal/mol relative to the absence of TFP. In this set of titrations, the *Span* was positive for ratios of 0, 1:1, 2:1 TFP:CaM. At ratios above 2:1, the fluorescence intensity decreased in response to an increase in calcium. For ease of comparison of medians and

slopes of the titration, the signal for the titrations conducted at ratios of 3:1, 4:1 and 8:1 TFP:CaM are shown inverted.

The inversion of the C-domain Tyr signal may be attributed to multiple factors linked to TFP binding to CaM. One possible scenario that may alter the fluorescence properties of the C-domain Tyr signal may be TFP increasing the exposure of Tyr residues to the solvent in a Ca^{2+} -dependent manner resulting in quenching of the Tyr signal. Another possibility may be that dipole-dipole interactions occur between TFP and Tyr as well as between neighboring TFP that result in quenching of the Tyr signal upon Ca^{2+} addition. These possibilities are complicated due to the stoichiometry of TFP binding to CaM, as each ratio of TFP:CaM carries the potential to exhibit a unique fluorescence signal. Even further complicating the nature of the Tyr signal is the heterogeneous mix of CaM bound to TFP at varied ratios at the intermediate [TFP] examined. In these cases the fluorescent signal of CaM likely is the result of an ensemble of CaM in complex with TFP at various ratios based on the $6 \mu\text{M}$ [CaM] and $1\text{-}5 \mu\text{M}$ K_d of TFP previously reported (Massom et al., 1990b).

The domain-specific effects of TFP on calcium binding to CaM_{1-148} were complex, and suggested that the domains had intrinsic differences in affinity for TFP, and possibly stoichiometry of TFP binding. The NMR-monitored TFP titrations of CaM_{1-148} suggested that TFP might bind to an interface between domains, as well as a hydrophobic cleft in each domain. To attempt to simplify these linked binding processes, each half-CaM domain fragment (CaM_{1-80} and CaM_{76-148}) was studied independently. Each one contains a pair of EF-hands that retain (a) cooperative calcium binding energetics, and (b) secondary and tertiary structure nearly identical to that of full-length CaM.

Effect of TFP on Calcium Binding to CaM_{1-80}

Equilibrium calcium titrations of the CaM_{1-80} fragment (N-domain) were performed to examine the effect that TFP has on the calcium affinity of sites I and II in

the absence of the C-domain. Analysis of a calcium titration in the absence of TFP (**Figure 2.8a**, blue) showed that the total free energy of ΔG_2 of calcium binding to sites I and II was -12.91 kcal/mol (**Table 2.1**)

As was observed for CaM₁₋₁₄₈, the effect of TFP on the apparent free energy of calcium binding to sites I and II changed in magnitude in a nonlinear manner between ratios of 1:1 and 8:1 TFP:CaM (**Figure 2.8a**). At a ratio of 1:1, the apparent ΔG_2 was -11.12 kcal/mol, almost 2 kcal/mol less favorable than for CaM alone. This ratio of TFP:CaM₁₋₈₀ induced a smaller change than had been observed for calcium binding to sites I and II of CaM₁₋₁₄₈. A TFP:CaM₁₋₈₀ ratio of 2:1 reduced the calcium affinity further, such that ΔG_2 was -10.70 kcal/mol; this was the largest effect that TFP was observed to have on CaM₁₋₈₀, as shown in the bar graph inset of $\Delta\Delta G_2$ values in **Figure 2.8a**. A TFP:CaM₁₋₈₀ ratio of 3:1 had a slightly greater, but nearly identical effect, to a ratio of 4:1, consistent with its effect on sites I and II in CaM₁₋₁₄₈ at these ratios. The most striking difference was observed at the ratio of 8:1 TFP:CaM. Calcium binding to sites I and II remained less favorable by 0.63 kcal/mol for CaM₁₋₈₀. This indicated that the C-domain was necessary for the favorable effect (-0.54 kcal/mol) of TFP on sites I and II in CaM₁₋₁₄₈ that had been observed at an 8:1 TFP:CaM ratio.

Effect of TFP on Calcium Binding to CaM₇₆₋₁₄₈

To examine the effect that TFP had on the calcium affinity of sites III and IV in the absence of the N-domain, the free energy of calcium binding to the pair of sites in CaM₇₆₋₁₄₈ was determined. In the absence of TFP, ΔG_2 was determined to be -14.47 kcal/mol (Table 2. 1) (**Figure 2.8b**). A ratio of 1:1 TFP:CaM₇₆₋₁₄₈ led to a decrease in affinity (the apparent ΔG_2 was less favorable by 1.67 kcal/mol). This was smaller than the change (2.56 kcal/mol) observed for calcium binding to the same sites in CaM₁₋₁₄₈. The difference of almost 1 kcal/mol is greater than the largest standard deviation (0.26 kcal/mol) observed for a single condition. The calcium-dependent change in

fluorescence intensity was positive, as it was in the absence of TFP. However, the absolute magnitude of the intensity was lower (data not shown).

At ratios of 2:1 (red) and 3:1 (black) TFP:CaM₇₆₋₁₄₈, non-monotonic calcium-dependent changes in fluorescence intensity signals were observed. The first inflection was an increase in intensity, like that observed for calcium titrations conducted at a ratio of 1:1 TFP:CaM. Representative normalized data sets are shown in **Figure 2.8b**. The second inflection was a decrease in intensity; both the first and second transitions are shown in **Figure 2.9** for the ratios of 2:1 and 3:1. Apparent free energies were estimated using piece-wise analysis of the two transitions, as described below.

As shown in **Figure 2.8b**, at ratios of 4:1 and 8:1 TFP:CaM₇₆₋₁₄₈, a greater decrease in the calcium affinity of sites III and IV was observed than had been seen at the same ratio of TFP:CaM for these sites in CaM₁₋₁₄₈. Presumably this relates to the absence of the N-domain and interdomain sites as locations for TFP binding. Also, as had been observed for CaM₁₋₁₄₈, the *Span* observed for the calcium-dependent change in fluorescence intensity was negative. For ease of comparing medians and slopes of the titrations, the normalized titrations at these two conditions were inverted in **Figure 2.8b**. The slope of the calcium titration at a ratio of 8:1 TFP:CaM was notably more shallow than those at other molar ratios of TFP:CaM. This may arise from a change in cooperativity and/or may represent a mixed population of ligation states: CaM₇₆₋₁₄₈ saturated with varying numbers of TFP.

Piecewise Analysis of Biphasic Calcium Titrations of

CaM₇₆₋₁₄₈

The calcium titrations conducted at ratios of 2:1 and 3:1 TFP:CaM (**Figure 2.9**) are comprised of two phases with a sharp transition between them. Because the asymptotes for each phase were not well defined, it was not possible to determine an

independent maximum for the upward-trending signal, or minimum for the downward-trending signal by fitting the data to **Equation 2.3**.

Instead, to estimate the apparent free energy of calcium binding, the fluorescence signal was normalized to the maximal observed intensity, and the value of *Span* was set equal to 1.0. Using that approach, the apparent free energies of Ca^{2+} binding were -12.78 kcal/mol (at ratio of 2:1) and -13.02 kcal/mol at a ratio of 3:1. The corresponding estimates of $\Delta\Delta G_2$ for calcium binding to sites III and IV are shown in the solid bars in the inset of **Figure 2.8b**. These values were similar to what had been observed at 1:1 TFP:CaM. (The maximal fluorescence intensity for the increasing phase must be at least as high the value observed, but could be higher. If it were under-estimated, this approach would also under-estimate the effect of TFP by estimating a median calcium concentration lower than the actual value and therefore closer to the value in the absence of TFP.)

A similar approach was applied to analysis of the decreasing signal recorded at ratios of 2:1 (red) and 3:1 (black) TFP:CaM₇₆₋₁₄₈. The net downward deflection was fixed to be as large as that for the increasing phase. Using this approach, the apparent free energies were -9.55 (a ratio of 2:1) and -10.65 kcal/mol (at 3:1). The dashed bars shown in the inset of **Figure 2.8b** represent the value of $\Delta\Delta G_2$ values obtained assuming that the net change in affinity is equal to the effect represented by the decreasing fluorescent intensity. If the value of the *Span* of this transition were not as large as the increasing phase, this assumption would err on the side of reporting a weaker calcium-binding affinity (i.e., a median concentration for the titration that would be higher than the actual one).

The presence of multi-phasic fluorescence signals, changes in direction of calcium-dependent response of steady-state fluorescence, and differing free energies of Ca^{2+} binding as a function of [TFP] provide strong evidence for the existence of populated intermediates that have different fluorescence signals. All estimates of calcium

binding affinity in the presence of TFP are denoted as apparent free energies to draw attention to the complexity of analysis of multiple, partial ligation states.

Discussion

The studies presented here address the nature of TFP binding to apo and calcium-saturated CaM, and the allosteric effects of TFP on calcium binding to the non-equivalent domains of CaM. Their combined effects on conformational switching of this essential regulatory protein are of interest because of the ubiquitous practice of applying drugs to cell cultures to disrupt CaM-mediated pathways of calcium-dependent signal transduction.

Two TFP molecules bind to apo CaM

Although some reports have suggested that TFP binds to apo CaM (Matsushima et al., 2000) (Matsushima et al., 2007) most have not supported this premise (Massom et al., 1990a; Massom et al., 1991; Tanokura and Yamada, 1985). However, stoichiometric TFP titrations of apo CaM₁₋₁₄₈ (460 μ M) monitored by NMR showed it to be saturated by two TFP molecules, with preferential binding to the C-domain; studies of calcium binding to TFP-saturated apo CaM demonstrated that TFP reduced calcium affinity. This was similar to the effect of peptides derived from individual protein targets, such as those containing IQ-motifs, that bind preferentially to apo CaM (Martin and Bayley, 2004; Theoharis et al., 2008). They also have the thermodynamic property of decreasing the calcium-binding affinity of the EF-hand sites of CaM. However, to our knowledge, this is the first time such behavior has been reported for a drug binding to CaM.

In a unique high resolution study of apo CaM bound to a peptide representing an IQ-motif (from myosin V (Houdusse et al., 2006)), the C-domain of CaM adopted the “semi-open” tertiary conformation (**Figure 2.2**). The interface between the peptide and the C-domain buries more surface area than does the peptide interaction with N-domain which is in the “closed” conformation. In other structures of CaM:peptide complexes,

the C-domain has been observed to adopt multiple conformations, depending on the nature and number of ligand(s) (calcium and/or protein) bound (**Figure 2.2**). Solution studies of apo CaM alone have shown that the C-domain has a lower fraction of ordered secondary structure and is less thermodynamically stable than the N-domain (Masino et al., 2000; Sorensen and Shea, 1998). These findings indicate that, under apo conditions, fluctuation between a “closed” and “semi-open” conformation is more energetically favorable for the C-domain than for the N-domain, consistent with TFP binding preferentially to the C-domain. It is also possible for either apo N- or C-domain to sample the “open” conformation. However, favorable tertiary constraints within each domain provide an energetic barrier for this transition. Thus, the population of this conformation of apo CaM will be low.

Small-angle x-ray scattering (SAXS) data indicate that TFP binds to both apo and $(\text{Ca}^{2+})_4$ -CaM. However, the radius of gyration of each ensemble is different ($(\text{Ca}^{2+}$ -CaM:TFP = $20.5 \pm 0.3 \text{ \AA}$, versus apo CaM $20.5 \pm 0.3 \text{ \AA}$) (Matsushima et al., 2000; Matsushima et al., 2007), suggesting that the dominant tertiary structure and stoichiometry of TFP binding are not identical for both apo and $(\text{Ca}^{2+})_4$ -CaM. Because both of these differ from apo CaM alone (radius of gyration = $21.5 \pm 0.3 \text{ \AA}$) which preferentially samples the “closed” conformation, we hypothesize that TFP binds preferentially to the “semi-open” conformation of the 4-helix bundle domains of apo CaM. At any specific level of TFP, the fraction of apo CaM having TFP bound to a “semi-open” domain will be determined by the energy of isomerization reactions needed for conformational rearrangements, the energy of TFP binding to CaM and concentration of TFP.

Additional evidence that TFP recognizes different sites within apo and $(\text{Ca}^{2+})_4$ -CaM comes from comparing ^{15}N -HSQC spectra of TFP-saturated apo and $(\text{Ca}^{2+})_4$ -CaM (**Figure 2. 10**). The spectra differ at most positions, meaning that the local chemical environments of most amide bonds in the CaM backbone are dissimilar. Note that the

changes observed in this study appear to be considerably larger than those observed by Matsushima et al. (Matsushima et al., 2007); although a direct comparison cannot be made because chemical shifts due to TFP binding were not quantified in that study.

It would be attractive to determine a high-resolution structural model of TFP bound to apo CaM. This would allow us to determine residues participating in the drug-protein interfaces and interhelical angles of each 4-helix bundle domain. However, it is beyond the scope of this study. Instead, a computational approach (*AutoDock Vina*) (Trott and Olson, 2010) was used to identify an ensemble of preferred binding sites for TFP on the apo C-domain of CaM in a “closed”, “semi-open” and “open” conformation (**Figure 2. 11a**).

For each tertiary structure, an overlay of the 20 models that were most favorable energetically are shown. The single sulfur atom in each computationally docked TFP molecule is shown as an enlarged sphere. The color of that sphere corresponds to the predicted free energy of binding (darkest green corresponds to most favorable positions); the range of predicted energies of the models is shown in the bar below. The docking results of *AutoDock Vina* have been validated in control experiments in which a ligand was extracted from a known complex and then successfully re-docked in a similar orientation by *AutoDock Vina* as was observed experimentally (Trott and Olson, 2010). *AutoDock Vina* was also validated for use in TFP binding to CaM, by extracting TFP from the x-ray structure of $(\text{Ca}^{2+})_4$ -CaM bound to 1 TFP and then allowing *AutoDock Vina* to determine where the extracted TFP molecule would bind. Upon completion of docking simulation, it was observed that the predicted TFP binding site was less than Å away from the experimentally determined TFP binding site (**Figure 2.11a**).

For the “closed” C-domain (based on 1DMO), the preferred binding locations of TFP were on the exterior surface near the first and second helix of the domain, and near the highly acidic calcium-binding sites III and IV; predicted free energies for this set of models ranged from -6 to -5.1 kcal/mol. For the “semi-open” C-domain (based on 2IX7),

there were two preferred binding locations: one was in the shallow cleft between the pairs of helices in the 4-helix bundle and the other was near site III. Free energies of TFP binding to the “semi-open” domain ranged from -6.8 to -6.0 kcal/mol.

The “open” C-domain has only been observed in structures of calcium-saturated CaM. However, it may be sampled at a very low frequency by apo CaM. Therefore, TFP binding to an apo “open” C-domain was modeled by removing the calcium ions from two “open” tertiary structures of calcium-saturated CaM that differed in their side chain orientations. One set of coordinates was taken from a structure of $(\text{Ca}^{2+})_4$ -CaM bound to a peptide (2HQW) and another was from a structure of $(\text{Ca}^{2+})_4$ -CaM bound to 4 TFP (1LIN). In both cases, the most favorable binding site for TFP was located deep in the hydrophobic pocket between the pairs of helices.

The “open” conformation is the only one for which there are high resolution structures showing the location(s) of TFP bound to the C-domain. The position of TFP at site A of 1LIN.pdb (see **Figure 2.3**) is shown in magenta for comparison to the models. This is the site that is occupied in all three of the crystallographic structures of TFP bound to $(\text{Ca}^{2+})_4$ -CaM. For the 20 models having the lowest energy, the sulfur atom in each computationally docked TFP molecule was within 1 Å of the location where it had been observed experimentally in 1LIN, suggesting that the interhelical angles and surface residues are necessary and sufficient to provide a binding site for TFP in the absence of calcium.

This prediction of sites of TFP binding to the “closed” and “semi-open” conformations of a single domain does not explore additional possible sites that might exist in full-length CaM in pockets created by the juxtaposition of the two domains. However, the models suggest that TFP binding to the “semi-open” form has the potential of interfering with calcium binding. Note that all of these calculations have CaM account only for TFP binding, and not for the energy required for conformational isomerization. That barrier exists, in part, because the “open” form of each 4-helix bundle domain of

(Ca²⁺)₄-CaM exposes more hydrophobic surface to solvent than the “closed” or “semi-open” conformations (Houdusse et al., 2006; Kuboniwa et al., 1995).

Four TFP bind to (Ca²⁺)₄-CaM

Residue-specific titrations monitored by NMR (**Figure 2.5**) showed the stoichiometry of TFP binding to (Ca²⁺)₄-CaM was 4:1 in agreement with a SAXS study (Matsushima et al., 2000), an HPLC study (Massom et al., 1991)(Massom et al., 1990c), and one of the three crystallographic structures 1LIN.pdb (Vandonselaar et al., 1994a). The stoichiometry of 4 contrasts with two other crystallographic structures of CaM:TFP (1CTR.pdb, 1A29.pdb, see **Figure 2.3**), and a recent computational study that concluded that TFP binds only to the C-domain of (Ca²⁺)₄-CaM (Kovesi et al., 2008). Although NMR is a powerful method for precisely determining the stoichiometry of binding, the observed spectral changes report on changes in chemical environment that may arise from local binding, or a global conformational change. Thus, it is challenging to determine the location of individual binding sites when multiple ligands bind. It was evident that TFP binding perturbed amide resonances in both domains of (Ca²⁺)₄-CaM, consistent with TFP binding to each, as depicted in the superposition shown in **Figure 2.3d** (sites A, B, C, and D). The majority of residues in slow exchange mapped to the C-domain, indicating that this domain of (Ca²⁺)₄-CaM contained the site with highest affinity for TFP. This observation, coupled with the locations of residues in both domains that undergo fast exchange, indicates that a hierarchy of 4 TFP-binding sites is present in (Ca²⁺)₄-CaM. Interpreted according to the positions of TFP in 1LIN.pdb, it appeared that two TFP binding sites with different affinities exist in the C-domain, that a third low-affinity site is present in the N-domain, and that a fourth site (also of low affinity) bridges the two domains (**Figure 2.3**). Although a hierarchy of TFP binding sites was identified in this work, it was not possible to distinguish a preferential order of binding order to the low-affinity sites.

Interdomain Interactions

For any protein binding 4 ligands, there are 5 macroscopic ligation states (0, 1, 2, 3, 4 ligand:protein). Thus, in principle, it might be possible to titrate $(\text{Ca}^{2+})_4\text{-CaM}$ with TFP and monitor 4 independent transitions corresponding to individual TFP-binding sites as has been done for calcium binding to 4 sites in CaM (Jaren et al., 2002), (*Martin et al., 1986; Starovasnik et al., 1992*). For residues of CaM affected by a single TFP molecule, a monotonic transition between a “free” and “bound” state might be observed. For each residue that experienced only those two chemical environments, a stoichiometric titration would show (a) a linear transition, if in fast exchange, or (b) reciprocal changes in intensity for pairs of peaks (one diminishing, one increasing), if in slow exchange. Similarly, if there were 4 sites with identical affinity, all residues affected by TFP binding would titrate identically over the range of 0 to 4 equivalents of TFP added.

However, it is also possible that intermediate ligation states adopt distinct conformers with unique biophysical properties. A residue that responds to TFP binding at multiple sites has the potential to experience a different environment in each, and therefore show a nonlinear response to TFP binding as monitored by NMR or fluorescence. In HSQC spectra, this was observed for a subset of residues (**Figure 2.6b**) that experienced at least three chemical environments and sampled at least one intermediate conformation. These residues responding to multiple TFP-binding sites are most likely located at the interface between the N- and C-domains, or between TFP binding sites within a single domain (**Figure 2.6d**). The crystallographic structure of 4 TFP molecules bound to $(\text{Ca}^{2+})_4\text{-CaM}$ shows that, at their closest approach, TFP binding sites A, B, and C are in close proximity ($\sim 4 \text{ \AA}$) to each other, while the TFP molecule bound at site D is $\sim 9 \text{ \AA}$ away from site C (**Figure 2.3d**). This constellation would allow for unique chemical environments to be sampled as TFP sequentially fills its 4 binding sites, and would lead to changes in the chemical environment of adjacent TFP binding sites.

This type of biphasic response of CaM resonances was observed previously in calcium titrations monitored by ^{15}N -HSQC that showed that several residues within the linker region between domains of CaM experienced three distinct chemical environments (Jaren et al., 2002). These residues were in slow exchange between 0 and 2 equivalents of calcium (i.e., during saturation of the C-domain), and were in fast exchange between 2 and 4 (i.e., during saturation of the N-domain). Similar biphasic responses were observed in drug titrations of cardiac Troponin C (a related calcium-binding EF-hand protein) (Kleerekoper et al., 1998). Nonlinear peak shifts due to the significant population of an intermediate state have also been observed for other proteins such as the phosphorylated kinase-inducible activation domain (pKID) of the transcription factor camp response element-binding protein (CREB) binding to subdomain of CREB Binding Protein (CBP) termed KIX (Sugase et al., 2007).

Further evidence for domain interactions was provided by the behavior of residues Glu11 and Glu 14, located in the first helix of the N-domain of $(\text{Ca}^{2+})_4\text{-CaM}$. It was expected that these residues would respond to TFP binding to the N-domain itself based on their location and proximity to a target peptide or drug observed in 17 $(\text{Ca}^{2+})_4\text{-CaM-peptide}$ or drug complexes (Ataman et al., 2007). In those, both Glu11 and Glu14 were within 4.5 Å of the peptide or drug interacting with $(\text{Ca}^{2+})_4\text{-CaM}$. Over the full range of 0 to 4 TFP molecules binding to $(\text{Ca}^{2+})_4\text{-CaM}$, these residues exhibited a biphasic response to TFP binding (**Figure 2.6d**), initially increasing with a maximum at 2 TFP:CaM. As observed in TFP-CaM structures shown in **Figure 2. 2**, these residues are located between TFP-binding sites in the N- and C-domain of $(\text{Ca}^{2+})_4\text{-CaM}$ which positions them to respond to saturation of all TFP binding sites. The highest affinity TFP-binding site in $(\text{Ca}^{2+})_4\text{-CaM}$ is in the C-domain, assumed to be TFP-site A (**Figure 2.3d, Figure 2. 1a**). Glu11 and Glu14 are also < 4 Å from TFP-site B which is comprised primarily of C-domain residues. This hints at the possibility that the response of Glu11 and Glu14 from 0 to 2 relates to occupancy of sites A and B, and the response

from 2 to 4 indicates occupancy of sites C and D. But, other models of hierarchical binding are also consistent with the titrations.

Effects of TFP on the Calcium Affinity of CaM

Most proteins known to be regulated by CaM contain a BAA motif (basic amphipathic alpha-helix) that binds to CaM, and causes an increase in the calcium affinity of CaM. Thermodynamic linkage requires that the BAA motif bind to $(Ca^{2+})_4$ -CaM with higher affinity than it binds to apo CaM in order to increase the Ca^{2+} -binding affinity of CaM. A subset of CaM-target interactions—typically those between CaM and targets bearing IQ-motifs—lead to a reduction in the calcium affinity of CaM, due to the higher affinity of these targets for apo CaM (Bahler and Rhoads, 2002; Cui et al., 2003; Martin and Bayley, 2004; Mori et al., 2003; Putkey et al., 2003). A peptide (Nav1.2 IQp) representing the IQ-motif from the Voltage-Dependent Sodium Channel Nav1.2 has been shown to have a negligible effect on calcium binding to the N-domain of CaM, while significantly lowering calcium-binding affinity of the C-domain (Theoharis et al., 2008). To satisfy thermodynamic constraints, an IQ-motif with this property has a higher affinity for the apo C-domain than the $(Ca^{2+})_2$ -C-domain.

Like an IQ-motif, at most concentrations studied in this study, TFP diminished the calcium-binding affinity of both domains of CaM. TFP binds with lower absolute affinity to apo CaM (Massom et al., 1990a) than does Nav1.2 IQp (Theoharis et al., 2008). However, like Nav1.2 IQp, it has a higher relative affinity for the C-domain of apo CaM (**Figure 2.5**) than for the N-domain. This is consistent with the observation that the C-domain exhibited a larger TFP-induced decrease in calcium affinity than the N-domain (**Figure 2. b**).

If $(Ca^{2+})_4$ -CaM had not bound TFP at all, or bound TFP more weakly than apo CaM but at the same sites, then the major allosteric effect of TFP would be to decrease calcium affinity by binding preferentially to apo CaM. The magnitude of the TFP effect

would increase monotonically until CaM was saturated with TFP. In this way, its effect on CaM would be analogous to that of 2,3-BPG reducing oxygen binding affinity by binding preferentially to deoxy hemoglobin (Ackers, 1979; Arnone, 1972; Benesch and Benesch, 1967). Mammalian adaptation to high altitudes depends on this mechanism of promoting oxygen release from hemoglobin under the low-oxygen conditions of human tissues (Martin et al., 1975). However, this mechanism of negative allosteric regulation would not explain how the effect of TFP on calcium binding reversed direction (**Figures 2.7 and 2.8**) when the ratio of TFP:CaM increased from 1 to 8. Several possible explanations were considered.

The reversal of the initially negative allosteric effect of TFP on calcium binding by CaM might be explained if higher levels of total added TFP did not actually represent higher soluble concentrations. For example, the effective concentration of TFP might drop if it formed micelles that would compete with CaM as a sink for additional TFP. However, that micelle-sink model contradicts several observations in this study (Caetano et al., 2003; Caetano and Tabak, 2000). For example, a prediction of that model is that increasing TFP would ameliorate the initially negative effect until all of it was drawn into micelles and the calcium-binding affinity of CaM returned to that observed in the absence of TFP. Instead, an increase in TFP ultimately increased the calcium-binding affinity of N-domain of CaM, rather than returning it to the values in the absence of TFP. An additional contrary observation was that the direction of calcium-dependent changes in fluorescence signal changed over the course of the TFP titration: monotonically increasing in the absence of TFP, and monotonically decreasing at the 8:1 ratio, showing that TFP was still associated with CaM at the 8:1 ratio. Finally, in all CaM-TFP samples (including the millimolar CaM samples used in NMR studies), there was no visual evidence of turbidity that would indicate the formation of a significant population of micelles.

Another possibility is that the mechanism of allosteric reversal depends on TFP changing the relative populations of “closed”, “semi-open”, and “open” tertiary structures of apo CaM. Given that both calcium and TFP have micromolar affinity for CaM, and that two calcium ions are needed to drive each domain to adopt the “open” state in the absence of TFP, it may be that 2 TFP molecules are needed to drive the conformational change of opening each domain. The NMR-monitored TFP titrations showed only two TFP bound to apo CaM₁₋₁₄₈, but they did so sequentially with higher affinity for the C-domain. In contrast, multiple TFP may bind cooperatively to “open” calcium-saturated domains. Thus, as the level of TFP increases above 1:1, there is a chance for more than 1 to bind to a single domain. TFP may promote the “open” conformation of apo CaM in a manner similar to that of BAA-motif peptides that bind with high affinity to the hydrophobic surfaces exposed upon Ca²⁺ binding (Ataman et al., 2007; Meador et al., 1993). That would then increase calcium-binding affinity. This biphasic binding shares some features with that observed for an IQ-motif from neuromodulin studied by Persechini and colleagues (Black et al., 2006).

Comparison of the calcium-binding free energies for sites I and II in CaM₁₋₁₄₈ and CaM₁₋₈₀ at a ratio of 8:1 TFP:CaM revealed that the calcium affinity does not become more favorable than that in the absence of TFP. This difference may result from the loss of a TFP-binding site that bridges the N- and C-domains of CaM₁₋₁₄₈ with contributions from both (**Figure 2.12**). Calcium titrations of CaM₇₆₋₁₄₈ at ratios of 2:1 and 3:1 TFP:CaM₇₆₋₁₄₈ resulted in multiphasic fluorescence signals attributed to a mix of different TFP-bound CaM₇₆₋₁₄₈ complexes that each have unique properties. This is consistent with kinetic studies of calcium release from (Ca²⁺)₄-CaM mixed with TFP (Martin et al., 1985). The different species in solution are sufficiently populated at ratios of 2:1 and 3:1 TFP:CaM₇₆₋₁₄₈ to exhibit multiple signals, but are not abundant at the ratios of 1:1, 4:1, and 8:1. For the ratios of 2:1 and 3:1, the maximum value of raw fluorescent intensity was approximately a third of that observed for the calcium titrations conducted at 1:1 and

4:1 ratios of TFP:CaM₇₆₋₁₄₈. All of these observations are consistent with a mechanism of drug action that specifies that the apo C-domain has a higher affinity for TFP than does the apo N-domain, and that TFP binding interferes with calcium binding by inhibiting the conformational switch from “semi-open” to “open” conformation.

Unlike Ca²⁺-titration curves at 1, 2, 3, or 4:1 TFP:CaM₇₆₋₁₄₈, a significant decrease in cooperativity of Ca²⁺ binding to CaM₇₆₋₁₄₈ was observed at a ratio of 8:1 (**Figure 2.8**). This decrease in cooperativity may be a result of multiple factors such as TFP uncoupling Ca²⁺-binding sites III and IV such that one site has a more favorable Ca²⁺-binding affinity than the other and/or a heterogeneous mix of TFP bound CaM₇₆₋₁₄₈ species in solution.

These interpretations are summarized in a simplified isomerization and binding model shown in **Figure 2.11b**. Distinct tertiary conformations of CaM provide unique TFP binding interfaces with different relative affinities for TFP. A “closed” domain of apo CaM is depicted as having no interaction with TFP. A “semi-open” apo domain has hydrophobic residues located sufficiently near the perimeter of the canonical target binding pocket (blue patches) to interact with TFP. With TFP bound at these positions, the domain responds as it does when interacting with the Ile-Gln dipeptide found in a canonical IQ-motif of ion channels. Calcium binding is sufficient to switch the tertiary structure from the “semi-open” to “open” conformation, exposing hydrophobic residues (indicated by blue patches) located deep in the hydrophobic cleft. The “open” conformation may also be sampled by apo CaM, allowing TFP at high concentrations to bind there. In either case, TFP bound at the blue sites may have an allosteric effect on calcium affinity more like that of a BAA motif peptide which consistently buries an aromatic group in the FLMM pocket of the C-domain (Ataman et al., 2007; Yamniuk and Vogel, 2004).

Summary

This study of the allosteric regulation of calcium binding to CaM by TFP demonstrates that there is considerable complexity in the interactions between these two ligands of CaM. TFP interacts with distinct interfaces available in the dominant tertiary conformations of apo and $(\text{Ca}^{2+})_4\text{-CaM}$. These are likely to be primarily a “semi-open” state in the ensemble of conformations that are energetically sampled by apo CaM, and the “open” conformation for $(\text{Ca}^{2+})_4\text{-CaM}$. TFP lowered the calcium binding affinity of sites III and IV in the C-domain of CaM_{1-148} more than sites I and II in the N-domain, indicating that the apo C-domain has a higher affinity for TFP. TFP titrations monitored by NMR showed differences in the stoichiometry and location of binding of TFP to apo and $(\text{Ca}^{2+})_4\text{-CaM}$; this is the driving force behind the non-monotonic allosteric effect of TFP on the calcium affinity of CaM.

This analysis suggests that despite the significant number of calcium-dependent processes regulated by $(\text{Ca}^{2+})_4\text{-CaM}$, it is equally important to consider interactions of target proteins with apo CaM when testing drugs similar to TFP. This broadens the interpretation of a widely used approach of bathing cells *in vitro* in a TFP-containing solution with the goal of disrupting pathways that are regulated by $(\text{Ca}^{2+})_4\text{-CaM}$.

Like other anti-psychotic drugs, TFP can cause the debilitating side-effect of tardive dyskinesia (Lahti et al., 1993). The etiology of this is not completely understood, but believed to originate from hypersensitive dopamine receptors, which have been shown to be regulated by CaM under both apo and calcium-saturating conditions (Liu et al., 2007; Woods et al., 2008). Apo and $(\text{Ca}^{2+})_4\text{-CaM}$ have also been shown to regulate numerous ion channels responsible for the propagation of nerve impulses (Ataman et al., 2007; Schumacher et al., 2004; Shah et al., 2006; van Petegem et al., 2005). It is possible that TFP alters physiological processes by disrupting interactions of apo CaM with these receptors.

Given the large number of signaling pathways that CaM regulates, CaM itself is not a promising target for drug design. However, an interface between CaM and a particular target may offer more selectivity. This study of allosteric interactions between calcium and TFP suggests that interactions between channels and the “semi-open” form of CaM may be an especially attractive target for future drug testing.

Table 2.1 TFP Effects on Calcium Binding to CaM

Sites I and II of CaM ₁₋₁₄₈				Sites III and IV of CaM ₁₋₁₄₈		
[TFP]	Ratio	ΔG_2^a	$\Delta\Delta G_2^b$		ΔG_2^a	$\Delta\Delta G_2^b$
-		-13.05 ± 0.06	-		-15.00 ± 0.06	-
6 μ M	1:1	-10.97 ± 0.11	2.08		-12.39 ± 0.07	2.61
12 μ M	2:1	-11.57 ± 0.11	1.48		-12.76 ± 0.18	2.24
18 μ M	3:1	-11.60 ± 0.16	1.45		-13.23 ± 0.20	1.77
24 μ M	4:1	-12.47 ± 0.04	0.58		-13.07 ± 0.09	1.83
48 μ M	8:1	-13.59 ± 0.08	-0.54		-13.99 ± 0.06	0.99

Sites I and II of CaM ₁₋₈₀				Sites III and IV of CaM ₇₆₋₁₄₈		
[TFP]	Ratio	ΔG_2^a	$\Delta\Delta G_2^b$		ΔG_2^a	$\Delta\Delta G_2^b$
-		-12.91 ± 0.09	-		-14.47 ± 0.12	-
6 μ M	1:1	-11.12 ± 0.03	1.79		-12.80 ± 0.08	1.67
12 μ M	2:1	-10.70 ± 0.17	2.21		-12.78 ± 0.20^c -9.55 ± 0.31^d	1.69 4.92
18 μ M	3:1	-11.37 ± 0.05	1.54		-13.02 ± 0.18^c -10.65 ± 0.27^d	1.45 3.82
24 μ M	4:1	-11.56 ± 0.05	1.35		-11.56 ± 0.05	2.91
48 μ M	8:1	-12.28 ± 0.18	0.63		-12.71 ± 0.21	1.76

^a ΔG_2 (kcal/mol) represents apparent total free energy indicated $\text{TFP}]_{\text{total}}/[\text{CaM}]_{\text{total}}$ ratio

^b $\Delta\Delta G_2 = \Delta G_2^{\text{app}} (\text{TFP Added}) - \Delta G_2 (\text{TFP Absent})$

^c Values determined from initial phase of increasing fluorescent signal.

^d Values determined from decreasing fluorescent signal at higher [calcium].

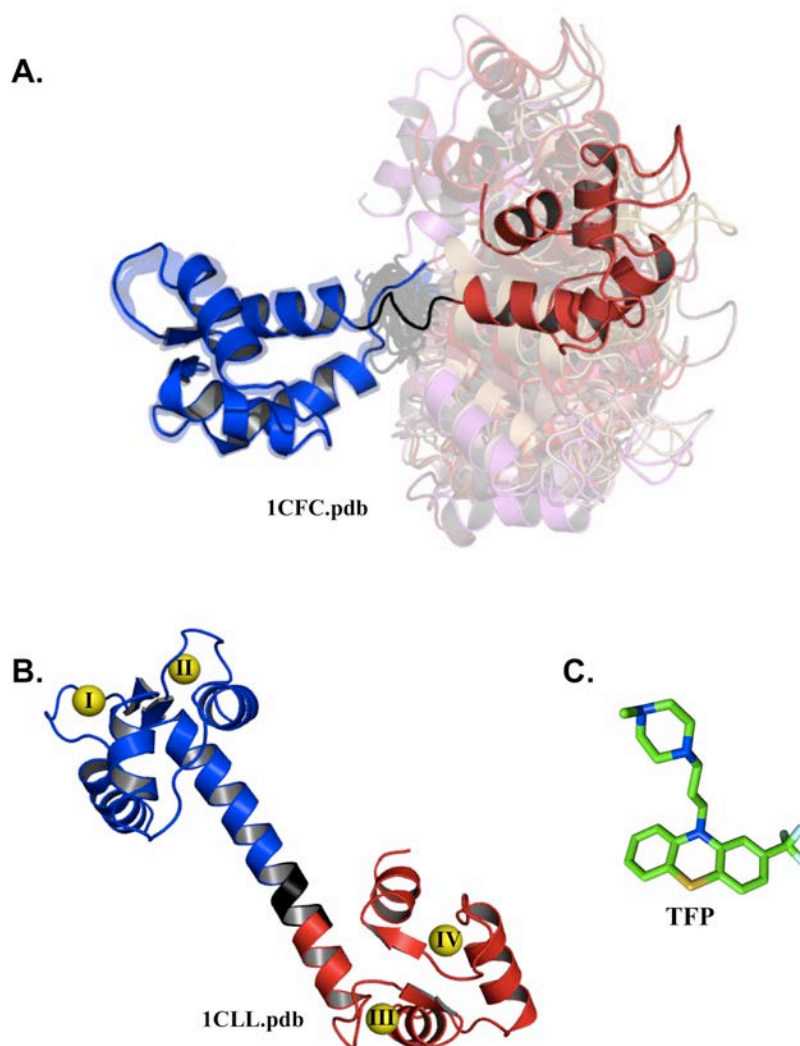


Figure 2.1: Structures of apo CaM, (Ca²⁺)₄-CaM, and Trifluoperazine.
 A: Superposition of solution structure models of CaM (1CFC.pdb) determined by NMR. Alignment minimized the difference between models with respect to the N-domain (residues 1-75 in blue), illustrating flexibility of interdomain linker (residues 76-80 in black) and range of positions adopted by C-domain (residues 81-148 in red). A single model is highlighted to reveal the tertiary structures of the apo N- and C-domains B: (Ca²⁺)₄-CaM structure determined crystallographically (1CLL.PDB); backbone colored as in panel A. Ca²⁺ ions (yellow) are bound at sites I and II in the N-domain, and at sites III and IV in the C-domain. C: Chemical structure of the antipsychotic drug Trifluoperazine (TFP; green), with sulfur atom in yellow and fluorine atoms in light blue. Users/nmr_mike/Thesis/Chapter_II/Figure2_1.jpg

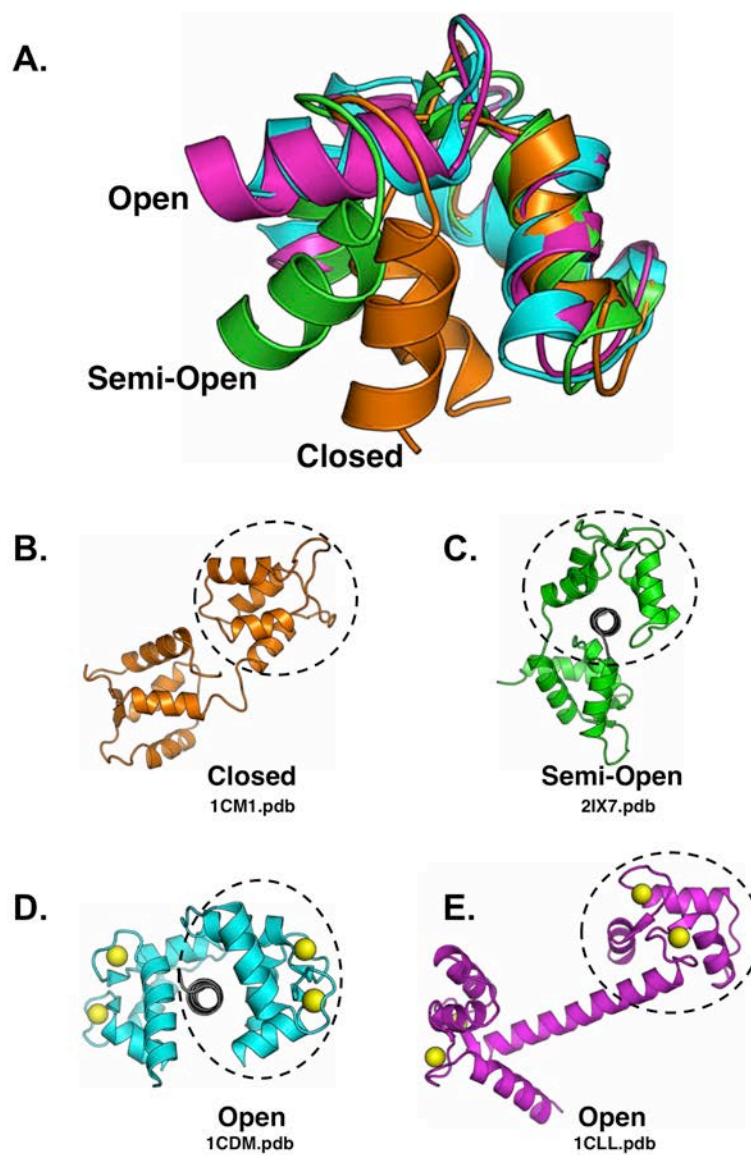


Figure 2.2: Superposition of 3 Tertiary structures of the C-domain of CaM. Examples of the “closed” (1CFC.pdb–orange), “semi-open” (2IX7.pdb–green) and “open” (1CDM.pdb–aqua and 1CLL.pdb–magenta) conformations of the C-domain of CaM are aligned according to the positions of the F and G (second and third) helices of the domain. Structures of the corresponding full-length CaM is shown below. Users/nmr_mike/Thesis/Chapter_II/Figure2_2.jpg

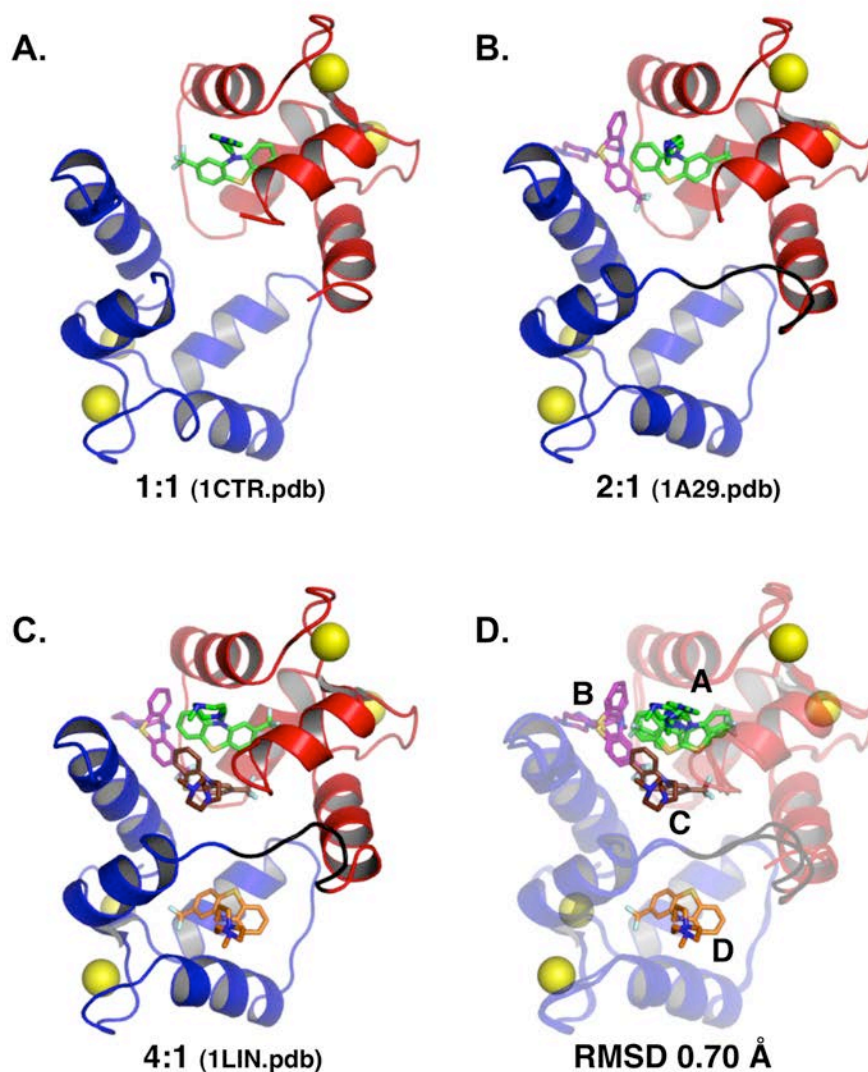


Figure 2.3: Structures of $(\text{Ca}^{2+})_4\text{-CaM}$ bound to TFP. Individual panels show crystallographically derived structures of TFP: $(\text{Ca}^{2+})_4\text{-CaM}$ complexes, with drug:protein ratios of 1:1 (1CTR.pdb), 2:1 (1A29.pdb), and 4:1 (1LIN.pdb), as well as a structural superposition of these three structures, with the TFP-binding sites labeled A (green), B (magenta), C (brown) and D (orange). TFP-binding sites A and B are located in the C-domain (backbone red), site C bridges the two domains, and site D is located in the N-domain (backbone blue).
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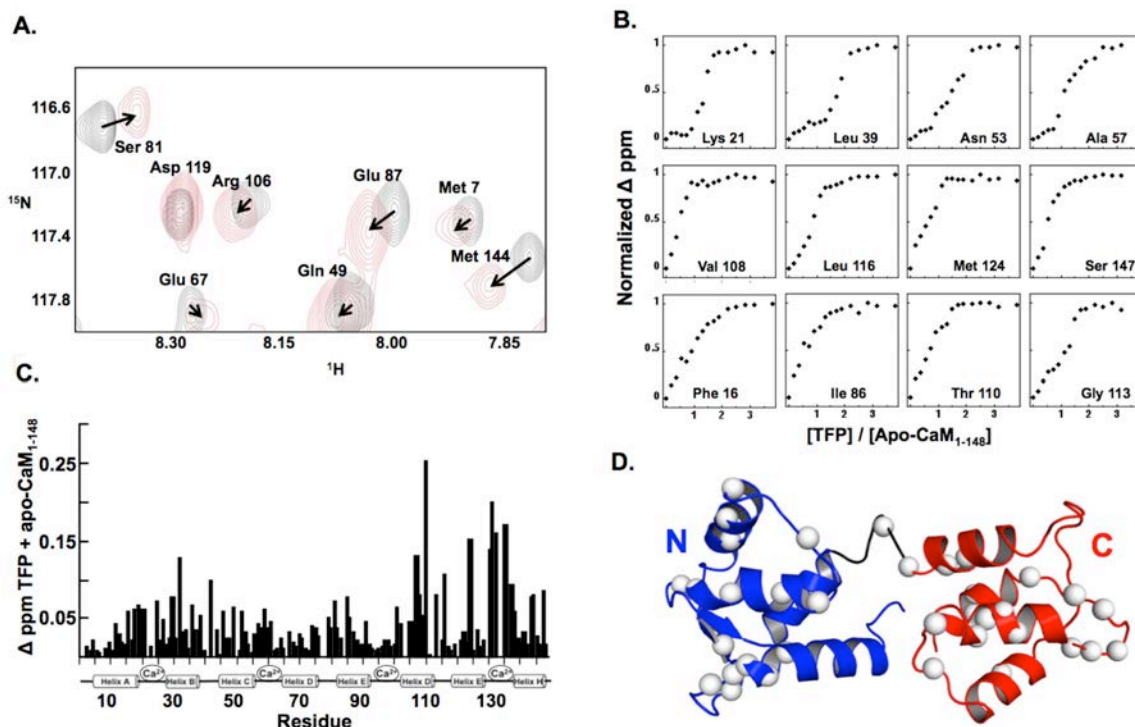


Figure 2.4: ^{15}N -HSQC-monitored TFP titration of uniformly ^{15}N -labeled apo PCaM
 A: Comparison of subset of ^{15}N -HSQC spectra for apo PCaM (black) and TFP-saturated apo PCaM (red); arrows indicate change in resonance positions over the course of the TFP titration. B: Normalized TFP-induced chemical shifts of individual representative residues of apo CaM. C: Bar graph of net chemical shift per residue caused by TFP saturation of CaM. D: Location of each apo CaM residue whose chemical shift was perturbed > 0.05 ppm by TFP saturation (white spheres); backbone modeled as that of apo CaM (1DMO.pdb). Solution conditions: 10% D_2O , 10 mM imidazole, 100 mM KCl, 50 μM EDTA, 5mM, pH 6.5 at 22°C.

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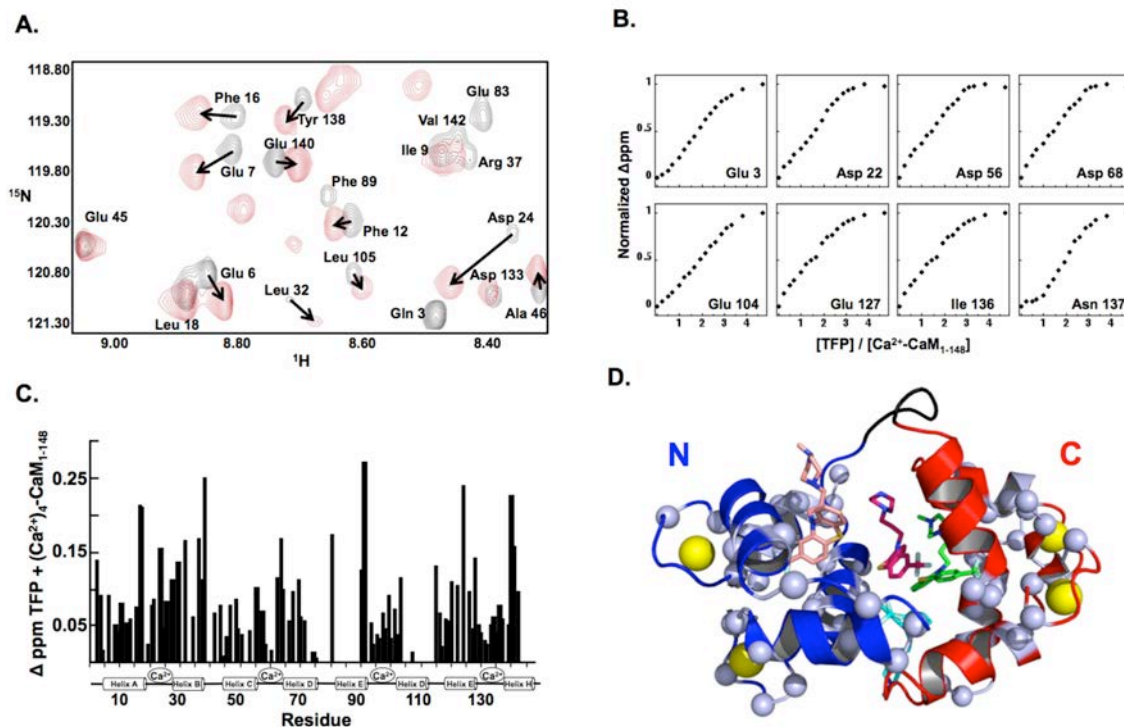


Figure 2.5: ^{15}N -HSQC-monitored TFP titration of uniformly ^{15}N -labeled (Ca^{2+})₄-CaM. A: Comparison of (Ca^{2+})₄-CaM (black) and TFP-saturated (Ca^{2+})₄-CaM (red) ^{15}N -HSQC spectra, arrows indicate change in resonance positions over the course of the TFP titration. B: Normalized TFP-induced chemical shifts of individual representative residues of (Ca^{2+})₄-CaM. C: Bar graph of net chemical shift per residue caused by TFP saturation of CaM. D: Location of each (Ca^{2+})₄-CaM residue whose chemical shift was perturbed > 0.05 ppm by TFP saturation (white spheres); backbone modeled according to the structure of TFP bound to (Ca^{2+})₄-CaM at a 4:1 ratio (1LIN.pdb). Solution conditions: 10% D₂O, 10 mM imidazole, 100 mM KCl, 50 μM EDTA, 5mM CaCl₂, pH 6.5 at 22°C. Users/nmr_mike/Thesis/Chapter_II/Figure2_5.jpg

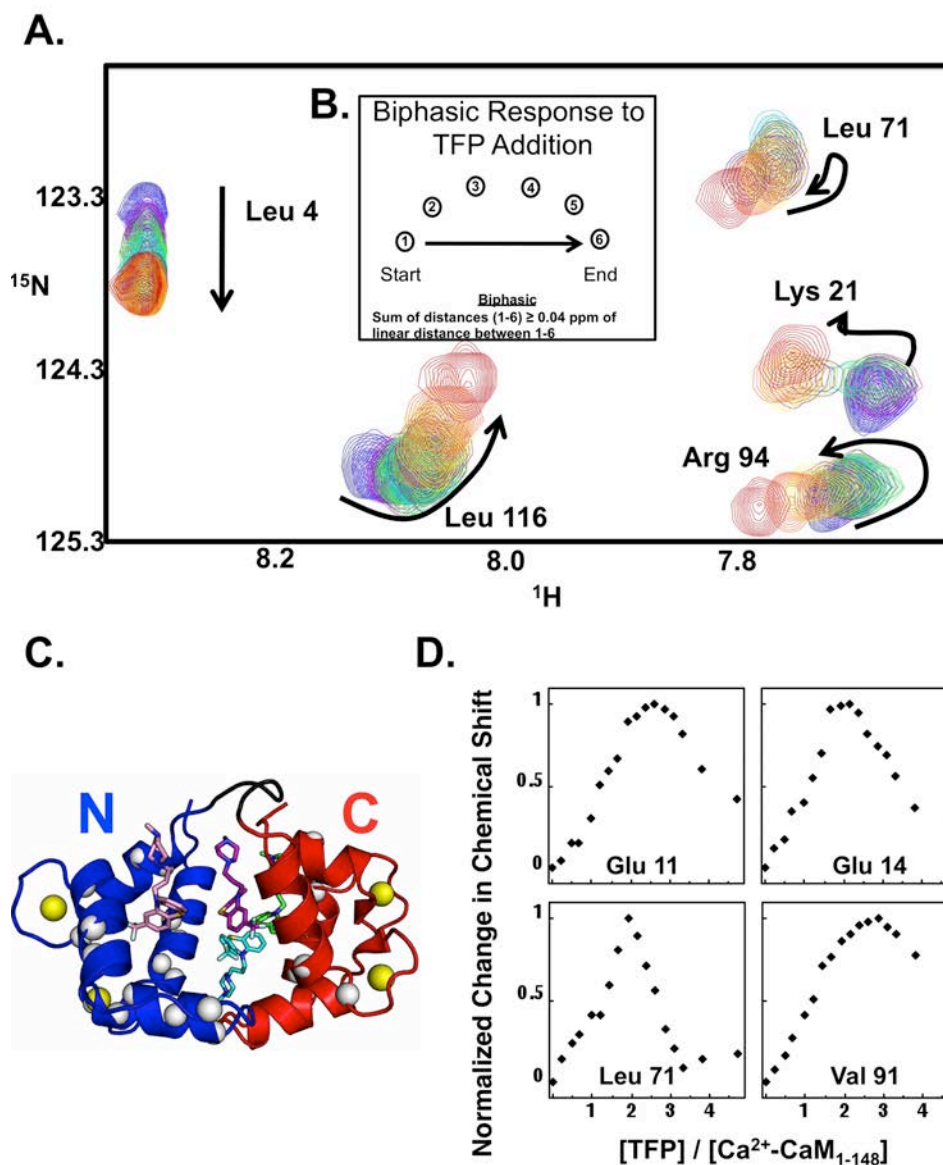


Figure 2.6: Multiple chemical environments observed upon TFP titration of $(\text{Ca}^{2+})_4\text{-CaM}$. A: ^{15}N -HSQC spectrum of uniformly ^{15}N labeled $(\text{Ca}^{2+})_4\text{-CaM}$ titrated with TFP, where arrows represent the movement of each resonance from its initial position. B: Schematic diagram of quantitative criterion for classification of biphasic chemical shift. C: Locations of select residues that underwent a biphasic response upon TFP addition, mapped onto the structure of TFP bound to $(\text{Ca}^{2+})_4\text{-CaM}$ at a 4:1 ratio (1LIN.pdb). D: Normalized chemical shift plots for select individual residues deemed to undergo a biphasic response upon TFP titration of $(\text{Ca}^{2+})_4\text{-CaM}$. Solution conditions: 10% D_2O , 10 mM imidazole, 100 mM KCl, 50 μM EDTA, 5mM CaCl_2 , pH 6.5 at 22°C. Users/nmr_mike/Thesis/Chapter_II/Figure2_6.jpg

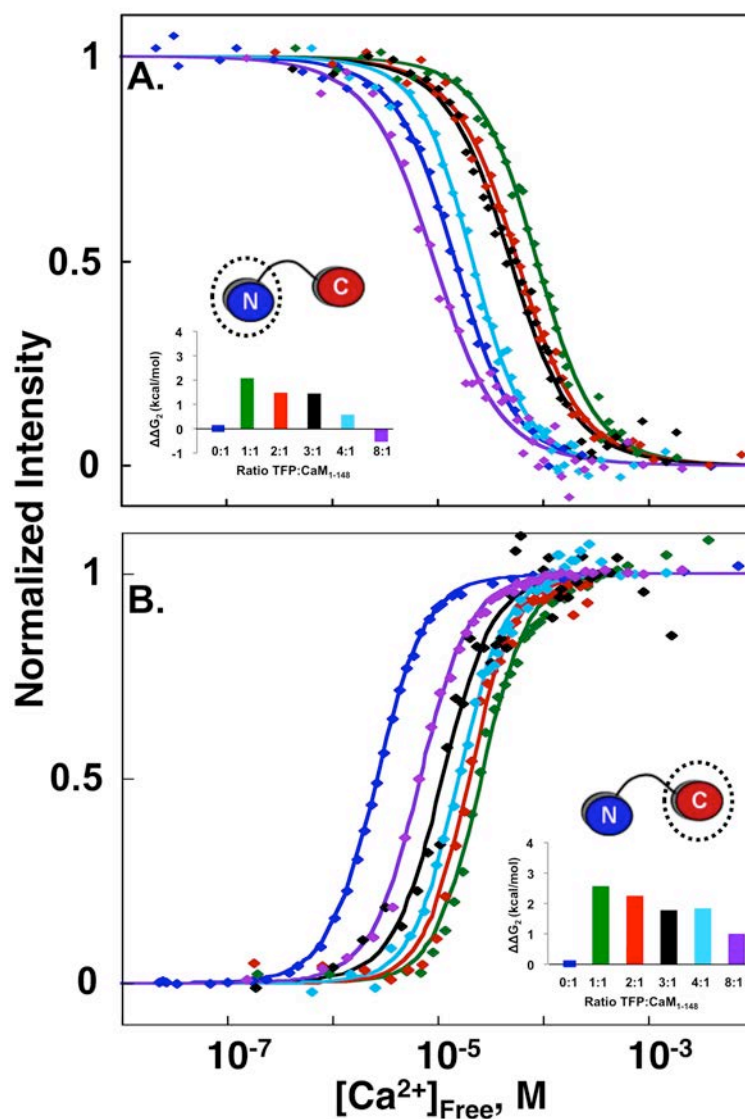


Figure 2.7: Effect of TFP on calcium binding to CaM₁₋₁₄₈. Equilibrium calcium titrations of CaM (6 μ M) were conducted in the presence of 0 (blue), 6 (green, 1:1), 12 (red, 2:1), 18 (black, 3:1), 24 (cyan, 4:1), or 48 μ M (purple, 8:1 TFP:CaM) TFP, and were monitored using the intrinsic fluorescence of CaM. A: phenylalanine fluorescence (250 nm_{ex} and 280 nm_{em}). B: tyrosine fluorescence (277 nm_{ex} and 320 nm_{em}). In B, for 3:1, 4:1 and 8:1 TFP:CaM, the raw signal decreased; it is shown inverted to facilitate comparisons. Solid curves were simulated according to Equation 2.3 and free energies in Table 1; bar graph insets represent $\Delta\Delta G_2$ values in Table 1. Solution conditions: 50 mM HEPES, 100 mM KCl, 5 mM KCl, 0.05 mM EGTA, 1 mM MgCl₂, and 6 nM Oregon Green (pH 7.4) at 22°C. Users/nmr_mike/Thesis/Chapter_II/Figure2_7.jpg

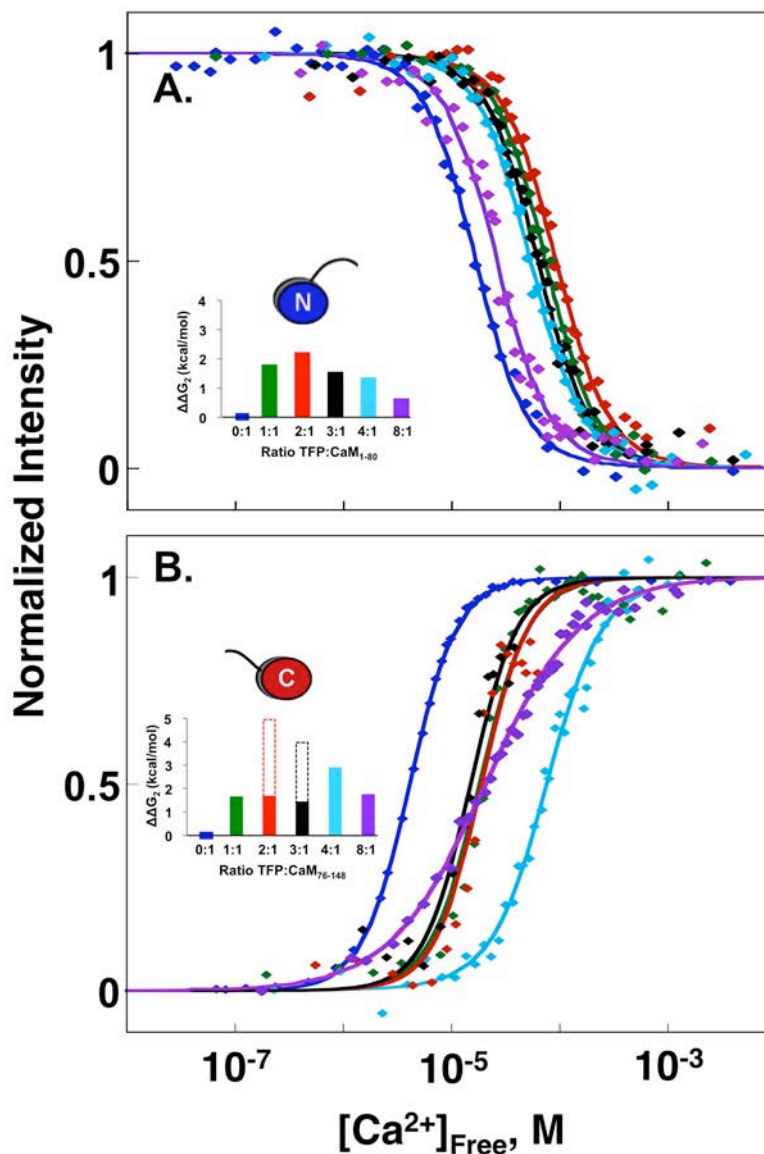


Figure 2.8: Effect of TFP on the calcium binding affinity of CaM₁₋₈₀ and CaM₇₆₋₁₄₈. Equilibrium calcium titrations of CaM (6 μ M) were conducted in the presence of 0 (blue), 6 (green, 1:1), 12 (red, 2:1), 18 (black, 3:1), 24 (cyan, 4:1), or 48 μ M (purple, 8:1 TFP:CaM) TFP, and were monitored using the intrinsic fluorescence of CaM. A: phenylalanine fluorescence (250 nm_{ex} and 280 nm_{em}). B: tyrosine fluorescence (277 nm_{ex} and 320 nm_{em}). In B, for 2:1 and 3:1 TFP:CaM, only the first transition is shown. In B, for 4:1 and 8:1 TFP:CaM, the raw signal decreased; it is shown inverted to facilitate comparisons. Solid curves were simulated according to Equation 2.3 and free energies in Table 1; bar graph insets represent $\Delta\Delta G_2$ values in Table 1. Solution conditions were 50 mM HEPES, 100 mM KCl, 5 mM KCl, 0.05 mM EGTA, 1 mM MgCl₂, and 6 nM Oregon Green (pH 7.4) at 22°C.
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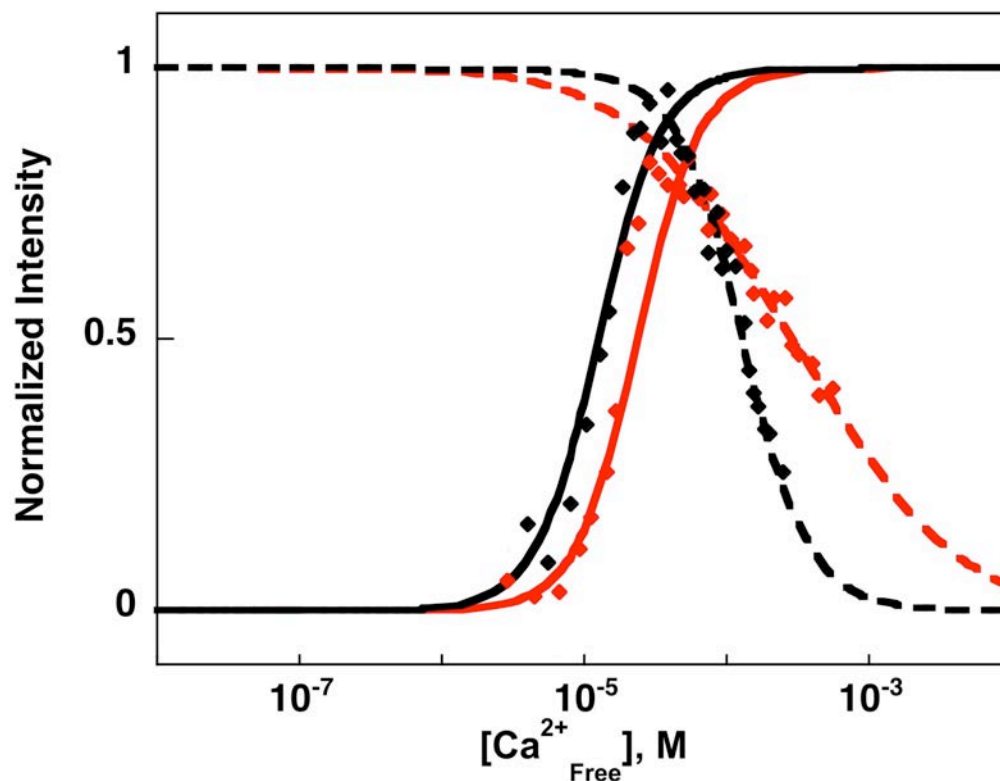


Figure 2.9: Biphasic fluorescence response to calcium binding at intermediate TFP. Effect of TFP on calcium titration of CaM₇₆₋₁₄₈ at 12 μ M (red, 2:1 TFP:CaM) and 18 μ M (black, 3:1 TFP:CaM) monitored using the intrinsic tyrosine fluorescence of CaM (277 nm_{ex} and 320 nm_{em}). Evidence for multiple species, and piecewise analysis described in *Results*. Solid curves for calcium-dependent increase in fluorescence intensity were simulated according to Equation 2.3 and free energies in Table 1; dashed curves correspond to decrease in fluorescence intensity.
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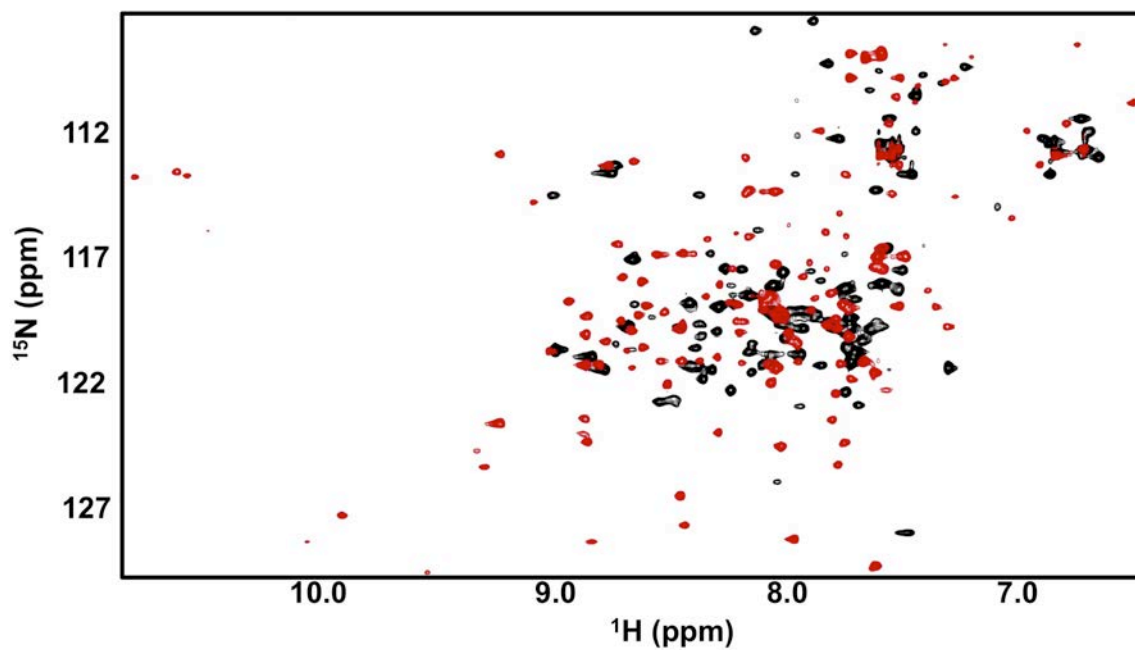


Figure 2.10: Comparison of TFP-saturated apo CaN (Ca^{2+})₄-CaM₁₋₁₄₈. Overlay of ^{15}N -HSQC spectra of apo CaM (black, 2 TFP:CaM) and (Ca^{2+})₄-CaM (red, 4 TFP:CaM). Few peaks overlap, indicating significantly different chemical environments for backbone amides in the structures of apo and (Ca^{2+})₄-CaM saturated by TFP.
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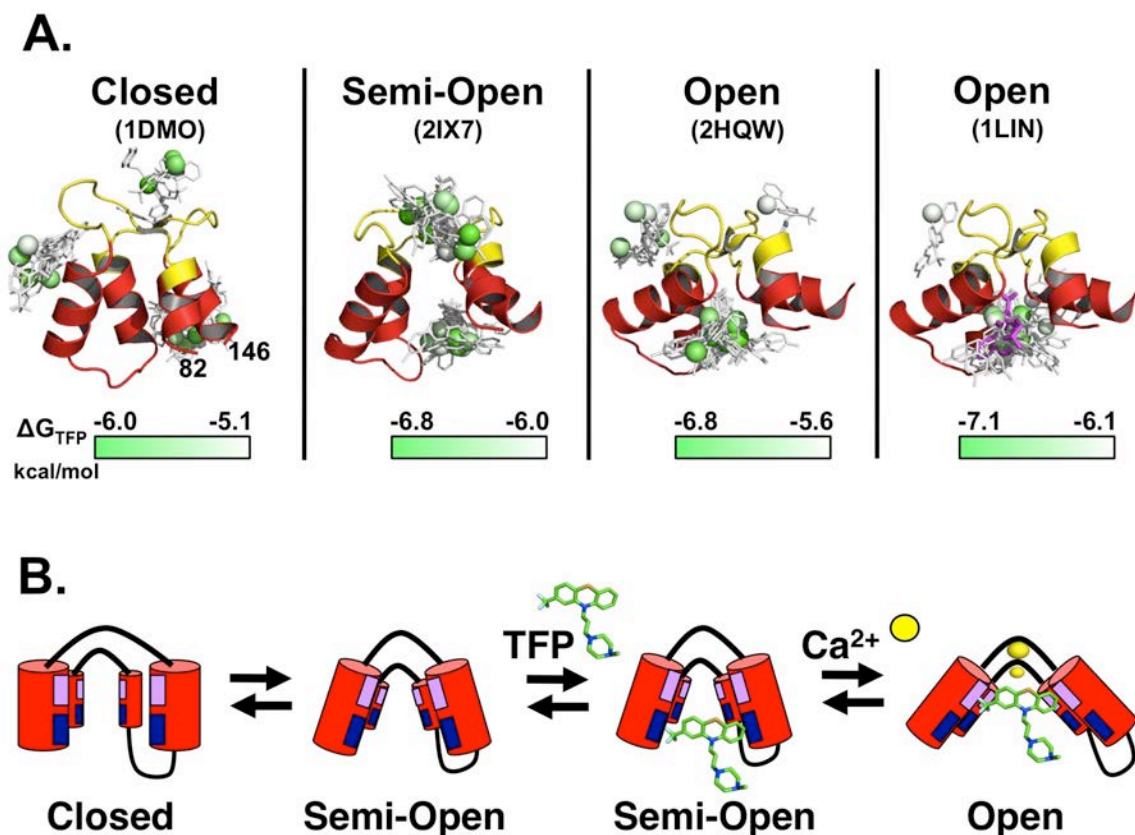


Figure 2.11: Docking and models of TFP binding to the C-domain of CaM

A. TFP Docking to alternative tertiary conformations of apo CaM. Ribbon diagrams of C-domain fragments (residues 82 to 146) represent the “closed” (1DMO.pdb), “semi-open” (2IX7.pdb) and “open” (2Hqw, 1LIN.pdb) conformations. Calcium was removed from 2Hqw and 1LIN. *AutoDock Vina 1.0.3*⁴² predicted positions of TFP binding; 20 models having lowest free energy are shown as sticks. The single sulfur atom of each TFP is shown as a sphere; green corresponds to the most favorable free energy of binding; white is the least favorable. Color thermometer below each set of models indicates the range of energies predicted. The TFP molecule observed at site A of 1LIN.pdb is shown in magenta. Residues in calcium-binding sites are yellow; arrows are included only to orient the viewer to chain direction. **B.** Model of conformational transition of apo C-domain in equilibrium between a “closed” and “semi-open” conformation. Binding of TFP to the blue patches accessible in the “semi-open” conformation is energetically more favorable than binding to “closed” form. TFP binding to the blue occludes hydrophobic patches shown in purple of the apo C-domain that are otherwise exposed to solvent. An “open” conformation is adopted upon calcium binding, whether alone or also bound to a drug or protein target exposing hydrophobic patches shown in purple.

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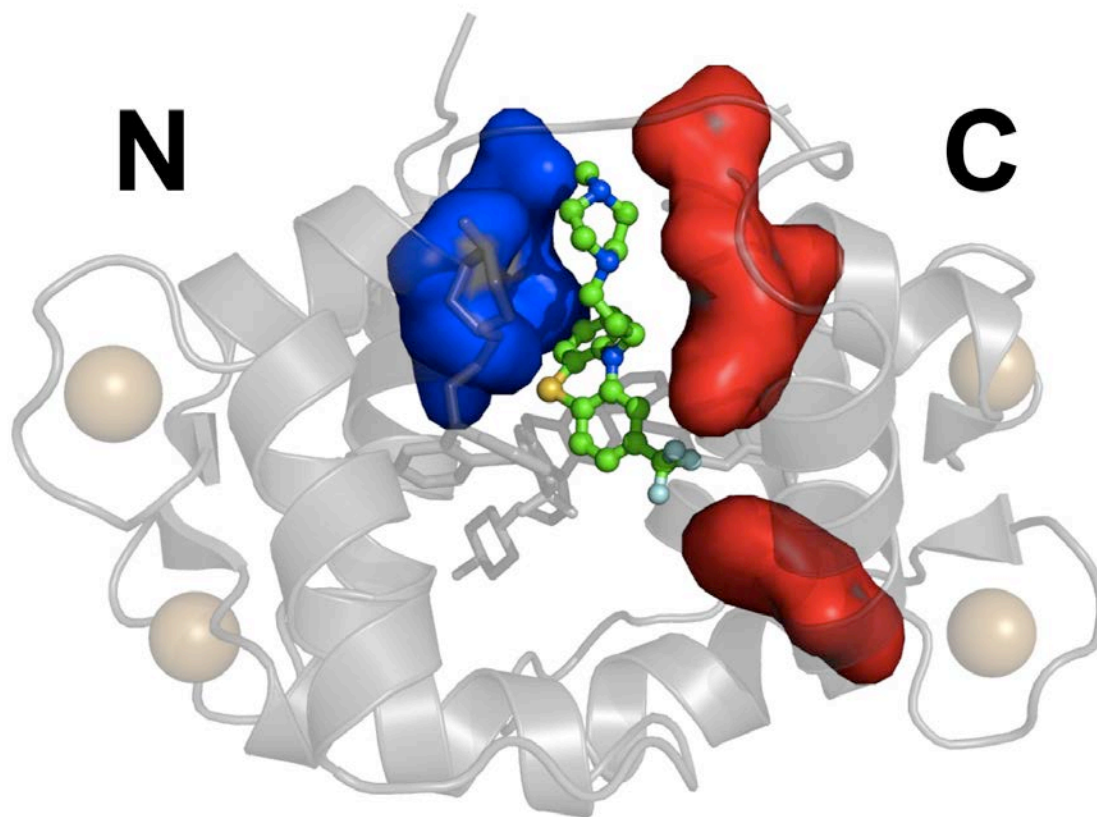


Figure 2.12: Interdomain interactions mediated by TFP
Based on the crystal structure with 4:1 TFP:(Ca²⁺)₄-CaM (1LIN.pdb), the trifluoperazine molecule shown in ball-and-stick (green) with fluorine, sulphur, and nitrogen atoms in cyan, yellow, and blue respectively) interacts with residues in both the calcium-saturated N-domain (blue) and C-domain (red). Those within 4 Å were residues 8, 11, 72, 92, 144, 145, TFP 1, and TFP 2. (Ca²⁺)₄-CaM backbone (gray), 4 calcium ions (yellow spheres), and three other TFP (gray sticks) are shown.
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CHAPTER III

BINDING OF TRIFLUOPERAZINE TO THE C- DOMAIN OF CAM

Introduction

A difficulty encountered in **Chapter II** studies involving TFP binding to apo and $(\text{Ca}^{2+})_4\text{-CaM}$ was found in the added complexity of TFP binding to CaM at stoichiometries greater than 1:1. Binding of TFP at ratios greater than 1:1 complicates structural analysis of TFP induced effects upon CaM as it is difficult to attribute observed changes in CaM to TFP binding at one site, as opposed to another. Further complicating analysis are the allosteric linkages that exist between Ca^{2+} and TFP binding sites if structural changes within CaM can be induced by either direct TFP binding to an area of CaM, or allosterically via propagated change. These complicating factors prompted us to examine how TFP interacted with an isolated C-domain fragment of CaM (CaM_{76-148}).

CaM_{76-148} was appealing to us as studies performed in **Chapter II** showed that TFP preferentially interacted with the C-domain of apo and $(\text{Ca}^{2+})_4\text{-CaM}_{1-148}$. Choosing to work with isolated CaM_{76-148} reduces that complexity of allosteric interactions that occur between domains of CaM_{1-148} upon TFP binding while still retaining similar Ca^{2+} -binding properties as found in the C-domain of CaM_{1-48} . There is also a common TFP binding site within the C-domain of crystal structures of TFP bound to $(\text{Ca}^{2+})_4\text{-CaM}_{1-148}$ where a significant difference is observed in TFP binding orientation that we would like to resolve why it exists.

The crystal structures of TFP bound to $(\text{Ca}^{2+})_4\text{-CaM}$ differ in the stoichiometry of TFP per molecule of $(\text{Ca}^{2+})_4\text{-CaM}$ (**Figure 2.1c**) (Cook et al., 1994; Vandonselaar et al., 1994a; Vertessy et al., 1998b). These structures also differ in the orientation of the trifluoromethyl group of the TFP molecules common to all 3 TFP bound $(\text{Ca}^{2+})_4\text{-CaM}$ structures. In the structures of TFP bound $(\text{Ca}^{2+})_4\text{-CaM}$ at ratios of 2:1 and 4:1 the

trifluoromethyl group is inserted into the hydrophobic pocket of the C-domain, while in the 1:1 structure it is flipped 180° (**Figure 3.1**). There is no clear explanation as to why the trifluoromethyl would adopt one orientation over the other, as the backbone and side chain conformations of all 3 TFP bound $(\text{Ca}^{2+})_4$ -CaM structures are nearly indistinguishable (**Figure 3.2**).

Chapter III builds upon thermodynamic studies performed in **Chapter II** to more closely examine the molecular constraints required for TFP binding to apo and $(\text{Ca}^{2+})_2$ -CaM₇₆₋₁₄₈. This chapter presents structural studies of apo and $(\text{Ca}^{2+})_2$ -CaM₇₆₋₁₄₈ using NMR and X-ray crystallography respectively. Using NMR spectroscopy we have assigned backbone and side chain nuclei of apo CaM₇₆₋₁₄₈ in the absence and presence of a 1:1 ratio of TFP, allowing for quantification of the residue specific changes in chemical shift. X-ray crystallography was used to determine the structure of TFP bound to $(\text{Ca}^{2+})_2$ -CaM₇₆₋₁₄₈. Findings reported here reveal residue specific changes and interactions associated with TFP binding to apo and $(\text{Ca}^{2+})_2$ -CaM₇₆₋₁₄₈.

Materials and Methods

Protein Overexpression

IPTG-induced CaM overexpression was performed using transformed *E. coli* BL21(DE3) cells containing the recombinant pT7-7 vector expressing the C-domain of *Rattus Norvegicus* CaM. Proteins were overexpressed in Luria-Bertani broth. CaM was then purified as previously described by Putkey et al. (Putkey et al., 1985). The recombinant proteins were 97-99% pure as judged by silver-stained SDS-PAGE gels. Protein concentrations were determined by UV spectroscopy of protein denatured with NaOH or native at pH 7.4 (Crouch and Klee, 1980).

Crystallography Materials and Methods

Crystallization of TFP-bound to $(\text{Ca}^{2+})_2\text{-CaM}_{76-148}$ was performed by adding a 10-fold molar excess of TFP to 500 μl of $\sim 10\text{mg/ml}$ of CaM_{76-148} in 50mM HEPES, 100mM KCl, 1mM MgCl_2 , 5mM NTA, 50 μM EGTA, pH 7.4, with 500 μl of 200mM potassium thiocyanate, 20% polyethylene glycol 3350, pH 6.64 (Solution PEG 62 Qiagen) as a hanging drop in a 96-well tray. The tray was incubated at 15° C for ~ 8 months, at which time a single rod shaped crystal was observed. The crystal was cryo-protected with mother liquor containing 10% ethylene glycol prior to being flash-frozen at 100 K. Data were collected on this crystal at 100 K at the 4.2.2 synchrotron beamline at the Advanced Light Source at the Ernest Orlando Lawrence Berkeley National Laboratory, with a 150 mm crystal-to-detector distance and the assistance of Jay Nixx (beam-line manager). The program d*TREK was used to analyze and scale the data (Pflugrath, 1999). The monoclinic crystals diffracted to a resolution of 1.9 Å and were of the space group P2_1 . Molecular replacement was performed using the extracted C-domain of TFP bound $(\text{Ca}^{2+})_4\text{-CaM}_{1-148}$ (1LIN.pdb) as a template with the program Phaser (Read, 2001). TFP and Ca^{2+} were removed from the template prior to use in molecular replacement. Refinement was performed using the program Refmac5 of the CCP4 program suite (Murshudov et al., 1997). Coot was used for molecular visualization and model building (Emsley and Cowtan, 2004). Ca^{2+} and TFP were modeled into clearly visible electron density, water molecules were finally added to the structure using Coot, followed by manual editing. Structure validation was performed using the WhatIf Web Server (<http://swift.cmbi.ru.nl/servers/html/index.html>).

Overexpression and Purification of Isotope Enriched

CaM

All isotopes were obtained from Cambridge Isotope Laboratories (Andover, MA). IPTG-induced CaM overexpression was performed using transformed BL21(DE3) cells

containing the recombinant pET vector expressing the C-domain of *Paramecium* CaM (a gene generously provided by C. Kung, University of Wisconsin, Madison, WI). ^{15}N -labeled proteins were overexpressed in minimal medium, using 2 g/L unlabeled glucose as a carbon source and 1g/L $^{15}\text{NH}_4\text{Cl}$ as the sole nitrogen source. Double labeled (^{13}C - and ^{15}N -) proteins were produced using 2g/L ^{13}C -glucose as the sole carbon source and 1 g/L $^{15}\text{NH}_4\text{Cl}$ as the sole nitrogen source. CaM was then purified as previously described by Putkey et al. The recombinant proteins were 97-99% pure as judged by silver-stained SDS-PAGE. Protein concentrations were determined by UV spectroscopy of protein denatured with NaOH or native at pH 7.4 (Beaven and Holiday, 1952).

Assignment of Backbone and Side chain Resonances

The NMR spectra apo CaM₇₆₋₁₄₈ ± TFP were collected at 25 °C on a Bruker Avance II 500 or 800 NMR spectrometer. The ^1H , ^{15}N , and ^{13}C resonances of the backbone were assigned using triple resonance experiments (HNCA, HN(CO)CA, HNCACB, HN(CO)CACB, HNCOC, and HN(CA)CO) (Yamazaki et al., 1994) with the uniformly ^{15}N and ^{13}C -labeled CaM in complex with unlabeled TFP. $^1\text{H}_\alpha$ resonances were assigned from an ^{15}N -edited TOCSY spectrum using an uniformly ^{15}N -labeled protein (Clare and Gronenborn, 1994) and from HA(CACO)NH experiment using an uniformly ^{15}N and ^{13}C -labeled sample. The side chain signals were assigned from 3D H(CCO)NH-TOCSY, C(CO)NH-TOCSY, HCCH-TOCSY, ^{15}N -edited TOCSY, and ^{15}N or ^{13}C -edited NOESY spectra (Clare and Gronenborn, 1994; Fesik and Zuiderweg, 1988).

Quantification of ^{15}N -apo CaM₇₆₋₁₄₈ Chemical Shifts due to TFP Addition

To determine the change in chemical shift upon TFP binding to apo CaM₇₆₋₁₄₈, chemical-shift changes in both the ^1H and ^{15}N dimensions were quantified using the modified Pythagorean theorem previously described by Jaren et al., 2002 shown in **Equation 3.1**.

$$\Delta ppm = \sqrt{(\Delta^1 H ppm)^2 + (0.10134 \cdot \Delta^{15} N ppm)^2} \quad (3.1)$$

In this equation, Δppm refers to the linear change of a specific resonance peak from its initial starting position in apo CaM₇₆₋₁₄₈.

Results

Structure of TFP Bound (Ca²⁺)₂-CaM₇₆₋₁₄₈

X-ray crystallography studies of TFP bound to (Ca²⁺)₂-CaM₇₆₋₁₄₈ were refined to 2.1 Å resolution with statistical measures of the goodness-of-fit listed in **Table 1**. It is important to point out that the % completeness value (80.92%) reported in this table is not ideal. The reason for this lower than expected value is due to the presence of an ice ring in the data that needed to be removed to properly index and scale the dataset resulting in the loss of some of the diffraction data. Although this value is not ideal, it was sufficient to provide a usable electron density map for model building.

The structure revealed that there were 2 (Ca²⁺)₂-CaM₇₆₋₁₄₈ and 4 TFP molecules per asymmetric unit (**Figure 3.3a**). The conformations adopted by the 2 (Ca²⁺)₂-CaM₇₆₋₁₄₈ chains were very similar to each other with an all atom RMSD of 0.46 Å (**Figure 3.3b**). The interhelical angles adopted between helices E-F of TFP bound (Ca²⁺)₂-CaM₇₆₋₁₄₈ chains A and B were 79.0° and 82.9° respectively, while interhelical angles of 89.7° (chain A) and 88.4° (chain B) were observed for helices G-H. Consistent with other structures of (Ca²⁺)₄-CaM₁₋₁₄₈ either with or without a bound target, both chains of TFP bound (Ca²⁺)₄-CaM₇₆₋₁₄₈ adopt an “open” domain conformation.

As previously stated and shown in **Figure 3.3b**, 4 TFP molecules were found within the asymmetric unit. Both (Ca²⁺)₂-CaM₇₆₋₁₄₈ chains share a common TFP-binding site located within each of their hydrophobic pockets. Although each (Ca²⁺)₂-CaM₇₆₋₁₄₈ chain has a TFP molecule bound at a common position, the orientation of TFP within the hydrophobic pocket is different dependent upon the (Ca²⁺)₂-CaM₇₆₋₁₄₈ chain examined (**Figure 3.4**). Examination of residues within hydrophobic pockets of (Ca²⁺)₂-CaM₇₆₋₁₄₈

that are within 4 Å of TFP that may account for the 180° flip of TFP between $(\text{Ca}^{2+})_2$ -CaM₇₆₋₁₄₈ chains indicate that most of the residues are unchanged with the exception of M144 (**Figure 3.5**).

In addition to the 2 TFP molecules that were observed to bind in the hydrophobic pockets of $(\text{Ca}^{2+})_2$ -CaM₇₆₋₁₄₈, 2 additional TFP molecules were observed within the asymmetric unit. Analysis of contacts within 4 Å of these TFP molecules indicate that they largely interact with other TFP molecules and made few interactions with $(\text{Ca}^{2+})_2$ -CaM₇₆₋₁₄₈, compared to TFP molecules found within the hydrophobic clefts of $(\text{Ca}^{2+})_2$ -CaM₇₆₋₁₄₈ (**Figure 3.6**).

This structural study unequivocally shows that TFP-binding to $(\text{Ca}^{2+})_2$ -CaM₇₆₋₁₄₈ does not alter the “open” backbone conformation of $(\text{Ca}^{2+})_2$ -CaM₇₆₋₁₄₈ observed in the absence of TFP. Previous studies conducted in **Chapter II**, indicated that a binding interface used by the D-domain of apo CaM₁₋₁₄₈ for TFP binding was distinct from those of apo CaM₁₋₁₄₈ alone and that of TFP-bound $(\text{Ca}^{2+})_2$ -CaM₇₆₋₁₄₈. To more closely examine the apo CaM₇₆₋₁₄₈ TFP binding interface in-depth solution NMR experiments were required.

TFP-Induced Changes of apo CaM₇₆₋₁₄₈ as Monitored by ¹⁵N-HSQC Spectroscopy

Due to the inherent flexibility found within the Ca^{2+} -binding loops of apo CaM₇₆₋₁₄₈, solution NMR methods were used to examine how TFP interacts with apo CaM₇₆₋₁₄₈ at the structural level. **Figure 3.7** shows an overlay of spectra of apo CaM₇₆₋₁₄₈ without and with TFP at a 1:1 apo CaM₇₆₋₁₄₈:TFP ratio. TFP binding to apo CaM₇₆₋₁₄₈ was observed to be in fast exchange on the NMR time scale due to its weak (~1 μM) binding affinity. TFP binding induced significant chemical shift perturbations of apo CaM₇₆₋₁₄₈ amide resonances, as shown in **Figure 3.7**. ¹⁵N-HSQC peak assignments of apo CaM₇₆₋₁₄₈ in the presence and absence of TFP were determined via 3-dimensional NMR

experiments described in the materials and methods resulting in ~95% assignment of backbone resonances.

Quantification of chemical shifts of apo CaM₇₆₋₁₄₈ amide resonances due to TFP binding resulted in an average chemical shift of 0.047 ppm. Individual residue chemical shift values are shown in **Figure 3.8** where it can be observed that TFP binding did not shift apo CaM₇₆₋₁₄₈ resonances in a uniform manner, but rather causes shifts at unique positions. Of the 73 amino acids that comprise CaM₇₆₋₁₄₈, 21 were observed to have a chemical shift greater than 0.05 ppm upon TFP addition. Although the structure of TFP-bound apo CaM₇₆₋₁₄₈ is unknown, it is likely that TFP binds to an exposed hydrophobic patch of apo CaM₇₆₋₁₄₈ comprised of hydrophobic residues identified in **Figure 3.8**. Computational docking of TFP to the C-domain of CaM described in **Chapter II** predicted that TFP bound within the shallow hydrophobic cleft of a “semi-open” conformation composed of similar hydrophobic residues identified here.

The location and magnitude of these chemical shifts have been mapped onto the solution structure (1F71.pdb) of apo CaM₇₆₋₁₄₈ (**Figure 3.8**), where it can be observed that although sequentially distant in primary sequence, many of the TFP perturbed residues are located near each other spatially. As shown in **Figure 3.8**, apo CaM₇₆₋₁₄₈ helices F-H as well as the Ca²⁺ binding loops contain resonances that are significantly (>0.05ppm) perturbed, indicating that these regions are either directly or allosterically perturbed upon TFP binding.

Dynamics of TFP Bound apo CaM₇₆₋₁₄₈ Monitored with T₂ Relaxation Spectroscopy

The change in overall size of apo CaM₇₆₋₁₄₈ upon TFP binding was investigated using T₂ NMR relaxation experiments. Comparison of average amide T₂ relaxation times of apo CaM₇₆₋₁₄₈ with (147 ± 79.90 msec) and without TFP (161 ± 59.46 msec), indicate that TFP binding to apo CaM₇₆₋₁₄₈ causes an increase in hydrodynamic radius. Shown in

Figure 3.9, are the calculated individual T_2 amide relaxation times for free and TFP bound apo CaM₇₆₋₁₄₈. As expected the N- and C-termini of both TFP free and TFP bound apo CaM₇₆₋₁₄₈ have significantly longer T_2 relaxation times indicative of their lack of defined secondary structure. Consistent with the analysis of ¹⁵N-HSQC spectra of TFP binding to apo CaM₇₆₋₁₄₈ presented in this chapter, the T_2 relaxation times apo CaM₇₆₋₁₄₈ of residues within helices F-G, and Ca²⁺ binding loops of CaM are increased upon TFP addition (**Figure 3.10**).

Discussion

The C-domains of apo and (Ca²⁺)₄-CaM can bind to their targets using a variety of conformations described previously in **Chapter II**. To simplify structural studies of TFP interacting with the C-domain of apo and (Ca²⁺)₄-CaM₁₋₁₄₈, the C-domain fragment (CaM₇₆₋₁₄₈) was used. NMR studies presented in this chapter address residue-specific changes in apo CaM₇₆₋₁₄₈ upon TFP addition. The crystal structure of TFP bound (Ca²⁺)₂-CaM₇₆₋₁₄₈ consolidates differing observations of the location of the trifluoromethyl group of the TFP common in all observed structures, as well as provides for the first time a molecular basis for the trifluoromethyl group location.

Protein Crystallography of (Ca²⁺)₂-CaM₇₆₋₁₄₈-TFP

Complex

Crystallization of TFP bound to (Ca²⁺)₂-CaM₇₆₋₁₄₈ revealed that TFP binds within the hydrophobic cleft of the isolated C-domain in a similar manner to TFP binding to (Ca²⁺)₄-CaM₁₋₁₄₈. In the asymmetric unit, 4 TFP molecules were found. The TFP molecules were numbered based on their order from left to right when chain A when Chain A is position on the left and chain B on the right as depicted in **Figure 3.3**. TFP #2 and TFP #3 are likely crystallization artifacts, as both of these TFP molecules made few interactions with either chain A or B compared to TFP bound within the clefts of (Ca²⁺)₂-CaM₇₆₋₁₄₈ (**Figure 3.6**). Examination of the asymmetric unit reveals that these bound

TFP molecules function to create a continuous array of TFP molecules between chains A and B (**Figure 3.3a**). This array is similar to the way TFP molecules are positioned in the 4:1 TFP-(Ca²⁺)₄-CaM₁₋₁₄₈ structure, although the (Ca²⁺)₂-CaM₇₆₋₁₄₈ domains are rotated 180° relative to each other (**Figure 2.1c, Figure 3.3**). This rotation is likely the result of the lack of covalent linkage between the C-domain fragments which would otherwise be present in (Ca²⁺)₄-CaM₁₋₁₄₈.

Protein chains A and B have a common TFP molecule found in each of their hydrophobic pockets although the location of the TFP's trifluoromethyl groups are flipped 180° relative to each other (**Figure 3.4**). Previous structures of TFP bound (Ca²⁺)₄-CaM₁₋₁₄₈ report that the difference in location of the trifluoromethyl group at the common TFP binding site was reported as improperly assigned electron density (Cook et al., 1994; Vandonselaar et al., 1994a; Vertessy et al., 1998a). These structures were determined at 2.45 (1:1), 2.74 (2:1), and 2.0 Å (4:1) resolution, and only contained one molecule of (Ca²⁺)₄-CaM₁₋₁₄₈ per unit cell. The electron density of the TFP binding site in our 1.9 Å resolution structure which contains two molecules of (Ca²⁺)₂-CaM₇₆₋₁₄₈ per asymmetric unit clearly indicates that the trifluoromethyl group at the common TFP binding site can occupy both orientations (**Figure 3.11**).

Superposition of chain A or B of (Ca²⁺)₂-CaM₇₆₋₁₄₈ with the TFP bound C-domain of (Ca²⁺)₄-CaM₁₋₁₄₈ (1CTR.pdb and 1LIN.pdb) shows that the backbone of the C-domain of CaM in CaM₇₆₋₁₄₈, and CaM₁₋₁₄₈ are similar (RMSD < 0.9Å). However, differences were observed between TFP-bound (Ca²⁺)₂-CaM₇₆₋₁₄₈, and (Ca²⁺)₄-CaM₁₋₁₄₈ at the location of TFP binding, and location of the trifluoromethyl groups (Chain A vs. 1:1 CaM₁₋₁₄₈:TFP and Chain B vs. 4:1 CaM₁₋₄₈:TFP) (**Figure 3.12**). Superposition of chains A and B revealed that nearly identical conformations were adopted by both chains as well as the residues within 4Å of the common TFP binding sites in both chains (**Figure 3.4, 3.5**). Slight differences in the structure of (Ca²⁺)₂-CaM₇₆₋₁₄₈ formed by chains A and B were seen outside of the hydrophobic cleft. Shown in **Figure 3.13** is a Chimera

(Pettersen et al., 2004) generated all-atom morph between chains A and B which depicts that magnitude of change between chains.

The most significant difference between chain A and B is the orientation of the side chain of M144 which appears to act as a selection gate to control whether the trifluoromethyl group can be accommodated within the hydrophobic pocket of $(\text{Ca}^{2+})_2$ -CaM₇₆₋₁₄₈ (**Figure 3.5**). Variability in the side chain conformation of M144 is consistent with other structures of $(\text{Ca}^{2+})_4$ -CaM₁₋₁₄₈ bound to drugs or peptides. Side chain methyl dynamics measurements of M144 indicate that it is highly dynamic with respect to other CaM Met residues (Chen et al., 1993; Ehrhardt et al., 1995). In a comparison of 17 compact CaM-drug or CaM-peptide complexes, a tetrad of CaM residues termed FLMM (L92, L105, M124, M144) was found to consistently contact the ligand for the C-domain in all complexes examined (Ataman et al., 2007). Of the FLMM tetrad residues, the side chain of M144 was the most variable in its closest approach distance when interacting with the hydrophobic anchor residue of the ligand, as well as most variable in terms of amino acid identity in 102 CaM sequences (Ataman et al., 2007). Similar variability conservation of a Met residue in the FLMM tetrad at position 144 was observed in a computational study in which the hydrophobic cleft of CaM was redesigned to improve target binding affinity (Shifman and Mayo, 2002; Shifman and Mayo, 2003). The binding site for TFP within the hydrophobic cleft of the C-domain of CaM is not well defined based on the dynamic nature of Met 144 as its conformation solely selects the orientation of TFP that can be accommodated by the hydrophobic cleft. These observations indicate that the dynamic nature of Met144 results in multiple modes of TFP binding that are used by the C-domain of $(\text{Ca}^{2+})_4$ -CaM.

This structure helps to resolve controversy within the field when comparing the location of the common TFP binding site found in each of the full length structures of TFP bound to $(\text{Ca}^{2+})_4$ -CaM₁₋₁₄₈ at 1:1, 2:1, and 4:1 ratios. It shows that TFP employs 2 modes of binding to the C-domain of CaM, as both orientations of the trifluoromethyl

group are observed within the same unit cell. Previous studies with $(\text{Ca}^{2+})_4\text{-CaM}_{1-148}$ have shown 2 different TFP binding orientations in separate protein crystals obtained from differing crystallization solutions, whereas our findings are from a single crystal and therefore cannot be a result of difference in solution conditions. This structure also indicates that both conformations of TFP have similar binding affinities for the C-domain as protein crystallization requires molecules to adopt low energy conformations to form an ordered crystal lattice. If one of the CaM conformations observed had a significantly more favorable energy than the other, it would be unlikely for both a high and low energy conformation to co-crystallize together.

TFP-Induced Changes of apo CaM₇₆₋₁₄₈ as Monitored by ¹⁵N-HSQC Spectroscopy

To examine the effect of TFP upon individual residues of apo CaM₇₆₋₁₄₈ we have used ¹⁵N-HSQC chemical shift mapping to determine TFP induced changes in the chemical environments of residues within apo CaM₇₆₋₁₄₈. **Figure 3.7** shows an overlay of ¹⁵N-HSQC spectra of apo CaM₇₆₋₁₄₈ recorded in the absence and presence of TFP, where the quantified average change in chemical shift upon TFP addition to apo CaM₇₆₋₁₄₈ was 0.047 ppm. If TFP interacted in a nonspecific manner with apo CaM₇₆₋₁₄₈, all residues would have individual Δppm values of ~ 0.047 , but comparison of the average Δppm with individual residue values indicates that TFP selectively interacts with specific residues of apo CaM₇₆₋₁₄₈ (**Figure 3.8**). This observation is in agreement with observations made in **Chapter II** that TFP interacts with a selective set of apo CaM C-domain residues. TFP binding to apo CaM₇₆₋₁₄₈ perturbs residues located near the hydrophobic cleft of apo CaM₇₆₋₁₄₈ as well as residues within the Ca^{2+} -binding loops of sites III and IV of apo CaM₇₆₋₁₄₈ (**Figure 3.8**). This result provides direct evidence of allosteric linkage between TFP and Ca^{2+} binding to CaM₇₆₋₁₄₈, consistent with Ca^{2+} titration data in **Chapter II**,

which indicated that TFP at a 1:1 ratio lowered the Ca^{2+} -binding affinity of CaM_{76-148} by 2.56 kcal/mol.

TFP-Induced Changes of apo CaM_{76-148} as Monitored by T_2 Relaxation

Comparison of average T_2 relaxation times of apo CaM_{76-148} and TFP bound apo CaM_{76-148} (161 ± 59 and 147 ± 79 msec, respectively) indicates that TFP binding to apo CaM_{76-148} induces a conformational change within apo CaM_{76-148} . Due to the inverse relationship between T_2 time and molecular weight ($\uparrow T_2$ time = \downarrow molecular tumbling), TFP binding causes apo CaM_{76-148} to tumble at a slower rate in solution (Palmer, 2001). This conformational change is interpreted to be due to TFP binding resulting in a “semi-open” conformation of apo CaM_{76-148} that was previously in a “closed” conformation. The average T_2 values of apo CaM_{76-148} and TFP bound apo CaM_{76-148} also indicated that both molecules are found as monomers in solution. It may also possibly however unlikely that the change in T_2 time after TFP addition could be a result of aggregation, though visual inspection of the sample indicated that no precipitate was present. An orthogonal approach proposed for future studies to verify the absence of precipitate in the NMR sample would be to use dynamic light scattering or analytical ultracentrifugation to verify

Table 3.1: Data collection and refinement statistics of $(\text{Ca}^{2+})_2\text{-CaM}_{76-148}\text{:TFP}$ complex

Data Collection Parameters	
Temperature (K)	100
Wavelength (Å)	1.033
Space Group	P 21
Unit cell parameters	
a, b, c (Å)	24.59, 85.54, 35.35
α , β , δ (Å)	90.00, 93.03, 90.00
Resolution (Å)	22.181 – 2.1
I/ σ	6.0
Completeness (%)	80.92
R _{merge} (%)	0.091
Redundancy	3.53
Refinement Details	
Resolution (Å)	2.1
R _{work} /R _{free} (%)	1
Number of atoms	2194
Protein	1969
Ligand	139
Water	86
B-factor average (Å ²)	
Protein (main chain)	28.55
TFP	33.00
Water	31.08
RMS deviation from ideal geometry	
Bond lengths (Å)	0.009
Bond angles (°)	1.277
Dihedral angles (°)	22.020
Planarity (°)	0.004
Chirality (°)	0.100
Ramachandran plot (% residues)	
Most Favored	100
Additionally allowed	0
Disallowed	0

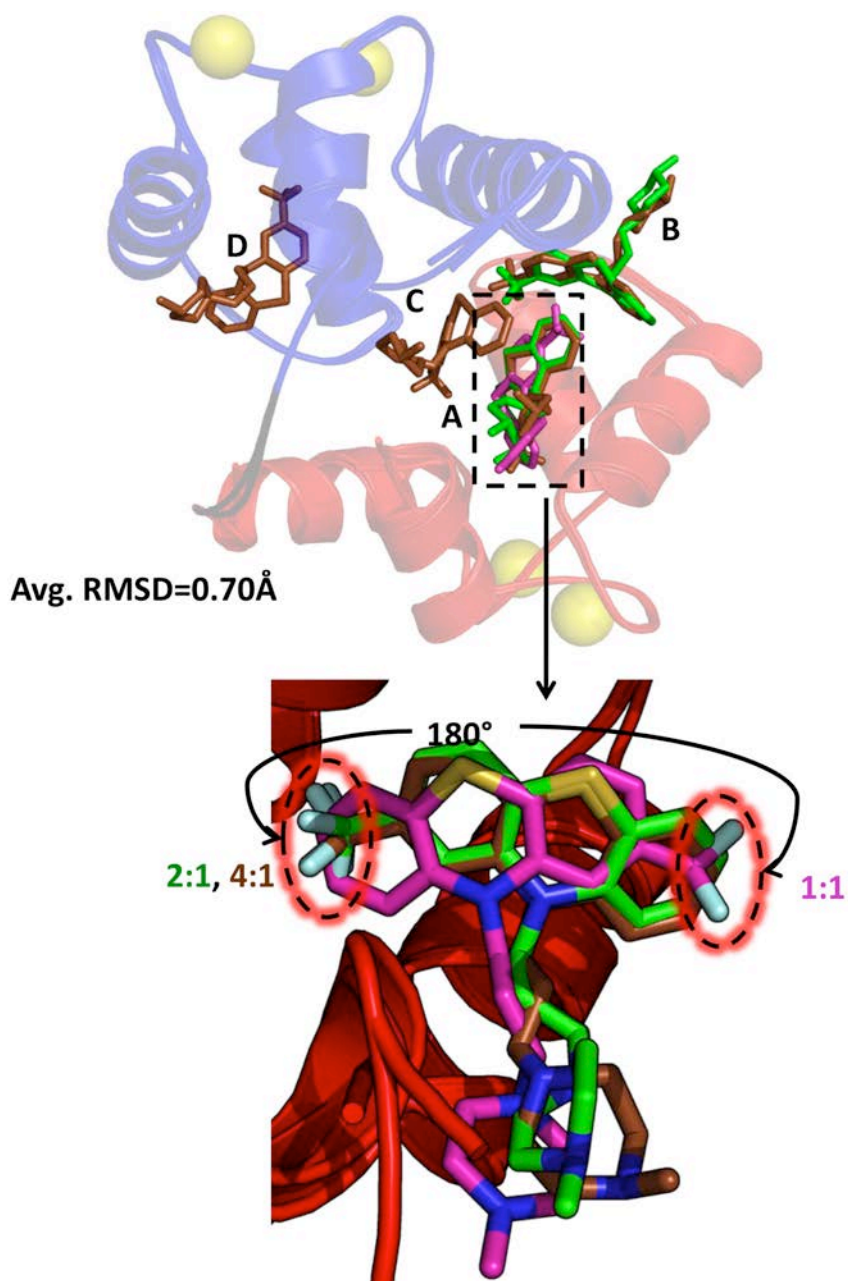


Figure 3.1: Superposition of TFP:(Ca²⁺)₄-CaM₁₋₁₄₈ structures
 Superposition of crystallographically derived structures of TFP:(Ca²⁺)₄-CaM₁₋₁₄₈ with drug:protein ratios of 1:1 (1CTR.pdb), 2:1 (1A29.pdf), and 4:1 (1LIN.pdb). The TFP-binding sites are labeled A (green), B (magenta), C (brown) and D (orange). TFP-binding sites A and B are located in the C-domain (backbone red), site C bridges the two domains, and site D is located in the N-domain (backbone blue). The 180° flip of TFP molecules at TFP binding site A (dashed square) results in the trifluoromethyl group of TFP being found in 2 locations (dashed ovals).
 Users/nmr_mike/Thesis/Chapter_III/Figure3_1.jpg

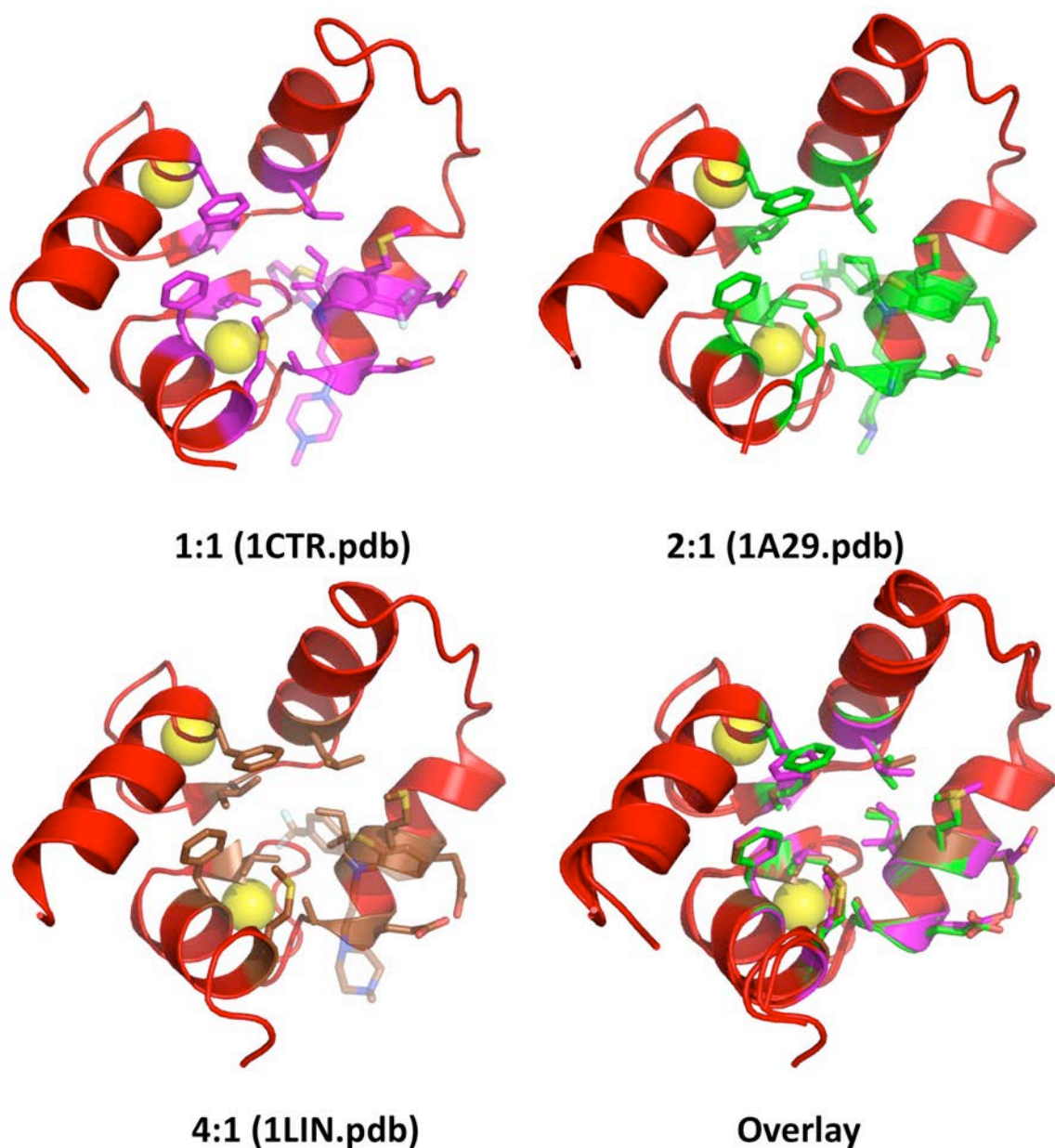


Figure 3.2: Side chain Orientations of $(\text{Ca}^{2+})_4\text{-CaM}_{1-148}$ when binding TFP at site A. A-C: Isolated residues of $(\text{Ca}^{2+})_4\text{-CaM}_{1-148}$ C-domain (red) bound to TFP within 4 Å of TFP bound at site A (transparent sticks) with Ca^{2+} ions shown as yellow spheres. D: Superposition of isolated C-domains of 1:1, 2:1, and 4:1 TFP: $(\text{Ca}^{2+})_4\text{-CaM}_{1-148}$ shown in panels A-C to illustrate that nearly identical binding interfaces and side chain conformations are used when binding TFP at site A. Users/nmr_mike/Thesis/Chapter_III/Figure3_2.jpg

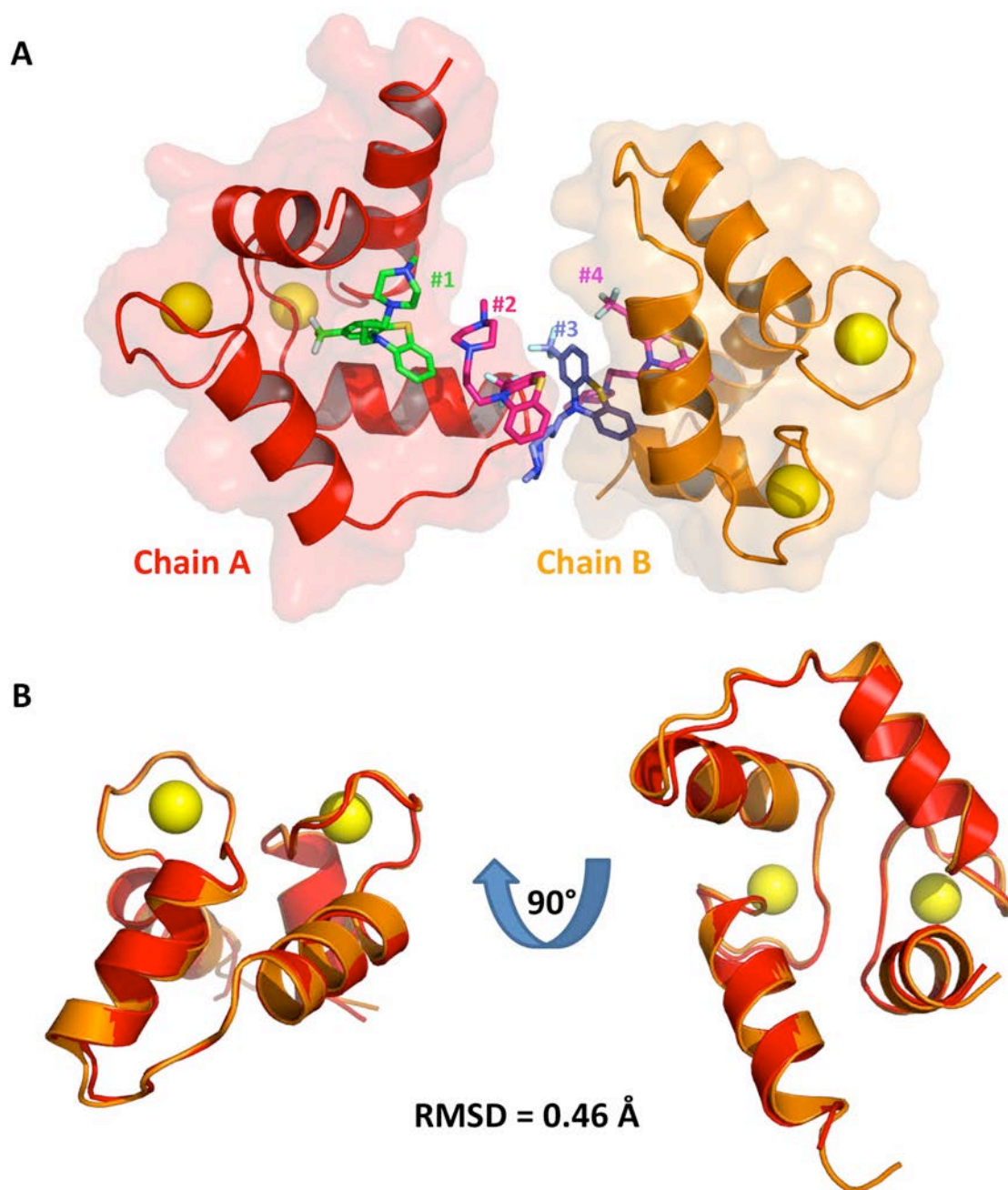


Figure 3.3: Crystal structure of TFP bound $(\text{Ca}^{2+})_2\text{-CaM}_{76-148}$
 A: $(\text{Ca}^{2+})_2\text{-CaM}_{76-148}$ chains A and B are shown in red, and orange respectively while TFP molecules is depicted in sticks, Ca^{2+} are yellow spheres. B: Superposition of $(\text{Ca}^{2+})_2\text{-CaM}_{76-148}$ chains A and B shown in red and orange respectively.
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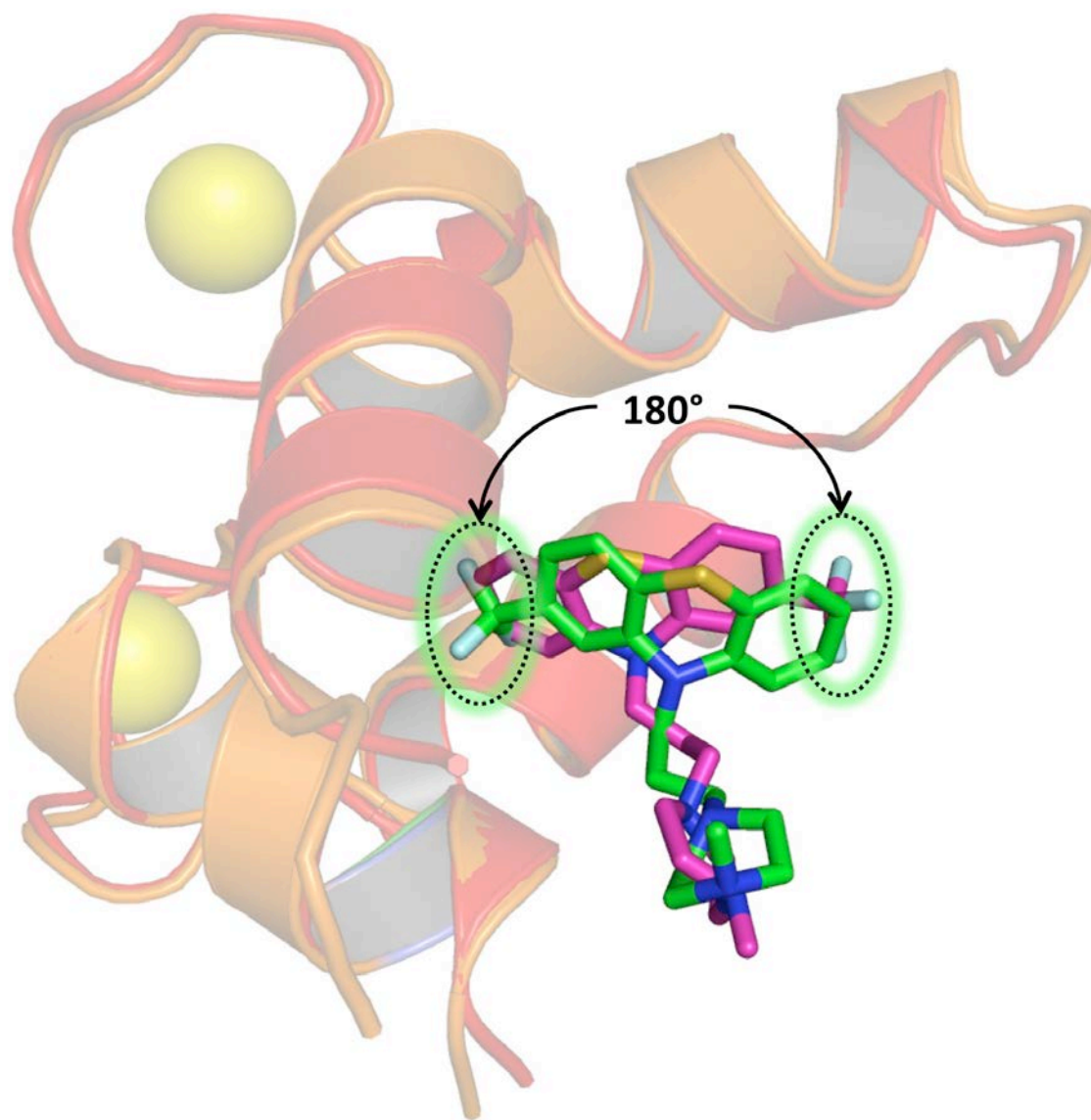


Figure 3.4: Comparison of common TFP binding sites within $(\text{Ca}^{2+})_2\text{-CaM}_{76-148}$. Superposition of TFP (sticks) bound to $(\text{Ca}^{2+})_2\text{-CaM}_{76-148}$ chains A and B colored red and orange respectively. The TFP molecules associated with chains A and B are colored green and magenta respectively where the 180° flip in the location of the trifluoromethyl of TFP is indicated by dashed circles.

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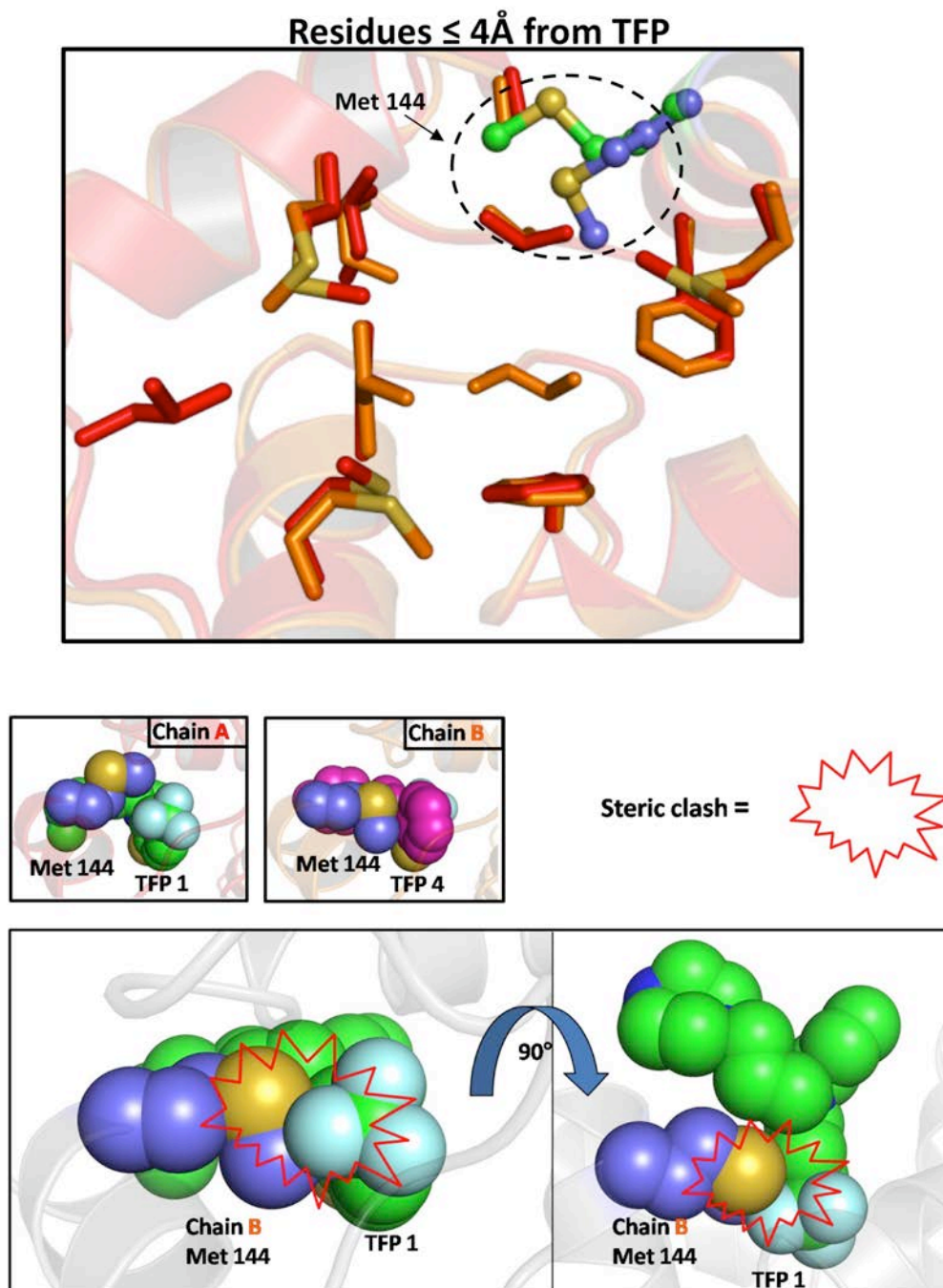
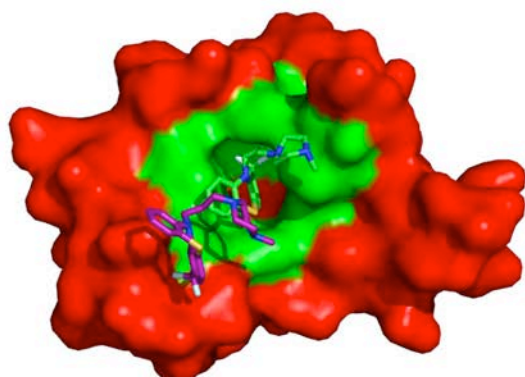


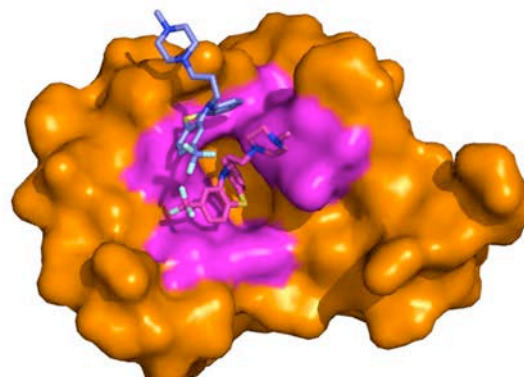
Figure 3.5: Met 144 gating determines TFP orientation

A: Superposition of $(\text{Ca}^{2+})_2\text{-CaM}_{76-148}$ chains A and B colored red and orange respectively, where residues within 4\AA of TFP are shown as sticks. Shown in ball and stick are the conformations adopted by Met 144 of chains A and B colored green and slate respectively. B: Space-filling representation of Met 144 conformations in relation to TFP. Modeled steric clash between Met 144 side chain conformation found in chain B when TFP #1 from chain A is superimposed into the binding site of TFP 4. Users/nmr_mike/Thesis/Chapter_III/Figure3_5.jpg



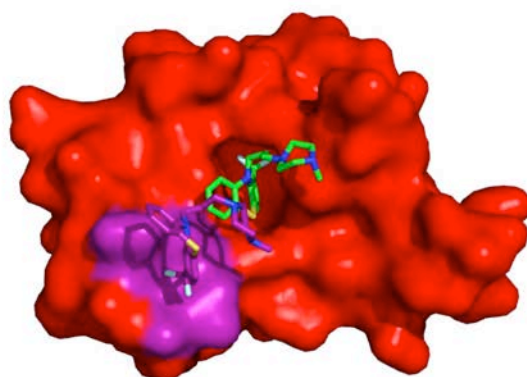
Chain A TFP #1 Contacts $\leq 4\text{\AA}$

F 92 L105
M 109 M 124
M 124 I 125
A 128 V136
V 136 F 141
M 144 M145



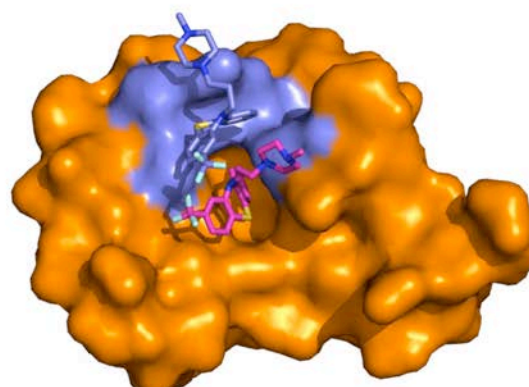
Chain B TFP #4 Contacts $\leq 4\text{\AA}$

F 92 L105
M 109 M 124
A 128 M 144
M145



Chain A TFP #2 Contacts $\leq 4\text{\AA}$

E 124 M124
E 127 A 128
M144



Chain B TFP #3 Contacts $\leq 4\text{\AA}$

M 109 L 112
E 114

Figure 3.6: Contacts made by TFP molecules with $(\text{Ca}^{2+})_2\text{-CaM}_{76-148}$. The molecular surface of $(\text{Ca}^{2+})_2\text{-CaM}_{76-148}$ chains A and B are shown as red and orange respectively, where portions are colored according to the respective TFP examined to indicate areas of $(\text{Ca}^{2+})_2\text{-CaM}_{1-148}$ that are within 4\AA .
Users/nmr_mike/Thesis/Chapter_III/Figure3_6.jpg

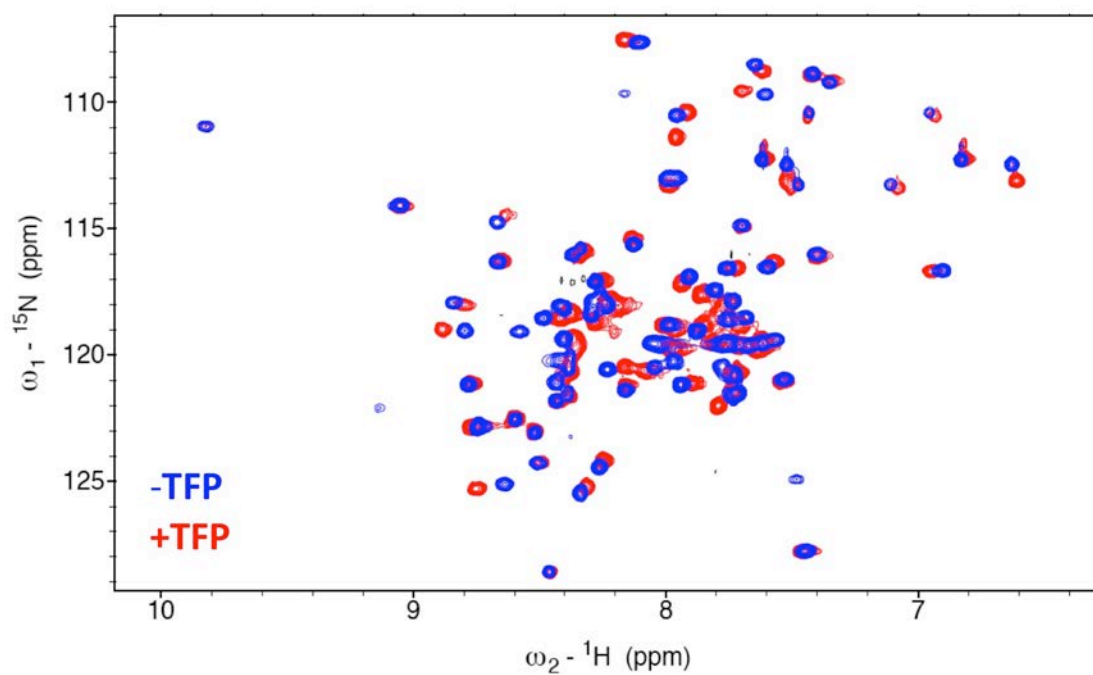


Figure 3.7: ¹⁵N-HSQC spectrum of apo CaM₇₆₋₁₄₈ showing the effect of TFP binding
Overlaid ¹⁵N-HSQC spectra of apo CaM₇₆₋₁₄₈ alone (blue) and in the presence of TFP
(red)
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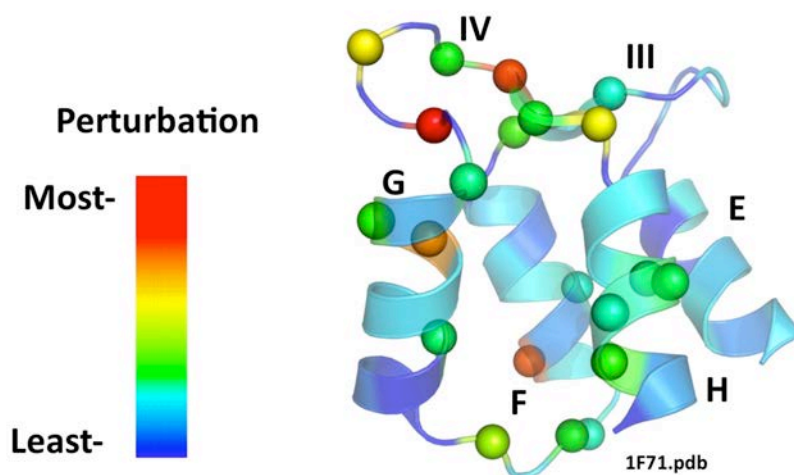
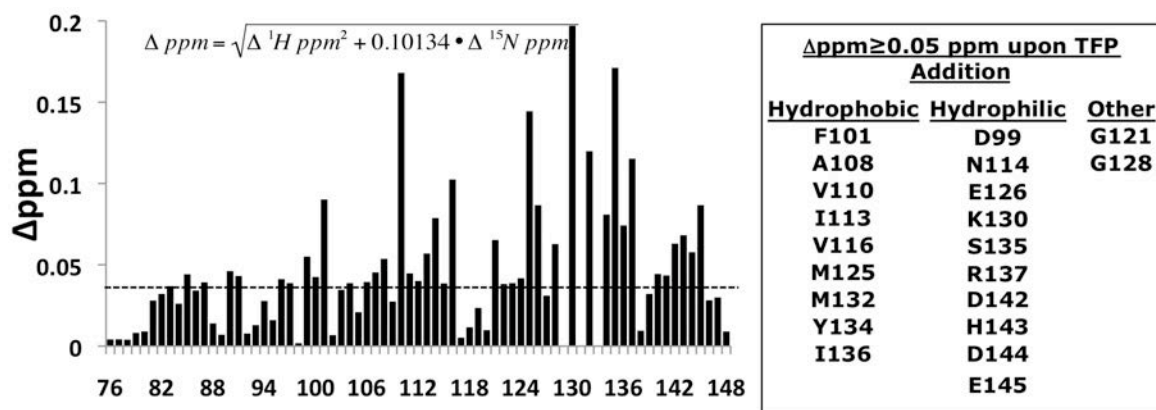
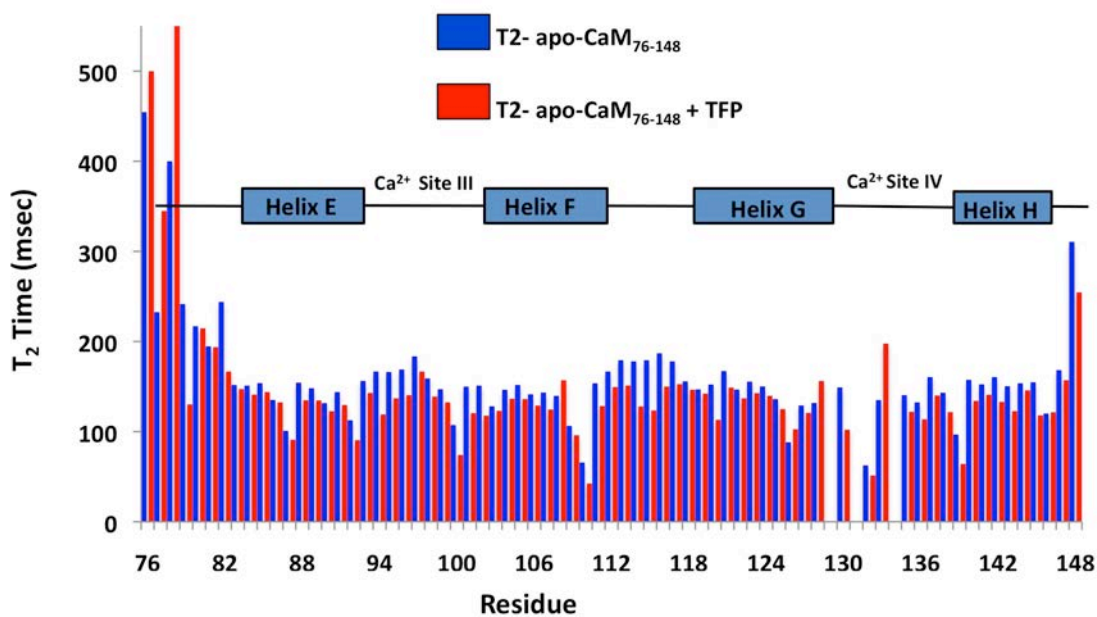


Figure 3.8: Change in amide chemical shift of apo CaM₇₆₋₁₄₈ upon TFP addition. The difference in chemical shift value (Δppm) is plotted for each assigned apo CaM₇₆₋₁₄₈ residue (absent bar represents unassigned residue), dashed line represents average value of TFP induced change in chemical shift of 0.047 ppm. The apo CaM₇₆₋₁₄₈ structure is shown as a cartoon with the C α atoms of residues whose quantified chemical shift upon TFP addition was greater than 0.05 ppm shown as spheres.

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	w/out TFP	w/ TFP
Average T ₂	161 ± 59	147 ± 79
Median T ₂	152	135

↓ T₂ time = ↑ Hydrodynamic radius

TFP addition decreases T₂ time

Figure 3.9: T₂ Values for apo CaM₇₆₋₁₄₈ in the absence and presence of TFP
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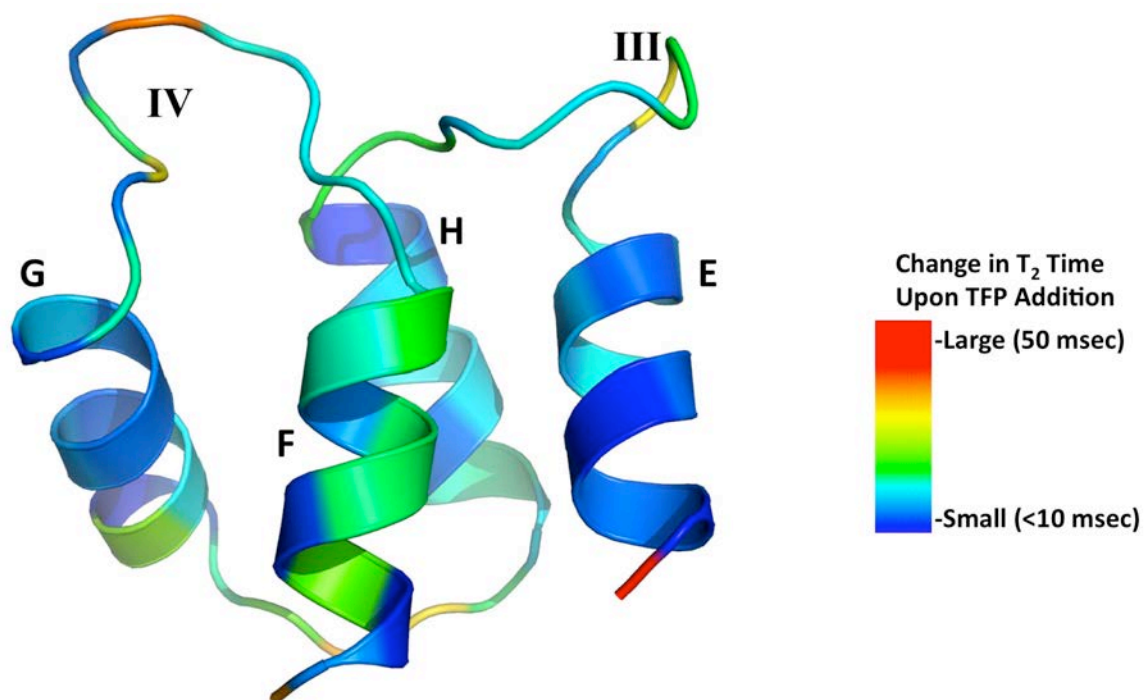


Figure 3.10: Change in T_2 values of apo CaM₇₆₋₁₄₈ upon TFP addition mapped upon the structure of apo CaM₇₆₋₁₄₈. Apo CaM₇₆₋₁₄₈ is shown as a cartoon with labeled helices and Ca²⁺ binding sites. Users/nmr_mike/Thesis/Chapter_III/Figure3_10.jpg

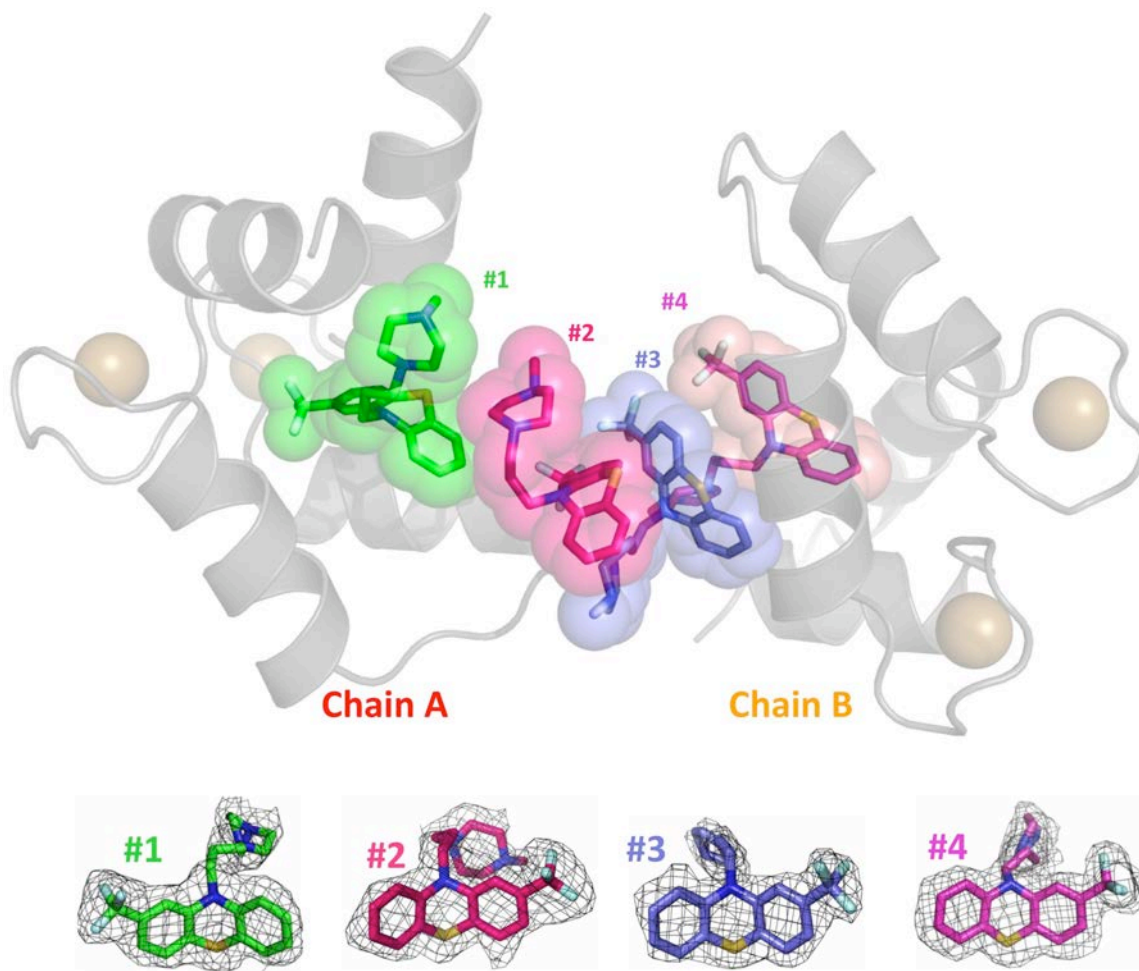


Figure 3.11: Electron density of TFP binding sites in $(\text{Ca}^{2+})_2\text{-CaM}_{76-148}$. The asymmetric unit of $(\text{Ca}^{2+})_2\text{-CaM}_{76-148}$ bound to TFP where chains A and B are shown in red and orange respectively, while TFP are shown as sticks. The electron density for TFP molecules in upper panel is displayed at 2σ over each modeled TFP (lower panel). [Users/nmr_mike/Thesis/Chapter_III/Figure3_11.jpg](https://www.researchgate.net/publication/312111111)

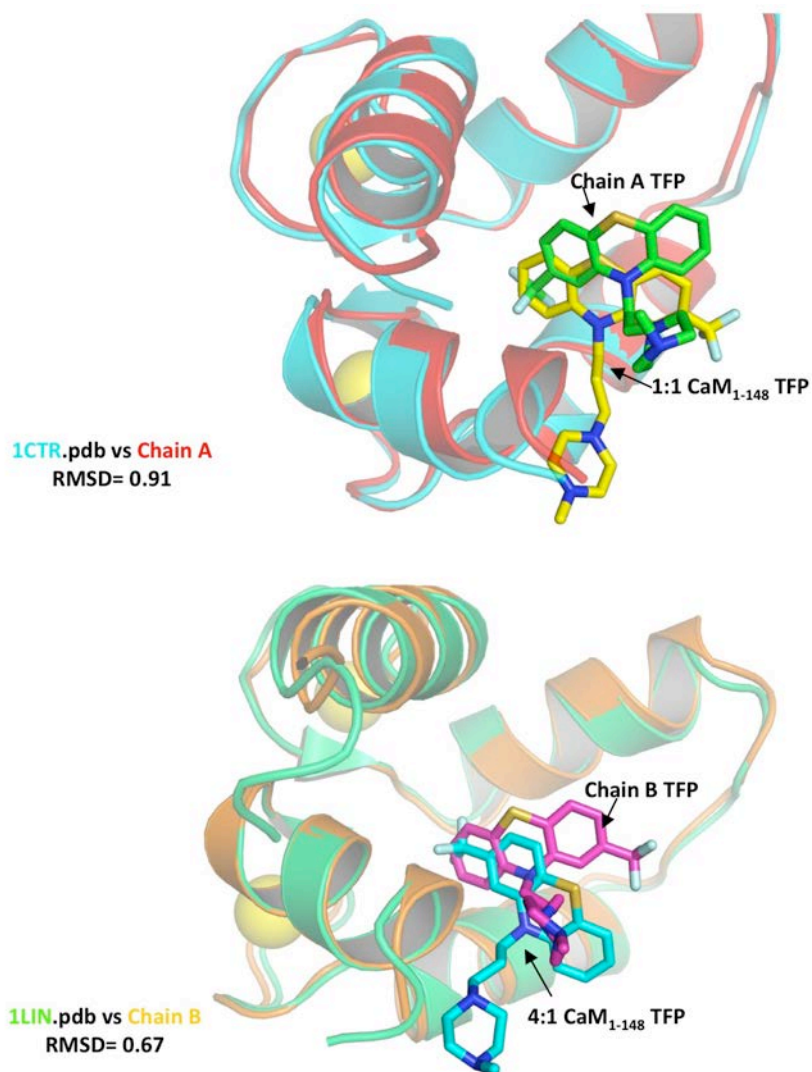


Figure 3.12: Superposition of common TFP binding sites in $(\text{Ca}^{2+})_2\text{-CaM}_{76-148}$ and $(\text{Ca}^{2+})_4\text{-CaM}_{1-148}$. Superpositions of Isolated C-domains of TFP: $(\text{Ca}^{2+})\text{-CaM}_{1-148}$ at 1:1 and 4:1 ratios shown in cyan and light green respectively onto TFP bound $(\text{Ca}^{2+})\text{-CaM}_{76-148}$ chains A and B colored red and orange respectively.
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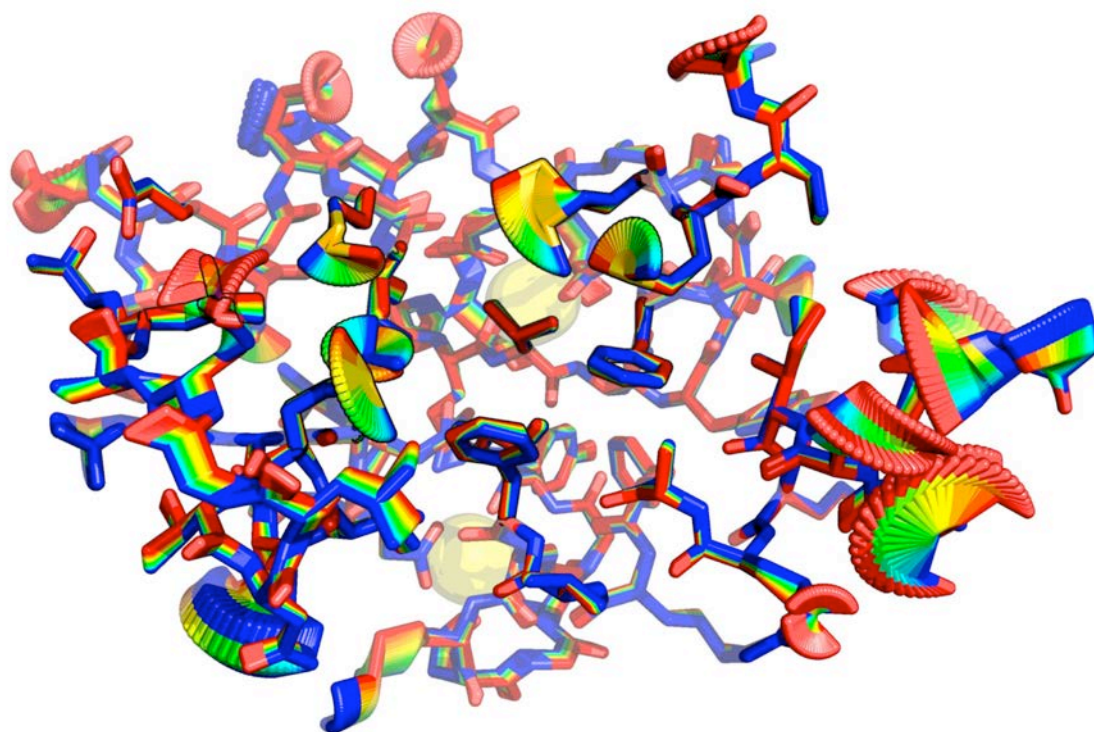


Figure 3.13: Structural morph between chain A and B or TFP-bound $(Ca^{2+})_2$ -CaM₇₆₋₁₄₈. UCSF Chimera generated morph between chain A (red) and chain B (blue), indicating areas of similarity and divergence between chains.
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CHAPTER IV
BINDING OF Na_v1.2_{IQP} TO THE C-DOMAIN OF
APO CAM

Introduction

The voltage-dependent sodium channel type II (Na_v1.2) is necessary for the propagation of action potentials along unmyelinated axons that are found primarily in neuronal and muscle tissues (Schaller and Caldwell, 2003; Trimmer and Rhodes, 2004). Na_v1.2 has one α -subunit (260 kDa) and one or more β -subunits (33–36 kDa each) (Catterall, 2000b). The α -subunit is comprised of four homologous transmembrane domains (I–IV) that are the pore-forming entity of the channel, as well as intracellular domains that play a vital role in control of channel gating (Cormier et al., 2002; Herzog et al., 2003; Mantegazza et al., 2001). Inactivation or closing of Na_v1.2 requires an interaction between two regions of the α -subunit: the intracellular loop that connects domains III and IV and the C-terminal tail (Catterall, 2000a; Cormier et al., 2002; Deschênes et al., 2001; Herzog et al., 2003).

The membrane-proximal half of the C-terminal tail of Na_v1.2 has been modeled to contain six α -helices. Patch clamp studies have shown that deletion of the putative sixth α -helix region slows recovery from inactivation by maintaining the channel in a closed, inactivated state (Cormier et al., 2002; Herzog et al., 2003; Mantegazza et al., 2001). A sequence (residues 1901 to 1927, Na_v1.2_{IQP}) within the putative sixth α -helix also contains a classical calmodulin-binding IQ-motif (IQxxxBGxxxB), where X is any amino acid, and B is either Arg or Lys (**Figure 4.1a**) (Mori et al., 2003; Mori et al., 2000). IQ-motifs found in many proteins such as myosins, neuronal growth proteins, and ion channels have been characterized to bind preferentially to apo (calcium-depleted) calmodulin (CaM) (Bayley et al., 2003; Gerendasy et al., 1995; Houdusse et al., 2006; Liu and Storm, 1990; Mori et al., 2003; Mori et al., 2000; Mori et al., 2004; Shah et al.,

2006). CaM is a small intracellular calcium sensor that is essential to many eukaryotic signal transduction pathways (Klee, 1988; Wang et al., 1980; Yagi et al., 1990; Yap et al., 2000).

CaM has two homologous domains (N- and C-domains) that are 4-helix bundles connected by a flexible linker (**Figure 4.1b**) (Babu et al., 1988; Evans and Shea, 2009; Sorensen and Shea, 1997). Each domain binds two Ca^{2+} ions cooperatively in neighboring EF-hand motifs, giving rise to a total of 4 bound Ca^{2+} ions per molecule of CaM (Pedigo and Shea, 1993; Sorensen et al., 2002b). Although the two domains are similar in sequence and structure, the N-domain binds Ca^{2+} more weakly than the C-domain in the absence of allosteric effectors (Newman and Shea, 2006; Newman, 2008; Vanscyoc and Shea, 2001b). Changes in intracellular Ca^{2+} levels are linked to many cellular events by the effect of Ca^{2+} on CaM: binding triggers conformational changes that expose hydrophobic surfaces in both domains of CaM, altering the free energy of association with many target proteins (Crivici and Ikura, 1995; Sorensen and Shea, 1997). Several high resolution structures of $(\text{Ca}^{2+})_4$ -CaM bound to an IQ-motif show that it adopts a compact ellipsoidal conformation. That conformation has also been observed when calcium-saturated CaM is bound to target regions of most known cyclases, phosphatases, cytoskeletal motors and ion channels (Tjandra et al., 1999). Besides the IQ-motif, the most common CaM-binding sequence is a Basic Amphipathic Alpha-helix (BAA) motif, where the domains of $(\text{Ca}^{2+})_4$ -CaM adopt an “open” conformation to interact with hydrophobic “anchor” residues located within the BAA motif (**Figure 4.1c**) (Tjandra et al., 1999). There are only two available high-resolution structures showing the apo C-domain of CaM bound to a peptide (**Figure 4.1d, 4.1e**): CaM bound to a fragment of myosin V (2IX7) or of the SK channel (1G4Y). Only one of those (2IX7) contains an IQ-motif. However, in both complexes, the apo C-domain of CaM adopted a “semi-open” conformation in which interhelical angles are smaller than those of $(\text{Ca}^{2+})_4$ -CaM (Houdusse et al., 2006; Schumacher et al., 2001).

All ten human isoforms of the voltage-dependent sodium channel contain a single IQ-motif necessary for regulation by CaM (Yu and Catterall, 2003). Previous studies conducted by the Shea laboratory (Theoharis et al., 2008) showed that both apo and calcium-saturated CaM bind to Na_v1.2_{IQp} with high affinity (i.e., dissociation constants near nanomolar). However, the two homologous domains of CaM have different roles when interacting with Na_v1.2_{IQp} (Theoharis et al., 2008). Circular dichroism and fluorescence spectroscopy showed that Na_v1.2_{IQp} binds preferentially to the C-domain of apo CaM which selectively lowers the calcium-binding affinity of sites III and IV, while having little effect on sites I and II.

To examine the molecular basis of the preferential binding of apo CaM to Na_v1.2_{IQp}, and the roles of each domain of CaM, we used heteronuclear NMR to determine residue-specific responses of CaM to binding the IQ-motif. HSQC spectra showed that, under apo conditions, the C-domain of CaM was necessary and sufficient to bind Na_v1.2_{IQp}, while resonances corresponding to the N-domain of CaM were not affected. NMR studies of ¹³C-¹⁵N-labeled CaM were then used to determine a set of high-resolution models of a C-domain fragment of apo CaM (CaM₇₆₋₁₄₈) bound to Na_v1.2_{IQp}. This set of structures revealed that apo CaM₇₆₋₁₄₈ adopts a “semi-open” conformation when bound to Na_v1.2_{IQp} similar to the C-domains of either apo CaM, CaM-like proteins, or essential light chain (ELC) when bound to IQ-motifs in myosin (Houdusse et al., 2006; Swindells and Ikura, 1996; Terrak et al., 2003). It also demonstrated the importance of two Tyr residues in the sequence of Na_v1.2_{IQp}; these are conserved in most IQ-motifs of all isoforms of voltage-dependent sodium channels. Although both Na_v1.2_{IQp} and CaM are very polar, the interface was dominated by hydrophobic interactions. This was consistent with results from NaCl titrations of the apo CaM-IQ complex which remained unperturbed in the presence of elevated levels of NaCl (up to 650 mM).

Although Na_v1.2_{IQp} tightly associates with (Ca²⁺)₄-CaM, NMR studies showed that participating residues changed dramatically, consistent with an interaction interface

like that found for a peptide bound to an “open” domain of CaM. However, for both apo and calcium-saturated CaM, ^{15}N -HSQC and fluorescence anisotropy revealed that the N-domain of CaM did not interact with $\text{Na}_v1.2_{\text{IQp}}$. This suggests that its preferred binding site lies elsewhere in the sequence $\text{Na}_v1.2$. Analysis of conserved contact residues in all available high-resolution structures of apo CaM, apo ELC, or apo CaM-like proteins in complex with canonical IQ-motifs suggested that all of the IQ motifs bind with a similar polarity, having the amino-terminus closer to the EF-hand, and the C-terminus closer to the G-H hand of that domain.

Materials and Methods

Overexpression and Purification of Calmodulin

All isotopes were obtained from Cambridge Isotope Laboratories (Andover, MA). IPTG-induced CaM overexpression was performed using transformed *E. coli* BL21(DE3) cells containing the recombinant pT7-7 vector expressing either the full length or the C-domain of *Paramecium* CaM (a gift from C. Kung, University of Wisconsin, Madison, WI). ^{15}N -labeled proteins were overexpressed in minimal medium, using 2 g/L unlabeled glucose as a carbon source and 1g/L $^{15}\text{NH}_4\text{Cl}$ as the sole nitrogen source. ^{13}C and ^{15}N -labeled proteins were produced using 2g/L ^{13}C -glucose as the sole carbon source and 1 g/L $^{15}\text{NH}_4\text{Cl}$ as the sole nitrogen source. CaM was then purified as previously described by Putkey et al (Putkey et al., 1985). The recombinant proteins were 97-99% pure as judged by silver-stained SDS-PAGE gels. Protein concentrations were determined by UV spectroscopy of protein denatured with NaOH (Beaven and Holiday, 1952) or native at pH 7.4 (Crouch and Klee, 1980).

Prepared $\text{Na}_v1.2_{\text{IQp}}$

A peptide ($\text{Nav}1.2_{\text{IQp}}$:KRKQEEVSAIVIQRAYRRYLLKQKVKK 3.36kDa) representing residues 1901-1927 of the α -subunit of $\text{Na}_v1.2$ was custom-synthesized by

the GenScript Corporation (Scotch Plains, NJ). The peptide was evaluated to be at least 95% pure by HPLC analysis and MALDI-TOF mass spectrometry.

^{15}N -HSQC Monitored $\text{Na}_v1.2_{\text{IQP}}$ and Ca^{2+} Titration of CaM_{1-148}

^{15}N -HSQC spectroscopy was used to monitor titration of $\text{Na}_v1.2_{\text{IQP}}$ into ^{15}N -labeled apo CaM_{1-148} (400 μM) in 10 mM D_4 -imidazole, 100 mM KCl, 50 μM D_{16} -EDTA, 0.01% NaN_3 , pH 6.5. Upon $\text{Na}_v1.2_{\text{IQP}}$ saturation of apo CaM_{1-148} , CaCl_2 was titrated into the sample to a final concentration of 5mM. $\text{Na}_v1.2_{\text{IQP}}$ and Ca^{2+} saturation of CaM_{1-148} were confirmed by plateaus in the intensity of peaks corresponding to either the $\text{Na}_v1.2_{\text{IQP}}:\text{apo CaM}_{1-148}$ or $\text{Na}_v1.2_{\text{IQP}}:(\text{Ca}^{2+})_4\text{-CaM}_{1-148}$.

Preparation of $^{13}\text{C}/^{15}\text{N}$ - $\text{CaM}_{76-148}:\text{Na}_v1.2_{\text{IQP}}$ Complex

A ^{15}N -HSQC monitored titration of $^{12}\text{C}/^{14}\text{N}$ - $\text{Na}_v1.2_{\text{IQP}}$ into $^{13}\text{C}/^{15}\text{N}$ -apo CaM_{76-148} was performed to ensure saturation of apo CaM_{76-148} by $\text{Na}_v1.2_{\text{IQP}}$ as determined by a plateau in the intensity of peaks corresponding the apo $\text{CaM}_{76-148}:\text{Na}_v1.2_{\text{IQP}}$ complex and the disappearance of peaks corresponding to apo CaM_{76-148} alone. The sample was then applied to a 200 mL G50 Superdex column. Complex containing fractions were pooled, buffer exchanged into 10 mM D_4 -imidazole, 100 mM KCl, 50 μM D_{16} -EDTA, 0.01% NaN_3 , pH 6.5 and concentrated to 1.5 mM. Nitrogen-based experiments were conducted in the previously described buffer conditions in 90% H_2O / 10% D_2O , while carbon-based experiments were conducted in 100% D_2O . All spectra with the exception of amide exchange were conducted with Shigemi (Allison Park, PA) microscale NMR tubes whose magnetic susceptibility was matched to D_2O .

¹⁵N-HSQC Monitored Amide Exchange of ¹⁵N-CaM76-148:Na_v1.2_{IQP} Complex

A 480 μ l sample of 1.5 mM ¹⁵N-CaM₇₆₋₁₄₈:Na_v1.2_{IQP} complex in 10 mM D₄-imidazole, 100 mM KCl, 50 μ M D₁₆-EDTA, 0.01% NaN₃, pH 6.5 that had been prepared in H₂O was lyophilized in a Speed-Vac Model VG-5 (Savant). After lyophilization the complex was then resuspended with 480 μ l of 99.96% D₂O (Cambridge Isotope Laboratories, Andover, MA) from a freshly broken ampoule. The pH of the sample was not adjusted after addition of D₂O, and assumed to be 7.4 as HMQC spectra taken of the sample prior to lyophilization in H₂O were identical to that of the D₂O spectrum. The resuspended sample was placed immediately in a Bruker 500 MHz Avance II spectrometer and data acquisition began within 5 minutes. ¹⁵N-HSQC spectra were acquired at 30 minute intervals for 23 hours. Spectra were processed with NMRPipe (Delaglio et al., 1995b) and individual peak intensities were determined with Sparky (Goddard and Kneller). Peak intensity decays times were fit to a mono-exponential decay model (**Equation 4.1**) using the Solver function within Microsoft Excel.

$$I = I_0 e^{-(t-t_0)/\tau} + b \quad (4.1)$$

In this equation, I represents the intensity at time t , I_0 is the initial intensity at time zero, t_0 is time zero, τ is the apparent lifetime, and b is a constant representing the offset intensity of the baseline.

NMR Spectroscopy for Structure Determination

The NMR spectra were collected at 25 °C on a Bruker Avance II 500 or cryo-probe equipped 800 MHz NMR spectrometers. The ¹H, ¹⁵N, and ¹³C resonances of the apo CaM₇₆₋₁₄₈ backbone were assigned using triple resonance experiments (HNCA, HN(CO)CA, HNCACB, HN(CO)CACB, HNCO, and HN(CA)CO) (Yamazaki et al., 1994) with the uniformly ¹⁵N and ¹³C-labeled apo CaM₇₆₋₁₄₈ in complex with unlabeled Na_v1.2_{IQP}. ¹H _{α} resonances were assigned from an ¹⁵N-edited TOCSY spectrum using a

uniformly ^{15}N -labeled protein (Clore and Gronenborn, 1994) and from HA(CACO)NH experiment using a uniformly ^{15}N and ^{13}C -labeled sample. The side chain signals were assigned from 3D H(CCO)NH-TOCSY, C(CO)NH-TOCSY, HCCH-TOCSY, ^{15}N -edited TOCSY, and ^{15}N or ^{13}C -edited NOESY spectra (Clore and Gronenborn, 1994; Fesik and Zuiderweg, 1988). The unlabeled $\text{Na}_v1.2_{\text{IQP}}$ resonances were assigned from several 2D doubly ^{14}N and ^{12}C -filtered NOESY and TOCSY spectra acquired with mixing times of 80 to 120 ms for NOESY and 26 to 46 ms for TOCSY (Ikura and Bax, 1992; Vuister et al., 1994). The intermolecular NOEs were assigned from the ^{13}C -edited and ^{12}C , ^{14}N -filtered 3D NOESY spectrum acquired with a mixing time of 140 ms (Vuister et al., 1994). All NMR spectra were processed with the NMRPipe program (Delaglio et al., 1995a) and analyzed using NMRView (Johnson and Blevins, 1994) and Sparky (Goddard and Kneller).

Apo CaM₇₆₋₁₄₈ :Na_v1.2_{IQP} Structure Calculations

Structures of the apo CaM₇₆₋₁₄₈:Na_v1.2_{IQP} complex were generated using a torsion-angle molecular dynamics protocol (Karimi-Nejad et al., 1998; Stein et al., 1997) with the CNS program (Brunger et al., 1998). Structure calculations employed a total of 1812 NMR-derived distance restraints from the analysis of 3D ^{15}N - and ^{13}C -resolved NOESY spectra acquired with a mixing time of 120 ms (Fesik and Zuiderweg, 1988). The NOE-derived distance restraints were given upper bounds of 3.0, 4.0, 5.0, and 6.0 Å based upon the measured NOE intensities. From an analysis of the amide exchange rates measured from a series of $^{15}\text{N}/^1\text{H}$ N HSQC spectra recorded after the addition of D_2O , 47 hydrogen bonds from the α -helices were included in the structural calculations. In addition, 44 ϕ and ψ angular restraints derived from an analysis of the C, N, C $_{\alpha}$, H $_{\alpha}$, and C $_{\beta}$ chemical shifts using the TALOS program (Cornilescu et al., 1999a; Shen et al., 2009b) were included in the structural calculations. A square-well potential was employed to constrain the NMR-derived distance restraints with F_{NOE} set to 150 and 50

kcal mol⁻¹ Å⁻² during the stages of high temperature and slow-cooling Torsion Angle Dynamics (TAD) and the final stage of conjugate gradient minimization, respectively. Force constants of 100 and 200 kcal mol⁻¹ rad⁻² were applied to all torsional restraints during the stage of high temperature TAD and the rest stages of structural calculations, respectively.

Quantification of chemical shifts due to Na_v1.2_{IQP} or Ca²⁺ addition

To determine the change in chemical shift upon Na_v1.2_{IQP} binding to apo CaM₇₆₋₁₄₈, or Ca²⁺ binding to apo CaM₁₋₁₄₈, chemical-shift changes in both the ¹H and ¹⁵N dimensions were quantified using the modified Pythagorean theorem shown in **Equation 4.2**, as described previously (Jaren et al., 2002).

$$\Delta ppm = \sqrt{(\Delta^1 H ppm)^2 + (0.10134 \cdot \Delta^{15} N ppm)^2} \quad (4.2)$$

In this equation, Δppm refers to the linear change of a specific resonance peak from its initial starting position in the reference spectrum.

NaCl Titration of apo CaM-Na_v1.2_{IQP} Complex

Fluorescence anisotropy was used to examine the [NaCl] dependence of Na_v1.2_{IQP} binding to either apo CaM₇₆₋₁₄₈ or apo CaM₁₋₁₄₈. A Fluorolog 3 (Jobin Yvon, Horiba) spectrofluorimeter, equipped with dual auto-assembly Glan-Thompson polarizers was used to monitor the anisotropy change of a fluorescein-labeled Na_v1.2_{IQP} in 50mM HEPES, 1mM MgCl₂, 5mM NTA, 50μM EGTA, pH 7.4 as it was titrated at 25 °C to a 1:1.2 ratio of Na_v1.2_{IQP} to either apo CaM₇₆₋₁₄₈ or apo CaM₁₋₁₄₈. The complex was then titrated with 50mM HEPES, 5M NaCl, 1mM MgCl₂, 5mM NTA, 50μM EGTA, pH 7.4. The anisotropy of fluorescein-labeled Na_v1.2_{IQP} was monitored using λ_{ex} 498 nm and λ_{em} 520 nm with 2 nm excitation and 10 nm emission bandpasses. Anisotropy (r) was calculated as shown in **Equation 4.3**, described previously (Akyol et al., 2004)

$$r = \frac{I_{VV} - G \cdot I_{VH}}{I_{VV} + 2G \cdot I_{VH}} \quad (4.3)$$

where I_{VV} and I_{VH} are the intensities of vertically or horizontally emitted light upon vertical excitation respectively, and G is the instrument correction factor ($G = I_{HV} / I_{HH}$).

Results

The C-domain of apo CaM₁₋₁₄₈ is Necessary and Sufficient to Bind Na_v1.2_{IQP}

Previous studies showed that both CaM₁₋₁₄₈ and a C-domain fragment (CaM₇₆₋₁₄₈) exhibit tight, calcium-independent binding to Na_v1.2_{IQP} ($K_d \leq 10$ nM), whereas an N-domain fragment of CaM (CaM₁₋₈₀) binds weakly ($K_d \sim 1$ mM), regardless of calcium concentration (Theoharis et al., 2008). To understand the structural differences at interfaces responsible for these affinities, ¹⁵N-HSQC spectra were collected of apo CaM₁₋₁₄₈ ± Na_v1.2_{IQP} for comparison to spectra of apo CaM₁₋₈₀ and apo CaM₇₆₋₁₄₈:Na_v1.2_{IQP} complex. When spectra of apo CaM ± Na_v1.2_{IQP} were overlaid (**Figure 4.2a**), many of the observed peaks for the apo CaM₁₋₁₄₈:Na_v1.2_{IQP} complex were shifted relative to apo CaM₁₋₁₄₈ alone. Analyzing spectral overlap indicated that the chemical environment of the amide groups of the C-domain were all perturbed upon Na_v1.2_{IQP} addition, while those of the N-domain of apo CaM₁₋₁₄₈ were unperturbed by the peptide.

When spectra of the apo CaM₇₆₋₁₄₈:Na_v1.2_{IQP} complex (**Figure 4.2b**) or apo CaM₁₋₈₀ (**Figure 4.2c**) were compared to the spectrum of apo CaM₁₋₁₄₈ bound to Na_v1.2_{IQP}, it was seen that the spectra were additive. Nearly all of the peaks in the samples of apo CaM₇₆₋₁₄₈:Na_v1.2_{IQP} and apo CaM₁₋₈₀ alone overlaid upon the full set of those observed for the apo CaM₁₋₁₄₈:Na_v1.2_{IQP} complex. These results indicated that the interaction of apo CaM₁₋₁₄₈ with Na_v1.2_{IQP} was mediated solely via the C-domain because the spectrum of apo CaM₁₋₁₄₈:Na_v1.2_{IQP} complex could be reproduced by combining the spectra of apo CaM₁₋₈₀ and apo CaM₇₆₋₁₄₈:Na_v1.2_{IQP} complex as seen in **Figure 4.2d**.

Previous studies of Na_v1.2_{IQp} binding to CaM showed similar affinities for apo CaM₇₆₋₁₄₈ or apo CaM₁₋₁₄₈ indicating that the interaction of Na_v1.2_{IQp} with apo CaM is mediated solely by the C-domain. These previous data, coupled with the new residue-specific observation that the N-domain of apo CaM₁₋₁₄₈ was unperturbed by Na_v1.2_{IQp} binding prompted using NMR to determine the structure of apo CaM₇₆₋₁₄₈ bound to Na_v1.2_{IQp}. Relative to using full-length CaM this complex reduced spectral complexity while still capturing the complete high-affinity binding interface that would exist in the apo CaM₁₋₁₄₈:Na_v1.2_{IQp} complex.

Comparison and quantification of ¹⁵N-HSQC spectra of apo CaM₇₆₋₁₄₈ ± Na_v1.2_{IQp} indicated significant perturbation (average Δppm for all residues was 0.51) upon addition of Na_v1.2_{IQp} (**Figure 4.3a and 4.3b**). It is of interest to point out that that Na_v1.2_{IQp} addition dramatically increased dispersion of ¹⁵N-HSQC peaks (**Figure 4.3a**) by breaking the symmetry of similar chemical environments of homologous residues in the paired EF-hands of apo CaM₇₆₋₁₄₈. Mapping of chemical shift perturbations upon the structure of apo CaM₇₆₋₁₄₈ bound to Na_v1.2_{IQp} indicated that the highest chemical shift perturbation (average Δppm of 1.47) was located in the loop region (residues 108-117) between helices F and G which is in close proximity to the Gln (Q) of the IQ-motif (**Figure 4.3c**).

Hydrogen/Deuterium Backbone Amide Exchange

Analysis of hydrogen/deuterium (H-D) backbone amide exchange of the apo CaM₇₆₋₁₄₈:Na_v1.2_{IQp} complex was performed to observe the location of persistent hydrogen bonds. These were used as structural restraints, and were analyzed to determine apparent amide exchange rates that correlate with protein packing and flexibility. A total of 30 amide resonances were detected ~30 minutes after addition to D₂O to a lyophilized sample of the apo CaM₇₆₋₁₄₈:Na_v1.2_{IQp} complex. Their identities were included as hydrogen bonding restraints for structure calculations. Apparent amide exchange rates (listed in **Table 4.1**) were calculated from peak decays such as those

shown in **Figure 4.4a**, fit to a mono-exponential decay (Eq. 1, **Figure 4.4b**) and mapped onto the structure of the apo CaM₇₆₋₁₄₈:Na_v1.2_{IQp} complex (**Figure 4.4c**). The residues with observable exchange rates indicate persistent secondary structure elements. These locations in combination with residue-specific dihedral angles calculated via TALOS (Cornilescu et al., 1999b; Shen et al., 2009a) from backbone chemical shifts were consistent with the pattern of secondary structure elements found in a 4-helix bundle with two anti-parallel β -sheets.

Determination of apo CaM₇₆₋₁₄₈:Na_v1.2_{IQp} Structure

The structural statistics of the final apo CaM₇₆₋₁₄₈:Na_v1.2_{IQp} complex structure are presented in **Table 4.2**. Favorable linewidths and dispersion exhibited in the NMR spectra greatly facilitated chemical shift assignment and NOE analysis as compared to analysis of CaM alone. Nearly complete ¹H, ¹³C, and ¹⁵N resonance assignments were obtained for apo CaM₇₆₋₁₄₈ when bound to Na_v1.2_{IQp}. This structure was determined based on 1974 unambiguous restraints comprised of 457 intra-residue, 298 short (+1), 300 medium (2-4), 331 long (5+), 245 intra-peptide, 188 intermolecular, 46 hydrogen bonds, and 109 dihedral angles. The distribution of restraints was in agreement with the occurrences of residues in a secondary structure element, where a higher number of restraints were observed in α -helical and β -sheet regions than in the loop regions that contain Ca²⁺-binding sites III and IV. **Figure 4.5a** presents a superposition of the family of 20 lowest energy structures of Na_v1.2_{IQp} bound to apo CaM₇₆₋₁₄₈ that best satisfied the experimental restraints. The interhelical angles as calculated by UCSF Chimera (Pettersen et al., 2004) between helices E-F and G-H of these structures were $77.4^\circ \pm 1.7$ and $78.0^\circ \pm 2.3$ respectively.

The structure of the apo CaM₇₆₋₁₄₈:Na_v1.2_{IQp} complex was well defined, as reflected by the low value of the ensemble RMSD ($0.65 \pm 0.06 \text{ \AA}$) and is of good quality, as indicated by the Ramachandran statistics and energetic terms listed in **Table 4.2**.

Frayed termini were observed at the N- and C-termini of apo CaM₇₆₋₁₄₈ and Na_v1.2_{IQp} in **Figure 4.5a**. This was attributed to the absence of observable NOEs for these residues, and lack of assignments for peptide residues 1901-1903, 1922-1923, and 1926-1927.

Interaction Interface of Na_v1.2_{IQp} with apo CaM₇₆₋₁₄₈

The hydrophobic interface between apo CaM₇₆₋₁₄₈ and Na_v1.2_{IQp} was well defined as judged by the positions of interacting residues shown in **Figure 4.5b**. The change in solvent accessible hydrophobic surface between apo CaM₇₆₋₁₄₈ and Na_v1.2_{IQp} was calculated using GETAREA (Fraczkiewicz and Braun, 1998). Upon binding the peptide, 1393 Å² (751 Å² Na_v1.2_{IQp}, 642 Å² apo CaM₇₆₋₁₄₈) of solvent accessible hydrophobic surface area was buried. In apo CaM₇₆₋₁₄₈, a subset of hydrophobic residues that consisted of A88, V91, F92, L112 and M145 accounted for 43% of the buried surface, while in Na_v1.2_{IQp}, hydrophobic residues V1911, I1912, Y1916, Y1919, and L1920 accounted for 53.5% of the buried surface area.

In addition to hydrophobic interactions, favorable electrostatic interactions were observed between apo CaM₇₆₋₁₄₈ and Na_v1.2_{IQp}. Shown in **Figure 4.5c** are the electrostatic surface potentials calculated by APBS (Baker et al., 2001) for coordinates of apo CaM₇₆₋₁₄₈ and Na_v1.2_{IQp} corresponding to their conformation in the complex. Examination of the electrostatic potentials of solvent-accessible regions clearly showed that charge complementarity is present between the negatively charged apo CaM₇₆₋₁₄₈ and the positively charged Na_v1.2_{IQp}.

All contacts between apo CaM₇₆₋₁₄₈ and Na_v1.2_{IQp} that were within 4.5 Å were tabulated using the program *Contacts of Structural Units (CSU)* (Sobolev et al., 1999) and are shown in **Figure 4.6a**. The interactions of the IQ residues (I1912 and Q1913) with apo CaM₇₆₋₁₄₈ were of special interest due to their highly conserved nature in IQ-motifs as shown in **Figure 4.1a** and could be investigated on the basis of numerous NOEs such as those shown in **Figure 4.6b** and **4.6c**. As indicated in the CSU analysis and

displayed in **Figure 4.6d**, I1912 inserted directly into the shallow hydrophobic pocket of apo CaM₇₆₋₁₄₈. As shown in **Figure 4.6e**, Q1913 is positioned to form hydrogen bonds with backbone atoms of residues L112 and E114 that are in the loop connecting helices F and G of apo CaM₇₆₋₁₄₈. A comparison of CaM sequences in 102 eukaryotic species (Ataman et al, Supplementary Table 1) showed that residue E114 and its preceding residue, G113, were identical in all species; while L112 was highly conserved (found in 91 of 102 sequences).

Effect of Ca²⁺ upon apo CaM₁₋₁₄₈:Na_v1.2_{IQp} Complex

¹⁵N-HSQC NMR spectroscopy was used to examine whether apo CaM₁₋₁₄₈ bound to Na_v1.2_{IQp} undergoes a structural transition as a result of Ca²⁺ binding. Shown in **Figure 4.7a** are spectral overlays of apo CaM₁₋₁₄₈ (red) and (Ca²⁺)₄-CaM₁₋₁₄₈ (black). Comparison of these indicated a significant change in the chemical environment of apo CaM₁₋₁₄₈ resonances within the apo CaM₁₋₁₄₈:Na_v1.2_{IQp} complex upon Ca²⁺ addition. To determine whether the chemical shifts observed upon Ca²⁺ addition to the apo CaM₁₋₁₄₈:Na_v1.2_{IQp} complex are due to Ca²⁺ induced release of Na_v1.2_{IQp}, the spectrum of free (Ca²⁺)₄-CaM₁₋₁₄₈ was compared to the spectrum of the (Ca²⁺)₄-CaM₁₋₁₄₈:Na_v1.2_{IQp} complex (**Figure 4.7a**). Lack of spectral overlap in **Figure 4.7a** indicated that both apo CaM₁₋₁₄₈ and (Ca²⁺)₄-CaM₁₋₁₄₈ were bound to Na_v1.2_{IQp}, but that their structures were significantly different.

Shown in **Figures 4.7b, 4.7c, and 4.7d** are plots showing the HSQC signal for selected resonances of the N- and C-domain of apo CaM₁₋₁₄₈ when bound to Na_v1.2_{IQp} at successively higher levels of calcium during a titration. Shown in **Figure 4.7b** are residues of the N-domain of apo CaM₁₋₁₄₈ that were uniformly in fast-exchange over the course of the calcium titration. The normalized change in chemical shift plotted against the equivalents of calcium added indicated that most residues within the N-domain titrated fully between 0 and 2 Ca²⁺ equivalents.

Observation of changes in position and intensity of C-domain peaks during the calcium titration (**Figure 4.7c**) revealed that they were in intermediate or slow exchange. For those, calcium-dependent change in intensities of peaks were used to determine their relative populations over the course of the titration. For apo CaM₁₋₁₄₈ bound to Na_v1.2_{IQP}, peaks corresponding to the C-domain of the apo state broadened beyond the limit of detection after addition of 2 Ca²⁺ equivalents (**Figure 4.7c**). Although amide backbone assignments for the C-domain of (Ca²⁺)₄-CaM₁₋₁₄₈ bound to Na_v1.2_{IQP} are unknown, peaks that appeared at the midpoint of the Ca²⁺ titration are likely to correspond to residues located within the C-domain of (Ca²⁺)₄-CaM₁₋₁₄₈. When the intensities of these peaks were plotted against the ratio of [Ca²⁺]/[CaM₁₋₁₄₈:Na_v1.2_{IQP}], increases in peak intensity were observed at higher calcium stoichiometries than were chemical shift changes. These data indicated that when bound to Na_v1.2_{IQP} the N- and C-domains of CaM₁₋₁₄₈ bind 2 Ca²⁺ per domain, where the N-domain has a slightly more favorable Ca²⁺-binding affinity than the C-domain, consistent with species population simulations conducted previously (Theoharis et al., 2008).

Effect of Ca²⁺ upon Molecular Size of CaM₁₋₁₄₈:Na_v1.2_{IQP} Complex

To examine whether Ca²⁺ binding causes the N-domain of the (Ca²⁺)₄-CaM₁₋₁₄₈ to collapse onto Na_v1.2_{IQP} forming a compact ellipsoidal structure, CaCl₂ was titrated into a solution of preformed apo CaM₁₋₁₄₈:Na_v1.2 complex and monitored with fluorescence anisotropy. As shown in **Figure 4.8a**, apo CaM₁₋₁₄₈ bound to the fluoresceinated Na_v1.2_{IQP} stoichiometrically at a 1:1 ratio. Following complex formation, the apo CaM-IQ complex was titrated with 10 mM CaCl₂ in matching buffer (**Figure 4.8a**). No significant decrease in fluorescence anisotropy was observed, indicating that (Ca²⁺)₄-CaM₁₋₁₄₈ maintains the hydrodynamic behavior of apo CaM₁₋₁₄₈ bound to fl-Na_v1.2_{IQP}.

This suggested that $(\text{Ca}^{2+})_4\text{-CaM}$ does not adopt a compact ellipsoidal structure when bound to $\text{Na}_v1.2_{\text{IQp}}$ on the basis of raw anisotropy.

Effect of Na^+ upon $\text{Na}_v1.2_{\text{IQp}}$ Binding to CaM

$\text{Na}_v1.2$ and other proteins in the plasma membrane experience a large fluctuation in Na^+ ion concentration during $\text{Na}_v1.2$ gating. Thus, it was of interest to determine whether NaCl affected the affinity of $\text{Na}_v1.2_{\text{IQp}}$ for apo or calcium-saturated CaM. Shown in **Figure 4.8b** are NaCl titrations of complexes of $\text{Na}_v1.2_{\text{IQp}}$ binding apo CaM_{76-148} or apo CaM_{1-148} as monitored by fluorescence anisotropy to examine the NaCl dependence of $\text{Na}_v1.2_{\text{IQp}}$ dissociation. Both apo CaM_{76-148} and apo CaM_{1-148} bound to the fluoresceinated $\text{Na}_v1.2_{\text{IQp}}$ stoichiometrically at a 1:1 ratio. Saturation of $\text{Na}_v1.2_{\text{IQp}}$ was ensured by the addition of a slight excess of either apo CaM_{76-148} or apo CaM_{1-148} , then the complex was titrated with NaCl in matching buffer to cover a range from 0 to 650 mM NaCl (**Figure 4.8b**). This NaCl range was used in attempt to dissociate apo CaM from $\text{Na}_v1.2_{\text{IQp}}$, but proved unsuccessful as the cuvette volume limited the amount of NaCl solution that could be added resulting in 650 mM being the maximum NaCl examined. In NaCl titrations of complexes of $\text{Na}_v1.2_{\text{IQp}}$ bound to apo CaM_{76-148} and apo CaM_{1-148} , the anisotropy of $\text{Na}_v1.2_{\text{IQp}}$ was unchanged by NaCl, which indicated a negligible effect on $\text{Na}_v1.2_{\text{IQp}}$ dissociation from CaM at the final concentration of peptide and protein.

Discussion

The voltage-dependent sodium channel $\text{Na}_v1.2$ contains a CaM-binding IQ-motif that is required for proper regulation of its physiological function. The high-resolution solution structure of apo CaM_{76-148} bound to $\text{Na}_v1.2_{\text{IQps}}$ presented here demonstrates that apo CaM_{76-148} adopts a “semi-open” conformation when bound to $\text{Na}_v1.2_{\text{IQp}}$, and that it binds such that the F-G loop interacts directly with the glutamine residue of the IQ motif. The structure of this apo $\text{CaM}_{76-148}:\text{Na}_v1.2_{\text{IQp}}$ complex provides a foundation for studying

other isoforms of voltage-dependent ion channels that interact preferentially with the C-domain of apo CaM.

A “semi-open” apo C-domain Binds Na_v1.2_{IQp}.

Shown in **Figure 4.9a** are superpositions of the backbone of (a) apo CaM₇₆₋₁₄₈ bound to Na_v1.2_{IQp} determined in this study (b) the C-domains of two apo CaM₁₋₁₄₈ molecules bound to two neighboring IQ-motifs found in myosin V (2IX7.pdb), (c) apo CaM-like proteins bound to IQ-motifs of MYO2P (1M46.pdb, 1M45.pdb, and 1N2D.pdb) and (d) ELC bound to myosin heavy chain. A feature common to all of these structures is that each set of paired EF-hands of the C-domain adopts a “semi-open” conformation when bound to its respective IQ-motif. Another feature common to all structures shown in **Figure 4.9a** is that residues located at positions “0” and “1” (defined in **Figure 4.1a**) of the IQ-motif interact with a similar subset of residues in the apo C-domain. These conserved structural features are also reflected in the sequence conservation of the IQ-motif at these positions **Figure 4.1a**.

At position “0” we observed that a hydrophobic residue (often a branched chain) is necessary to insert into the exposed hydrophobic core of the “semi-open” C-domain, burying what would otherwise be solvent exposed hydrophobic surface. The absolute conservation of Gln at position “1” of canonical IQ-motifs is reflected by hydrogen bonding interactions made between the carboxamide of Gln at position “1” and the backbone of residues found within the loop connecting helices F and G as shown in **Figure 4.6e**. Formation of this hydrogen bonding network between Gln at position “1” and the backbone of the loop residues account for the large amide chemical shift at this position observed in the ¹⁵N-HSQC spectrum shown in **Figure 4.3a, 4.3b, and 4.3c**.

A “Semi-open” apo C-domain is used to Bind a Non IQ- motif Containing Target

There is an example of 1 apo C-domain bound to a peptide that does not have an IQ-motif in the structure of partially Ca^{2+} -saturated CaM bound to a peptide derived from the small conductance potassium channel (SK-Channel) (Schumacher et al., 2001). When $\text{Na}_v1.2_{\text{IQp}}$ bound apo CaM_{76-148} , and the apo C-domain of CaM bound to the SK-channel peptide (SK_p) were overlaid (**Figure 4.9b**), both adopt “semi-open” domain conformations. This structural similarity indicates that an IQ-motif is sufficient but not necessary to induce a “semi-open” apo C-domain conformation. Additionally the structural similarity in **Figure 4.9b**, suggests that a “semi-open” C-domain conformation is not exclusive to IQ-motifs alone but quite possibly all motifs that bind to the C-domain of apo CaM.

Although the SK-Channel does not contain an IQ-motif, it shares key features seen in IQ-motif bound structures of apo CaM, CaM-like proteins, and ELC. **Figures 4.9a**, and **4.9b**, show that a carboxamide-containing side chain of Gln (IQ-motif) or Asn (SK_p) are required to form a hydrogen bond with the backbone of loop residues located between the F and G helices of CaM. This conserved hydrogen-bonding network between the side chain carboxamide and loop backbone of CaM helps to determine the α -helical register of the IQ-motif or SK_p relative to the apo C-domain.

The similarity in α -helical register between the IQ-motif and SK_p was also apparent in a conserved hydrophobic interaction between either I1912 ($\text{Na}_v1.2_{\text{IQp}}$) or L428 (SK_p) and the core of the apo C-domain (**Fig 4.9b**). Due to the polarity of the N- and C-terminus of their respective motifs, I1912 or L428 differ in primary sequence position relative to the highly conserved carboxamide side chain used to hydrogen bond to the backbone of residues L112 and E114 of CaM (**Figure 4.9b**). For ease in describing the reversal in polarity between $\text{Na}_v1.2_{\text{IQp}}$ and SK_p relative to the C-domain, we have termed the interactions of $\text{Na}_v1.2_{\text{IQp}}$ and SK_p with apo CaM_{76-148} as NF-G_C and

${}_C\text{F-G}_N$ respectively. These terms result from positioning the 4-helix bundle of the C-domain as depicted in **Figure 4.9c**, where dependent upon polarity of the peptide N- and C-terminus relative to helices F and G of CaM either a ${}_N\text{F-G}_C$ or ${}_C\text{F-G}_N$ orientation is adopted.

When an ${}_N\text{F-G}_C$ orientation is adopted, as in the interaction of apo CaM with $\text{Na}_v1.2_{\text{IQp}}$, the conserved hydrophobic residue (I1912) that inserts in the core of apo CaM_{76-148} precedes the conserved carboxamide-containing residue (Q1913). However if the peptide polarity is reversed as seen the ${}_C\text{F-G}_N$ orientation of SK_p relative to the C-domain of apo-CaM, the residue homologous to I1912 of $\text{Na}_v1.2_{\text{IQp}}$, (L428) is located 2 amino acids away on the C-terminal side of the carboxamide-containing residue (N426).

Canonical IQ-motifs Bind to apo CaM using Similar Orientations

Shown in **Figure 4.10** are structures of apo and $(\text{Ca}^{2+})_4\text{-CaM}$ bound to IQ-motif containing peptides derived from various targets, where the directionality of the target interaction with CaM is depicted with right or left arrows respectively. Depicted in **Figure 4.9a**, are conserved interactions made by all apo CaM or apo CaM-like proteins when bound to IQ-motifs. These conserved features consist of a hydrophobic interaction at IQ-motif position “0” of the peptide contacting the core of CaM as well as the hydrogen bond network between the ultra-conserved Gln residue at position “1” and CaM loop residues L112 and E114 or their CaM-like protein equivalent. If these conserved interactions are maintained throughout all canonical apo CaM binding motifs then all canonical IQ-motifs must bind to the C-domain of apo CaM in a ${}_N\text{F-G}_C$ orientation.

Examination of the binding orientation of $(\text{Ca}^{2+})_4\text{-CaM}$ bound to IQ-motif containing structures is less clear as structures of $(\text{Ca}^{2+})_4\text{-CaM}$ bound to the IQ-motifs of $\text{Ca}_v2.1$ and 2.3 differ as to the peptide orientation relative to $(\text{Ca}^{2+})_4\text{-CaM}$ (**Figure 4.10**). $\text{Ca}_v2.1$ and 2.3 peptides of differing length as well as different crystallization conditions

were used to obtain these structures which may account for this discrepancy (Fallon et al., 2009; Fallon et al., 2005; Halling et al., 2009; Houdusse et al., 2006; Kim et al., 2008; Mori et al., 2008; van Petegem et al., 2005). We propose apo and $(\text{Ca}^{2+})_4$ -CaM bind to $\text{Na}_v1.2_{\text{IQp}}$ and canonical IQ-motifs where a Gln is located at position “1” in a $_{\text{NF-GC}}$ orientation. The alternative would require CaM to release from its anti-parallel orientation to the IQ-motif upon Ca^{2+} influx, and reorient itself into a parallel arrangement for binding to occur.

Role of the N-domain of CaM_{1-148} in $\text{Na}_v1.2$ Regulation

^{15}N -HSQC and fluorescence anisotropy monitored Ca^{2+} titration studies shown in **Figures 4.7a** and **4.8a**, indicated that the N-domain of both apo and $(\text{Ca}^{2+})_4$ - CaM_{1-148} do not interact with $\text{Na}_v1.2_{\text{IQp}}$. These observations are consistent with previous studies that indicated that a region consisting of residues 1913-1938 in $\text{Na}_v1.2$ contains a second CaM-binding BAA-motif ($\text{Na}_v1.2_{\text{BAA}}$) (Mori et al., 2003; Mori et al., 2000). Based on results presented here, we propose the model presented in **Figure 4.9d**.

In this model, only the “semi-open” C-domain of apo CaM interacts with the IQ-motif of $\text{Na}_v1.2$, while under Ca^{2+} -saturating conditions both CaM N- and C-domains adopt “open” conformations when they bind to the BAA and IQ-motifs of $\text{Na}_v1.2$ respectively. This model is supported by observations in **Figure 4.2** where only the C-domain CaM_{1-148} is perturbed upon $\text{Na}_v1.2_{\text{IQp}}$ addition, while **Figure 4.3** shows that the apo C-domain adopts a “semi-open” conformation when bound to $\text{Na}_v1.2_{\text{IQp}}$. Support for the proposed $(\text{Ca}^{2+})_4$ - CaM_{1-148} interactions with $\text{Na}_v1.2$ are drawn from **Figure 4.7** which indicated CaM_{1-148} binds 4 Ca^{2+} ions and **Figure 4.8a**, which showed that the N-domain of $(\text{Ca}^{2+})_4$ - CaM_{1-148} does not interact with $\text{Na}_v1.2_{\text{IQp}}$ by collapsing upon it after Ca^{2+} addition. These observations coupled with studies by Mori et al. (Mori et al., 2003) which showed that $(\text{Ca}^{2+})_4$ - CaM_{1-148} bound to both $\text{Na}_v1.2_{\text{IQp}}$ and $\text{Na}_v1.2_{\text{BAA}}$, while only

apo CaM₁₋₁₄₈ bound to Na_v1.2_{IQP} provide strong evidence for the model proposed in **Figure 4.9d** .

Future studies will focus on uncovering the mechanistic role that CaM plays in Na_v1.2 regulation under Ca²⁺-saturating conditions as well as determination of structures of larger intracellular regions of Na_v1.2 in complex with CaM. To accomplish these goals, studies to establish the location of the N-domain binding site within Na_v1.2, coupled with structure determination of (Ca²⁺)₄-CaM in complex with a peptide that contains both the N-domain binding site and Na_v1.2_{IQP} are proposed.

Table 4.1: Apparent amide exchange rates of apo CaM₇₆₋₁₄₈ when bound to Na_v1.2_{IQP}

Residue	Rate*	Error*
86	69.1	0.9
87	138.9	1.5
88	162.9	1.7
89	180.6	2.4
90	122.9	1.1
91	16.7	0.3
92	185.6	2.0
93	43.8	0.8
99	85.6	0.9
100	382.6	5.5
103	247.4	26.2
105	3.6	0.5
106	30.9	0.4
108	70.7	1.1
109	125.1	2.4
110	17.9	0.4
111	141.4	7.8
112	20.2	0.4

Residue	Rate*	Error*
116	15.1	0.4
117	21.0	0.8
118	92.1	6.3
120	25.9	0.4
121	243.7	3.1
122	233.0	2.4
123	57.4	1.0
124	384.2	7.1
125	965.9	36.0
126	128.7	2.9
136	34.2	1.0
138	54.9	1.6
139	6.4	0.2
140	17.7	0.2
142	1223.6	46.4
143	408.3	6.6
144	139.0	1.6
145	126.9	1.5
146	69.6	1.0

*Values reported in minutes

Table 4.2: Structural statistics and root-mean-square deviation for 20 structures of apo CaM₇₆₋₁₄₈:Na_v1.2_{IQP} complex

Structural statistics ^a	<SA>	<SA> _r
Rmsd from experimental distance restraints (Å) ^b		
All (1865)	0.008 ± 0.001	0.007
CaM intra-residue (457)	0.007 ± 0.002	0.005
CaM sequential (298)	0.005 ± 0.004	0.003
CaM medium range (300)	0.009 ± 0.002	0.008
CaM long range (331)	0.006 ± 0.001	0.005
Intra-peptide (245)	0.005 ± 0.002	0.003
CaM-peptide intermolecular (188)	0.012 ± 0.002	0.011
hydrogen bond (46)	0.016 ± 0.001	0.021
Rmsd from experimental torsional angle restraints (deg) ^c		
φ and ψ angles (109)	0.3 ± 0.03	0.2
CNS potential energies (kcal mol ⁻¹)		
E _{tot}	89 ± 4.8	80
E _{bond}	5 ± 0.5	1
E _{ang}	65 ± 2.9	62
E _{imp}	5 ± 0.6	4
E _{repel}	6 ± 1.2	6
E _{noe}	6 ± 1.2	5
E _{cdih}	1 ± 0.1	1
Cartesian coordinate rmsd (Å)	N, C _α , and C'	all heavy
<SA> vs. <SA> ^d	0.31 ± 0.05	0.95 ± 0.11

^aWhere <SA> is the ensemble of 20 NMR-derived solution structures of CaM/peptide; <SA> is the mean atomic structure; <SA>_r is the energy-minimized average structure. The CNS F_{repel} function was used to simulate van der Waals interactions using a force constant of 4.0 kcal mol⁻¹ Å⁻⁴ with the atomic radii set to 0.8 times their CHARMM values (Brooks et al., 1983)(Brooks, Brucoleri et al. 1983)

^bDistance restraints were employed with a square-well potential (F_{noe} = 50 kcal mol⁻¹ Å⁻²). Hydrogen bonds were given bounds of 1.8-2.4 Å (H-O) and 2.7-3.3 Å (N-O). No distance restraint was violated by more than 0.3 Å in any of the final structures.

^cTorsional restraints were applied with values derived from an analysis of the C', N, C_α, H_α, and C_β chemical shifts using the TALOS program. Force constant of 200 kcal mol⁻¹ rad² was applied for all torsional restraints.

^dRmsd for CaM protein residues 80-128 and 134-146 and peptide residues 1905-1920.

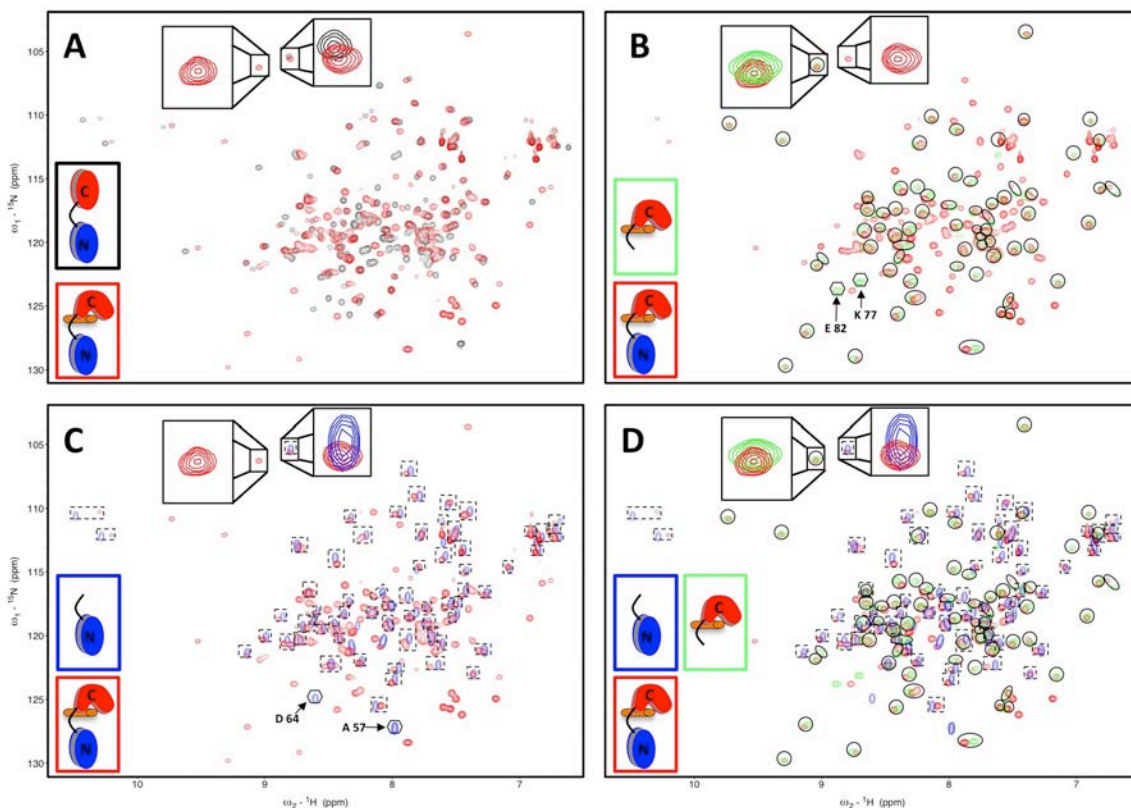


Figure 4.2: ^{15}N -HSQC Spectra of apo CaM₇₆₋₁₄₈ and apo CaM₁₋₁₄₈ bound to Nav1.2_{IQP}
A: Overlaid ^{15}N -HSQC spectra of apo CaM₁₋₁₄₈ alone (black) and apo CaM₁₋₁₄₈ bound to Nav1.2_{IQP} (red). **B:** Overlaid ^{15}N -HSQC spectra of apo CaM₇₆₋₁₄₈ (green) and apo CaM₁₋₁₄₈ when bound to Nav1.2_{IQP}. Overlapping resonances are indicated in solid circles, while non-overlapping residues of apo CaM₇₆₋₁₄₈ bound Nav1.2_{IQP} are indicated by solid hexagons. **C:** Overlaid ^{15}N -HSQC spectra of apo CaM₁₋₈₀ (blue) and apo CaM₁₋₁₄₈ bound to Nav1.2_{IQP} (red). Overlapping resonances are indicated by dashed squares, while non-overlapping apo CaM₁₋₈₀ resonances are indicated by solid hexagons. **D:** Overlaid ^{15}N -HSQC spectra of apo CaM₁₋₈₀ (blue), apo CaM₇₆₋₁₄₈ bound to Nav1.2_{IQP} (green), and apo CaM₁₋₁₄₈ bound to Nav1.2_{IQP} bound to Nav1.2_{IQP}. Resonances of apo CaM₁₋₈₀ which overlap with those of apo CaM₁₋₁₄₈ bound to Nav1.2_{IQP} are shown in dashed squares, while resonances of apo CaM₇₆₋₁₄₈ bound to Nav1.2_{IQP} that overlay onto apo CaM₁₋₁₄₈ bound to Nav1.2_{IQP} are shown in solid circles.

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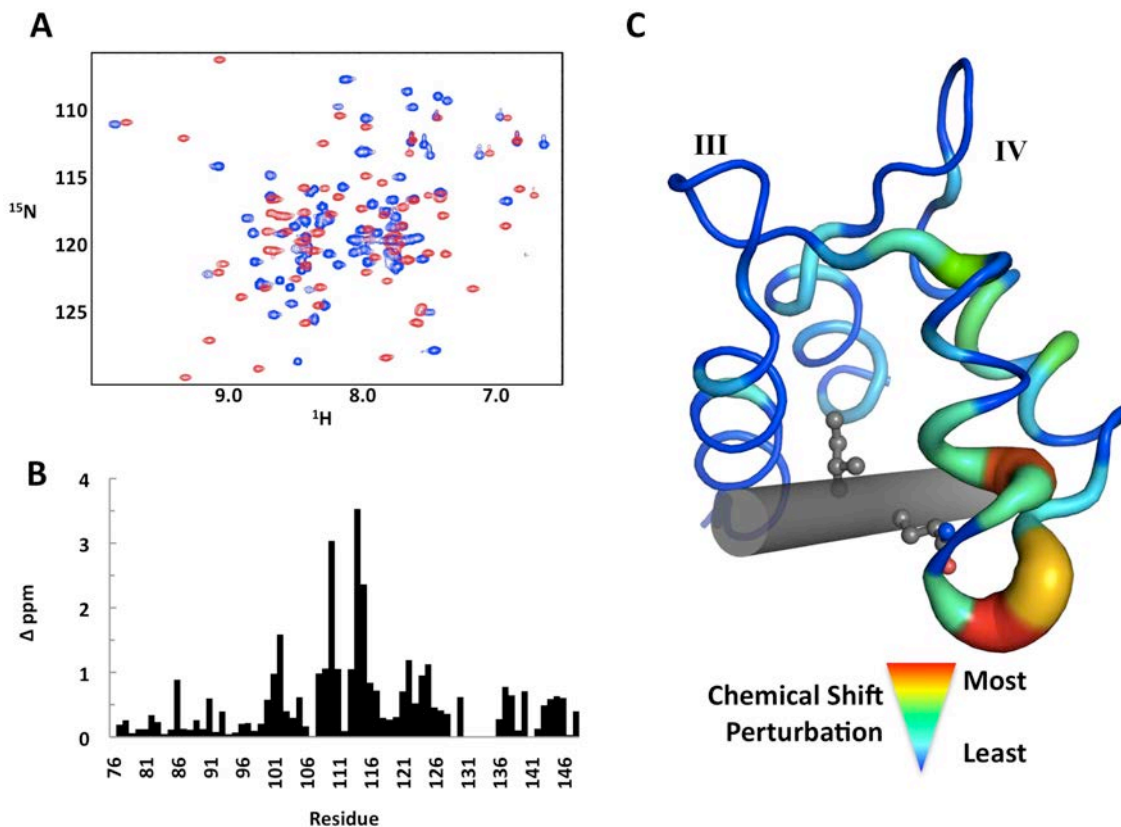


Figure 4.3: $\text{Na}_v1.2_{\text{IQp}}$ binding to apo CaM_{76-148} quantified by ^{15}N -HSQC spectroscopy
A: Overlaid ^{15}N -HSQC spectra of apo CaM_{76-148} (blue) and apo CaM_{76-148} bound to $\text{Na}_v1.2_{\text{IQp}}$ (red). **B:** Quantified apo CaM_{76-148} amide proton chemical shifts upon $\text{Na}_v1.2_{\text{IQp}}$ addition. **C:** Magnitude of $\text{Na}_v1.2_{\text{IQp}}$ induced chemical shift mapped to the structure of apo CaM_{76-148} bound to $\text{Na}_v1.2_{\text{IQp}}$ (gray rod) with I and Q residues of the IQ-motif shown in ball and stick.
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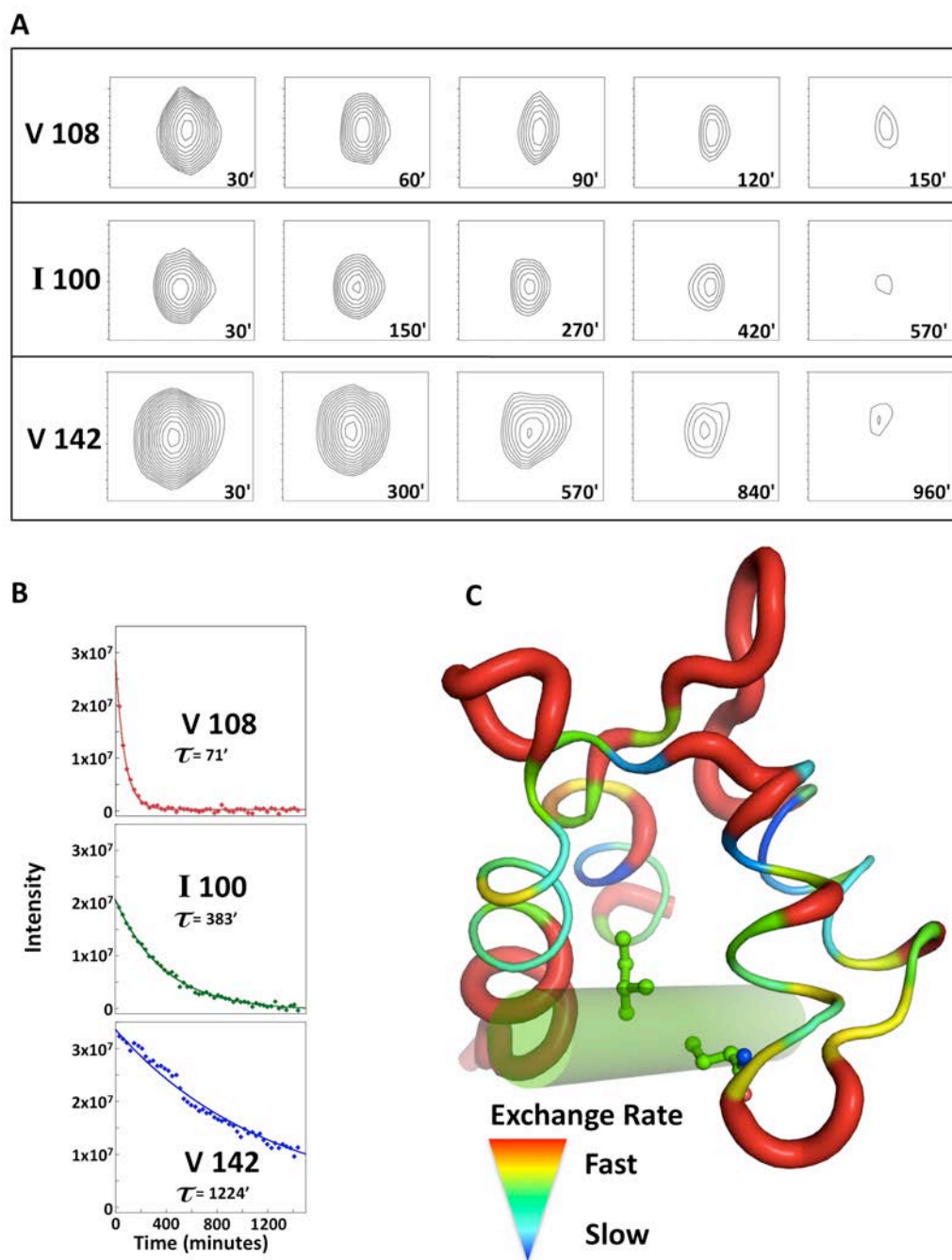


Figure 4.4: ^{15}N -HSQC-monitored amide exchange of apo $\text{CaM}_{76-148}:\text{Na}_v1.2_{\text{IQp}}$ complex
A: ^{15}N -HSQC monitored amide exchange series of select residues peak intensities as a function of exchange time corresponding to different exchange regimes. **B:** Fitted exchange curves and rates for residues shown in panel A. **C:** Magnitude of observed exchange rates mapped to the structure of apo CaM_{76-148} bound to $\text{Na}_v1.2_{\text{IQp}}$ (green rod) with I and Q residues of the IQ-motif shown in ball and stick.

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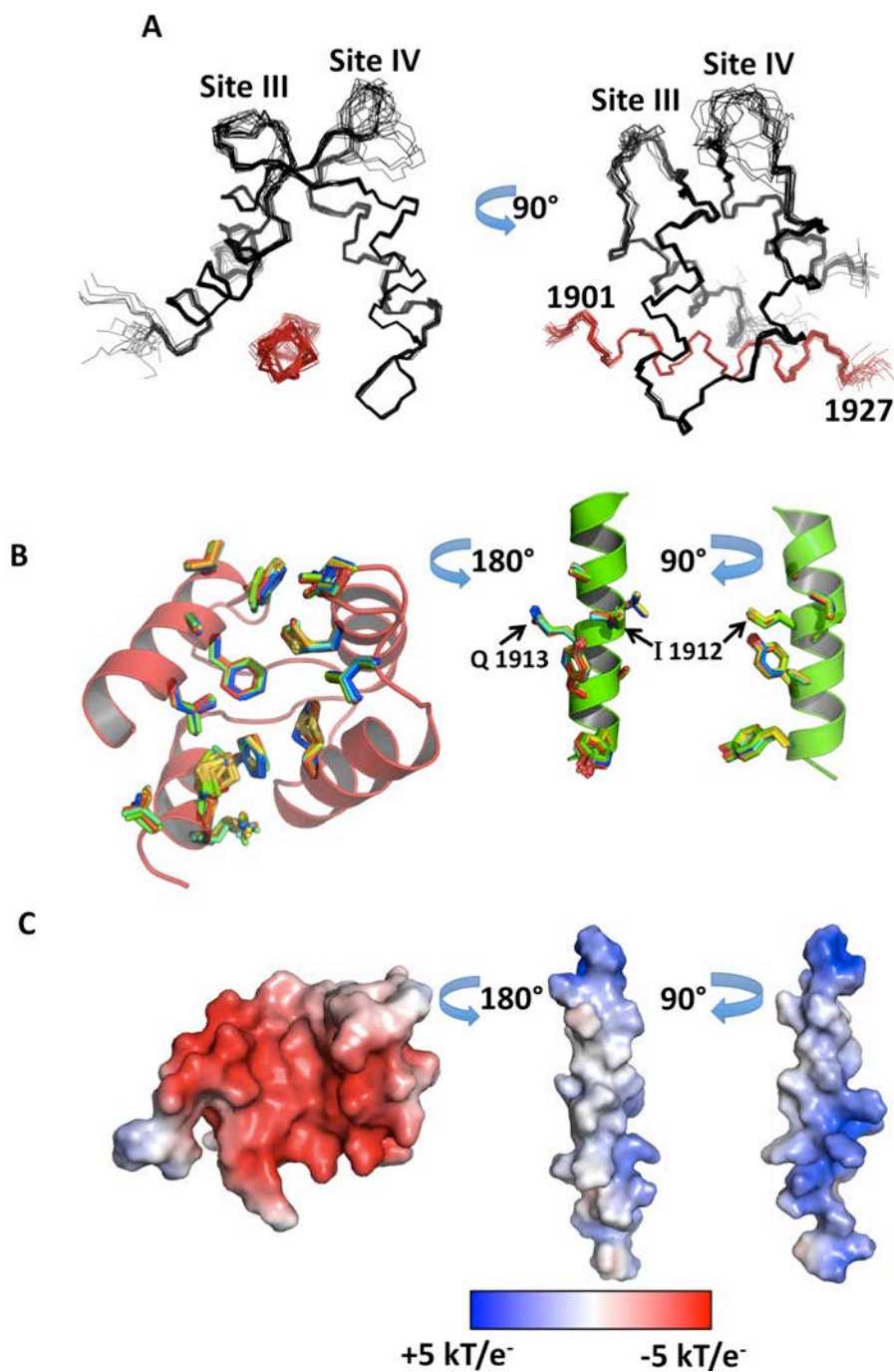


Figure 4.5: Solution structure of apo CaM₇₆₋₁₄₈ bound to Nav1.2_{IQP}
A: Ensemble of 20 lowest energy structures of apo CaM₇₆₋₁₄₈ (black) bound to Nav1.2_{IQP} (red) **B:** Hydrophobic interaction interfaces of 20 lowest energy structures of apo CaM₇₆₋₁₄₈ (red) bound to Nav1.2_{IQP} (green). **C:** Electrostatic potentials of apo CaM₇₆₋₁₄₈ (left), and Nav1.2_{IQP} (middle, and right) in 150mM NaCl, pH 6.5.
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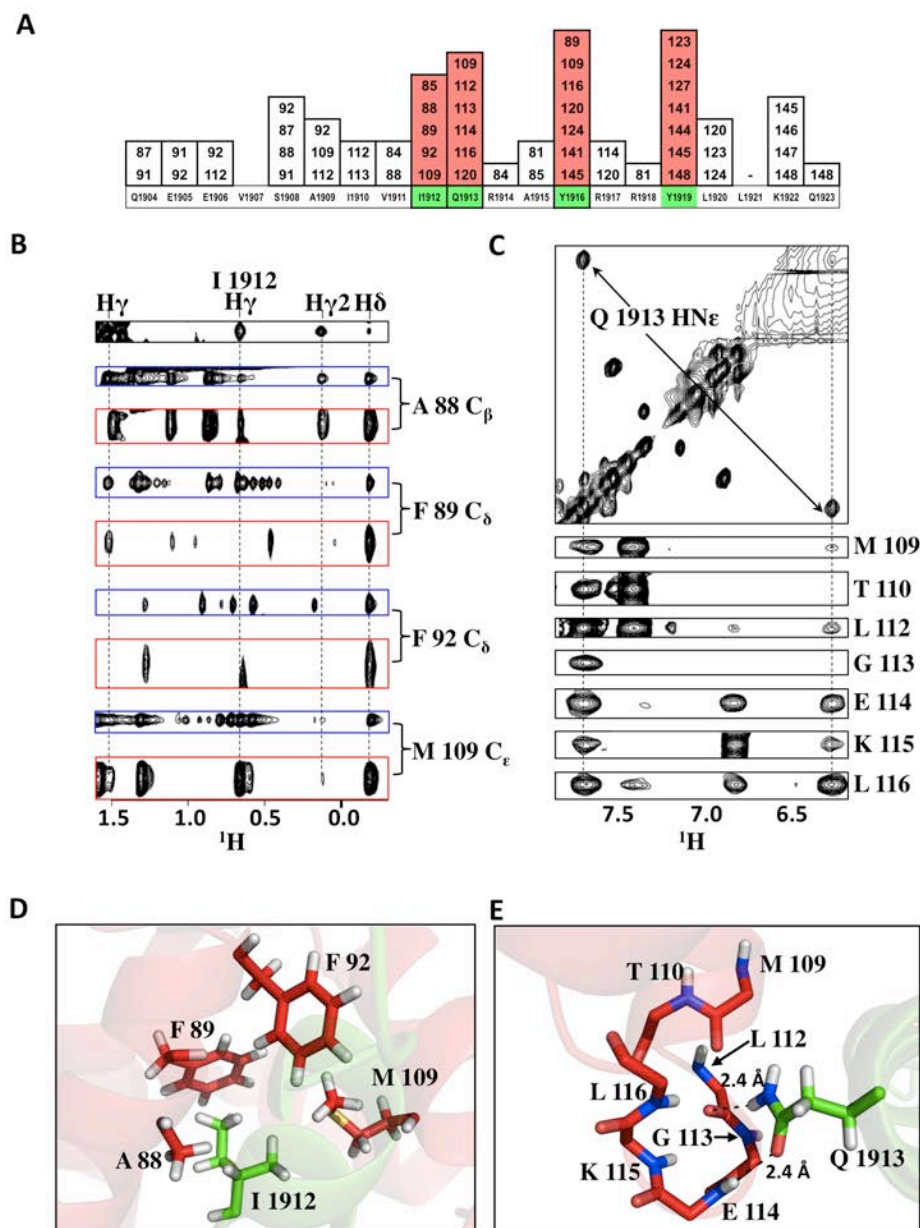


Figure 4.6: Binding interfaces of apo CaM₇₆₋₁₄₈ and Na_v1.2_{IQP}
A: apo CaM₇₆₋₁₄₈ residues (red) ≤ 4.5 Å of Na_v1.2_{IQP} (black) **B:** Select NOE peak between I1912 of Na_v1.2_{IQP} and apo CaM₇₆₋₁₄₈ where individual residue unfiltered and filtered ¹³C-NOESY spectral strips are outlined in blue and red respectively. **C:** Select NOE peaks between Q1913 of Na_v1.2_{IQP} and apo CaM₇₆₋₁₄₈, where individual ¹⁵N-NOESY spectral strips are outlined in black. **D:** Location of apo CaM₇₆₋₁₄₈ residues shown in panel B in relation to I1912. **E:** Location of apo CaM₇₆₋₁₄₈ residues shown in panel C in relation to Q1913, shown in dashed lines are hydrogen bonds made by the carboxamide of Q1913 to the backbone of residues L112, and E114.
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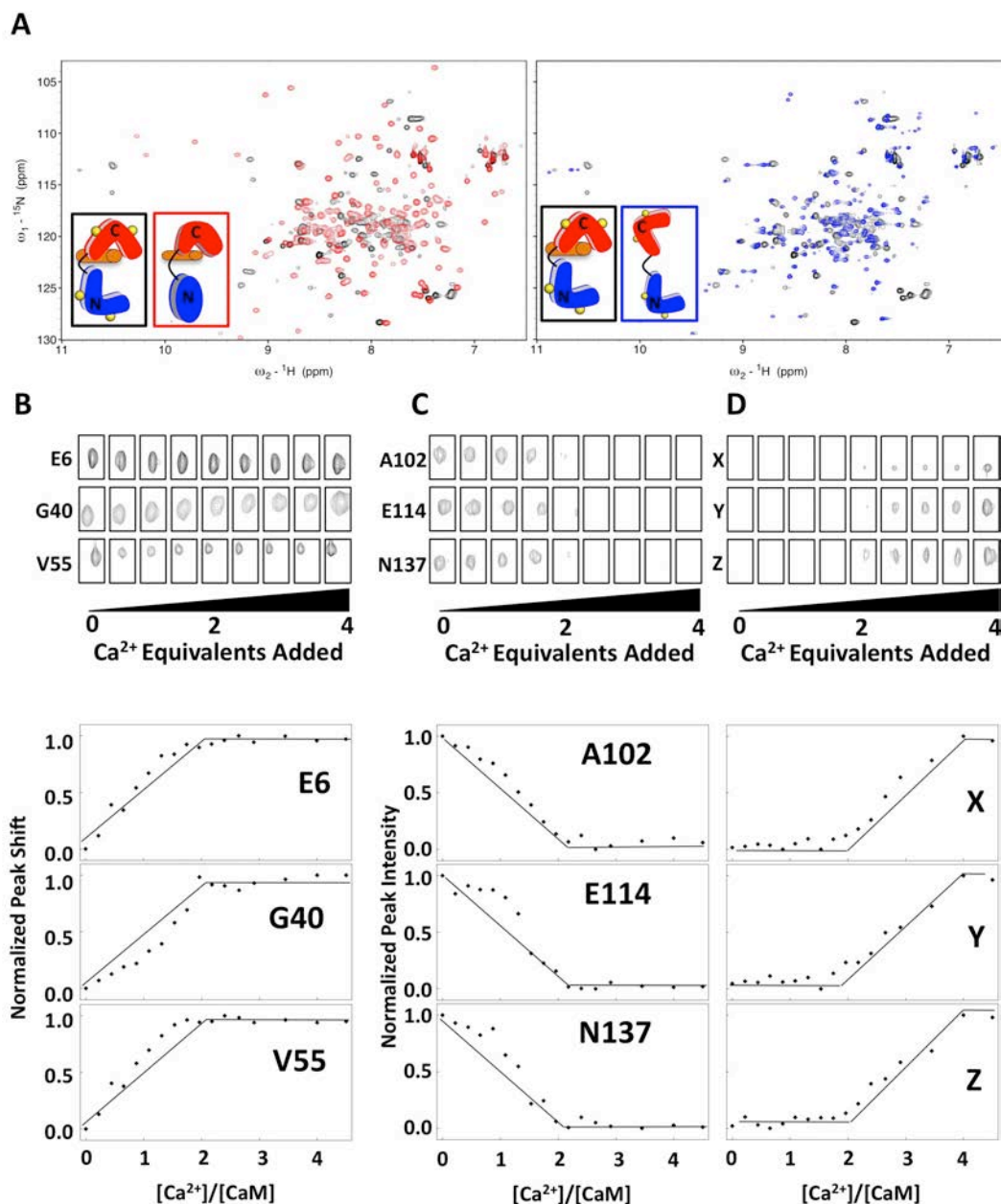


Figure 4.7: Effect of Ca^{2+} upon apo CaM_{76-148} when bound to $\text{Nav1.2}_{\text{IQp}}$
A: ^{15}N -HSQC spectral overlay of apo (red) and $(\text{Ca}^{2+})_4\text{-CaM}_{1-148}$ (black) bound to $\text{Nav1.2}_{\text{IQp}}$ (left). ^{15}N -HSQC spectral overlay of apo CaM_{1-148} (blue) and $(\text{Ca}^{2+})_4\text{-CaM}_{1-148}$ bound to $\text{Nav1.2}_{\text{IQp}}$ (right). **B:** ^{15}N -HSQC Monitored Ca^{2+} titration of select N-domain residues (upper) with quantified change in chemical shift as Ca^{2+} was added (lower). **C:** ^{15}N -HSQC Monitored Ca^{2+} titration of select C-domain residues (upper) with quantified change in chemical shift as Ca^{2+} was added (lower), lines are added to guide the eye. **D:** ^{15}N -HSQC Monitored Ca^{2+} titration of residues from the C-domain of CaM , but whose individual residue identity is unknown (upper) with quantified change in chemical shift as Ca^{2+} was added (lower)Users/nmr_mike/Thesis/Chapter_IV/Figure4_7.jpg

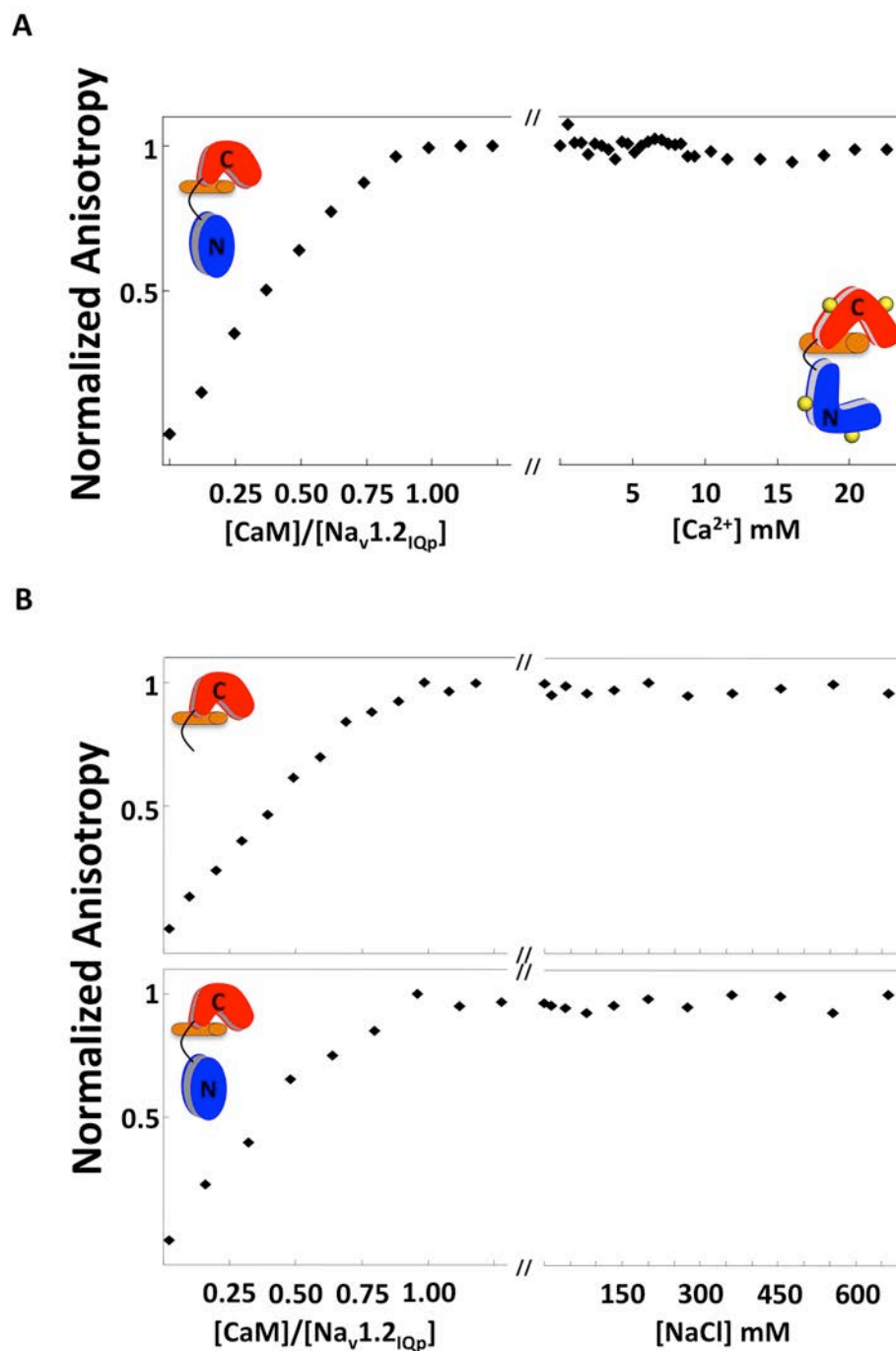


Figure 4.8: Fluorescence anisotropy monitored titration of apo CaM bound to Na_v1.2_{IQp}
A: Fluorescence anisotropy monitored apo CaM₁₋₁₄₈ titration of fluoresceinated Na_v1.2_{IQp} (left) followed by CaCl₂ titration of apo CaM₇₆₋₁₄₈:Na_v1.2_{IQp} complex (right).
B: Fluorescence anisotropy monitored of fluoresceinated Na_v1.2_{IQp} (left) with apo CaM followed by NaCl titration of the apo CaM:Na_v1.2_{IQp} complex (right).
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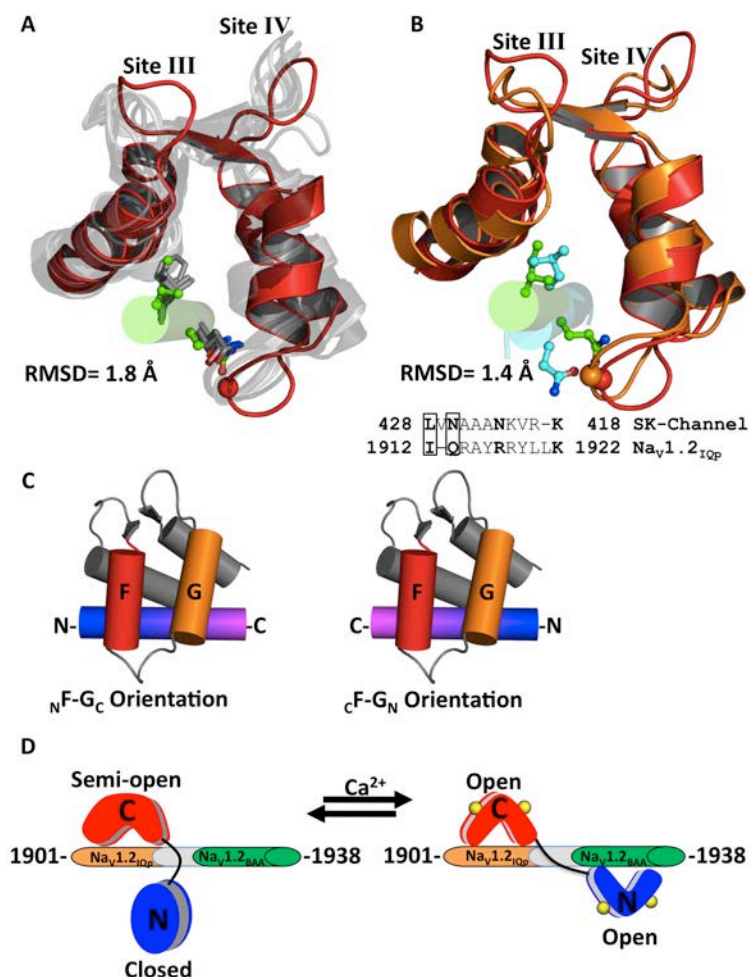


Figure 4.9: Superposition of target bound apo CaM and CaM-like protein C-domains
A: Superposition of apo CaM₇₆₋₁₄₈ (red):Na_v1.2_{IQP} (green rod) complex onto C-domains of apo CaM and apo CaM-like proteins (gray) bound to canonical IQ-motifs. I and Q residues of all IQ-motifs are shown in sticks, the C α atom of residue 113 is shown as a sphere as a point of reference. Pdb files used in overlay 2IX7, 1M45, 1M46, 1ND2, and 3JVT. **B:** Superposition of apo CaM₇₆₋₁₄₈ (red):Na_v1.2_{IQP} (green rod) complex and the C-domain of partially Ca²⁺-saturated CaM₁₋₁₄₈ (orange) bound to the SK-Channel peptide (cyan). Residues at positions I and Q of IQ-motif and their SK-channel homologs are shown in sticks, the C α atom of residue 113 is shown as a sphere as a point of reference. The aligned primary sequences of Na_v1.2_{IQP} and SK-channel are shown where residues at IQ-motif defining position are shown in bold, and essential IQ-motif residues or their homolog are boxed. **C:** Structural depiction of naming scheme used to describe peptide polarity when interacting with the C-domain of CaM. The N- and C-termini of a peptide (blue and magenta respectively) can orient themselves in 2 possible ways with respect to helices F (red) and G (orange) of the C-domain resulting in either a _NF-G_C or _CF-G_N orientation. **D:** Proposed model of CaM₁₋₁₄₈ interaction with the C-terminal tail of Na_v1.2 under apo and Ca²⁺-saturating conditions.
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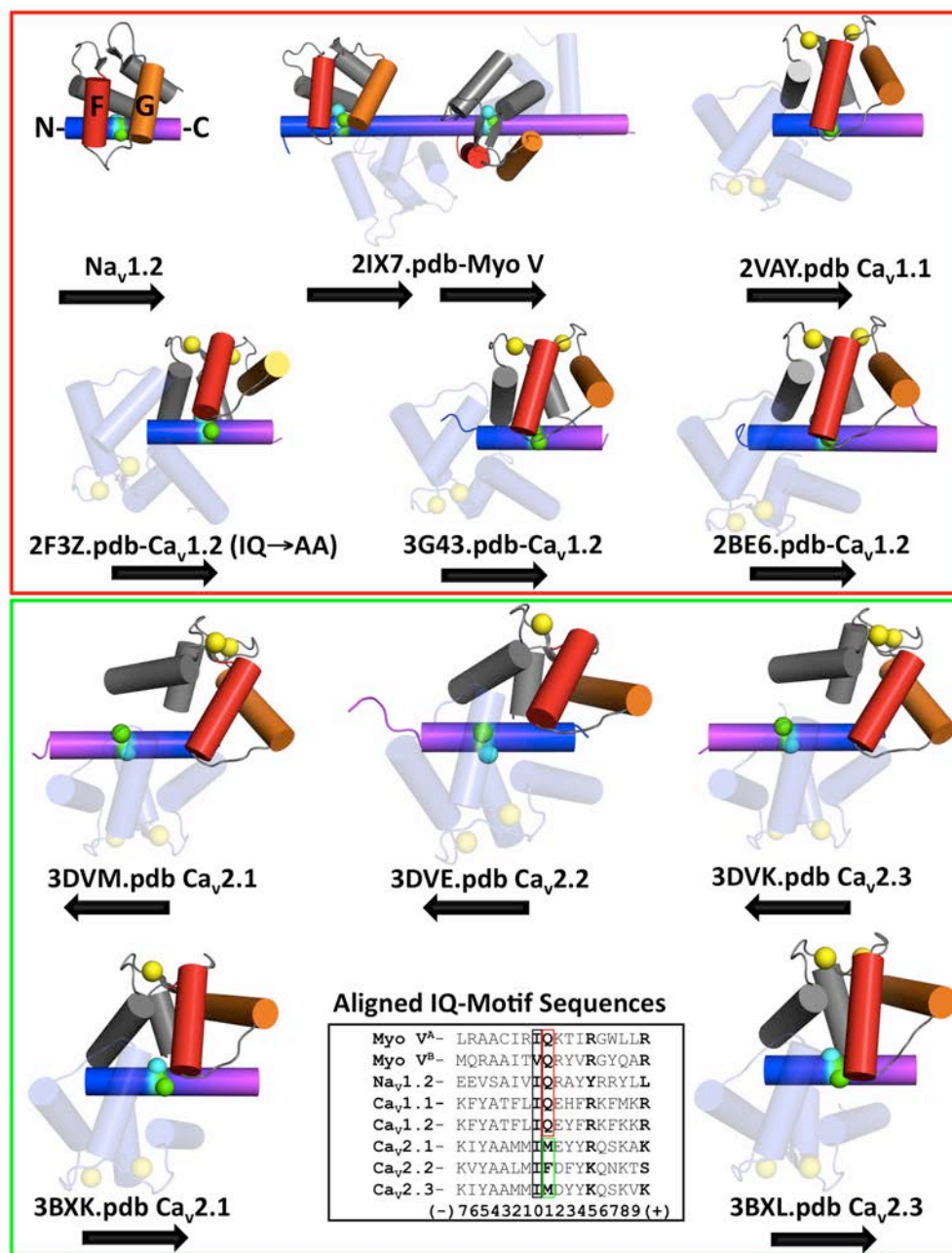


Figure 4.10: Directionality of IQ-motif binding to the C-domain of CaM
 The polarity of IQ-motif containing helices is indicated by the color gradient where the N-terminus is shown in blue and C-terminus in purple. CaM helices F and G are colored red and orange, while the I and Q of each IQ-motif are shown as spheres colored cyan and green respectively.

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CHAPTER V
INFLUENCE OF ELECTROSTATIC INTERACTIONS ON
Ca²⁺ AND TARGET BINDING BY CAM

Introduction

CaM is a Ca²⁺ sensor protein that is essential to eukaryotic signal transduction pathways (Pedigo et al., 1992). Changes in intra-cellular Ca²⁺ levels are linked to cellular events by the effect of Ca²⁺ on CaM: it triggers conformational changes that expose hydrophobic surfaces in both domains of CaM, altering its binding to many target proteins (VanScyoc and Shea, 2001a; Yagi et al., 1990). In addition to the widely accepted role of hydrophobic interactions in recognition and binding of targets by CaM, electrostatic interactions have an important role in these processes as well (André et al., 2004; Linse et al., 1988; Ogawa and Tanokura, 1984). At physiological conditions (pH 7, 100mM KCl) CaM is a negatively charged protein (pI=4) with net charges of -24 (apo) and -16 (Ca²⁺-saturated) while peptide derivatives from the CaMBD's of its targets typically carry a complementary positive charge (Na_v1.2_{IQp} = +8, CaMKII_p = +6). Compared to hydrophobic interactions alone, electrostatic interaction between CaM and its targets allow for recognition of each other at much greater distances, increasing the rate at which they associate (André et al., 2004; Antosiewicz and McCammon, 1995). The strength of electrostatic attraction between CaM and its targets is particularly interesting as it can be influenced in the cell due to spatial and temporal fluctuations in ionic concentrations due to naturally occurring currents through ion channels (Akyol et al., 2004; Cens et al., 2006; Levitan, 1999; Saimi and Kung, 2002).

Hydrophobic and electrostatic interactions are the major forces responsible for CaM-target recognition and binding affinity (Bayley et al., 1996; Tjandra et al., 1999; Yagi et al., 1990). Electrostatic interactions between CaM and its targets guide hydrophobic surfaces located on CaM and its targets into close proximity so that they can interact. Due to their α -helical character, CaMBD's contain a periodicity within their primary amino acid sequence

producing a hydrophobic face that is buried from the solvent upon binding to CaM and a basic face that remains solvent exposed. This periodicity has resulted in multiple CaM binding motifs that have been characterized based upon the numerical position of the hydrophobic residues within CaMBD's that insert into the hydrophobic pockets of the N- and C-domains of CaM (**Figure 5.1**). This periodicity can also be visualized in helical wheel diagrams of the CaMBDs of proteins regulated by CaM (**Figure 5.2**). The regions between hydrophobic “anchor” positions are enriched with positively charged residues creating charge complementarity between CaM and its target (Yamniuk and Vogel, 2004).

The extent to which electrostatic interactions contribute to CaM-target recognition and binding affinity may significantly change in the cell due to ion fluxes necessary for the generation and propagation of action potentials (Andre et al., 2006; Suizu et al., 1995). The ions that undergo the largest change in intra and extracellular concentration involved in this process are Na^+ , K^+ , Ca^{2+} , and Cl^- . Shown in **Figure 5.3** is a schematic illustrating the ionic strength and concentration gradients present in a resting cell (Hille, 2001).

Large changes in local intracellular ion concentration occur near ion channels as they open to allow their specific ions to pass through the membrane. Given that CaM is an intrinsic subunit of several known Na^+ , K^+ , and Ca^{2+} channels, fluctuations in the strength of electrostatic interaction may alter occupancy of CaM at CaMBDs on these channels (Ehlers et al., 1996; Halling et al., 2005; Pitt, 2005; Saimi and Kung, 1994; Theoharis et al., 2008). To investigate whether these fluctuations significantly alter the affinity of CaM binding to CaMKII_p or $\text{Na}_v1.2_{IQp}$, fluorescence anisotropy monitored titrations as a function of varied salt were performed.

Poisson-Boltzmann Equation

Electrostatic interactions can be observed between virtually all interacting macromolecules. Creighton has estimated that more than 20% of all amino acids in globular proteins are ionized at physiological pH (Creighton, 1993). Structural methods such as NMR,

and X-ray crystallography are able to provide the spatial arrangement of charged amino acid groups within a protein or protein-ligand complex, allowing for the calculation of electrostatic potentials that can be mapped to the macromolecular surface. Solution of Poisson-Boltzmann equation was first described by Gouy (1910) and Chapman (1913) (Chapman, 1913; Gouy, 1910) allows for the calculation of the electrostatic potential of a macromolecule throughout the calculated space, as well the distribution the local ions around the macromolecule.

The Poisson-Boltzmann equation (**Equation. 5.1**)

$$-\nabla \cdot \varepsilon(\chi)\nabla\phi(\chi) + \kappa^{-2}(\chi)\sinh\phi(\chi) = f(\chi) \quad (5.1)$$

is a second-order nonlinear elliptic partial differential equation that relates the electrostatic potential (ϕ) to the dielectric properties of the solute and solvent (ε), the ionic strength of the solution and the accessibility of ions to the solute interior (κ^{-2}), and the distribution of solute atomic partial charges (f). To expedite solution of the equation, the nonlinear PBE is often approximated by the linearized PBE (LPBE) by assuming $\sinh\phi(\chi) \approx \phi(\chi)$. Pioneering work in this area has been done by many groups including Honig and McCammon. Their studies allow the calculation of electrostatic potentials of macromolecules such as nucleic acids and proteins (Allison et al., 1988; Gilson and Honig, 1988; Honig and Nicholls, 1995; Sharp and Honig, 1990). In addition to visualization of electrostatic potentials of a macromolecule in solutions of varying ionic strength, solution of the linear or non-linear Poisson-Boltzmann equation allows for calculation of macromolecular solvation energies.

Determination of the solvation energies of the macromolecular complex and the individual components that comprise them can be used via linkage analysis to calculate the contribution of electrostatic interactions to overall binding affinity. Multiple ionic strengths can be used to examine the dependence of the electrostatic binding energy on solution ionic strength. This dependence can be determined by plotting the electrostatic binding energy (x-axis) versus the $\log[\text{salt}]$ (y-axis) and fitting a line to the points. The greater the slope of this plot, the more dependent the electrostatic component of binding and by default the overall binding energy of the associating macromolecules are on the ionic strength of the solution.

Materials and Methods

Poisson-Boltzmann calculations

Poisson-Boltzmann calculations were performed using the Adaptive Poisson-Boltzmann Solver (APBS) software package developed by Nathan Baker and coworkers (Baker et al., 2001) to determine the theoretical contribution of electrostatic interactions to the overall binding energy of $(\text{Ca}^{2+})_4\text{-CaM}$ for CaMKII_p. Structural coordinates for the $(\text{Ca}^{2+})_4\text{-CaM}$:CaMKII_p complex were obtained from the pdb file 1CM1.pdb, and missing side chains were added using Swiss-PDBViewer (Guex and Peitsch, 1997). From the previously mentioned file, two separate PDB files were created consisting of $(\text{Ca}^{2+})_4\text{-CaM}$, and CaMKII_p, in the conformation they were observed in the $(\text{Ca}^{2+})_4\text{-CaM}$:CaMKII_p complex structure. These files were then converted into PQR format and protonated based on their charge states at pH 7.4 using the PDB2PQR webserver (Sobczak et al., 2002).

Identical calculation grid center and length coordinates were used in calculating electrostatic potentials of each molecule or complex. The dimensions of the coarse grid were $x=60 \text{ \AA}$, $y=75 \text{ \AA}$, and $z=70 \text{ \AA}$ with a grid spacing of 0.5 \AA , while the fine grid was $x=41.5 \text{ \AA}$, $y=52 \text{ \AA}$, and $z=48 \text{ \AA}$ with a grid spacing of 0.25 , each was centered at $x=19.558$, $y=56.112$, and $z=74.867$. The temperature used in the calculation was 295K, while the dielectric constant values used for the protein and solvent were 4.00 and 78.54. The linearized form of the Poisson-Boltzmann equation was solved using single Debye-Huckel boundary conditions with cubic B-spline charge discretization and surface smoothing, while the spline window was set to 0.3 \AA .

The $[\text{NaCl}]$ or $[\text{KCl}]$ was varied in 50mM increments over a range between 0 and 2M, while the keeping the $[\text{CaCl}_2]$ and $[\text{MgCl}_2]$ constant to mimic the experimental setup to which these calculations were compared to. The atomic radii of the Ca^{2+} , Mg^{2+} , Cl^- , and $(\text{Na}^+ \text{ or } \text{K}^+)$ used in the calculation were 1.97 \AA , 1.60 \AA , 1.75 \AA , 1.86 \AA and 2.27 \AA respectively, while the solvent radius was 1.4 \AA . The atomic radii of the afore mentioned ions were calculated using APBS by iterative rounds of calculation of the solvation energy of the ion and radius adjustment

to match experimentally determined solvation values (Burgess, 1988). The electrostatic potentials of the $(Ca^{2+})_4$ -CaM:CaMKII_p complex, $(Ca^{2+})_4$ -CaM, and CaMKII_p were then calculated at each [NaCl or KCl]. These electrostatic potentials were then used to determine the electrostatic component of CaMKII_p binding to $(Ca^{2+})_4$ -CaM at each [NaCl or KCl] by subtracting the electrostatic energy values determined for $(Ca^{2+})_4$ -CaM and CaMKII_p from the value determined for the $(Ca^{2+})_4$ -CaM:CaMKII_p complex as shown in **Equation 5.2**.

$$\Delta G_{Electrostatic} = Complex - CaM_{Alone} - Na_v1.2_{IQp} \quad (5.2)$$

The resulting value from this calculation represents the energy of binding/complex formation.

Fluorescence Anisotropy Monitored Titrations of CaM Binding to Either CaMKII_p or Na_v1.2_{IQp} at Varied [NaCl or KCl]

Fluorescence anisotropy monitored titrations of CaM binding to either CaMKII_p or Na_v1.2_{IQp} at varied [NaCl or KCl] were performed to examine how these salts alter the binding affinity of CaM for these peptides. $(Ca^{2+})_4$ -CaM₁₋₁₄₈ binding to fluorescein-labeled CaMKII_p, or CaM₇₆₋₁₄₈ or CaM₁₋₁₄₈ binding to fluorescein-labeled Na_v1.2_{IQp} peptide under apo and Ca²⁺-saturating conditions were monitored using a Fluorolog 3 (Jobin Yvon, Horiba) spectrofluorimeter, equipped with dual auto-assembly Glan-Thompson polarizers. The anisotropy of the fluorescein labeled peptides were monitored using λ_{ex} 496 nm and λ_{em} 520 nm with 2nm excitation and 10nm emission bandpasses. Anisotropy (r) was calculated as shown in **Equation 5.3** as described previously (Akyol et al., 2004),

$$r = \frac{I_{VV} - G \cdot I_{VH}}{I_{VV} + 2G \cdot I_{VH}} \quad (5.3)$$

where I_{VV} and I_{VH} are the intensities of vertically- or horizontally-emitted light upon vertical excitation, respectively, and G is the instrument correction factor ($G = I_{HV}/I_{HH}$). Averages of three readings with a 1-sec integration time at each point were recorded. Samples of 100 nM Fl-

CaMKII_p or 1 μM FI-Na_v1.2_{IQp} in 50 mM HEPES, 100 mM KCl, 50 μM EGTA, 5 mM NTA, 1 mM MgCl₂, pH 7.4 in the absence (apo) or presence of 10 mM CaCl₂ (Ca²⁺-saturated) at 22 °C were titrated with concentrated apo or Ca²⁺-saturated CaM. At least three replicate titrations were conducted for each NaCl or KCl concentration examined as well as for apo and (Ca²⁺)₄-CaM₁₋₁₄₈. Titrations involving CaM₇₆₋₁₄₈ were less well determined and represent only single trials performed as part of exploratory measurements.

Analysis of K_d for CaM Binding to Na_v1.2_{IQp} or CaMKII_p

Affinity estimates of CaM for Na_v1.2_{IQp} or CaMKII_p were determined by fitting titration data to a one-site binding model using NONLIN (Johnson and Frasier, 1985). Fractional saturation of Na_v1.2_{IQp} or CaMKII_p was described by **Equation 5.4** as described previously (Akyol et al., 2004):

$$\bar{Y} = \frac{K_a [X_{\text{Free}}]}{1 + K_a [X_{\text{Free}}]} \quad (5.4)$$

where K_a represents the association constant for CaM binding to CaMKII or Na_v1.2_{IQp}, and $[X_{\text{free}}]$ is the free concentration of CaM in solution, as calculated from the independent variables (total concentration of X, $[X_{\text{total}}]$) and (total concentration of M, $[M_{\text{total}}]$) according to the quadratic equation described by **Equation 5.5** as described previously (Akyol et al., 2004):

$$[X_{\text{Free}}] = \frac{\sqrt{-b \pm b^2 - 4K_a(-X_{\text{Total}})}}{2K_a} \quad (5.5)$$

where $b = (1 + K_a[M_{\text{Total}}] - K_a[X_{\text{Total}}])$. Under equilibrium conditions, the concentration of Na_v1.2_{IQp} or CaMKII ($[M_{\text{total}}]$) was low relative to the K_d (dissociation constant, 1/K_a) of CaM binding to the peptide. The free concentration of CaM may be approximated by the total (i.e., $[X_{\text{free}}]$ at $[X_{\text{total}}]$). This allows for an accurate estimate of the association constant. Under stoichiometric conditions however, the ligand is limiting and $[X_{\text{free}}]$ is estimated iteratively in the nonlinear least squares function for saturation as the best solution to the difference between $[X_{\text{total}}]$ (calculated on the basis of the total ligand added) and $[X_{\text{bound}}]$ (calculated as the product

of $[M_{\text{total}}]$ and \bar{Y}). The value of a binding constant estimated in this way is highly correlated with the precision of the numerical value measured for $[M_{\text{total}}]$; therefore, the dissociation constant of $\text{Na}_v1.2_{\text{IQP}}$ or CaMKII_P for apo and Ca^{2+} -saturated CaM_{76-148} or CaM_{1-148} in **Tables 5.1 and 5.2** is reported as a limiting value. Experimental variations in the observed endpoints of individual titration curves were accounted for by **Equation 5.6** described previously (Akyol et al., 2004):

$$f(\mathbf{X}) = Y_{[\mathbf{X}]_{\text{low}}} + \bar{Y}_1 \cdot [(Y_{[\mathbf{X}]_{\text{high}}} - Y_{[\mathbf{X}]_{\text{low}}})] = \textit{Span} \quad (5.6)$$

where \bar{Y}_1 refers to average fractional saturation of $\text{Na}_v1.2_{\text{IQP}}$ or CaMKII and $Y_{[\mathbf{X}]_{\text{low}}}$ corresponds to the intrinsic fluorescence anisotropy of CaMKII or $\text{Na}_v1.2_{\text{IQP}}$ in the absence of CaM . The *Span* describes the magnitude and direction of signal change upon titration, which describes the difference between the high ($Y_{[\mathbf{X}]_{\text{high}}}$) and low ($Y_{[\mathbf{X}]_{\text{low}}}$) endpoints. The *Span* is positive for an increasing signal and negative for a decreasing signal.

In most titrations, the upper and lower endpoints were well defined experimentally. However, in the equilibrium titrations of $\text{Na}_v1.2_{\text{IQP}}$ with apo and $(\text{Ca}^{2+})_2\text{-CaM}_{76-148}$ at 300mM KCl the fluorescence anisotropy of $\text{Na}_v1.2_{\text{IQP}}$ did not reach a plateau at the final CaM concentration tested. To estimate the final anisotropy that might have been reached if $\text{Na}_v1.2_{\text{IQP}}$ had become saturated with apo and $(\text{Ca}^{2+})_2\text{-CaM}_{76-148}$, the end points of the same titrations done at 100 mM KCl were used. The endpoint ($Y_{[\mathbf{X}]_{\text{high}}}$) was fixed at this value in the nonlinear least squares analysis of the affected titrations.

To illustrate the degree of precision which we were able to place on the limiting K_d values of stoichiometric titrations reported in this chapter binding curves corresponding to K_d values of 1, 10, 25, and 50 nM were simulated and are shown in **Figure 5.5**. These curves were simulated assuming a 1 μM peptide concentration and plotted against either $\log [\text{CaM}]_{\text{total}}$, or $[\text{CaM}]/[\text{Peptide}]$. Plotted upon each graph are identical data points of a $(\text{Ca}^{2+})_4\text{CaM}_{1-148}$ titration of $\text{Na}_v1.2_{\text{IQP}}$ where the $[\text{Na}_v1.2_{\text{IQP}}]$ concentration was 1 μM to demonstrate how the representation of the same data points change dependent upon how the x-axis is displayed.

Results

Fluorescence Anisotropy Monitored CaM Titrations of Nav1.2_{IQP}

As shown in **Figures 5.4** and **5.6**, CaM titrations of Nav1.2_{IQP} monitored by fluorescence anisotropy were used to determine the binding affinities of CaM₇₆₋₁₄₈ and CaM₁₋₁₄₈ for Nav1.2_{IQP} under apo and Ca²⁺-saturating conditions at varied NaCl (CaM₁₋₁₄₈ only) and KCl concentrations. Reported in Table 5.2 are estimated binding affinities for CaM₇₆₋₁₄₈ and CaM₁₋₁₄₈ for Nav1.2_{IQP}. It should be noted that the value of CaM₇₆₋₁₄₈ titration of Nav1.2_{IQP} is the result of a single trial performed as part of initial exploratory study of salt effects on the affinity of Nav1.2_{IQP} binding to CaM₇₆₋₁₄₈.

A common feature present for all titrations of CaM₁₋₁₄₈ was that Nav1.2_{IQP} binding was observed to be in a stoichiometric binding regime regardless of the salt (NaCl, or KCl) used (**Figure 5.3**). Increasing the KCl concentration shifted Nav1.2_{IQP} binding to both apo and (Ca²⁺)₂-CaM₇₆₋₁₄₈ from a stoichiometric to equilibrium binding regime (**Figure 5.2**). The stoichiometry of Nav1.2_{IQP} binding to either CaM₇₆₋₁₄₈ or CaM₁₋₁₄₈ was determined to be 1:1 under both apo and Ca²⁺-saturating conditions. A decrease in Nav1.2_{IQP} binding affinity was observed as the concentration of salt was increased in all titrations (**Figures 5.1**, and **5.2**). Comparison of changes in free energy of Nav1.2_{IQP} binding at varied salt concentrations showed that NaCl induced larger changes in Nav1.2_{IQP} binding affinity than KCl (**Table 5.2**) for both apo and (Ca²⁺)₄-CaM₁₋₁₄₈.

Although additional trials of CaM₇₆₋₁₄₈ titrations of Nav1.2_{IQP} are needed to confirm the values of these initial findings, when changes in Nav1.2_{IQP} binding affinities at 100mM and 300mM KCl were compared for CaM₇₆₋₁₄₈ and CaM₁₋₁₄₈, larger changes are observed for CaM₇₆₋₁₄₈ than CaM₁₋₁₄₈ (**Table 5.2**).

Fluorescence Anisotropy Monitored CaM Titrations of CaMKII_p

As shown in **Figure 5.3** CaM titrations of CaMKII_p monitored by fluorescence anisotropy were used to determine the binding affinity of (Ca²⁺)₄-CaM₁₋₁₄₈ for CaMKII_p at varied NaCl and KCl concentrations. Reported in **Table 5.1** are estimated binding affinities of CaMKII_p for (Ca²⁺)₄-CaM₁₋₁₄₈. A common feature present for all titrations was that CaMKII_p binding was observed to be in a stoichiometric binding regime regardless of the salt (NaCl, or KCl) used (**Figure 5.3**). The stoichiometry of CaMKII_p to either (Ca²⁺)₄-CaM₁₋₁₄₈ was determined to be 1:1. Though stoichiometric, a decrease in CaMKII_p binding affinity was observed as the concentration of either NaCl or KCl was increased. Comparison of changes in free energy of CaMKII_p binding at varied salt concentrations showed that NaCl induced larger changes in CaMKII_p binding affinity than KCl (**Table 5.1**) for (Ca²⁺)₄-CaM₁₋₁₄₈.

Electrostatic Binding Energy of CaMKII for (Ca²⁺)₄-CaM₁₋₁₄₈ Calculated via APBS

To predict the effect of NaCl and KCl on the binding affinity of CaMKII_p for (Ca²⁺)₄-CaM₁₋₁₄₈, Poisson-Boltzmann calculations were performed at varied NaCl and KCl concentrations (25 mM–1000 mM). Both NaCl and KCl lowered the binding affinity of CaMKII_p for (Ca²⁺)₄-CaM in a concentration dependent manner shown in **Figure 5.4a**. Though both NaCl and KCl lowered the affinity of CaMKII_p for (Ca²⁺)₄-CaM, NaCl induced a greater change in affinity at concentrations ranging from 25-400 mM than KCl (**Figure 5.4a**). Comparison of $\Delta\Delta G_{\text{NaCl-KCl}}$ values of CaMKII_p binding at salt concentrations greater than 400 mM indicated a convergence in the degree to which Na⁺ or K⁺ reduce the binding affinity of CaMKII_p (**Figure 5.4a**). Shown in **Figure 5.4b** are structures of (Ca²⁺)₄-CaM₁₋₁₄₈ and CaMKII_p whose solved exposed surfaces are colored according to their electrostatic potential at varied NaCl or KCl concentrations.

Discussion

CaM is a negatively charged protein ($pI=4$) that binds to CaMBD sequences that are enriched in positive charge. The interaction of charged particles in solution can be lessened by electrostatic shielding by the addition of salt. Results presented in **Chapter V** indicate that increasing the concentration of NaCl or KCl lowered the binding affinity of targets for CaM. Examination of values determined for CaMKII_p and Na_v1.2_{IQp} binding to CaM at the same NaCl or KCl concentration indicated that the Na⁺ ion induced a larger effect on CaMKII_p or Na_v1.2_{IQp} binding affinity than the K⁺ ion.

Poisson-Boltzmann calculations of CaMKII_p binding to (Ca²⁺)₄-CaM₁₋₁₄₈ indicated that both NaCl and KCl reduced the binding affinity of CaMKII_p for (Ca²⁺)₄-CaM, with NaCl having a larger effect than KCl up to 400 mM. Calculation of NaCl and KCl effects on CaMKII_p binding to (Ca²⁺)₄-CaM₁₋₁₄₈ at concentrations greater than 400 mM showed a decrease in Na⁺ or K⁺ ion specific effects indicating a general ion effect applies at salt concentrations above 400 mM. The onset of a general salt effect as NaCl or KCl progress past 400 mM represents a saturation of (Ca²⁺)₄-CaM₁₋₁₄₈ and CaMKII_p surfaces with either salt, after which Na⁺ or K⁺ ion specific effects become less pronounced.

Results obtained in this chapter show that electrostatic interactions between CaM and its targets CaMKII_p and Na_v1.2_{IQp} are influenced by changes in solvent ionic strength, influencing their binding affinity for each other. A caveat of this observation is that although the peptide binding affinity of CaM was decreased by increasing the salt concentration, the binding affinity of CaM for CaMKII_p and Na_v1.2_{IQp} remained at levels where they would remain associated with CaM in the cell. Together these observations suggest that CaM and its targets have evolved binding surfaces whose molecular interfaces rely on interactions that are largely independent of changes in intracellular salt concentration.

Salt Dependence of $\text{Na}_v1.2_{\text{IQP}}$ Binding to apo and (Ca^{2+}) -CaM

Titration of $\text{Na}_v1.2_{\text{IQP}}$ binding to CaM at varied NaCl and KCl concentrations showed differential effects upon the $\text{Na}_v1.2_{\text{IQP}}$ binding affinity of CaM. In all cases titrations done in the presence of NaCl showed a greater decrease in the binding affinity of CaMKII_p or $\text{Na}_v1.2_{\text{IQP}}$ for CaM than those done in KCl. This effect can be directly attributed to differences between interactions of the Na^+ or K^+ ion and the CaM:target complex. Comparison of the radii of the Na^+ and K^+ ion indicate that the Na^+ ion has a greater charge density than the K^+ ion (Hille et al. 2001). This characteristic allows for stronger attraction, and greater access to negatively charged cavities found within CaM and either CaMKII_p or $\text{Na}_v1.2_{\text{IQP}}$ than the larger K^+ ion. These results suggest that as K^+ ions exchange for Na^+ ions during an action potential, the affinity of CaM for its targets decreases.

Comparison of salt effects on the binding affinity of $\text{Na}_v1.2_{\text{IQP}}$ for apo and $(\text{Ca}^{2+})_4$ -CaM₁₋₁₄₈ showed that the binding affinity of $\text{Na}_v1.2_{\text{IQP}}$ for apo CaM₁₋₁₄₈ was decreased to a greater extent than for $(\text{Ca}^{2+})_4$ -CaM₁₋₁₄₈ (**Table 5.2**). This effect is attributed to the electrostatic component of the overall $\text{Na}_v1.2_{\text{IQP}}$ binding affinity for apo CaM₁₋₁₄₈ comprising a greater percentage than $(\text{Ca}^{2+})_4$ -CaM. This difference in contribution of electrostatic interactions to the overall binding affinity is likely due to apo CaM₁₋₁₄₈ have a greater overall net negative charge of -24 compared to -16 of $(\text{Ca}^{2+})_4$ -CaM₁₋₁₄₈ due to Ca^{2+} binding.

Preliminary trials of both apo and $(\text{Ca}^{2+})_2$ -CaM₇₆₋₁₄₈ at 100 and 300 mM KCl showed a more pronounced change in the affinity of $\text{Na}_v1.2_{\text{IQP}}$ than what was observed for apo and $(\text{Ca}^{2+})_4$ -CaM₁₋₁₄₈. Additional trials are necessary to confirm the significance of these observed values. Based on the available data for CaM₇₆₋₁₄₈, addition of KCl caused a larger decrease in $\text{Na}_v1.2_{\text{IQP}}$ affinity for both apo and $(\text{Ca}^{2+})_2$ -CaM₇₆₋₁₄₈ than apo and $(\text{Ca}^{2+})_4$ -CaM₁₋₁₄₈ with the largest decrease in $\text{Na}_v1.2_{\text{IQP}}$ binding affinity seen in apo CaM₇₆₋₁₄₈ (**Table 5.2**). The greater effect of KCl upon CaM₇₆₋₁₄₈ than CaM₁₋₁₄₈ may be attributed to KCl induced structural changes specific to CaM₇₆₋₁₄₈ that are not found in CaM₁₋₁₄₈. This differential effect is consistent with previous

studies that showed that when separated, the domains of CaM do not behave in an identical manner as when they are covalently linked (Sorensen et al., 2002a; Sorensen and Shea, 1998).

Evaluation of calculated versus experimentally observed changes in the binding affinity of CaMKII_p for (Ca²⁺)₄-CaM indicates qualitative effects similar to those described previously for Na_v1.2_{IQp} binding. In both experimental and calculated measurements, increasing NaCl or KCl concentrations decreased the affinity of CaMKII_p for (Ca²⁺)₄-CaM. Quantitative comparison of calculated values determined for CaMKII_p binding to (Ca²⁺)₄-CaM showed that the salt dependence of CaMKII_p binding to (Ca²⁺)₄-CaM is overestimated. Shown in **Figure 5.9** are the salt dependence slopes of experimentally observed and calculated changes in CaMKII_p binding affinity for (Ca²⁺)₄-CaM. This difference between observed and predicted results represents a significant overestimate in the calculated salt dependence of CaMKII_p binding to (Ca²⁺)₄-CaM.

Comparison of APBS calculated salt dependence slopes of (Ca²⁺)₄-CaM binding to CaMKII_p (6.7-6.8) with that calculated for DNA binding to the Lambda repressor DNA binding domain (4.6) indicates a significantly greater salt dependence in the CaM-CaMKII_p system (Sharp et al., 1995). Like CaMKII_p binding to CaM, both DNA and the lambda repressor DNA binding domain are oppositely charged macromolecules indicating a significant contribution of electrostatic forces to the overall binding affinity (Sharp et al., 1995). It is of interest to note that unlike CaM binding CaMKII_p, the structure of the DNA binding domain of the lambda repressor does not undergo a large conformational change upon DNA binding (Beamer and Pabo, 1992; Pervushin et al., 1996).

The calculated overestimate in salt dependence of CaMKII_p binding to (Ca²⁺)₄-CaM is likely due to multiple factors. APBS calculations do not take into account conformational changes that are known to occur within (Ca²⁺)₄-CaM and CaMKII_p upon binding. CaM is a very flexible molecule whose conformation changes it from an extended conformation in the unbound state to a collapsed structure when bound to a target (**Figure 1.4**). CD data of CaMKII_p peptide in the absence of CaM indicates that it does not form a persistent α -helix in solution (Shifman

and Mayo, 2002). Another factor that may also contribute the discrepancy between observed and predicted salt dependence values are structural differences that may be present at one salt concentration but not at others.

The discrepancy between predicted salt effects from calculated and observed values suggest that conformational changes occur upon peptide binding or salt-induced structural changes of CaMKII_p to (Ca²⁺)₄-CaM, that are not captured by the structures used in these calculations. These results further demonstrate that a folding transition must occur upon CaMKII_p (unstructured) binding to (Ca²⁺)₄-CaM (extended) and the final compact ellipsoidal structure of (Ca²⁺)₄-CaM bound to CaMKII_p (**Figure 1.4**) (Tse et al., 2007).

Table 5.1: Calculated effect of salt on CaMKII_p binding to (Ca²⁺)₄-CaM₁₋₁₄₈

[Salt], mM	K _d (KCl)	K _d (NaCl)	Fold Difference (K _d KCl/ K _d NaCl)
50	10 nM (-10.8*)	12 nM (-10.7*)	0.83
100	14 nM (-10.6*)	16 nM (-10.5*)	0.88
200	29 nM (-10.2*)	47 nM (-9.9*)	0.62
300	55 nM (-9.8*)	85 nM (-9.5*)	0.65

*Values reported in kcal/mol

Table 5.2: Effect of salt on $\text{Na}_v1.2_{\text{IQP}}$ binding to apo and $(\text{Ca}^{2+})\text{-CaM}$

CaM ₁₋₁₄₈						
[Salt], mM	apo K_d (KCl)	(Ca^{2+}) K_d (KCl)	apo/ Ca^{2+}	apo K_d (NaCl)	(Ca^{2+}) K_d (NaCl)	apo/ Ca^{2+}
100	≤ 10 nM ($\leq -10.8^*$)	≤ 10 nM ($\leq -10.8^*$)	1	≤ 1 nM ($\leq -10.8^*$)	≤ 51 nM ($\leq -9.8^*$)	0.02
300	≤ 85 nM ($\leq -9.5^*$)	≤ 62 nM ($\leq -9.7^*$)	1.37	≤ 122 nM ($\leq -9.3^*$)	≤ 170 nM ($\leq -9.1^*$)	0.71

CaM ₇₆₋₁₄₈						
[Salt], mM	apo K_d (KCl)	(Ca^{2+}) K_d (KCl)	apo/ Ca^{2+}	apo K_d (NaCl)	(Ca^{2+}) K_d (NaCl)	apo/ Ca^{2+}
100	≤ 1 nM ($\leq -10.8^*$)	≤ 1 nM ($\leq -10.8^*$)	1	--	--	--
300	1 ± 4 μM ($-8.1^* \pm 0.2$)	2.8 ± 3 μM ($-7.5^* \pm 0.3$)	0.36	--	--	--

*Values reported in kcal/mol

Table 5.3: Calculated effect of salt on CaMKII_P binding to (Ca²⁺)₄-CaM₁₋₁₄₈

[Salt], mM	$\Delta G_{\text{Electrostatic}}^*$ (KCl)	$\Delta G_{\text{Electrostatic}}^*$ (NaCl)
0	-11.5	-11.5
25	-9.6	-9.3
50	-8.2	-7.8
75	-7.2	-6.8
100	-6.4	-5.9
150	-5.2	-4.6
200	-4.3	-3.7
300	-3.0	-2.4
400	-2.1	-1.5
500	-1.5	-0.9
600	-1.0	-0.4
700	-0.6	0.1
800	-0.2	0.4
900	0.1	0.7
1000	0.3	0.9

*Values reported in kcal/mol

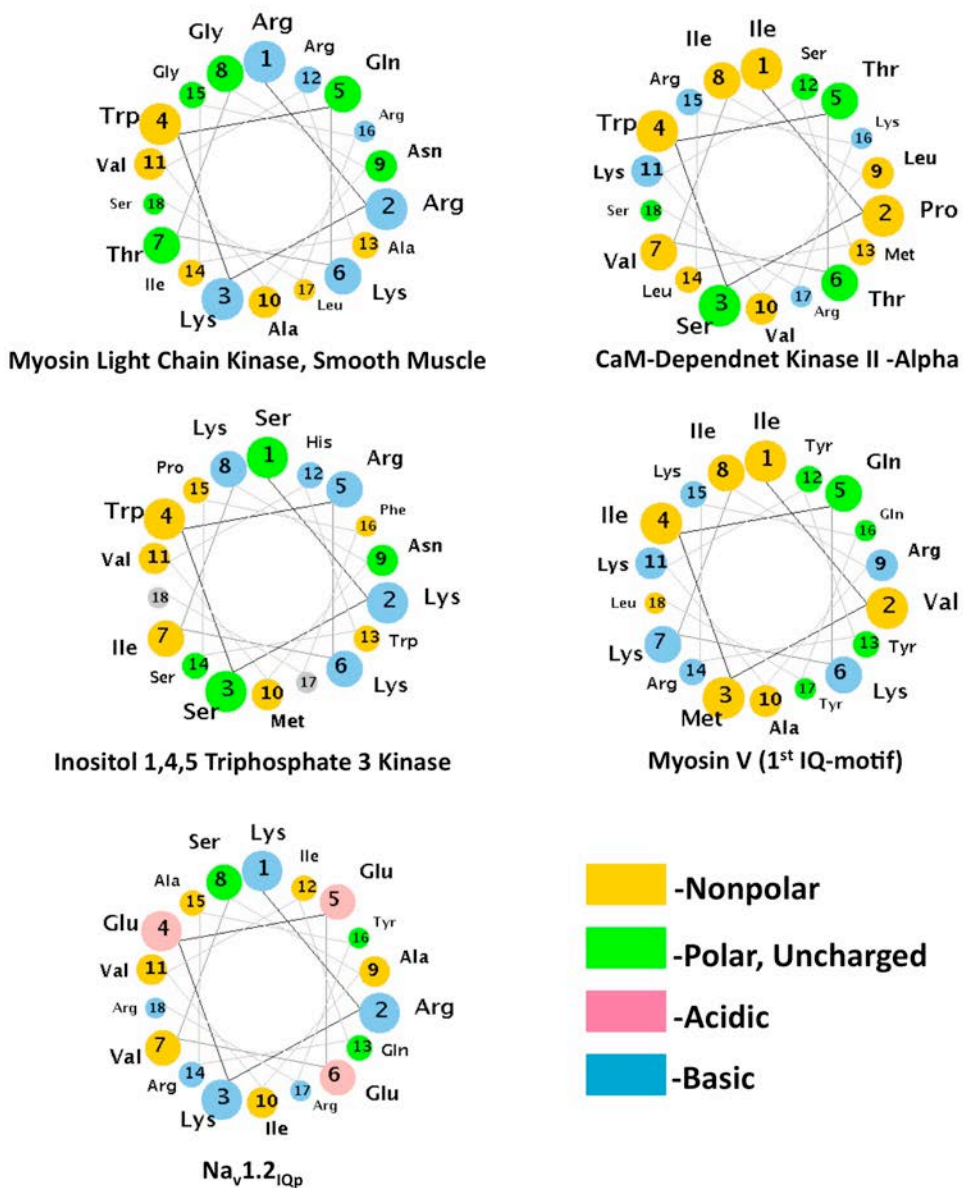


Figure 5.2: Helical Wheel Diagrams of CaM-Binding Domains
 Helical wheel diagrams generated using University of Virginia applet
<http://cti.itc.virginia.edu/~cmg/Demo/wheel/wheelApp.html> for various Ca²⁺-dependent and Ca²⁺-independent CaMBDs where nonpolar residues are colored according to the scheme depicted above.
 Users/nmr_mike/Thesis/Chapter_V/Figure5_2.jpg

[Salt] (mM) in Cellular Environment

Ion	Outside (mM)	Inside (mM)	Ratio (Out/In)
Na ⁺	145.0	15.0	9.67
K ⁺	4.0	155.0	0.03
Ca ²⁺	1.5	0.1-0.01	15.0 - 150
Mg ²⁺	2.0	0.5	4.0
Cl ⁻	123.0	4.2	29.29

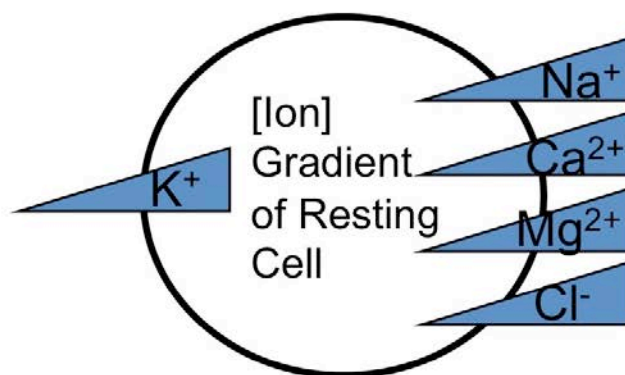


Figure 5.3: Intracellular and extracellular ion concentrations of a resting cell
 Intracellular and extracellular ionic strengths of Na⁺, K⁺, Ca²⁺, Mg²⁺, and Cl⁻ ions are shown (top), while their concentration gradients across the membrane are qualitatively shown below. Upon membrane depolarization ion channels open to allow for ions to pass into or out of the cell dependent upon their concentration gradients (Hille et al. 2001).
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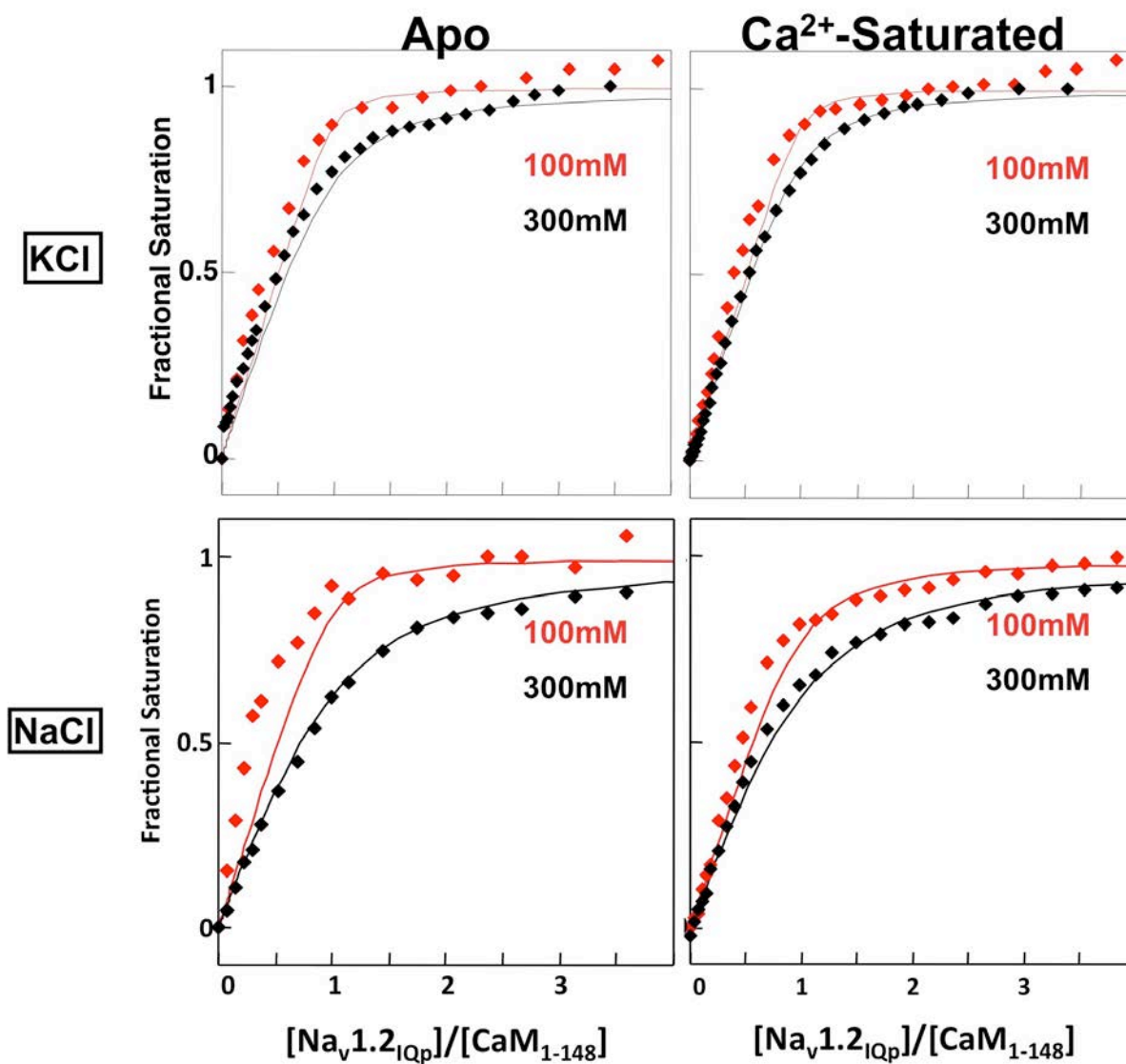


Figure 5.4: Apo or $(\text{Ca}^{2+})_4\text{-CaM}_{1-148}$ titration of $\text{Nav}1.2_{\text{IQp}}$ at varied [Salt]
 Titrations of $\text{Nav}1.2_{\text{IQp}}$ with apo or $(\text{Ca}^{2+})_4\text{-CaM}_{76-48}$ are shown. Titrations were conducted at 5 mM CaCl_2 or 50 μM EGTA for Ca^{2+} -saturating or apo trials respectively. Shown in red are trials performed at 100 mM [Salt], while trials shown in black were performed at 300 mM [Salt]
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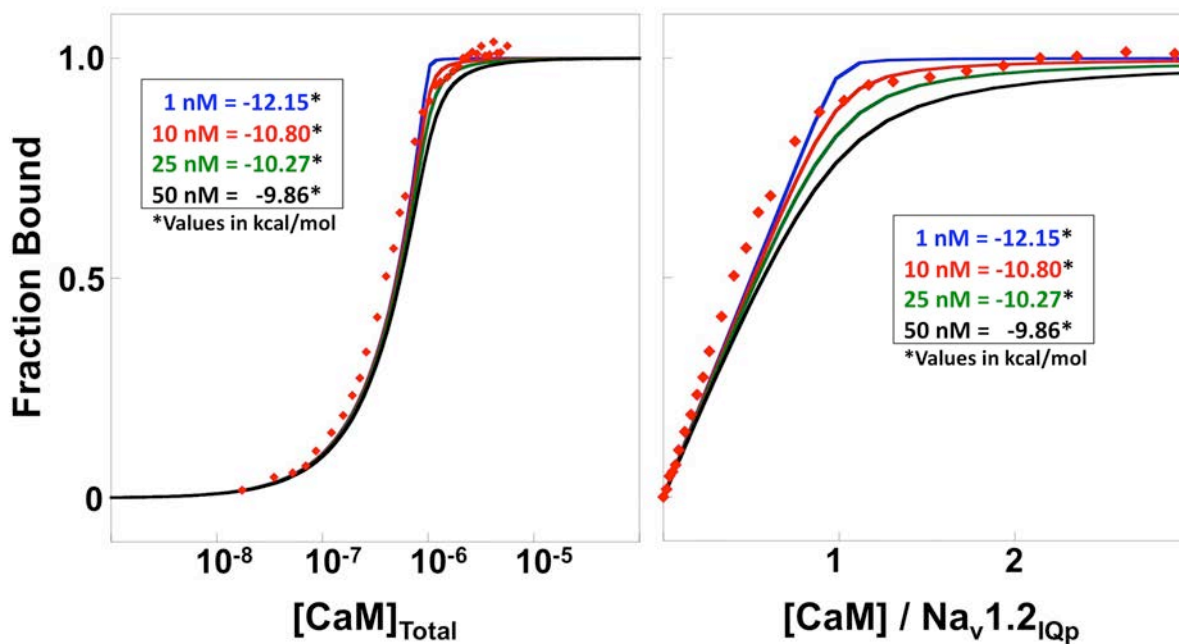


Figure 5.5: Simulated fitting curves used to determine Na_v1.2_{IQp} binding affinity for CaM. Both plots display identical data plotted in two different formats. Na_v1.2_{IQp} binding to (Ca²⁺)₄-CaM₁₋₁₄₈ (red diamonds) is shown on the left where the log [CaM]_{Total} is plotted on the x-axis, while on the x-axis of the titration on the right the stoichiometry of [CaM] / [Na_v1.2_{IQp}] is shown. Shown in solid lines are simulated binding curves at varied affinities/binding energies where the [Na_v1.2_{IQp}] was set to 1 μM. Note that when stoichiometric data is plotted on a log scale x-axis typically used for equilibrium binding curves that the difference between one possible fit or another is significantly reduced.

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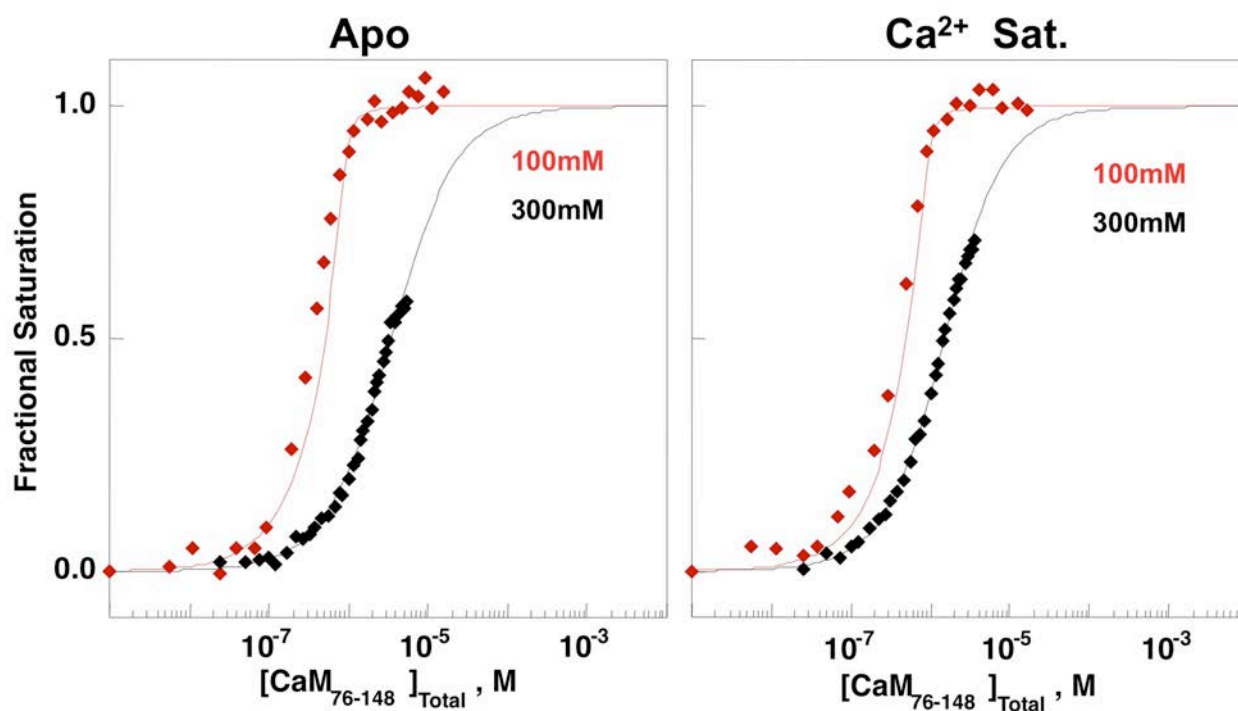


Figure 5.6: Apo or (Ca²⁺)₂-CaM₇₆₋₁₄₈ titration of Na_v1.2_{IQp} at varied [KCl]
 Titrations of Na_v1.2_{IQp} with apo or (Ca²⁺)₄-CaM₇₆₋₄₈ are shown. Titrations were conducted at 5 mM CaCl₂ or 50 μM EGTA for Ca²⁺-saturating or apo trials respectively. Shown in red are trials performed at 100 mM KCl, while trials shown in black were performed at 300 mM KCl.
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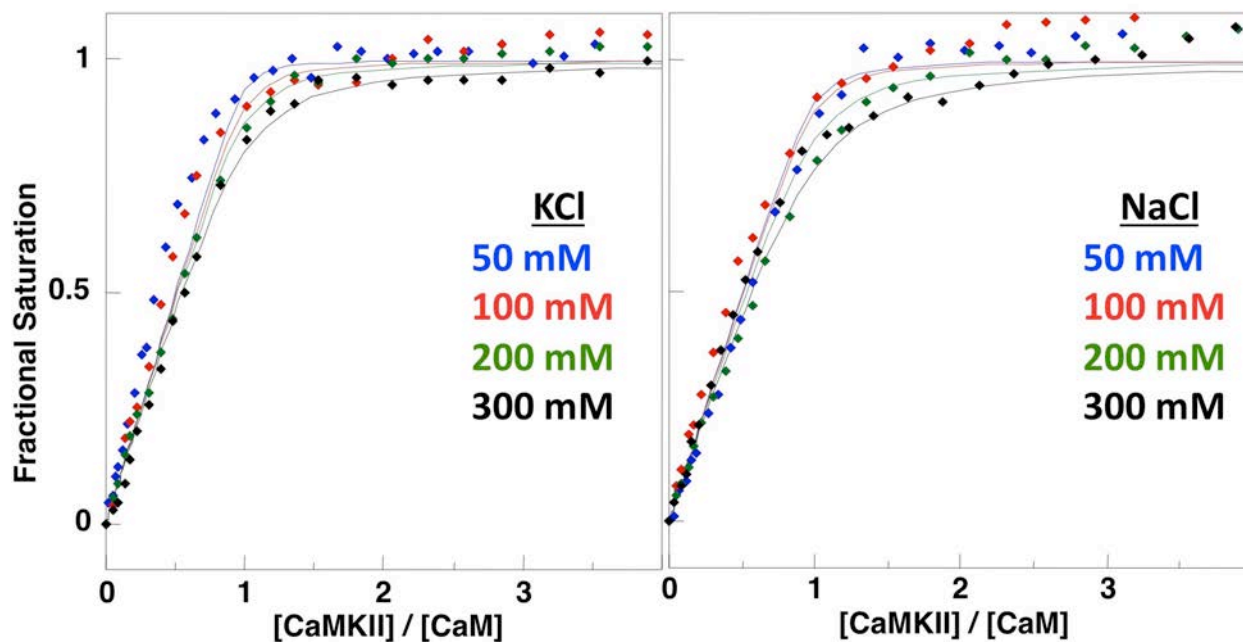


Figure 5.7: $(\text{Ca}^{2+})_4\text{-CaM}_{1-148}$ titration of CaMKII_p at varied $[\text{KCl}]$ and $[\text{NaCl}]$
 Titrations of CaMKII_p with $(\text{Ca}^{2+})_4\text{-CaM}_{1-148}$ are shown. Titrations were conducted at CaCl_2 concentrations of 5mM at 50 mM (blue), 100 mM (red), 200 mM (green), and 300 mM (black) NaCl or KCl .

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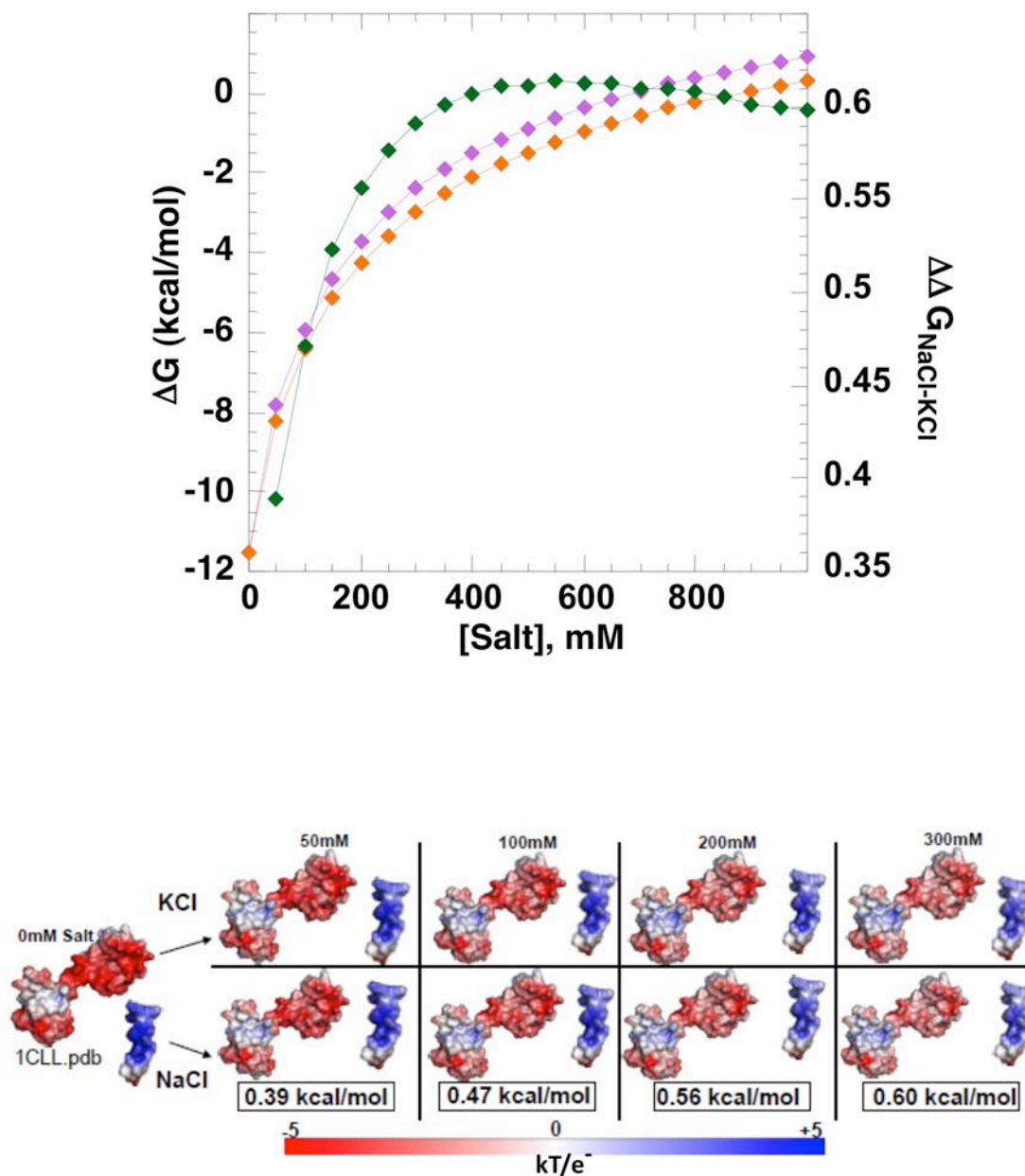


Figure 5.8: Calculated electrostatic binding energies of CaMKII_p for (Ca²⁺)₄-CaM₁₋₁₄₈. **A:** Plotted change in CaMKII_p affinity for (Ca²⁺)₄-CaM₁₋₁₄₈ as a function of salt. Plotted on the primary Y-axis are APBS calculated electrostatic binding energies of CaMKII_p for (Ca²⁺)₄-CaM₁₋₁₄₈ as a function of [NaCl] (purple) or [KCl] (orange). Plotted on the secondary Y-axis is the difference between APBS calculated NaCl and KCl binding affinities at similar concentrations (green). **B:** Electrostatic surface potentials of (Ca²⁺)₄-CaM₁₋₁₄₈ and CaMKII_p at varied NaCl and KCl concentrations. Positive surface is colored in blue, while negative surface is shown in red.

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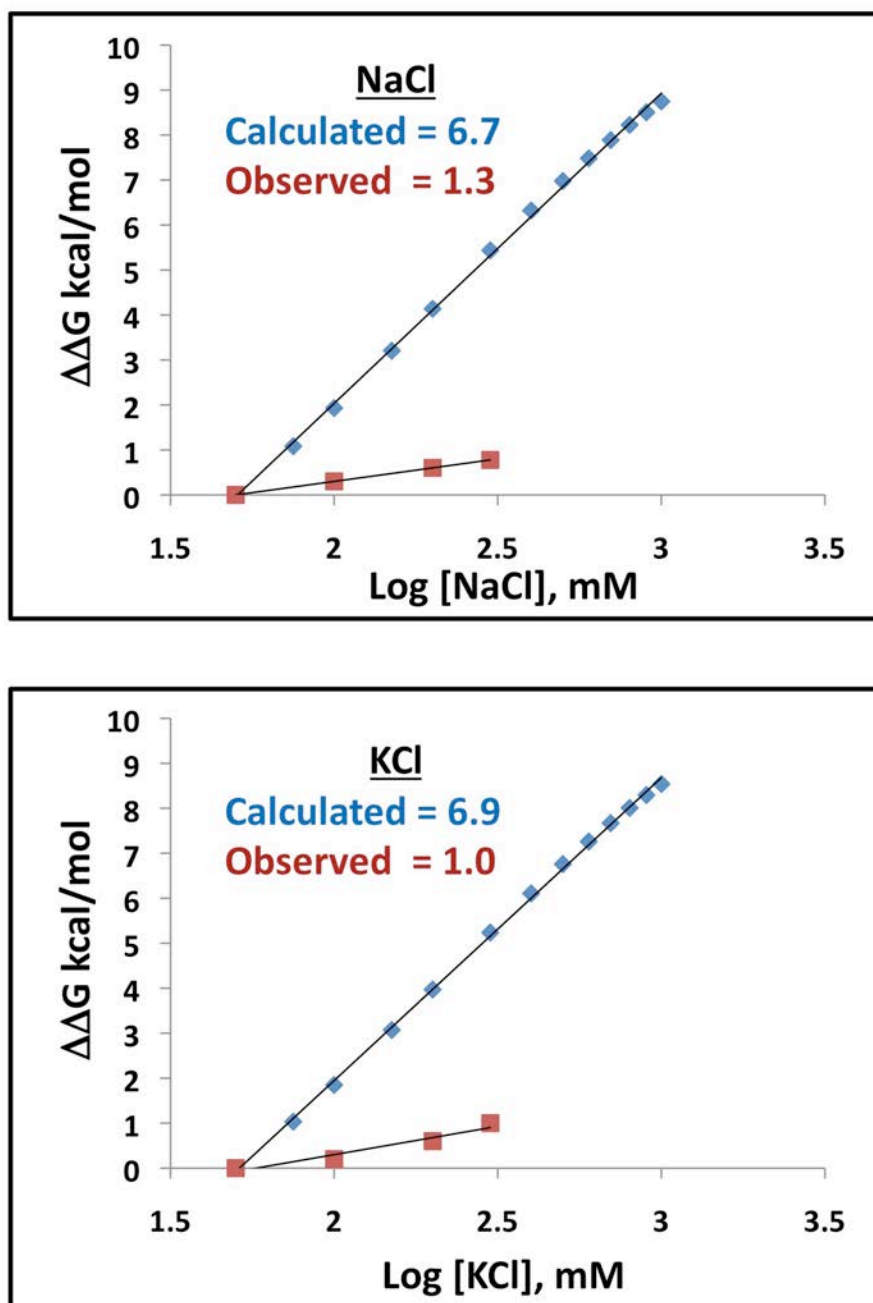


Figure 5.9: Comparison of calculated and observed changes in CaMKII_p binding to (Ca²⁺)₄-CaM₁₋₁₄₈ as a function of salt
 Comparison of observed (red) and calculated (light blue) changes in the binding affinity of CaMKII_p for (Ca²⁺)₄-CaM₁₋₁₄₈ as a function of either [NaCl] (upper plot) or [KCl] (lower plot).
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CHAPTER VI

SUMMARY AND FUTURE DIRECTIONS

Thesis Summary

The interaction of apo CaM with targets is an emerging area of interest in the CaM research field where historically it was a widely held belief that only $(Ca^{2+})_4$ -CaM bound and activated intracellular targets (Cheung, 1980; Klee and Haiech, 1980; Siegel, 1973). Numerous targets such as myosins, ion channels, and growth-associated proteins have been shown to interact with apo CaM via an IQ-motif (Black et al., 2005; Liu and Storm, 1990; Martin and Bayley, 2004; Shah et al., 2006). This thesis investigates how the anti-psychotic drug TFP, and $Na_v1.2_{IQp}$ interact with apo CaM at the molecular level to better understand how apo CaM recognizes and binds such diverse targets. A model of how CaM interacts with both of these targets is presented in **Figure 6.1**.

TFP Binding Induces a Biphasic Response in the Ca^{2+} -Binding

Affinity of CaM

Nearly all targets that bind to CaM either naturally occurring or synthetic “tune” its Ca^{2+} -binding affinity in some manner (Peersen et al., 1997). Targets such as CaMKII_p, Calcineurin, and the Ryanodine receptor increase the Ca^{2+} -binding affinity of CaM, while IQ-motif containing proteins such as $Na_v1.2$ lower the Ca^{2+} -binding affinity of CaM (Evans and Shea, 2009; Newman and Shea, 2006; Quintana et al., 2005; Theoharis et al., 2008). Unlike these previously mentioned protein targets of CaM, the anti-psychotic drug TFP was shown in Chapter II to both increase and decrease the Ca^{2+} -binding affinity of CaM dependent upon the concentration of TFP examined. To our knowledge this is the first time this behavior has been observed with any allosteric effector of CaM.

The basis of the observed biphasic response can be found in the stoichiometry of TFP binding to CaM. We determined that there are 2 TFP binding sites (one per domain) in apo CaM, while there are 4 TFP binding sites in $(Ca^{2+})_4$ -CaM. The TFP induced biphasic response in

Ca²⁺-binding affinity results from each TFP site having a different binding affinity for apo and (Ca²⁺)₄-CaM as shown in Chapter II. TFP at ratios of 1:1 and 2:1 have a more favorable binding affinity for apo CaM than (Ca²⁺)₄-CaM, and thus via allosteric linkage cause the Ca²⁺-binding affinity of CaM to become less favorable. TFP:CaM ratios of 3:1 and 4:1, are achieved in the Ca²⁺-bound state only, because of this the Ca²⁺-bound state of CaM is selectively stabilized over the apo state resulting in a more favorable Ca²⁺-binding affinity. To further probe the basis of this biphasic response, TFP titrations of (Ca²⁺)₄-CaM were performed. We observed biphasic responses in the chemical shift of residues located at the interface between the N- and C-domain of CaM as well as in residues in close (< 4 Å) proximity to multiple TFP binding sites, corroborating the biphasic response seen in Ca²⁺-binding measurements.

CaM uses Distinct Interfaces to Bind TFP under apo and Ca²⁺-Saturating Conditions

Ca²⁺-titrations of CaM at multiple ratios of TFP performed in Chapter II, indicated that the Ca²⁺-binding affinity of the C-domain of CaM was affected to a greater extent than that of the N-domain. It was clear in ¹⁵N-HSQC spectra taken of TFP bound to apo and (Ca²⁺)₄-CaM that TFP induced non-equivalent chemical shifts which were dependent upon the Ca²⁺-ligation state of CaM. Consistent with Ca²⁺-titration data, quantification of N- and C-domain ¹⁵N-HSQC chemical shifts of apo and (Ca²⁺)₄-CaM indicated that in both cases the C-domain of CaM was perturbed to a greater extent than the N-domain. These observations prompted structural studies using the isolated C-domain fragment (CaM₇₆₋₁₄₈) performed in Chapter III to gain insight into the differing molecular interfaces used by apo and (Ca²⁺)₂-CaM₇₆₋₁₄₈ identified in Chapter II.

The crystal structure of TFP bound to (Ca²⁺)₂-CaM₇₆₋₁₄₈ directly reveals the molecular binding interface and conformation used by (Ca²⁺)₂-CaM₇₆₋₁₄₈ when binding TFP. In this structure (Ca²⁺)₂-CaM₇₆₋₁₄₈ adopts an “open” domain conformation observed in all structures of (Ca²⁺)₄-CaM alone or when bound to a target. The TFP binding site is located within the hydrophobic pocket of (Ca²⁺)₂-CaM₇₆₋₁₄₈ and contacts the FLMM tetrad of residues, both

common features seen in all structures of $(\text{Ca}^{2+})_2\text{-CaM}_{76-148}$ bound to protein or drug targets (Ataman et al., 2007). A unique feature of this structure was that 2 chains of $(\text{Ca}^{2+})_2\text{-CaM}_{76-148}$ were found within the unit cell, and formed a pseudo ellipsoidal structure typically observed when the N- and C-domain of $(\text{Ca}^{2+})_4\text{-CaM}_{1-148}$ bind a target. A desirable future set of experiments would be to determine the ^1H , ^{15}N , and ^{13}C assignments of TFP-bound $(\text{Ca}^{2+})_2\text{-CaM}_{76-148}$ to use NMR spectroscopy to observe changes in dynamics upon TFP binding, and compare these changes with areas of chains A and B that differ structurally. This comparison will serve as an orthogonal approach to validate the structural changes observed between chains A and B of $(\text{Ca}^{2+})_2\text{-CaM}_{76-148}$ when bound to TFP, as well as allow for T_1 and T_2 experiments to verify its oligomeric state.

NMR studies of residue-specific changes induced upon TFP binding to apo CaM_{76-148} performed in **Chapter III** indicated that significant structural changes occurred within apo CaM_{76-148} as TFP binds. These residue-specific changes were then mapped to a structure of apo CaM_{76-148} determined in the absence of TFP and clustered onto helices F-H as well as the Ca^{2+} -binding loops. Hydrophobic residues dominate TFP binding perturbed resonances within helices F-H, while polar residues are perturbed in the Ca^{2+} -binding loops. The chemical shifts induced upon TFP addition to apo CaM_{76-148} are most likely a result of 2 different phenomena: 1) direct TFP binding or 2) allosteric linkage. Due to the chemical properties of TFP (hydrophobic small molecule) we hypothesize that TFP interacts directly with hydrophobic residues contained within helices F-H, while it allosterically alters the chemical shifts of the Ca^{2+} -binding loops. Evidence to support this claim was shown in **Chapter II** in which TFP was shown to allosterically alter the Ca^{2+} -binding affinity of CaM_{76-148} , and that it is unlikely that TFP directly competes with Ca^{2+} within the Ca^{2+} -binding loops of CaM_{76-148} .

Although TFP significantly perturbed apo CaM_{76-148} upon binding, the magnitude of change in chemical shifts was much smaller than those observed for TFP binding to C-domain of CaM_{1-148} , suggesting a smaller conformational change occurs under apo conditions. The only conformation observed of the C-domain of apo CaM or CaM-like proteins when bound to a

target has been that of the “semi-open” conformation. We hypothesize that a “semi-open” conformation is used by apo CaM₇₆₋₁₄₈ when binding TFP, as this conformation (consistent with magnitude of ¹⁵N-HSQC shifts) requires less structural rearrangement than the “open” conformation. The adoption of a “semi-open” conformation is also evidenced by the slight decrease in T₂ relaxation time upon addition of TFP to apo CaM₇₆₋₁₄₈, indicative of a slight increase in molecular size. Future studies are proposed to determine changes in T₂ relaxation rates upon TFP binding to (Ca²⁺)₂-CaM₇₆₋₁₄₈. These values would help to confirm that the rate of molecular tumbling due to changes in the hydrodynamic radius of TFP-bound apo CaM₇₆₋₁₄₈ is faster than that of TFP-bound (Ca²⁺)₂-CaM₇₆₋₁₄₈. It is anticipated that the average T₂ rate of apo CaM > TFP-bound apo CaM₇₆₋₁₄₈ > TFP-bound (Ca²⁺)₂-CaM₇₆₋₁₄₈.

To definitively confirm that CaM uses distinct interfaces to bind TFP under apo and Ca²⁺-saturating conditions high resolution 3 dimensional structures are required. To this end we have determined the (Ca²⁺)₂-CaM₇₆₋₁₄₈ adopts an “open” conformation via x-ray crystallography. The future goal of the work described in Chapter III is to determine the solution structure of TFP bound to apo CaM₇₆₋₁₄₈. These NMR data have been collected and are awaiting assignment and analysis for calculation the structure of TFP bound to apo CaM₇₆₋₁₄₈.

Conservation of Carboxamide-Containing Side chains in apo CaM Binding Motifs

Studies presented in Chapter IV indicated the “semi-open” domain conformation is used by the C-domain of CaM and CaM-like proteins when interacting with targets under apo conditions. The structure of the SK-channel bound to a partially Ca²⁺-saturated CaM shows that an IQ-motif is not necessary for binding of the C-domain of apo CaM (Figure 4.9) (Schumacher et al., 2004). The one structural feature observed that was conserved by all apo CaM binding motifs was a Gln or Asn residue whose carboxamide-containing side chain was used to make 2 hydrogen bonds to atoms found in the backbone of residues 112 and 114 located within the loop connecting helices F and G of CaM (Figure 4.9).

To examine the necessity of this hydrogen bonding network for apo C-domain binding additional studies are proposed in which mutations of either Gln→Glu ($\text{Na}_v1.2_{\text{IQp}}$) or Asn→Asp (SK-Channel) will be made at the Q position of the IQ-motif followed by fluorescence anisotropy monitored binding studies. These mutants would test whether 2 hydrogen bonds are required for apo binding of the C-domain of CaM. The aforementioned study (dependent upon the results) could be repeated except instead of Gln→Glu or Asn→Asp mutations, a Glu or Asn →Ala mutation would be made that completely removes the observed hydrogen bond network between apo CaM and the target peptide.

Changes in the Electrostatic Environment of CaM Alters its Interaction with Targets

Studies shown in **Chapter V** indicate that increasing the concentration of NaCl or KCl in solution lowers the affinity of CaM for its target, due to the screening of electrostatic interactions between CaM and the peptide target. Perturbation of the local electrostatic environment of CaM can also be performed by introducing point mutations that introduce, remove or reverse the intrinsic charge of an amino acid. Structural changes due to this type of perturbation of the electrostatic environment of CaM, will be examined with CaM mutants D95G and H135R. These mutants were identified via a genetic screen of viable *Paramecia* that exhibited abnormal chemotactic behavior resulting from impaired ion channel function (Kung et al., 1992; Ling et al., 1992). Exploratory studies of D95G and H135R have identified these over-reactive mutants as ideal candidates for structural studies. $(\text{Ca}^{2+})_2\text{-CaM}_{76-148}$ H135R was crystallized and diffracted to 2.2 Å, while the ^{15}N -HSQC spectrum of D95G $(\text{Ca}^{2+})_2\text{-CaM}_{76-148}$ bound to $\text{Na}_v1.2_{\text{IQp}}$ showed excellent peak dispersion (Figure 6.2).

Future endeavors to determine to these high-resolution structures will allow for direct observations to be made of the effect of D95G or H135R on CaM alone or when in complex with $\text{Na}_v1.2_{\text{IQp}}$. Either of these structures would be the first of their kind to have been determined from the genetic screen of viable *Paramecium* mutants. These structures would lay the

groundwork for structural based conclusions to be made as to the molecular basis of the altered chemotactic behavior observed in *Paramecium* mutants.

Future Studies

Changes in Met 144 Dynamics upon Binding $(\text{Ca}^{2+})_2\text{-CaM}_{76-148}$

The residue M144 has previously been identified to adopt the most variable conformations when bound to a target of the FLMM tetrad residues, as well as being the least conserved FLMM tetrad residue in 102 CaM sequences from other species. Studies presented in Chapter III indicated that the side chain dynamics of M144 play a key role in selecting the orientation of TFP within the hydrophobic cleft of $(\text{Ca}^{2+})_2\text{-CaM}_{76-148}$. To investigate this observation further, ^{13}C -methyl relaxation experiments are proposed to examine changes in relaxation rates of $(\text{Ca}^{2+})_2\text{-CaM}_{76-148}$ upon binding TFP (Palmer, 2001; Wand, 2001). This data would help to clarify at a finer level of detail the amount of molecular motion present within the TFP-binding site of $(\text{Ca}^{2+})_2\text{-CaM}_{76-148}$ as x-ray crystallography only provide static images.

This data could then be incorporated into small molecule docking simulations of drugs to CaM in which highly dynamic residues would be allowed to be flexible, instead of being held rigid as is often done for all residues in docking simulations. This treatment would allow for a more realistic docking simulation to be run, yet be less computationally taxing than allowing all-atom flexibility. On a larger scale, incorporation of *in vitro* data into virtual docking simulations of lead compounds against hub-proteins such as CaM, may benefit pharmaceutical discovery in which off target effects are not desired. Virtual screening of a compound library against every known protein structure is considerably more computationally expensive than screening against a select set of hub-proteins who regulate a myriad of downstream targets. *In silico* identification of compounds that alter the function of hub-proteins such as CaM would save in the cost of bringing a compound to clinical trials by identifying a drug or drugs whose interaction with a network of regulatory pathways could result in an undesired off-target effect.

Determination of Binding Site for N-domain of $(Ca^{2+})_4$ -CaM on $Na_v1.2$

Studies presented in **Chapter IV** determined that the N-domain of $(Ca^{2+})_4$ -CaM does not bind to $Na_v1.2_{IQp}$, indicating that there is likely to be binding site for the N-domain of $(Ca^{2+})_4$ -CaM located in another region of $Na_v1.2$. Future studies will build off of the observations made in Chapter IV by using a peptide array. This array will be composed of overlapping 30-residue peptide sequences derived from the cytoplasmic face of $Na_v1.2$. Following initial identification of a putative N-domain $Na_v1.2$ binding sequence ($Na_v1.2_N$) FRET will be used to confirm that donor-acceptor labeled peptides corresponding to $Na_v1.2_{IQp}$, and $Na_v1.2_N$ are capable of simultaneously binding to the N- and C-domain of $(Ca^{2+})_4$ -CaM. Lastly, comparison of patch clamp currents of $Na_v1.2$ channel mutants in the area of $Na_v1.2_N$, as well as the use of CaM mutants deficient in N-domain Ca^{2+} -binding will be used to verify functional role binding to the N-domain of $(Ca^{2+})_4$ -CaM.

Circular Permutation of Charge Residues in $Na_v1.2_{IQp}$

An IQ-motif peptide derived from another sodium channel variant ($Na_v1.5_{IQp}$) is homologous (78%) to $Na_v1.2_{IQp}$ yet binds with a 16-fold lower affinity to Apo-CaM and 200-fold lower affinity to $(Ca^{2+})_4$ -CaM than $Na_v1.2_{IQp}$ (Shah et al., 2006). The most notable difference between the two to account for these differences is their charge distribution and overall net charge (**Figure 6.3**). To investigate the role that charge distribution has on the affinity of apo and Ca^{2+} -CaM, charged residues of the $Na_v1.2_{IQp}$ will be mutated. Initially charged residues of the $Na_v1.2_{IQp}$ which are not conserved when compared to $Na_v1.5_{IQp}$ will be mutated to their $Na_v1.5_{IQp}$ counterpart. The effect of the mutant $Na_v1.2_{IQp}$ on its affinity for CaM and effect on Ca^{2+} binding by CaM will then be measured to clarify how apo- and $(Ca^{2+})_4$ -CaM interact. Pinpointing and then mutating key areas of the $Na_v1.2_{IQp}$ -CaM interface will explain on a molecular level how the Ca^{2+} -affinity of the domains of CaM are “tuned” to perform their physiological functions when bound to IQ-motifs of sodium channel variants (**Figure 1.8**)

***In Vitro* Evolution of apo CaM Binding IQ-motifs**

A fascinating aspect of the IQ-motif (Figure 4.1a) is that with the exception of the Q, it is similar in amino acid composition and spacing with a 1-8-14 BAA motif known to only interact with $(\text{Ca}^{2+})_4$ -CaM (Yap et al., 2000). Although the IQ-motif as a whole has been characterized to preferentially interact with apo CaM, exceptions to this statement have been observed for IQ motifs found in $\text{Na}_v1.5$ and $\text{Ca}_v1.2$ ion channels in which more favorable IQ-motif binding is observed for $(\text{Ca}^{2+})_4$ - than apo CaM (Shah et al., 2006). To explore the sequence requirements of target binding to CaM, phage display will be used to evolve peptide target sequences or varying affinity to apo and $(\text{Ca}^{2+})_4$ -CaM to.

The selection pool will be composed of a phage library in which unconserved positions of the IQ-motif will be fully randomized at positions indicated with an “X” in the following sequence [x(Q/N)xxx(K/R)xxxx(K/R)], while positions in parenthesis will only carry a binary randomization. Bio-panning will first be carried out against immobilized CaM where initial wash steps will include EGTA to insure enrichment of apo CaM binding sequences. In the last biopanning selection round prior to sequencing, the phage library will be split and selected against either apo or $(\text{Ca}^{2+})_4$ -CaM using either an EGTA or Ca^{2+} final wash step. After phage sequencing, comparison of these two groups will be made to identify sequences (if any) that only bind to apo CaM, as well as those that are capable of binding both apo and $(\text{Ca}^{2+})_4$ -CaM. This process will be repeated again with the exception that initial bio-panning will be done in the presence of Ca^{2+} instead of EGTA, to evolve IQ-motif sequences that preferentially bind $(\text{Ca}^{2+})_4$ -CaM (Figure 6.4) The final result will be 4 pools of CaM binding motifs consisting of sequences that: only bind $(\text{Ca}^{2+})_4$ -CaM, bind apo and $(\text{Ca}^{2+})_4$ -CaM with preference for $(\text{Ca}^{2+})_4$ -CaM, bind apo and $(\text{Ca}^{2+})_4$ -CaM with preference for apo CaM, and sequences that only bind apo CaM.

The results from these experiments would help to resolve uncertainties as to what the sequence determinants of IQ-motifs are that allow them to tune their affinity for the multiple

Ca²⁺-ligation states of CaM. Comparison of *in vitro* evolutionary results with those of phylogenetic trees of IQ-motifs evolved *in vivo* will also provide insight into what evolutionary pressures produced naturally occurring IQ-motifs whose evolution are a result of selection pressure of multiple simultaneous factors.

Examination of Naturally Occurring Na_v1.2_{IQp} R1902C

Mutation on CaM

The sodium channel variants contain naturally occurring mutations distributed along multiple areas of the channel sequence that are genetically linked to human disorders. One such mutation R1902C is located within the IQ-motif of Na_v1.2 and is associated with familial autism (Weiss et al., 2003). Future studies are proposed to determine if the R1902C mutation alters previously determined Ca²⁺-binding affinity values of CaM when bound to Wt-Na_v1.2, as well as to determine how this mutation alters the binding affinity of Na_v1.2 to CaM. Structural studies (SAXS, NMR, or X-ray crystallography) are proposed in combination with other biophysical measurements and techniques (CD, analytical ultracentrifugation, and stokes radius) to determine how R1902C may alter the CaM-Na_v1.2_{IQp} complex. It is well known that binding of targets “tune” the Ca²⁺-binding affinity of CaM, these investigations may provide the 1st example of how alteration of tuned affinities of CaM result in a human disorder. A better understanding of the fundamental molecular basis of this disorder may aid in improved treatment therapies for individuals carrying the R1902C mutation.

Structure Determination of Larger Intracellular Fragments of

Na_v1.2 in Complex with CaM

Structural studies involving CaM bound to peptides derived from target proteins have proven useful in the dissection of the molecular interactions made upon binding. Although valuable, these studies fail to capture the overall structural changes that occur within the entire macromolecular complex that result in their function. This lack of overall mechanistic detail is likely to be most pronounced in IQ-motif containing targets such as Na_v1.2 where CaM is an

intrinsic subunit under both apo and Ca^{2+} -saturating conditions. Ideally, to fully understand how CaM regulates $\text{Na}_v1.2$ gating to allow passage of Na^+ ions across cellular membranes, structures of $\text{Na}_v1.2$ bound to CaM in all of their physiologically relevant states are required. Given that the determination of the structure of $\text{Na}_v1.2$ represents a significant challenge that has yet to be overcome by structural biologists, it is unlikely that in the near future structures of apo and $(\text{Ca}^{2+})_4$ -CaM bound to $\text{Na}_v1.2$ will be determined. A more tractable path to uncover a deeper mechanistic understanding of how $\text{Na}_v1.2$ is regulated by CaM beyond that of peptide studies might be found in structural studies involving intracellular domains of $\text{Na}_v1.2$ identified to bind CaM. The IQ-motif of $\text{Na}_v1.2$ is found within a larger domain of $\text{Na}_v1.2$ theorized to be composed of 6 α -helices, while studies shown in **Chapter 4** indicated that a region outside of $\text{Na}_v1.2_{\text{IQp}}$ interacted with the N-domain of CaM. For these reasons, future structural studies are proposed to determine the structure of apo and $(\text{Ca}^{2+})_4$ -CaM bound to $\text{Na}_v1.2_{\text{IQp}}$ in the context of the entire C-terminal domain of $\text{Na}_v1.2$. It is also proposed that dependent upon studies to determine the portion of $\text{Na}_v1.2$ that binds the N-domain of CaM, that this segment be included as well. These proposed studies would help to establish spatial constraints upon where intracellular regions of $\text{Na}_v1.2$ are located under apo and Ca^{2+} -saturating conditions.

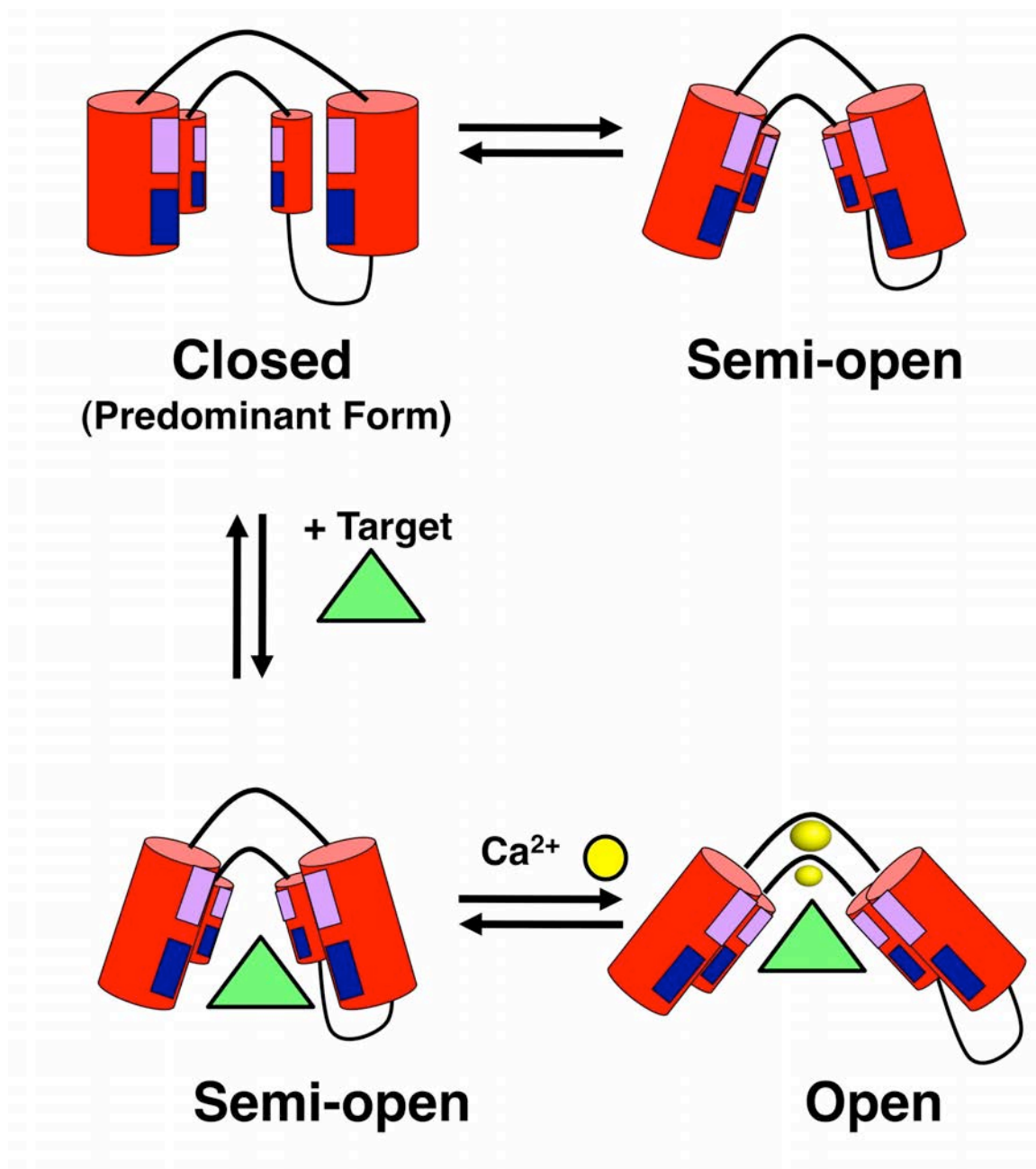


Figure 6.1: Model of apo and $(\text{Ca}^{2+})_4$ -CaM C-domain conformations used when binding targets. The C-domain of CaM is represented as red cylinders which contain two distinct binding surfaces. The purple surface located near the perimeter of the hydrophobic cleft of CaM is used for target (green triangle) binding under apo conditions, while the light purple surface is used when the C-domain is Ca^{2+} -saturated.

Users/nmr_mike/Thesis/Chapter_VI/Figure6_1.jpg

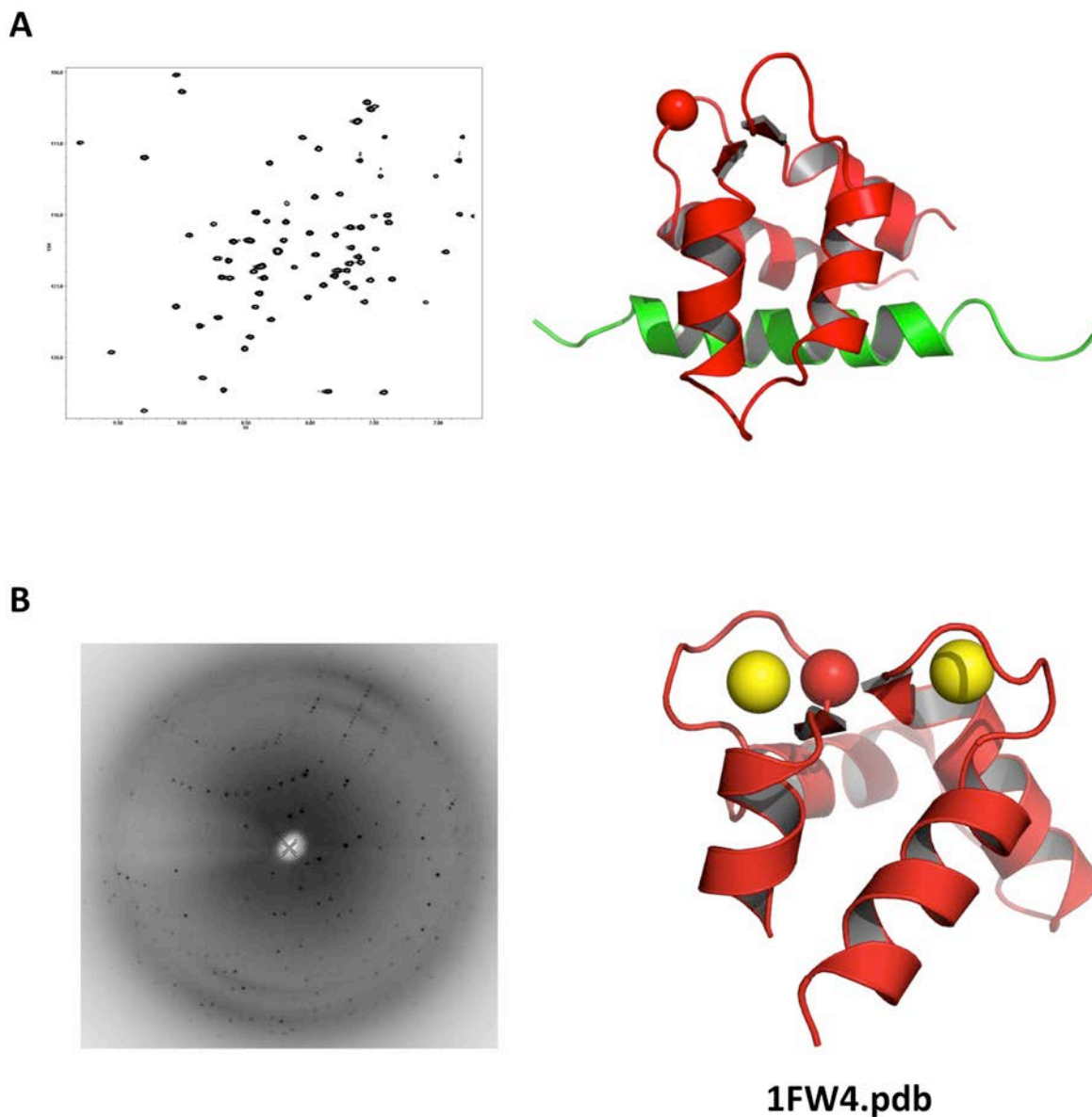


Figure 6.2: ^{15}N -HSQC Spectrum and diffraction image of PCaM Mutants

A: ^{15}N -HSQC spectrum of D95G apo CaM₇₆₋₁₄₈ (red) Bound to Nav1.2IQP (green), where the location of the D95G mutation is indicated by a red sphere. **B:** Diffraction image of H135R (Ca²⁺)₂-CaM₇₆₋₁₄₈. The location of the H135R mutation is shown as a red sphere on the structure of Wt-CaM₇₈₋₁₄₈ (red) where Ca²⁺-ions are colored yellow.
Users/nmr_mike/Thesis/Chapter_VI/Figure6_2.jpg

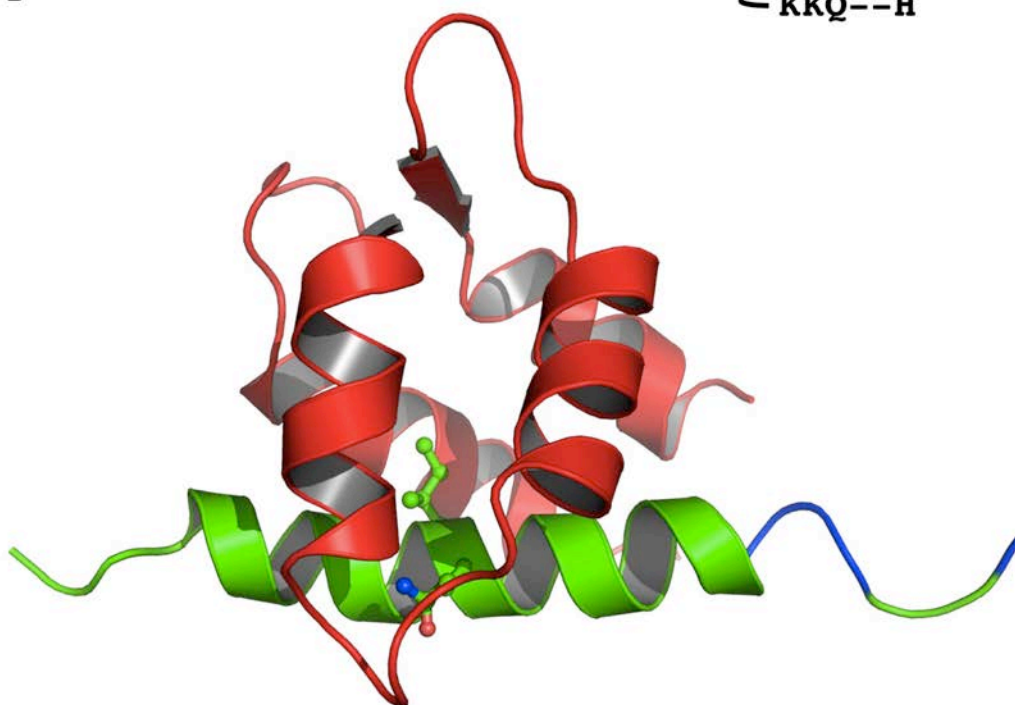
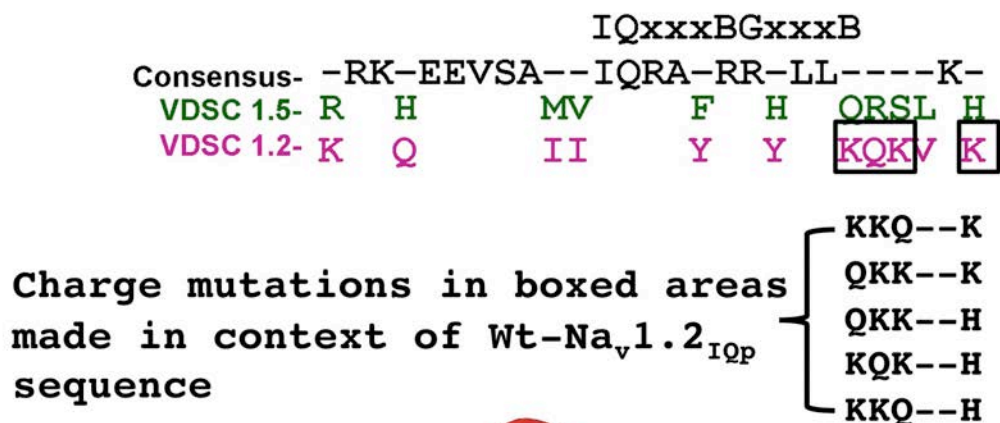


Figure 6.3: Sequence alignment of $\text{Na}_v1.2$ and $\text{Na}_v1.5$ IQ-motifs
 The IQ-motifs of $\text{Na}_v1.2$ and $\text{Na}_v1.5$ are shown in green and magenta respectively. Boxed areas represent positions that differ in charge, sequences below represent circular permutations of charged residues of $\text{Na}_v1.2_{\text{IQP}}$ that will be examined via Ca^{2+} titration and fluorescence anisotropy to monitor changes in Ca^{2+} and peptide-binding affinity of CaM.
 Users/nmr_mike/Thesis/Chapter_VI/Figure6_3.jpg

Starting Phage Library
 $x(N/Q)xxx(K/R)xxx(K/R)$
 Where x =any amino acid

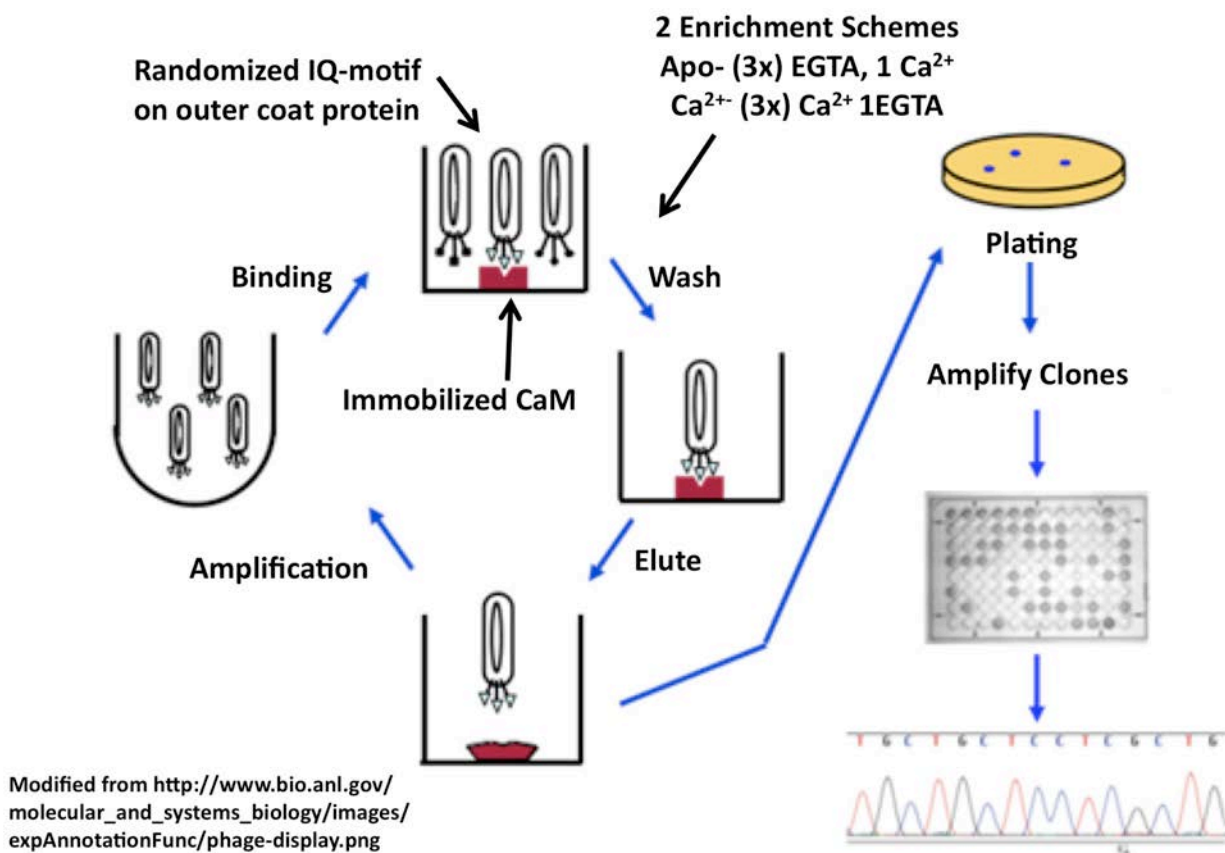


Figure 6.4: Schematic of phage display selection procedure
 An initial phage library will be generated and enriched to produce IQ-motifs which either selectively bind to apo or (Ca²⁺)₄-CaM.
 Users/nmr_mike/Thesis/Chapter_VI/Figure6_4.jpg

APPENDIX A
FORTRAN FUNCTION FOR FITTING FLUORESCENCE
ANISOTROPY DATA TO A SIMPLE LANGMUIR
BINDING ISOTHERM

The following Fortran function was used in nonlinear least squares analysis of the fluorescence anisotropy data presented in **Chapter II** and **Chapter IV**. Fits using this function were used to determine the association constant of CaM for a synthetic peptide represent the CaM-binding domain of Na_v1.2_{IQp} or CaMKII_p. This equation fits to a one-site binding model (**Equation A.1**), using the total concentration of CaM at each point.

$$\bar{Y} = \frac{K_a[X]}{1 + K_a[X]} \quad (\text{A.1})$$

Function FX(Ans, X, ydum, ierr, n)

```
*
* fxYvsXt1K.f
*
* Function for fitting ligand binding to macromolecule
* when total ligand, rather than free ligand,
* concentration is known.
*   Model of binding: Monomer binding to equal and
*   independent site(s).
*   Number of Sites: Set by ans parameter ans(3)
*   Original function name: &FNWT1 <831018.1019>
```

```
* The ANS vector has the following form:
* ANS(1) = macromolecule total concentration
* ANS(2) = Association binding constant for ligand to
*   macromolecule
* ANS(3) = N, number of independent sites on
*   macromolecule
* ANS(4) = Endpoint at Low X
* ANS(5) = Endpoint at High X
*
```

Real Ans(5), K, MKS

F = 1.E-7*X ! free ligand is approximated as a
fraction of total

MKS = Ans(1)*Ans(2)*Ans(3)

K = Ans(2)

```

EndLowX = Ans(4)
EndHighX = Ans(5)
Span = EndHighX - EndLowX

```

```

Do 10 i = 1, 500
gf = -X + F + MKS*F/(1.+K*F)
gf1 = 1. + MKS/(1.+K*F)/(1.+K*F)
fold = f
F = F -gf/gf1

```

```

If(Abs(Fold/F - 1.).lt.1.E-5) Go to 11

```

```

10 Continue

```

```

11 Fx = EndLowX + Span * F*K/(1.+F*K)

```

```

Return
End

```

```

SUBROUTINE START(NAME,MAXP,DNAME,MAXD,MAXV)

```

```

C
C   THIS ROUTINE IS USED TO SET THE VARIABLE NAMES FOR C   THE
PARAMETERS. THESE NAMES MUST CORRESPOND TO
C   THE VARIABLES USED IN THE FX ROUTINE.

```

```

CHARACTER*8 NAME(5),NAMES(5)
CHARACTER*8 DNAME(1),DNAMES(1)
DATA NAMES/'[Mtotal]','Ka (M) ','# sites ','EndLowX','EndHighX'/
DATA DNAMES/'dG@22C'/

```

```

C
MAXV=1
C   maxv is the # of independent variables for the fit.
C   This is usually set to 1. However, to fit multiple C sets of data it could be set to 2. The
second vector C   of X values can then be used to specify which
C   data set is actually being fit. If this is greater
C   than 1 then X in the FX routine must be dimensioned
C   appropriately.

```

```

C
MAXP=5
C   maxp is the # of parameters in model being fit by
C   these routines.
DO 10 I=1,MAXP
10 NAME(I)=NAMES(I)
C   NAME is the names of the actual fitting parameters.

```

```
MAXD=1
```

```
C MAXD specifies the number of derived parameters. A  
C derived parameter is simply a quantity that you wish  
C evaluated from the fitting answers after the fit is  
C finished. By using this feature the program will  
C complete the actual error propagation for the derived  
C parameters. In the current example, we calculate the  
C free energy from the Kd.
```

```
c
```

```
DO 110 I=1,MAXD
```

```
110 DNAME(I)=DNAMES(I)
```

```
c DNAME is the names of the derived parameters.
```

```
RETURN
```

```
END
```

```
subroutine derive(old,new)
```

```
C This routine is used to map the OLD (fitted)  
C parameters into the desired derived (NEW) parameters.
```

```
c
```

```
real old(3),new(1)
```

```
new(1) = -0.58646 * 2.303 * alog10(old(2)) !RT at 22C
```

```
return
```

```
end
```

APPENDIX B

FORTRAN FUNCTION FOR ANALYSIS OF

CALCIUM-DEPENDENT CHANGES IN FLUORESCENCE

THAT DESCRIBE

FILLING OF ONLY TWO SITES

The following Fortran function was used for nonlinear least squares analysis to determine the free energies of calcium binding to CaM from calcium-dependent changes in fluorescence during equilibrium calcium titrations. This function uses a model-independent two-site equation (**Equation B.1**) that describes potentially heterogeneous, cooperative sites.

$$\bar{Y} = \frac{K_1[X] + 2K_2[X]^2}{2(1 + K_1[X] + K_2[X]^2)} \quad (\text{B.1})$$

```

FUNCTION FX(ANS,X,Y,INDEX,N)
*****
* FX2s2K.mpf
* Function file for use with nonlin (M.L.Johnson)
* For analysis of classical 2-site isotherms
* Resolves macroscopic binding energies at 2 sites
* Permits resolution of span and Yzero value
*
* Independent variable:  X = [Ligand] free
* Dependent variable:  Fractional saturation
* Parameters or ANS vector elements:
*   ANS(1) = TEMPERATURE IN CENTIGRADE
*   ANS(2) = Macrosopic Delta G for 1 bound
*   ANS(3) = Macrosopic Delta G for 2 bound
*   ANS(4) = Span
*   ANS(5) = Y infinity
*
*****
*
  DIMENSION ANS(5)
  REAL X, Span, Yzero, K1X, K2X2, T, RT
  REAL Ybar, FX
*
  DATA R /.001987/
*

```



```

T=273.15 + ANS(1)
RT=R*T
K1X = exp(-Ans(2)/RT)*X    !k1*X
K2X2 = exp(-Ans(3)/RT)*X*X    !k2*X
Ybar = ((K1X + 2.*K2X2)/(1. + K1X + K2X2))/2.
FX= Ybar * Ans(4) + Ans(5)
*
RETURN
END

SUBROUTINE START(NAME,MAXP,DNAME,MAXD,MAXV)
C
C THIS ROUTINE IS USED TO SET THE VARIABLE NAMES FOR C THE
PARAMETERS. THESE NAMES MUST CORRESPOND TO THE
C VARIABLES USED IN THE FX ROUTINE.
c
CHARACTER*8 NAME(5),NAMES(5)
CHARACTER*8 DNAME(3),DNAMES(3)
DATA NAMES/'Temp','dG1 ','dG2 ','Span','Yzero'/
DATA DNAMES/'dG12est', 'EndLowX', 'EndHighX'/
C
MAXV=1
C maxv is the # of independent variables for the fit.
C This is usually set to 1. However, to fit multiple
C set of data it could be set to 2. The second vector
C of X values can then be used to specify which
C data set is actually being fit. If this is greater
C than 1 then X in the
C FX routine must be dimensioned appropriately.
C
MAXP=5
C MAXP is the # of parameters in model being fit by
C these routines.
DO 10 I=1,MAXP
10 NAME(I)=NAMES(I)
C NAME is the names of the actual fitting parameters.

MAXD=3
C MAXD specifies the number of derived parameters. A
C derived parameter is simply a quantity that you wish
C evaluated from the fitting answers after the fit is
C finished. By using this feature the program will
C complete the actual error propagation for the derived
C parameters. In the current

```

C example, we calculate the area under the exponential
C decay.

C

DO 110 I=1,MAXD

110 DNAME(I)=DNAMES(I)

C DNAME is the names of the derived parameters.

RETURN

END

subroutine derive(old,new)

C This routine is used to map the OLD (fitted)

C parameters into the

C desired derived (NEW) parameters.

C

real old(5),new(1)

DATA R /.001987/

T=273.15 + old(1)

RT = R * T

* Calculate estimate of dG cooperativity for equal

* intrinsic affinities

new(1) = -RT*ALOG(4.) + old(3) - 2. * old(2)

* Calculated endpoint at low [X]

new(2) = 1. * old(5)

* Calculated endpoint at high [X]

new(3) = old(4) + old(5)

return

end

APPENDIX C

NMR ASSIGNMENTS OF APO PCaM₇₆₋₁₄₈Amide Assignments of Apo PCaM₇₆₋₁₄₈

Below are apo PCaM amide assignments determined in 10 mM D₄-imidazole, 100 mM

KCl, 50 μM D₁₆-EDTA, 0.01% NaN₃, pH 6.5, 298.15 K

M76 N-H	128.582	8.462	N111 N-H	119.645	7.672
K77 N-H	123.092	8.519	L112 N-H	119.617	7.748
E78 N-H	122.549	8.597	G113 N-H	107.639	8.107
Q79 N-H	120.502	8.385	E114 N-H	119.557	8.048
D80 N-H	121.813	8.432	K115 N-H	121.092	8.432
S81 N-H	116.038	8.367	L116 N-H	121.217	7.731
E82 N-H	122.846	8.743	T117 N-H	114.079	9.052
E83 N-H	118.113	8.406	D118 N-H	121.178	8.786
E84 N-H	118.816	7.986	D119 N-H	117.112	8.278
L85 N-H	121.173	7.940	E120 N-H	120.834	7.729
I86 N-H	117.864	8.277	V121 N-H	120.580	8.232
E87 N-H	116.921	7.909	D122 N-H	119.401	8.403
A88 N-H	120.969	7.529	E123 N-H	119.626	7.775
F89 N-H	114.879	7.698	M124 N-H	119.043	7.876
K90 N-H	118.026	8.235	I125 N-H	119.076	8.579
V91 N-H	116.669	6.904	R126 N-H	117.879	7.731
F92 N-H	116.018	7.402	E127 N-H	116.569	7.755
D93 N-H	121.526	7.719	A128 N-H	120.458	7.777
R94 N-H	124.448	8.264	I130 N-H	123.259	8.378
D95 N-H	116.327	8.667	G132 N-H	109.669	8.165
G96 N-H	110.529	7.958	D133 N-H	120.198	8.420
N97 N-H	117.934	8.84	H135 N-H	119.500	8.038
G98 N-H	110.966	9.823	I136 N-H	122.114	9.138
L99 N-H	119.423	7.568	N137 N-H	125.122	8.639
I100 N-H	114.759	8.669	Y138 N-H	124.926	7.481
S101 N-H	119.045	8.797	E139 N-H	125.473	8.337
A102 N-H	124.274	8.505	E140 N-H	119.621	7.759
A103 N-H	118.398	8.295	F141 N-H	120.484	8.045
E104 N-H	119.537	7.616	V142 N-H	118.556	8.484
L105 N-H	121.404	8.16	R143 N-H	118.629	7.754
R106 N-H	117.768	8.266	M144 N-H	117.439	7.805
H107 N-H	118.521	7.682	M145 N-H	119.581	8.026
V108 N-H	120.27	7.97	V146 N-H	113.034	7.991
M109 N-H	115.629	8.129	S147 N-H	116.521	7.595
T110 N-H	113.005	7.959	K148 N-H	127.771	7.448

Carbonyl Assignments of apo PCaM₇₆₋₁₄₈

Listed below are carbonyl assignments of apo PCaM₇₆₋₁₄₈ determined in 10 mM D₄-imidazole, 100 mM KCl, 50 μM D₁₆-EDTA, 0.01% NaN₃, pH 6.5, 298.15 K

M76	175.843	T110	175.878
K77	176.488	N111	175.917
E78	176.426	L112	177.649
Q79	175.63	G113	174.786
D80	176.171	E114	176.405
S81	175.194	L116	177.529
E82	177.36	T117	175.34
E83	178.637	D118	178.109
E84	179.258	D119	178.825
L85	178.069	E120	179.412
I86	178.056	V121	177.593
E87	178.416	D122	179.17
A88	179.78	E123	178.217
F89	177.731	M124	178.732
K90	178.492	I125	177.245
V91	176.721	R126	179.317
F92	174.797	E127	177.072
D93	176.498	D129	177.253
R94	177.988	D131	177.201
D95	176.708	G132	174.328
G96	175.011	H135	172.347
N97	176.285	I136	175.07
G98	173.144	N137	175.495
L99	176.371	Y138	176.306
I100	174.419	E139	179.147
S101	175.432	E140	178.477
A102	178.856	F141	176.892
A103	181.134	V142	177.214
E104	177.798	R143	179.209
L105	179.066	M144	178.158
R106	178.531	M145	178.128
H107	177.462	V146	176.561
V108	178.552	S147	173.501
M109	178.812		

Methine, Methylene, and Methyl Assignments of apo PCaM₇₆₋₁₄₈

Listed below are methane, methylene and methyl assignments of apo PCaM₇₆₋₁₄₈ determined in 10 mM D₄-imidazole, 100 mM KCl, 50 μM D₁₆-EDTA, 0.01% NaN₃, pH 6.5, 298.15 K

M76 CA-HA 54.026 4.539	I86 CB-HB 37.149 2.211
M76 CB-HB2 33.188 2.094	I86 CD-HD 12.688 1.069
M76 CB-HB3 33.188 2.002	I86 CG1-HG12 30.141 2.024
M76 CE-HE 17.939 1.867	I86 CG1-HG13 30.088 1.401
M76 CG-HG2 31.831 2.592	I86 CG2-HG2 17.763 1.156
M76 CG-HG3 31.827 2.548	A88 CA-HA 54.926 4.145
K77 CA-HA 56.106 4.349	A88 CB-HB 18.011 1.32
K77 CE-CE## 41.814 2.997	F89 CA-HA 59.763 4.324
E78 CA-HA 56.633 4.264	F89 CB-HB2 39.428 2.611
D80 CA-HA 54.415 4.661	F89 CB-HB3 39.409 2.226
D80 CB-HB## 41.129 2.718	K90 CA-HA 59.18 3.7
D80 CB-HB2 41.081 2.729	K90 CB-HB2 32.741 1.991
D80 CB-HB3 41.038 2.687	K90 CB-HB3 32.64 1.877
S81 CA-HA 58.65 4.474	K90 CD-HD## 29.628 1.765
S81 CB-HB2 63.989 4.06	K90 CG-HG2 25.265 1.764
S81 CB-HB3 63.79 3.94	K90 CG-HG3 25.206 1.682
E82 CA-HA 59.02 4.018	V91 CA-HA 64.844 3.574
E82 CB-HB2 29.297 2.097	V91 CB-HB 31.772 1.726
E82 CB-HB3 29.306 2.07	V91 CG1-HG1 21.408 0.801
E82 CG-HG2 36.412 2.365	V91 CG2-HG2 20.515 0.421
E82 CG-HG3 36.417 2.266	F92 CA-HA 58.724 4.438
E83 CA-HA 59.49 3.977	F92 CB-HB2 39.368 3.514
L85 CA-HA 57.789 3.97	F92 CB-HB3 39.468 2.767
L85 CB-HB2 42.573 2.001	F92 CD1-HD1 131.629 7.274
L85 CB-HB3 42.557 1.56	F92 CE1-HE1 131.976 7.401
L85 CD1-HD1 24.388 0.89	D93 CA-HA 52.139 5.085
L85 CD2-HD2 23.75 0.869	D93 CB-HB2 39.778 3.332
L85 CG-HG2 27.42 1.76	D93 CB-HB3 39.772 2.559
I86 CA-HA 65.493 3.866	R94 CA-HA 57.935 4.101

R94 CD-HD##	42.91	3.274	L105 CB-HB2	41.77	1.696
D95 CA-HA	54.224	4.757	L105 CB-HB3	41.75	1.359
D95 CB-HB2	41.174	2.742	L105 CD1-HD1	25.349	
D95 CB-HB3	41.201	2.687	L105 CD2-HD2	24.757	0.167
G96 CA-HA2	46.848	3.932	L105 CG-HG	27.002	1.222
G96 CA-HA3	46.851	3.838	H106 CE-H32HE	137.655	8.193
N97 CA-HA	52.129	4.832	R106 CA-HA	60.12	3.663
N97 CB-HB2	38.905	3.073	R106 CB-HB3	29.621	1.482
N97 CB-HB3	38.939	2.672	R106 CD-HD2	42.986	3.19
G98 CA-HA3	45.381	3.74	R106 CD-HD3	42.922	3.033
G98 CA-HA3	45.865	3.292	R106 CG-HG3	29.144	1.484
L99 CA-HA	52.973	5.5	R106 CG-HG2?	29.207	1.713
L99 CB-HB2	44.812	1.797	H107 CA-HA	59.181	4.275
L99 CB-HB3	44.812	1.046	H107 CB-HB2	28.864	3.328
L99 CD1-HD1	25.469	0.738	H107 CB-HB3	28.858	3.14
L99 CD2-HD2	22.826	0.708	V108 CA-HA	66.35	3.505
L99 CG-HG2	26.267	1.384	V108 CB-HB	31.708	2.093
I100 CA-HA	59.281	5.016	V108 CG1-HG1	23.532	1.011
I100 CB-HB	41.977	2.204	V108 CG2-HG2	21.402	0.263
I100 CD-HD	14.411	0.71	M109 CA-HA	56.961	4.329
I100 CG1-HG12	25.584	1.52	M109 CB-HB2	31.296	2.258
I100 CG1-HG13	25.578	1.272	M109 CB-HB3	31.296	2.135
I100 CG2-HG2	17.996	0.903	M109 CE-HE	17.374	2.202
S101 CA-HA	57.526	4.922	M109 CG-HG2	32.859	2.753
S101 CB-HB2	65.023	4.322	M109 CG-HG3	32.852	2.673
S101 CB-HB3	65.02	3.988	T110 CA-HA	65.115	4.164
A102 CA-HA	55.483	3.788	T110 CB-HB	69.058	4.266
A102 CB-HB	18.068	1.344	T110 CG2-HG2	21.528	1.283
A103 CA-HA	54.944	3.992	N111 CA-HA	54.286	4.701
A103 CB-HB	18.187	1.337	N111 CB-HB2	39.08	2.821
E104 CA-HA	58.641	3.63	N111 CB-HB3	39.125	2.726
E104 CB-HB2	28.52	2.134	L112 CA-HA	55.384	4.345
E104 CB-HB3	28.522	1.225	L112 CB-HB2	42.892	1.783
E104 CG-HG2	37.255	2.04	L112 CB-HB3	42.887	1.651
E104 CG-HG3	37.25	1.827	L112 CD1-HD1	25.69	0.816
L105 CA-HA	59.169	3.779	L112 CD2-HD2	22.89	0.769

L112 CG-HG2 26.687 1.777	I125 CG2-HG2 17.145 0.845
G113 CA-HA2 46.011 3.933	L125 CG1-HG12 30.479 1.979
G113 CA-HA3 46.026 3.511	L125 CG1-HG13 30.447 0.774
L116 CA-HA 53.963 4.692	R126 CD-HD## 43.294 3.247
L116 CB-HB2 44.579 1.615	A128 CA-HA 52.147 4.421
L116 CB-HB3 44.593 1.551	A128 CB-HB 19.429 1.562
L116 CD1-HD1 26.492 0.888	D129 CB-HB2 39.988 2.982
L116 CD2-HD2 23.75 0.869	D129 CB-HB2 40.049 2.809
L116 CG-HG 27.617 1.628	D129 CB-HB3 40.049 2.74
T117 CA-HA 60.856 4.442	I130 CA-HA 62.644 4.161
T117 CB-HB 70.757 4.674	I130 CB-HB 38.806 1.945
T117 CG2-HG2 21.727 1.341	I130 CD-HD 13.725 0.9
D119 CA-HA 57.074 4.39	I130 CG1-HG12 27.454 1.521
D119 CB-HB2 40.334 2.671	I130 CG1-HG13 27.475 1.324
D119 CB-HB3 40.285 2.536	I130 CG2-HG2 17.999 1.035
E120 CB-HB2 30.254 2.41	D131 CA-HA 53.318 4.81
E120 CB-HB3 30.261 1.937	D131 CB-HB2 41.357 2.985
E120 CG-HG2 37.509 2.354	D131 CB-HB3 41.243 2.718
E120 CG-HG3 37.477 2.271	G132 CA-HA2 46.496 4.034
V121 CA-HA 67.25 3.683	G132 CA-HA3 46.496 3.768
V121 CB-HB 31.578 2.247	D133 CB-HB2 41.497 2.608
V121 CG1-HG1 24.35 1.079	D133 CB-HB3 41.529 2.517
V121 CG2-HG2 21.775 1.003	G134 CA-HA2 45.308 4.249
D122 CA-HA 57.763 4.369	H135 ?-? 119.032 7.047
D122 CB-HB2 40.291 2.853	H135 CA-HA 54.94 4.977
D122 CB-HB3 40.26 2.613	H135 CB-HB2 29.543 3.401
M124 CB-HB2 34.41 2.335	H135 CB-HB3 29.557 2.89
M124 CB-HG3 34.428 2.016	H135 CE-HE 135.857 8.536
M124 CE-HE 17.303 1.99	I136 CA-HA 59.509 4.269
M124 CG-HG2 32.631 2.788	I136 CB-HB 39.203 1.737
M124 CG-HG2 32.635 2.753	I136 CD-HD 12.541 0.652
M124 CG-HG3 32.546 2.439	I136 CG1-HG12 27.97 1.287
M124 CG-HG3 32.852 2.673	I136 CG1-HG13 27.977 1.161
I125 CA-HA 66.329 3.574	I136 CG2-HG2 18.174 0.605
I125 CB-HB 37.556 2.005	N137 CA-HA 52.269 5.155
I125 CD-HD 14.197 0.779	N137 CB-HB2 37.68 3.097

N137 CB-HB3 37.655 2.531	M144 CA-HA 58.32 4.033
Y138 CA-HA 59.34 4.092	M144 CB-HB2 32.76 2.113
Y138 CB-HB1 36.826 2.65	M144 CB-HB3 32.76 1.955
Y138 CB-HB2 36.775 2.302	M144 CE-HE 17.482 2.078
Y138 CD1-HD1 133.869 6.819	M144 CG-HG2 32.178 2.661
Y138 CE-HE1 117.708 6.768	M144 CG-HG3 32.21 2.447
E139 CA-HA 60.649 3.953	M145 CA-HA 58.762 3.875
E139 CB-HB2 28.668 2.209	M145 CB-HB2 33.11 2.071
E139 CB-HB3 28.738 2.005	M145 CB-HB3 33.035 1.943
E139 CG-HG3 37.118 2.093	M145 CE-HE 17.199 1.77
F141 CA-HA 61.997 4.011	M145 CG-HG2 32.139 2.355
F141 CB-HB2 39.654 3.1	M145 CG-HG3 32.132 2.26
F141 CB-HB3 39.623 2.761	V146 CA-HA 62.89 4.209
F141 CD1-HD1 131.395 6.872	V146 CB-HB 31.951 2.345
V142 CA-HA 67.249 3.516	V146 CG1-HG1 21.183 0.965
V142 CB-HB 31.689 2.332	V146 CG2-HG2 19.884 1
V142 CG1-HG1 25.319 1.382	S147 CA-HA 58.65 4.474
V142 CG2-HG2 22.474 1.141	S147 CB-HB2 63.989 4.06
R143 CA-HA 59.535 3.949	S147 CB-HB3 63.79 3.94

APPENDIX D

NMR ASSIGNMENTS OF APO PCAM₇₆₋₁₄₈ WHEN BOUND

TO TFP

Amide Assignments of Apo PCaM₇₆₋₁₄₈ When Bound to TFP

Listed below are amide assignments of apo PCaM₇₆₋₁₄₈ when bound to TFP determined in 10 mM D₄-imidazole, 100 mM KCl, 50 μM D₁₆-EDTA, 0.01% NaN₃, pH 6.5, 298.15 K

M76 N-H	128.617	8.46	M109 N-H	115.373	8.136
K77 N-H	123.08	8.523	T110 N-H	111.38	7.961
E78 N-H	122.524	8.6	N111 N-H	119.721	7.628
Q79 N-H	120.58	8.384	L112 N-H	119.595	7.708
D80 N-H	121.804	8.423	G113 N-H	107.539	8.163
S81 N-H	116.048	8.339	E114 N-H	119.757	7.972
E82 N-H	122.868	8.775	K115 N-H	120.826	8.405
E83 N-H	118.333	8.377	L116 N-H	122.02	7.793
E84 N-H	118.836	8.012	T117 N-H	114.12	9.055
L85 N-H	121.134	7.896	D118 N-H	121.143	8.775
I86 N-H	117.981	8.245	D119 N-H	117.067	8.255
E87 N-H	117.153	7.94	E120 N-H	120.779	7.737
A88 N-H	121.063	7.539	V121 N-H	120.522	8.167
F89 N-H	114.914	7.692	D122 N-H	119.364	8.365
K90 N-H	118.058	8.189	E123 N-H	119.488	7.811
V91 N-H	116.699	6.947	M124 N-H	118.796	7.843
F92 N-H	116.092	7.403	I125 N-H	118.633	8.442
D93 N-H	121.481	7.731	R126 N-H	118.474	7.792
R94 N-H	124.216	8.25	E127 N-H	116.551	7.724
D95 N-H	116.274	8.652	A128 N-H	120.665	7.718
G96 N-H	110.401	7.919	I130 N-H	123.24	8.181
N97 N-H	118.007	8.802	G132 N-H	108.708	8.232
G98 N-H	110.98	9.824	D133 N-H	119.653	8.362
L99 N-H	119.42	7.623	H135 N-H	119.098	8.204
I100 N-H	114.468	8.639	I136 N-H	121.575	9.187
S101 N-H	118.991	8.887	N137 N-H	125.286	8.753
A102 N-H	124.244	8.499	Y138 N-H	124.855	7.475
A103 N-H	118.712	8.283	E139 N-H	125.223	8.318
E104 N-H	119.823	7.641	E140 N-H	119.211	7.772
L105 N-H	121.203	8.158	F141 N-H	120.623	8.086
R106 N-H	117.825	8.227	V142 N-H	118.542	8.421
H107 N-H	118.841	7.713	R143 N-H	117.988	7.77
V108 N-H	119.781	7.988	M144 N-H	117.637	7.859

M145 N-H	118.995	7.964	S147 N-H	116.306	7.575
V146 N-H	113.306	7.992	K148 N-H	127.771	7.457

Carbonyl Assignments of apo PCaM₇₆₋₁₄₈ when Bound to TFP

Listed below carbonyl assignments of apo PCaM₇₆₋₁₄₈ when bound to TFP determined in
10 mM D₄-imidazole, 100 mM KCl, 50 μM D₁₆-EDTA, 0.01% NaN₃, pH 6.5, 298.15 K

M76	175.841	N111	175.673
K77	176.502	L112	177.628
E78	176.447	G113	174.646
Q79	175.639	E114	176.201
D80	176.227	K115	176.116
S81	175.209	L116	177.555
E82	177.4	T117	175.351
E83	178.681	D118	178.095
E84	179.204	D119	178.791
I86	178.026	E120	179.414
I86	178.099	V121	177.565
E87	178.458	D122	179.156
A88	179.766	E123	178.276
F89	177.72	M124	178.657
K90	178.513	I125	177.484
V91	176.734	R126	179.07
F92	174.885	E127	176.963
D93	176.538	D129	176.605
R94	177.91	D131	177.524
D95	176.782	G132	174.441
G96	175.014	G134	174.328
N97	176.314	H135	172.82
G98	173.079	I136	174.796
L99	176.477	N137	175.551
I100	174.584	Y138	176.227
A102	175.372	E139	179.025
A102	178.988	E140	178.621
A103	181.046	F141	176.843
E104	177.788	V142	177.269
L105	178.818	R143	179.099
R106	178.429	M144	179.156
H107	177.489	M144	178.197
V108	178.669	M145	178.188
M109	178.45	V146	176.502
T110	175.745	S147	173.534

Methine, Methylene, and Methyl Assignments of apo PCaM₇₆₋₁₄₈

When Bound to TFP

Below are listed methine, methylene, and methyl assignments of apo PCaM₇₆₋₁₄₈ when bound to TFP determined in 10 mM D₄-imidazole, 100 mM KCl, 50 μM D₁₆-EDTA, 0.01% NaN₃, pH 6.5, 298.15 K

M76 CA-HA	54.05	4.527	L85 CG-HG2	27.012	1.634
M76 CB-HB2	33.221	2.084	I86 CA-HA	65.55	3.854
M76 CB-HB3	33.148	1.996	I86 CB-HB	37.352	2.164
M76 CE-HE	17.427	1.948	I86 CD-HD	12.766	1.049
M76 CG-HG2	31.829	2.579	I86 CG1-HG12	30.024	2.01
M76 CG-HG3	31.832	2.536	I86 CG1-HG13	30.169	1.361
K77 CA-HA	56.219	4.348	I86 CG2-HG2	17.798	1.152
K77 CB-HB2	33.165	1.84	E87 CA-HA	59.08	3.924
K77 CB-HB3	33.179	1.77	E87 CB-HB##	29.131	2.085
K77 CD-HD##	28.884	1.685	E87 CG-HG##	36.242	2.382
K77 CE-K3HE##	41.838	2.989	A88 CA-HA	54.524	4.097
K77 CG-HG##	24.428	1.441	A88 CB-HB	17.96	1.33
E78 CA-HA	56.727	4.246	F89 CA-HA	59.668	4.319
E78 CB-HB2	29.859	2.066	F89 CB-HB2	39.348	2.63
E78 CB-HB3	30.061	1.948	F89 CB-HB3	39.323	2.208
E78 CG-HG2	36.213	2.434	F89 CD1-HD1	131.406	7.132
E78 CG-HG3	36.188	2.289	F89 CD2-HD2	60.378	7.094
Q79 CA-HA	56.075	4.318	F89 CE##-HE##	129.423	7.132
Q79 CB-HB##	29.478	2.031	F89 CE##-HE##	60.096	7.199
Q79 CG-HG2	33.766	2.366	K90 CA-HA	59.193	3.726
Q79 CG-HG3	33.699	2.106	K90 CB-HB2	32.643	1.989
D80 CA-HA	54.432	4.644	K90 CB-HB3	32.724	1.9
D80 CB-HB##	41.178	2.696	K90 CD-HD##	29.595	1.768
S81 CA-HA	58.741	4.469	K90 CE-HE2	41.75	3.047
S81 CB-HB2	63.949	4.032	K90 CE-HE3	41.719	3
S81 CB-HB3	63.982	3.938	K90 CG-HG2	25.299	1.77
E82 CA-HA	59.273	4.028	K90 CG-HG3	25.286	1.68
E82 CB-HB2	29.626	2.097	V91 CA-HA	64.841	3.581
E82 CG-HG2	36.45	2.368	V91 CB-HB	31.777	1.743
E82 CG-HG3	36.416	2.259	V91 CG1-HG1	21.423	0.802
E83 CA-HA	59.162	4.054	V91 CG2-HG2	20.51	0.427
L85 CA-HA	57.85	3.962	F92 CA-HA	58.753	4.401
L85 CB-HB2	42.323	1.9	F92 CB-HB2	39.405	3.5
L85 CB-HB3	42.302	1.556	F92 CB-HB2	39.426	2.779
L85 CD1-HD1	25.277	0.72	F92 CD##-HD##	131.635	7.269
L85 CD2-HD2	24.399	0.769	F92 CE##-HE##	131.681	7.335

D93 CA-HA	52.144	5.071	L105 CD1-HD1	25.113	0.221
D93 CB-HB2	39.786	3.319	L105 CD2-HD2	25.124	0.041
D93 CB-HB3	39.818	2.561	L105 CG-HG2	26.886	1.27
R94 CA-HA	57.972	4.102	R106 CA-HA	60.055	3.666
R94 CB-HB##	30.156	1.907	R106 CB-HB2	29.963	1.786
R94 CD-HD##	42.932	3.27	R106 CD-HD2	42.894	3.203
R94 CG-HG2	27.207	1.775	R106 CD-HD3	42.919	3.033
R94 CG-HG3	27.205	1.707	R106 CG-CG3	28.816	1.51
D95 CA-HA	54.148	4.759	R106 CG-HG2	28.807	1.615
D95 CB-HB2	41.045	3.017	H107 CA-HA	59.23	4.242
D95 CB-HB3	41.118	2.737	H107 CB-HB2	29.084	3.314
G96 CA-HA2	46.798	3.93	H107 CB-HB3	29.09	3.127
G96 CA-HA3	46.816	3.834	H107 CD2-HD2	119.691	6.954
N97 CA-HA	52.183	4.824	H108 CE1-HE1	137.785	8.139
N97 CB-HB2	38.929	3.065	V108 CA-HA	66.38	3.498
N97 CB-HB3	38.906	2.675	V108 CB-HB	31.774	2.083
G98 CA-HA2	45.866	3.927	V108 CG1-HG1	23.456	1.003
G98 CA-HA3	45.827	3.29	V108 CG2-HG2	21.278	0.27
S98 CA-HA	57.537	4.879	M109 CB-HB2	31.774	2.229
L99 CA-HA	53.013	5.469	M109 CB-HB3	31.764	2.128
L99 CB-HB2	44.722	1.77	M109 CE-HE	17.355	2.137
L99 CB-HB3	44.724	1.031	M109 CG-HG##	32.845	2.707
L99 CD1-HD1	25.402	0.732	T110 CA-HA	64.705	4.192
L99 CD2-HD2	22.79	0.701	T110 CB-HB	69.159	4.303
L99 CG-HG2	26.291	1.378	T110 CG2-HG2	21.588	1.283
I100 CA-HA	59.292	5.008	N111 CA-HA	54.298	4.751
I100 CB-HB	41.967	2.213	N111 CB-HB1	39.202	2.812
I100 CD-HD	14.397	0.707	N111 CB-HB2	39.109	2.675
I100 CG1-HG12	25.629	1.523	L112 CA-HA	55.354	4.349
I100 CG1-HG13	25.549	1.263	L112 CB-HB2	42.857	1.798
I100 CG2-HG2	17.982	0.908	L112 CB-HB3	42.84	1.639
S101 CB-HB2	65	4.31	L112 CD1-HD1	25.721	0.803
S101 CB-HB3	64.995	3.97	L112 CD2-HD2	22.901	0.742
A102 CA-HA	55.478	3.802	L112 CG-HG2	26.631	1.801
A102 CB-HB	17.787	1.355	G113 CA-HA1	45.664	3.896
A103 CA-HA	54.927	3.999	K115 CA-HA	56.504	4.247
A103 CB-HB	18.033	1.36	K115 CB-HB2	32.153	1.858
E104 CA-HA	58.644	3.656	K115 CB-HB3	32.105	1.778
E104 CB-HB2	28.574	2.152	K115 CD-HD##	28.873	1.669
E104 CB-HB3	28.498	1.27	K115 CE-HE2	41.838	2.989
E104 CB-HB3	28.15	1.167	K115 CE-HE3	41.833	2.949
E104 CG-HG2	37.234	2.064	K115 CG-HG2	24.671	1.44
E104 CG-HG3	37.239	1.843	K115 CG-HG3	24.735	1.378
L105 CA-HA	59.033	3.806	L116 CA-HA	53.885	4.687
L105 CB-HB2	41.681	1.737	L116 CB-HB2	44.468	1.582
L105 CB-HB3	41.734	1.346	L116 CB-HB3	44.481	1.535

L116 CD1-HD1	26.391	0.841	I130 CA-HA	62.093	4.185
L116 CD2-HD2	23.841	0.819	I130 CB-HB	38.854	1.879
L116 CG-HG2	27.508	1.605	I130 CD-HD	13.611	0.876
T117 CA-HA	60.749	4.454	I130 CG1-HG12	27.506	1.512
T117 CB-HB	70.798	4.674	I130 CG1-HG13	27.509	1.208
T117 CG2-HG2	21.722	1.33	I130 CG2-HG2	17.76	1.003
D118 CA-HA	57.565	4.261	D131 CA-HA	53.083	4.817
D118 CB-HB2	39.918	2.732	D131 CB-HB2	41.471	2.6
D118 CB-HB3	39.97	2.604	D131 CB-HB3	41.559	2.535
D119 CA-HA	57.077	4.388	G132 CA-HA2	45.958	3.927
D119 CB-HB2	40.284	2.843	G132 CA-HA3	46.512	3.807
D119 CB-HB2	40.328	2.668	G134 CA-HA2	45.958	3.971
D119 CB-HB3	40.136	2.609	G134 CA-HA3	45.93	3.539
E120 CA-HA	58.935	4.061	H135 CA-HA	54.892	5.046
E120 CB-HB2	30.186	2.402	H135 CB-HB2	30.039	3.439
E120 CB-HB3	30.188	1.918	H135 CB-HB3	30.044	2.895
E120 CG-HG2	37.51	2.349	H135 CD2-HD2	48.029	6.972
E120 CG-HG3	37.443	2.272	H135 CE-HE	136.126	8.512
V121 CA-HA	67.13	3.657	I136 CA-HA	59.621	4.29
V121 CB-HB	31.704	2.213	I136 CB-HB	40.137	1.653
V121 CG1-HG1	24.239	1.046	I136 CD-HD	13.201	0.635
V121 CG2-HG2	21.912	0.982	I136 CG1-HG13	28.187	1.231
D122 CA-HA	57.741	4.36	I136 CG2-HG2	17.917	0.601
D122 CB-HB2	40.106	2.857	N137 CA-HA	52.235	5.168
D122 CB-HB3	40.136	2.609	N137 CB-HB2	37.643	3.069
M124 CA-HA	59.365	4.099	N137 CB-HB3	37.608	2.53
M124 CB-HB2	34.19	2.352	Y138 CA-HA	59.328	4.09
M124 CB-HB3	34.195	2.046	Y138 CB-HB1	36.888	2.621
M124 CE-HE	17.178	1.989	Y138 CD1-HD1	133.914	6.811
M124 CG-HG2	32.69	2.783	Y138 CE##-HE##	117.731	6.757
M124 CG-HG3	32.694	2.454	E139 CA-HA	60.62	3.94
I125 CA-HA	65.855	3.586	E139 CB-HB##	28.807	1.993
I125 CB-HB	37.491	1.983	E139 CG-HG2	37.123	2.221
I125 CD-HD	13.894	0.764	E139 CG-HG3	37.112	2.094
I125 CG1-HG12	30.15	1.898	E140 CA-HA	58.264	4.051
I125 CG2-HG2	17.166	0.839	E140 CB-HB2	29.27	2.157
L125 CG1-HG13	30.109	0.836	E140 CB-HB3	29.27	2.113
R126 CA-HA	59.194	4.081	E140 CG-HG2	35.675	2.306
R126 CB-HB##	30.074	1.948	E140 CG-HG3	35.675	2.252
R126 CD-HD##	43.305	3.238	F141 CA-HA	61.791	4.085
R126 CG-HG2	27.782	1.8	F141 CB-HB1	39.512	3.077
R126 CG-HG3	27.801	1.629	F141 CB-HB2	39.655	2.876
A128 CA-HA	52.218	4.38	F141 CD##-HD##	131.425	6.935
A128 CB-HB	19.468	1.504	E142 CG-HG	36.238	2.317
D129 CB-HB2	40.003	2.938	V142 CA-HA	67.25	3.456
D129 CB-HB3	40.01	2.583	V142 CB-HB	31.77	2.306

V142 CG1-HG1 24.997 1.342
V142 CG2-HG2 22.313 1.091
R143 CA-HA 59.459 3.938
R143 CB-HB## 29.468 1.909
R143 CD-HD## 43.046 3.19
R143 CG-HG1 27.782 1.8
R143 CG-HG2 27.618 1.766
R143 CG-HG3 27.624 1.609
M144 CE-HE 16.911 2.076
M145 CA-HA 58.264 4.051
M145 CA-HA 58.156 4.048
M145 CB-HB2 32.523 2.122

M145 CB-HB3 32.523 2.008
M145 CE-HE 17.37 1.796
M145 CG-HG2 31.941 2.647
M145 CG-HG3 31.949 2.436
V146 CA-HA 62.975 4.176
V146 CB-HB 31.77 2.306
V146 CG1-HG1 21.169 0.94
V146 CG2-HG2 20.073 0.979
S147 CA-HA 58.637 4.474
S147 CB-HB2 63.801 4.03
S147 CB-HB3 63.816 3.938

APPENDIX E

NMR ASSIGNMENTS OF APO PCaM₇₆₋₁₄₈ WHEN BOUND TONa_v1.2_{IQP}Amide Assignments of apo PCaM₇₆₋₁₄₈ When Bound to Na_v1.2_{IQP}

Below are amide assignments of apo PCaM₇₆₋₁₄₈ when bound to Na_v1.2_{IQP} determined in

10 mM D₄-imidazole, 100 mM KCl, 50 μM D₁₆-EDTA, 0.01% NaN₃, pH 6.5, 298.15 K

K 77 N-H	123.132	8.744	N 111 N-H	116.488	7.438
E 78 N-H	121.824	8.518	L 112 N-H	119.883	7.745
Q 79 N-H	120.369	8.377	G 113 N-H	106.167	9.078
D 80 N-H	121.445	8.438	E 114 N-H	129.78	9.331
S 81 N-H	115.689	8.449	K 115 N-H	115.77	6.844
E 82 N-H	123.796	8.922	L 116 N-H	122.432	8.509
E 83 N-H	117.756	8.642	T 117 N-H	112.002	9.347
E 84 N-H	118.775	7.975	D 118 N-H	120.312	8.719
L 85 N-H	120.846	7.926	D 119 N-H	116.357	8.197
I 86 N-H	120.37	8.675	E 120 N-H	119.881	7.804
E 87 N-H	117.231	8.024	V 121 N-H	118.933	8.731
A 88 N-H	121.043	7.674	D 122 N-H	123.075	8.332
F 89 N-H	114.809	7.994	E 123 N-H	120.621	7.397
K 90 N-H	117.643	8.254	M 124 N-H	121.983	7.993
V 91 N-H	118.519	6.948	I 125 N-H	115.727	8.297
F 92 N-H	116.264	7.434	R 126 N-H	119.297	7.785
D 93 N-H	120.522	7.526	E 127 N-H	115.314	7.869
R 94 N-H	124.435	8.342	A 128 N-H	119.463	7.973
D 95 N-H	116.533	8.676	D 129 N-H	119.037	8.36
G 96 N-H	110.326	8.187	I 130 N-H	122.603	7.831
N 97 N-H	117.559	8.694	D 131 N-H	119.049	8.341
G 98 N-H	110.822	9.773	I 136 N-H	121.347	9.048
L 99 N-H	119.499	7.808	N 137 N-H	127.016	9.152
I 100 N-H	116.548	8.705	Y 138 N-H	123.19	7.193
S 101 N-H	121.978	9.084	E 139 N-H	125.718	8.437
A 102 N-H	129.122	8.793	E 140 N-H	117.759	7.417
A 103 N-H	119.588	8.441	V 142 N-H	118.912	8.578
E 104 N-H	120.377	7.773	R 143 N-H	117.124	7.715
L 105 N-H	119.693	8.467	M 144 N-H	119.249	7.771
R 106 N-H	117.512	8.445	M 145 N-H	117.727	7.857
H 107 N-H	118.473	7.72	V 146 N-H	111.161	7.993
V 108 N-H	117.819	8.588	S 147 N-H	116.623	7.633
M 109 N-H	112.375	8.307	K 148 N-H	128.318	7.839
T 110 N-H	103.758	7.448			

Carbonyl Assignments of apo CaM₇₆₋₁₄₈ When Bound to Na_v1.2_{IQP}

Carbonyl assignments of apo CaM₇₆₋₁₄₈ when bound to Na_v1.2_{IQP} determined in 10 mM

D₄-imidazole, 100 mM KCl, 50 μM D₁₆-EDTA, 0.01% NaN₃, pH 6.5, 298.15 K

K77	176.74	M109	178.198
E78	176.252	T110	174.894
Q79	175.402	N111	173.557
D80	175.692	L112	176.581
S81	175.062	G113	175.888
E82	177.566	E114	175.378
E83	178.959	K115	177.848
E84	179.375	L116	177.488
L85	178.762	T117	175.363
I86	177.765	D118	177.668
E87	178.713	D119	179.624
A88	178.924	E120	178.985
F89	178.68	V121	176.25
K90	179.155	D122	179.494
V91	176.29	E123	177.482
F92	174.239	M124	177.552
D93	176.327	I125	177.429
R94	178.028	R126	178.819
D95	176.859	E127	177.069
G96	174.912	D129	176.551
N97	176.096	H135	172.991
G98	172.933	I136	175.003
L99	176.203	N137	175.653
I100	174.971	Y138	176.121
S101	175.564	E139	178.662
A102	178.755	F141	177.211
A103	180.892	V142	177.111
E104	178.533	R143	178.945
L105	177.97	M144	178.357
R106	178.451	M145	178.73
H107	177.808	V146	176.602
V108	177.786	S147	173.944

Methine, Methyline, and Methyl Assignments of apo PCaM₇₆₋₁₄₈

When Bound to Na_v1.2_{IQp}

Methine, methyline, and methyl assignments of apo PCaM₇₆₋₁₄₈ when bound to Na_v1.2_{IQp} determined in 10 mM D₄-imidazole, 100 mM KCl, 50 μM D₁₆-EDTA, 0.01% NaN₃, pH 6.5, 298.15 K

M 76 CE-HE	16.854	2.102	I 86 CG2-HG2	17.788	1.131
K 77 CA-HA	56.437	4.347	E 87 CA-HA	59.354	3.902
K 77 CB-CB##	32.946	1.793	A 88 CA-HA	54.888	4.136
K 77 CD-HD##	29.12	1.734	A 88 CB-HB	18.184	1.55
K 77 CE-HE##	41.915	3.008	F 89 CA-HA	62.843	3.921
K 77 CG-HG##	24.663	1.438	F 89 CB-HB2	40.359	2.647
E 78 CA-HA	56.894	4.249	F 89 CB-HB3	40.343	1.773
E 78 CB-HB2	29.972	2.064	F 89 CD##-HD##	61.177	7.182
E 78 CB-HB3	30.005	1.938	F 89 CE##-HE##	59.427	7.07
E 78 CG-HG##	36.455	2.296	K 90 CA-HA	59.354	3.902
Q 79 CA-HA	56.126	4.356	K 90 CA-HA	59.16	3.893
Q 79 CB-HB2	29.972	2.064	K 90 CB-HB2	32.611	2.006
Q 79 CB-HB3	30.025	1.989	K 90 CB-HB3	32.606	1.876
Q 79 CG-HG##	33.793	2.345	K 90 CD-HD##	29.567	1.741
D 80 CA-HA	54.443	4.651	K 90 CE-HE2	41.858	3.033
D 80 CB-HB##	41.665	2.651	K 90 CE-HE3	41.859	2.97
S 81 CA-HA	58.02	4.525	K 90 CG-HG2	25.976	1.848
S 81 CB-HB2	64.513	4.125	K 90 CG-HG3	25.987	1.706
S 81 CB-HB3	64.523	3.935	V 91 CA-HA	65.174	3.458
E 82 CB-HB##	29.848	2.085	V 91 CB-HB	31.377	1.693
E 83 CA-HA	60.255	3.991	V 91 CG2-HG1	22.725	0.789
E 83 CB-HB##	29.327	2.103	V 91 CG3-HG3	20.718	0.162
E 83 CG-HG##	36.512	2.357	F 92 CA-HA	57.585	4.539
E 84 CA-HA	58.896	4.018	F 92 CB-HB2	39.169	3.512
E 84 CB-HB##	29.791	2.1	F 92 CB-HB3	39.219	2.774
E 84 CG-HG##	36.93	2.281	F 92 CD##-HD##	59.97	6.91
L 85 CA-HA	57.737	4.097	F 92 CE##-HE	58.473	7.353
L 85 CB-HB2	43.154	2.059	D 93 CA-HA	52.167	4.97
L 85 CB-HB3	43.151	1.612	D 93 CB-HB2	40.159	3.411
L 85 CD1-HD1	25.76	0.873	D 93 CB-HB3	40.139	2.546
L 85 CD2-HD2	25.097	0.864	R 94 CA-HA	58.078	4.109
L 85 CG-HG2	27.23	1.845	R 94 CB-HB##	30.031	1.942
I 86 CA-HA	66.046	3.877	R 94 CD-HD##	42.962	3.252
I 86 CB-HB	37.174	2.177	R 94 CG-HG##	27.016	1.831
I 86 CD-HD	12.797	1.003	D 95 CA-HA	54.664	4.793
I 86 CG1-HG12	30.389	2.047	D 95 CB-HB##	41.447	2.741
I 86 CG1-HG13	30.461	1.27	G 96 CA-HA2	46.658	3.956

G 96 CA-HA3	46.768	3.758	M 109 CA-HA	56.224	4.312
N 97 CA-HA	52.256	4.886	M 109 CB-HB2	36.698	2.334
N 97 CB-HB2	39.263	2.983	M 109 CB-HB3	36.752	1.628
N 97 CB-HB3	39.285	2.702	M 109 CE-HE	20.615	1.84
G 98 CA-HA2	45.652	3.957	M 109 CG-HG2	33.867	2.749
G 98 CA-HA3	45.685	3.228	M 109 CG-HG3	33.86	2.443
L 99 CA-HA	52.704	5.505	T 110 CA-HA	61.828	4.563
L 99 CB-HB2	45.339	1.725	T 110 CB-HB	70.765	4.349
L 99 CB-HB3	45.336	1.047	T 110 CG-HG2	21.743	1.207
L 99 CD1-HD1	22.895	0.683	N 111 CA-HA	54.695	4.881
L 99 CD2-HD2	25.365	0.646	N 111 CB-HB2	41.136	2.393
L 99 CG-HG2	26.453	1.334	N 111 CB-HB3	41.13	2.267
I 100 CA-HA	59.239	4.786	L 112 CA-HA	53.388	4.775
I 100 CB-HB	42.667	1.788	L 112 CB-HB2	45.617	1.861
I 100 CD-HD	15.572	0.92	L 112 CB-HB3	45.614	1.272
I 100 CG2-HG2	17.409	1.075	L 112 CD##-HD##	23.544	0.911
S 101 CA-HA	57.719	4.697	L 112 CG-HG2	27.676	1.622
S 101 CB-HB2	63.587	4.262	G 113 CA-HA2	46.878	3.863
S 101 CB-HB3	63.625	4.008	G 113 CA-HA3	46.986	3.771
A 102 CA-HA	55.492	3.843	E 114 CA-HA	54.797	4.299
A 102 CB-HB	18.18	1.328	E 114 CB-HB2	26.529	1.799
A 103 CA-HA	55.114	4.039	E 114 CB-HB3	26.453	1.741
A 103 CB-HB	18.146	1.382	E 114 CG-HG##	35.975	2.223
E 104 CA-HA	59.076	4.179	K 115 CA-HA	57.701	3.87
E 104 CB-HB##	29.503	2.144	K 115 CB-HB##	33.649	1.485
E 104 CG-HG##	36.222	2.349	K 115 CD-HD2	29.053	1.699
L 105 CA-HA	57.623	3.823	K 115 CD-HD3	29.248	1.547
L 105 CB-HB2	41.213	1.816	K 115 CE-HE2	41.715	2.91
L 105 CB-HB3	41.228	1.334	K 115 CE-HE3	41.693	2.861
L 105 CD1-HD1	22.555	0.806	K 115 CG-HG2	24.579	1.1
L 105 CD2-HD2	25.839	0.712	K 115 CG-HG3	24.449	0.871
L 105 CG2-HG2	26.702	1.375	L 116 CA-HA	54.279	4.627
R 106 CA-HA	59.785	3.606	L 116 CB-HB2	41.983	1.81
R 106 CB-HB##	29.986	1.826	L 116 CB-HB3	42.084	1.354
R 106 CD-HD2	42.751	3.31	L 116 CD1-HD1	21.997	1.014
R 106 CD-HD3	42.751	3.029	L 116 CD2-HD2	27.256	0.856
R 106 CG-HG2	28.302	1.556	L 116 CG-HG2	27.686	1.975
H 107 CA-HA	60.58	4.017	T 117 CA-HA	60.687	4.362
H 107 CB-HB2	30.409	3.361	T 117 CB-HB	70.956	4.725
H 107 CB-HB3	30.398	3.132	T 117 CG-HG2	21.772	1.328
H 107 CD2-HD2	48.793	6.917	D 118 CA-HA	57.84	4.305
H 107 CE-HE1	67.426	7.724	D 118 CB-HB2	40.455	2.67
V 108 CA-HA	66.455	3.494	D 118 CB-HB3	40.335	2.538
V 108 CB-HB	31.887	1.754	D 119 CA-HA	57.282	4.408
V 108 CG2-HG2	20.947	0.711	D 119 CB-HB2	40.455	2.67
V 108 CG3-HG3	21.421	0.575	D 119 CB-HB3	40.481	2.531

E 120 CB-HB2	32.048	2.611	G 134 CA-HA3	45.766	3.499
E 120 CB-HB3	32.073	1.924	H 135 CA-HA	54.87	5.06
V 121 CA-HA	64.698	3.836	H 135 CB-HB2	31.026	3.41
V 121 CB-HB	31.288	1.87	H 135 CB-HB3	31.125	2.853
V 121 CG2-HG2	21.027	1.052	H 135 CD2-HD2	49.618	6.895
V 121 CG3-HG3	24.696	1.003	H 135 CE-HE1	66.418	8.216
D 122 CA-HA	57.282	4.408	I 136 CA-HA	60.468	4.136
D 122 CB-HB2	39.829	2.915	I 136 CB-HB	40.203	1.873
D 122 CB-HB3	40.005	2.592	I 136 CD-HD	14.375	0.569
E 123 CA-HA	63.329	4.182	I 136 CG1-HG12	28.119	1.331
E 123 CB-HB2	28.887	2.377	I 136 CG1-HG13	28.143	0.848
E 123 CB-HB3	28.893	2.205	I 136 CG2-HG2	16.174	0.508
E 123 CG-HG##	35.453	2.509	N 137 CA-HA	52.167	4.97
M 124 CA-HA	59.792	3.825	N 137 CB-HB2	36.93	3.131
M 124 CB-HB2	33.536	1.925	N 137 CB-HB3	36.932	2.591
M 124 CB-HB3	33.692	1.668	Y 138 CA-HA	59.116	4.014
M 124 CE-HE	18.39	1.742	Y 138 CB-HB2	36.825	2.179
M 124 CG-HG2	32.28	1.844	Y 138 CB-HB3	37.076	2.021
M 124 CG-HG3	32.317	1.627	Y 138 CD##-HD##	62.918	6.645
I 125 CA-HA	63.412	3.383	Y 138 CE##-HE##	46.864	6.749
I 125 CB-HB	36.287	1.99	E 139 CA-HA	60.879	3.82
I 125 CD-HD	11.069	0.658	E 139 CB-HB##	28.866	1.953
I 125 CG1-HG1##	28.79	1.423	E 139 CG-HG##	36.825	2.179
I 125 CG2-HG2	18.1	0.869	F 141 CA-HA	59.991	4.585
R 126 CA-HA	59.299	4.009	F 141 CB-HB2	39.273	3.317
R 126 CD-HD2	43.24	3.21	F 141 CB-HB3	39.097	2.767
R 126 CD-HD3	43.248	3.117	F 141 CD##-HD##	59.734	7.13
E 127 CA-HA	58.126	4.176	F 141 CE##-HE##	57.622	7.053
E 127 CB-HB2	30.883	2.332	V 142 CA-HA	67.664	3.262
E 127 CB-HB3	30.852	2.232	V 142 CB-HB	31.555	2.278
E 127 CG-HG2	37.772	2.698	V 142 CG2-HG2	25.806	1.168
E 127 CG-HG3	37.758	2.334	V 142 CG3-HG3	22.962	1.101
A 128 CA-HA	53.076	3.968	R 143 CA-HA	57.999	4.162
A 128 CB-HB	18.732	0.593	R 143 CA-HA	60.068	4.16
D 129 CA-HA	52.572	4.757	R 143 CD-HD##	41.915	3.01
D 129 CB-HB2	39.339	2.919	R 143 CG-HG##	25.226	1.528
D 129 CB-HB3	39.412	2.234	M 144 CA-HA	58.905	4.018
I 130 CA-HA	62.973	3.967	M 144 CB-HB2	31.693	2.375
I 130 CB-HB	38.73	1.81	M 144 CB-HB3	31.665	2.176
I 130 CD-HD	13.605	0.906	M 144 CE-HE	16.9	1.929
I 130 CG1-HG12	27.904	1.562	M 144 CG-HG2	31.514	2.76
I 130 CG1-HG13	27.796	1.137	M 144 CG-HG3	31.665	2.176
I 130 CG2-HG2	17.571	0.953	M 145 CA-HA	59.428	3.71
D 133 CB-HB2	41.681	2.741	M 145 CB-HB2	34.527	1.931
D 133 CB-HB3	41.764	2.586	M 145 CB-HB3	34.549	1.652
G 134 CA-HA2	45.769	4.075	M 145 CE-HE	16.694	1.713

M 145 CG-HG2	31.426	1.802	V 146 CG3-HG3	21.369	0.902
M 145 CG-HG3	31.432	1.363	S 147 CA-HA	59.314	4.462
V 146 CA-HA	62.485	4.296	S 147 CB-HB2	64.113	4.019
V 146 CB-HB	31.693	2.375	S 147 CB-HB3	63.903	3.849
V 146 CG2-HG2	19.812	0.953			

Intramolecular NOE Assignments of Na_v1.2_{IQP}

Intramolecular NOE assignments of Na_v1.2_{IQP} determined in 10 mM D₄-imidazole, 100

mM KCl, 50 μM D₁₆-EDTA, 0.01% NaN₃, pH 6.5, 298.15 K

Q1904HA-Q1904HB#	4.06	2.108	Q1904HN-E1905HN	8.382	8.785
Q1904HA-E1905HN	4.06	8.783	Q1904HN-Q1904HB#	8.372	2.098
Q1904HA-Q1904HN	4.06	8.387	E1905HA-A1909HN	3.713	8.36
Q1904HA-V1907HN	4.063	7.776	E1905HA-E1906HN	3.714	7.905
Q1904HA-Q1904HG#	4.061	2.315	E1905HA-S1908HN	3.713	8.087
Q1904HA-Q1904HG#	4.057	2.044	E1905HA-E1905HG#	3.713	2.227
Q1904HA-Q1904HB#	4.062	1.932	E1905HA-E1905HN	3.713	8.782
Q1904HA-V1907HG##	4.063	1.064	E1905HA-E1905HG#	3.715	2.294
Q1904HA-V1907HG##	4.073	0.966	E1905HA-E1905HB#	3.714	2.054
Q1904HB#-Q1904HN	1.935	8.395	E1905HB#-E1905HA	2.055	3.712
Q1904HB#-Q1904HN	2.099	8.369	E1905HB#-E1905HN	2.056	8.783
Q1904HB#-Q1904HA	2.108	4.055	E1905HB#-E1905HG#	2.058	2.285
Q1904HB#-Q1904HA	1.93	4.061	E1905HB#-E1905HG#	2.058	2.219
Q1904HB#-Q1904HE2#	1.938	6.824	E1905HG#-E1905HA	2.213	3.709
Q1904HB#-Q1904HE2#	1.939	6.932	E1905HG#-E1905HA	2.284	3.713
Q1904HB#-Q1904HG#	2.109	2.312	E1905HG#-E1906HA	2.27	3.954
Q1904HB#-Q1904HG#	2.109	2.048	E1905HG#-E1905HN	2.232	8.782
Q1904HB#-Q1904HG#	1.932	2.313	E1905HG#-E1905HN	2.271	8.784
Q1904HB#-Q1904HG#	1.932	2.048	E1905HG#-E1905HB#	2.222	2.055
Q1904HE2#-Q1904HB#	6.825	1.942	E1905HG#-E1905HB#	2.274	2.053
Q1904HE2#-Q1904HE2#	7.364	7.156	E1905HN-Q1904HN	8.782	8.387
Q1904HE2#-Q1904HG#	6.827	2.031	E1905HN-E1905HA	8.784	3.712
Q1904HG#-Q1904HN	2.315	8.381	E1905HN-E1905HG#	8.783	2.213
Q1904HG#-Q1904HA	2.324	4.066	E1905HN-E1905HG#	8.785	2.29
Q1904HG#-Q1904HA	2.053	4.066	E1905HN-E1906HN	8.784	7.903
Q1904HG#-Q1904HN	2.058	8.377	E1905HN-E1905HB#	8.784	2.058
Q1904HG#-Q1904HB#	2.049	2.108	E1906HA-E1906HN	3.955	7.897
Q1904HG#-Q1904HB#	2.314	2.108	E1906HB#-V1907HN	2.269	7.79
Q1904HG#-Q1904HB#	2.048	1.935	E1906HB#-E1906HN	2.212	7.904
Q1904HG#-Q1904HB#	2.322	1.933	E1906HB#-E1906HN	2.284	7.901
Q1904HG#-Q1904HE2#	2.019	6.937	E1906HB#-E1906HG#	2.279	2.332
Q1904HG#-Q1904HE2#	2.031	6.824	E1906HB#-E1906HG#	2.218	2.331
Q1904HN-Q1904HB#	8.381	1.934	E1906HG#-E1906HA	2.338	3.964
Q1904HN-Q1904HG#	8.38	2.314	E1906HG#-E1906HN	2.342	7.913
Q1904HN-Q1904HG#	8.371	2.058	E1906HG#-E1906HA	2.214	3.951

E1906HG#-E1906HB# 2.335 2.216
 E1906HG#-E1906HB# 2.336 2.281
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 E1906HN-E1906HB# 7.904 2.277
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 V1907HB-V1907HG2# 2.154 0.973
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R1918HN-R1917HN 7.848 8.577
Y1919HA-L1920HN 4.301 8.554
Y1919HA-Y1919HB# 4.302 3.339
Y1919HA-Y1919HN 4.306 8.017
Y1919HA-Y1919HB# 4.307 3.198
Y1919HA-Y1919HD# 4.303 7.066
Y1919HB#-Y1919HA 3.34 4.304
Y1919HB#-Y1919HA 3.187 4.309
Y1919HB#-L1920HN 3.344 8.559
Y1919HB#-L1920HN 3.196 8.562
Y1919HB#-Y1916HN 3.208 9.451
Y1919HB#-Y1919HN 3.344 8.025
Y1919HB#-Y1919HN 3.19 8.029
Y1919HB#-Y1919HD# 3.197 7.072
Y1919HB#-Y1919HD# 3.344 7.067
Y1919HB#-Y1919HE# 3.348 6.968
Y1919HB#-Y1919HE# 3.203 6.968
Y1919HD#-Y1919HA 7.066 4.303
Y1919HD#-Y1919HN 7.068 8.025
Y1919HD#-L1920HN 7.07 8.56
Y1919HD#-Y1919HB# 7.068 3.342
Y1919HD#-Y1919HB# 7.075 3.197
Y1919HD#-Y1919HE# 7.062 6.967
Y1919HE#-Y1919HB# 6.976 3.345
Y1919HE#-Y1919HB# 6.968 3.2
Y1919HE#-Y1919HD# 6.969 7.068
Y1919HN-Y1919HA 8.02 4.307
Y1919HN-L1920HN 8.026 8.56
Y1919HN-Y1919HB# 8.024 3.344
Y1919HN-Y1919HB# 8.028 3.189
Y1919HN-Y1919HD## 8.024 7.068
L1920HA-L1921HN 3.566 7.55
L1920HA-Y1919HD# 3.57 7.064
L1920HA-L1920HG 3.565 1.686
L1920HA-L1920HN 3.569 8.557
L1920HA-L1920HB# 3.567 1.267
L1920HA-L1920HD1# 3.581 0.856
L1920HA-L1920HD2# 3.568 0.814
L1920HB#-L1920HA 1.269 3.573
L1920HB#-L1920HG 1.267 1.69
L1920HB#-Y1919HD# 1.26 7.066
L1920HB#-Y1919HE# 1.259 6.97
L1920HB#-L1920HD1# 1.263 0.85
L1920HB#-L1920HD2# 1.263 0.816
L1920HD1#-R1917HA 0.85 3.833
L1920HD1#-L1920HA 0.856 3.577
L1920HD1#-L1920HB# 0.854 1.262
L1920HD1#-L1920HD2# 0.854 0.814
L1920HD1#-L1920HG 0.853 1.691
L1920HD2#-L1920HA 0.817 3.569
L1920HD2#-L1920HB# 0.814 1.263
L1920HD2#-L1920HD1# 0.815 0.854
L1920HD2#-L1920HG 0.815 1.69
L1920HG-R1917HA 1.688 3.838
L1920HG-L1920HA 1.688 3.571
L1920HG-L1920HB# 1.689 1.263
L1920HG-L1920HD1# 1.676 0.849
L1920HG-L1920HD2# 1.691 0.817
L1920HN-Y1919HB# 8.559 3.343
L1920HN-Y1919HB# 8.569 3.201
L1920HN-L1920HA 8.557 3.569
L1920HN-Y1919HD# 8.56 7.068
L1920HN-L1920HG 8.557 1.689
L1920HN-L1921HN 8.556 7.553
L1920HN-Y1919HN 8.558 8.025
L1920HN-L1920HB# 8.558 1.262
L1920HN-L1920HD1# 8.56 0.851
L1920HN-L1920HD2# 8.558 0.813

L1921HA-L1921HG	4.063	1.275	L1921HN-L1921HG	7.554	1.266
L1921HA-L1920HN	4.073	8.571	L1921HN-L1920HN	7.552	8.558
L1921HA-L1921HN	4.069	7.552	L1921HN-L1921HB#	7.551	1.685
L1921HA-L1921HB#	4.068	1.685	L1921HN-L1921HD1#	7.56	0.857
L1921HA-L1921HD1#	4.07	0.842	L1921HN-L1921HD2#	7.55	0.804
L1921HA-L1921HD2#	4.077	0.818	V1925HA-V1925HB	3.921	2.139
L1921HB#-L1921HA	1.679	4.067	V1925HA-V1925HN	3.93	7.754
L1921HB#-L1921HN	1.683	7.552	V1925HA-V1925HG1#	3.914	1.01
L1921HB#-L1921HG	1.689	1.263	V1925HA-V1925HG2#	3.92	0.942
L1921HB#-L1921HD1#	1.698	0.859	V1925HB-V1925HA	2.136	3.916
L1921HB#-L1921HD2#	1.687	0.815	V1925HB-V1925HN	2.13	7.761
L1921HD1#-L1921HB#	0.848	1.69	V1925HB-V1925HG1#	2.136	1.017
L1921HD1#-L1921HD2#	0.849	0.813	V1925HB-V1925HG2#	2.116	0.927
L1921HD1#-L1921HG	0.848	1.264	V1925HG1#-V1925HA	1.012	3.921
L1921HD1#-L1921HN	0.845	7.55	V1925HG1#-V1925HB	1.012	2.139
L1921HD2#-L1921HB#	0.818	1.69	V1925HG1#-V1925HG2#	1.012	0.934
L1921HD2#-L1921HD1#	0.818	0.849	V1925HG1#-V1925HN	1.011	7.761
L1921HD2#-L1921HG	0.818	1.267	V1925HG2#-V1925HA	0.94	3.921
L1921HD2#-L1921HN	0.81	7.561	V1925HG2#-V1925HB	0.936	2.139
L1921HG-R1917HA	1.27	3.837	V1925HG2#-V1925HG1#	0.936	1.008
L1921HG-L1921HN	1.266	7.552	V1925HG2#-V1925HN	0.942	7.766
L1921HG-L1921HB#	1.267	1.69	V1925HN-V1925HA	7.763	3.921
L1921HG-L1921HD1#	1.264	0.85	V1925HN-V1925HB	7.767	2.136
L1921HG-L1921HD2#	1.263	0.816	V1925HN-V1925HG1#	7.761	1.01
L1921HN-L1921HA	7.552	4.071	V1925HN-V1925HG2#	7.761	0.94

APPENDIX F

RESTRAINT FILES USED FOR APO CAM₇₆₋₁₄₈:NA_{v1.2}IQP

STRUCTURE CALCULATION

Carbon NOE Restraint File

assign (residue 99 and name HA) (residue 137 and name HA) 3.00 1.20 0.00
assign (residue 116 and name HD1#) (residue 124 and name HE#) 5.00 3.20 0.50
assign (residue 93 and name HA) (residue 100 and name HA) 5.00 3.20 0.50
assign (residue 99 and name HA) (residue 100 and name HA) 5.00 3.20 0.50
assign (residue 89 and name HA) (residue 100 and name HD#) 5.00 3.20 0.50
assign (residue 93 and name HA) (residue 100 and name HD#) 5.00 3.20 0.50
assign (residue 100 and name HD#) (residue 100 and name HA) 5.00 3.20 0.50
assign (residue 100 and name HD#) (residue 105 and name HA) 5.00 3.20 0.50
assign (residue 104 and name HA) (residue 107 and name HA) 6.00 4.20 0.60
assign (residue 105 and name HA) (residue 106 and name HA) 5.00 3.20 0.50
assign (residue 105 and name HA) (residue 108 and name HA) 5.00 3.20 0.50
assign (residue 108 and name HG##) (residue 1909 and name HA) 6.00 4.20 0.60
assign (residue 109 and name HA) (residue 1909 and name HA) 5.00 3.20 0.50
assign (residue 109 and name HE#) (residue 1909 and name HA) 3.00 1.20 0.30
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assign (residue 105 and name HA) (residue 109 and name HG#) 6.00 4.20 0.60
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assign (residue 110 and name HB) (residue 116 and name HA) 6.00 4.20 0.50
assign (residue 112 and name HA) (residue 1906 and name HA) 6.00 4.20 0.60
assign (residue 112 and name HB#) (residue 1906 and name HA) 4.00 2.20 0.40
assign (residue 112 and name HD#) (residue 1906 and name HA) 4.00 2.20 0.40
assign (residue 113 and name HA#) (residue 1906 and name HA) 6.00 4.20 0.40
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assign (residue 110 and name HA) (residue 115 and name HD#) 5.00 3.20 0.50
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assign (residue 116 and name HD1#) (residue 121 and name HA) 5.00 3.20 0.50
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assign (residue 117 and name HG#) (residue 117 and name HA) 3.00 1.20 0.30
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assign (residue 123 and name HA) (residue 124 and name HB#) 5.00 3.20 0.50
assign (residue 102 and name HA) (residue 125 and name HA) 5.00 3.20 0.50

assign (residue 102 and name HA) (residue 125 and name HD#) 4.00 2.20 0.40
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assign (residue 91 and name HG##) (residue 1908 and name HA) 5.00 3.20 0.50
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assign (residue 92 and name HD#) (residue 1909 and name HA) 5.00 3.20 0.50
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assign (residue 92 and name HE#) (residue 1909 and name HA) 5.00 3.20 0.50
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assign (residue 99 and name HB#) (residue 135 and name HA) 5.00 3.20 0.50
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assign (residue 136 and name HG2#) (residue 138 and name HB#) 5.00 3.20 0.50
assign (residue 99 and name HB#) (residue 137 and name HA) 5.00 3.20 0.50
assign (residue 137 and name HA) (residue 137 and name HB#) 4.00 2.20 0.40
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assign (residue 138 and name HB#) (residue 139 and name HA) 4.00 2.20 0.40
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assign (residue 141 and name HA) (residue 141 and name HB#) 4.00 2.20 0.40
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assign (residue 142 and name HA) (residue 145 and name HB#) 5.00 3.20 0.50
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assign (residue 145 and name HA) (residue 145 and name HB#) 4.00 2.20 0.40

assign (residue 145 and name HB#) (residue 1919 and name HB#) 5.00 3.20 0.50
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assign (residue 145 and name HG#) (residue 1919 and name HB#) 5.00 3.20 0.50
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assign (residue 77 and name HE#) (residue 77 and name HB#) 5.00 3.20 0.50
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assign (residue 85 and name HA) (residue 1915 and name HB#) 4.00 2.20 0.40
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assign (residue 85 and name HD##) (residue 1915 and name HB#) 3.00 1.20 0.30
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assign (residue 86 and name HD#) (residue 139 and name HB#) 5.00 3.20 0.50
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assign (residue 88 and name HA) (residue 1908 and name HB#) 4.00 2.20 0.40
assign (residue 88 and name HA) (residue 88 and name HB#) 3.00 1.20 0.30
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assign (residue 91 and name HG##) (residue 1908 and name HB#) 4.00 2.20 0.40
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assign (residue 94 and name HG#) (residue 94 and name HB#) 3.00 1.20 0.30
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assign (residue 124 and name HG#) (residue 1916 and name HD#) 5.00 3.20 0.50
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assign (residue 127 and name HG#) (residue 1919 and name HD#) 5.00 3.20 0.50
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assign (residue 128 and name HB#) (residue 141 and name HD#) 5.00 3.20 0.50
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assign (residue 137 and name HA) (residue 138 and name HD#) 5.00 3.20 0.50

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assign (residue 138 and name HA) (residue 141 and name HD#) 5.00 3.20 0.50
assign (residue 89 and name HD#) (residue 138 and name HB#) 5.00 3.20 0.50
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assign (residue 144 and name HE#) (residue 1919 and name HD#) 5.00 3.20 0.50
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assign (residue 145 and name HG#) (residue 1919 and name HD#) 5.00 3.20 0.50
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assign (residue 86 and name HA) (residue 138 and name HD#) 5.00 3.20 0.50
assign (residue 86 and name HD#) (residue 138 and name HD#) 5.00 3.20 0.50
assign (residue 86 and name HG1#) (residue 138 and name HD#) 5.00 3.20 0.50
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assign (residue 89 and name HA) (residue 89 and name HD#) 3.00 1.20 0.30
assign (residue 89 and name HB#) (residue 89 and name HD#) 4.00 2.20 0.40
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assign (residue 89 and name HD#) (residue 92 and name HB#) 5.00 3.20 0.40
assign (residue 89 and name HD#) (residue 92 and name HD#) 3.00 1.20 0.30
assign (residue 94 and name HA) (residue 94 and name HD#) 3.00 1.20 0.30
assign (residue 94 and name HB#) (residue 94 and name HD#) 3.00 1.20 0.30
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assign (residue 100 and name HA) (residue 136 and name HD1#) 5.00 3.20 0.50
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assign (residue 100 and name HB) (residue 136 and name HD1#) 5.00 3.20 0.50
assign (residue 100 and name HD1#) (residue 1912 and name HD1#) 5.00 3.20 0.50

assign (residue 125 and name HB) (residue 136 and name HD1#) 5.00 3.20 0.50
assign (residue 125 and name HG1#) (residue 125 and name HD1#) 3.00 1.20 0.30
assign (residue 125 and name HG1#) (residue 136 and name HD1#) 4.00 2.20 0.40
assign (residue 125 and name HG2#) (residue 125 and name HD1#) 3.00 1.20 0.30
assign (residue 125 and name HG2#) (residue 136 and name HD1#) 4.00 2.20 0.40
assign (residue 129 and name HA) (residue 136 and name HD1#) 6.00 4.20 0.60
assign (residue 129 and name HB#) (residue 136 and name HD1#) 5.00 3.20 0.50
assign (residue 130 and name HA) (residue 130 and name HD1#) 5.00 3.20 0.50
assign (residue 130 and name HB) (residue 130 and name HD1#) 3.00 1.20 0.30
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assign (residue 130 and name HG1#) (residue 130 and name HD1#) 4.00 2.20 0.40
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assign (residue 136 and name HG1#) (residue 136 and name HD1#) 3.00 1.20 0.30
assign (residue 136 and name HG2#) (residue 136 and name HD1#) 3.00 1.20 0.30
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assign (residue 141 and name HE#) (residue 1912 and name HD1#) 5.00 3.20 0.50
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assign (residue 83 and name HB#) (residue 86 and name HD1#) 5.00 3.20 0.50
assign (residue 83 and name HG#) (residue 86 and name HD1#) 4.00 2.20 0.40
assign (residue 85 and name HA) (residue 1912 and name HD1#) 5.00 3.20 0.50
assign (residue 85 and name HB#) (residue 1912 and name HD1#) 5.00 3.20 0.50
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assign (residue 86 and name HA) (residue 86 and name HD1#) 4.00 2.20 0.40
assign (residue 86 and name HG1#) (residue 86 and name HD1#) 3.00 1.20 0.30
assign (residue 86 and name HG2#) (residue 86 and name HD1#) 3.00 1.20 0.30
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assign (residue 89 and name HB#) (residue 1912 and name HD1#) 5.00 3.20 0.50
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assign (residue 92 and name HE#) (residue 1912 and name HD1#) 5.00 3.20 0.50
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assign (residue 105 and name HA) (residue 105 and name HD2#) 4.00 2.20 0.40
assign (residue 105 and name HB#) (residue 105 and name HD2#) 3.00 1.20 0.30
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assign (residue 109 and name HE#) (residue 116 and name HD2#) 6.00 4.20 0.60
assign (residue 105 and name HD2#) (residue 109 and name HG#) 5.00 3.20 0.50
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assign (residue 110 and name HB) (residue 116 and name HD2#) 5.00 3.20 0.50
assign (residue 110 and name HG2#) (residue 116 and name HD2#) 3.00 1.20 0.30
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assign (residue 116 and name HA) (residue 116 and name HD2#) 4.00 2.20 0.40
assign (residue 116 and name HB#) (residue 116 and name HD2#) 4.00 2.20 0.40

assign (residue 116 and name HD1#) (residue 116 and name HD2#) 3.00 1.20 0.30
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assign (residue 116 and name HD2#) (residue 121 and name HG##) 3.00 1.20 0.30
assign (residue 123 and name HA) (residue 1920 and name HD##) 5.00 3.20 0.50
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assign (residue 105 and name HD2#) (residue 125 and name HB) 5.00 3.20 0.50
assign (residue 105 and name HD2#) (residue 125 and name HG1#) 5.00 3.20 0.50
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assign (residue 127 and name HG#) (residue 1920 and name HD##) 6.00 4.20 0.60
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assign (residue 81 and name HA) (residue 85 and name HD##) 5.00 3.20 0.50
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Nitrogen NOE Restraint File

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assign (residue 99 and name HD##) (residue 100 and name HN) 5.00 3.20 0.50
assign (residue 99 and name HN) (residue 100 and name HN) 5.00 3.20 0.50
assign (residue 100 and name HA) (residue 100 and name HN) 4.00 2.20 0.40
assign (residue 100 and name HB) (residue 100 and name HN) 4.00 2.20 0.40
assign (residue 100 and name HG2#) (residue 100 and name HN) 4.00 2.20 0.40
assign (residue 99 and name HN) (residue 137 and name HA) 5.00 3.20 0.50
assign (residue 98 and name HA#) (residue 99 and name HN) 5.00 3.20 0.50
assign (residue 99 and name HA) (residue 99 and name HN) 4.00 2.20 0.40
assign (residue 99 and name HB#) (residue 99 and name HN) 4.00 2.20 0.40
assign (residue 99 and name HD##) (residue 99 and name HN) 4.00 2.20 0.40
assign (residue 99 and name HG) (residue 99 and name HN) 4.00 2.20 0.40
assign (residue 98 and name HN) (residue 138 and name HE#) 5.00 3.20 0.50
assign (residue 96 and name HA#) (residue 98 and name HN) 5.00 3.20 0.50
assign (residue 97 and name HA) (residue 98 and name HN) 5.00 3.20 0.50
assign (residue 97 and name HB#) (residue 98 and name HN) 5.00 3.20 0.50
assign (residue 98 and name HN) (residue 99 and name HA) 5.00 3.20 0.50
assign (residue 98 and name HN) (residue 99 and name HB#) 5.00 3.20 0.50
assign (residue 98 and name HN) (residue 99 and name HD##) 5.00 3.20 0.50
assign (residue 98 and name HA#) (residue 98 and name HN) 4.00 2.20 0.40
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assign (residue 97 and name HN) (residue 98 and name HA#) 5.00 3.20 0.50
assign (residue 97 and name HN) (residue 99 and name HN) 5.00 3.20 0.50
assign (residue 96 and name HA#) (residue 97 and name HN) 4.00 2.20 0.40
assign (residue 97 and name HA) (residue 97 and name HN) 4.00 2.20 0.40

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 assign (residue 89 and name HD#) (residue 92 and name HN) 4.00 2.20 0.40
 assign (residue 92 and name HB#) (residue 92 and name HN) 4.00 2.20 0.40
 assign (residue 92 and name HA) (residue 92 and name HN) 3.00 1.20 0.30
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 assign (residue 88 and name HA) (residue 91 and name HN) 5.00 3.20 0.50
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 assign (residue 90 and name HG#) (residue 91 and name HN) 4.00 2.20 0.40
 assign (residue 91 and name HA) (residue 91 and name HN) 3.00 1.20 0.30
 assign (residue 91 and name HB) (residue 91 and name HN) 3.00 1.20 0.30
 assign (residue 91 and name HG##) (residue 91 and name HN) 4.00 2.20 0.40
 assign (residue 91 and name HN) (residue 92 and name HN) 3.00 1.20 0.30
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 assign (residue 88 and name HN) (residue 90 and name HN) 5.00 3.20 0.50
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 assign (residue 90 and name HE#) (residue 90 and name HN) 5.00 3.20 0.50
 assign (residue 90 and name HN) (residue 91 and name HA) 5.00 3.20 0.50
 assign (residue 90 and name HN) (residue 91 and name HG##) 4.00 2.20 0.40
 assign (residue 90 and name HN) (residue 92 and name HN) 5.00 3.20 0.50
 assign (residue 89 and name HB#) (residue 90 and name HN) 4.00 2.20 0.40
 assign (residue 90 and name HG#) (residue 90 and name HN) 4.00 2.20 0.40
 assign (residue 90 and name HN) (residue 91 and name HB) 5.00 3.20 0.50
 assign (residue 89 and name HA) (residue 90 and name HN) 4.00 2.20 0.30
 assign (residue 90 and name HA) (residue 90 and name HN) 3.00 1.20 0.30
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 assign (residue 89 and name HN) (residue 138 and name HD#) 5.00 3.20 0.50
 assign (residue 86 and name HN) (residue 89 and name HN) 5.00 3.20 0.50
 assign (residue 89 and name HN) (residue 91 and name HG##) 6.00 4.20 0.60
 assign (residue 89 and name HN) (residue 91 and name HN) 5.00 3.20 0.50
 assign (residue 89 and name HN) (residue 92 and name HN) 5.00 3.20 0.50
 assign (residue 88 and name HA) (residue 89 and name HN) 4.00 2.20 0.40
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 assign (residue 89 and name HA) (residue 89 and name HN) 3.00 1.20 0.30
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assign (residue 87 and name HN) (residue 88 and name HB#) 5.00 3.20 0.50
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 assign (residue 87 and name HN) (residue 88 and name HN) 3.00 1.20 0.30
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 assign (residue 86 and name HN) (residue 89 and name HD#) 5.00 3.20 0.50
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 assign (residue 86 and name HG1#) (residue 86 and name HN) 4.00 2.20 0.40
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 assign (residue 84 and name HG#) (residue 85 and name HN) 4.00 2.20 0.40
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 assign (residue 84 and name HN) (residue 85 and name HD##) 5.00 3.20 0.50
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 assign (residue 84 and name HB#) (residue 84 and name HN) 3.00 1.20 0.30
 assign (residue 84 and name HG#) (residue 84 and name HN) 3.00 1.20 0.30
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 assign (residue 83 and name HG#) (residue 83 and name HN) 4.00 2.20 0.40
 assign (residue 83 and name HN) (residue 84 and name HN) 4.00 2.20 0.40
 assign (residue 83 and name HA) (residue 83 and name HN) 3.00 1.20 0.30
 assign (residue 83 and name HB#) (residue 83 and name HN) 3.00 1.20 0.30
 assign (residue 82 and name HN) (residue 142 and name HG##) 5.00 3.20 0.50
 assign (residue 80 and name HB#) (residue 82 and name HN) 5.00 3.20 0.50
 assign (residue 81 and name HB#) (residue 82 and name HN) 5.00 3.20 0.50
 assign (residue 82 and name HN) (residue 146 and name HB) 5.00 3.20 0.50
 assign (residue 82 and name HN) (residue 84 and name HN) 5.00 3.20 0.50
 assign (residue 82 and name HN) (residue 83 and name HN) 4.00 2.20 0.40
 assign (residue 82 and name HA) (residue 82 and name HN) 3.00 1.20 0.30
 assign (residue 82 and name HN) (residue 83 and name HB#) 5.00 3.20 0.50
 assign (residue 81 and name HA) (residue 81 and name HN) 4.00 2.20 0.40

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assign ( residue 81 and name HN ) ( residue 82 and name HN ) 4.00 2.20 0.40
assign ( residue 81 and name HN ) ( residue 83 and name HB# ) 6.00 4.20 0.60
assign ( residue 80 and name HB# ) ( residue 81 and name HN ) 3.00 1.20 0.30
assign ( residue 81 and name HB# ) ( residue 81 and name HN ) 3.00 1.20 0.30
assign ( residue 80 and name HA ) ( residue 80 and name HN ) 4.00 2.20 0.40
assign ( residue 79 and name HA ) ( residue 80 and name HN ) 3.00 1.20 0.30
assign ( residue 80 and name HB# ) ( residue 80 and name HN ) 3.00 1.20 0.30
assign ( residue 79 and name HA ) ( residue 79 and name HN ) 4.00 2.20 0.40
assign ( residue 79 and name HB# ) ( residue 79 and name HN ) 4.00 2.20 0.40
assign ( residue 79 and name HG# ) ( residue 79 and name HN ) 5.00 3.20 0.40
assign ( residue 78 and name HA ) ( residue 79 and name HN ) 3.00 1.20 0.30
assign ( residue 77 and name HN ) ( residue 78 and name HN ) 5.00 3.20 0.50
assign ( residue 78 and name HA ) ( residue 78 and name HN ) 5.00 3.20 0.50
assign ( residue 78 and name HN ) ( residue 79 and name HG# ) 5.00 3.20 0.50
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assign ( residue 146 and name HB ) ( residue 148 and name HN ) 5.00 3.20 0.40
assign ( residue 140 and name HN ) ( residue 142 and name HB ) 6.00 4.20 0.60
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assign ( residue 81 and name HN ) ( residue 84 and name HG# ) 5.00 3.20 0.50
assign ( residue 82 and name HN ) ( residue 85 and name HN ) 5.00 3.20 0.50

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Hydrogen Bonding Restraint File

Hydrogen bonding restraint file derived from amide exchange data

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assign ( residue 114 and name N ) ( residue 1913 and name OE1 ) 2.90 0.20 0.40
assign ( residue 114 and name HN ) ( residue 1913 and name OE1 ) 1.80 0.00 0.60
assign ( residue 83 and name HN ) ( residue 79 and name O ) 1.80 0.00 0.60
assign ( residue 86 and name HN ) ( residue 82 and name O ) 1.80 0.00 0.60
assign ( residue 87 and name HN ) ( residue 83 and name O ) 1.80 0.00 0.60
assign ( residue 88 and name HN ) ( residue 84 and name O ) 1.80 0.00 0.60
assign ( residue 89 and name HN ) ( residue 85 and name O ) 1.80 0.00 0.60
assign ( residue 90 and name HN ) ( residue 86 and name O ) 1.80 0.00 0.60
assign ( residue 91 and name HN ) ( residue 87 and name O ) 1.80 0.00 0.60
assign ( residue 105 and name HN ) ( residue 101 and name O ) 1.80 0.00 0.60
assign ( residue 106 and name HN ) ( residue 102 and name O ) 1.80 0.00 0.60
assign ( residue 108 and name HN ) ( residue 104 and name O ) 1.80 0.00 0.60
assign ( residue 109 and name HN ) ( residue 105 and name O ) 1.80 0.00 0.60
assign ( residue 110 and name HN ) ( residue 106 and name O ) 1.80 0.00 0.60
assign ( residue 111 and name HN ) ( residue 107 and name O ) 1.80 0.00 0.60
assign ( residue 112 and name HN ) ( residue 108 and name O ) 1.80 0.00 0.60
assign ( residue 116 and name HN ) ( residue 112 and name O ) 1.80 0.00 0.60
assign ( residue 121 and name HN ) ( residue 117 and name O ) 1.80 0.00 0.60
assign ( residue 122 and name HN ) ( residue 118 and name O ) 1.80 0.00 0.60

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assign (residue 123 and name HN) (residue 119 and name O) 1.80 0.00 0.60
assign (residue 124 and name HN) (residue 120 and name O) 1.80 0.00 0.60
assign (residue 125 and name HN) (residue 121 and name O) 1.80 0.00 0.60
assign (residue 126 and name HN) (residue 122 and name O) 1.80 0.00 0.60
assign (residue 128 and name HN) (residue 124 and name O) 1.80 0.00 0.60
assign (residue 138 and name HN) (residue 98 and name O) 1.80 0.00 0.60
assign (residue 142 and name HN) (residue 138 and name O) 1.80 0.00 0.60
assign (residue 143 and name HN) (residue 139 and name O) 1.80 0.00 0.60
assign (residue 144 and name HN) (residue 140 and name O) 1.80 0.00 0.60
assign (residue 145 and name HN) (residue 141 and name O) 1.80 0.00 0.60
assign (residue 146 and name HN) (residue 142 and name O) 1.80 0.00 0.60
assign (residue 100 and name HN) (residue 136 and name O) 1.80 0.00 0.60
assign (residue 136 and name HN) (residue 100 and name O) 1.80 0.00 0.60
assign (residue 1913 and name HE2#) (residue 112 and name O) 1.80 0.00 0.60
assign (residue 114 and name HN) (residue 1913 and name OE1) 1.80 0.00 0.60
assign (residue 83 and name N) (residue 79 and name O) 2.90 0.20 0.40
assign (residue 86 and name N) (residue 82 and name O) 2.90 0.20 0.40
assign (residue 87 and name N) (residue 83 and name O) 2.90 0.20 0.40
assign (residue 88 and name N) (residue 84 and name O) 2.90 0.20 0.40
assign (residue 89 and name N) (residue 85 and name O) 2.90 0.20 0.40
assign (residue 90 and name N) (residue 86 and name O) 2.90 0.20 0.40
assign (residue 91 and name N) (residue 87 and name O) 2.90 0.20 0.40
assign (residue 105 and name N) (residue 101 and name O) 2.90 0.20 0.40
assign (residue 106 and name N) (residue 102 and name O) 2.90 0.20 0.40
assign (residue 108 and name N) (residue 104 and name O) 2.90 0.20 0.40
assign (residue 109 and name N) (residue 105 and name O) 2.90 0.20 0.40
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assign (residue 126 and name N) (residue 122 and name O) 2.90 0.20 0.40
assign (residue 128 and name N) (residue 124 and name O) 2.90 0.20 0.40
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assign (residue 143 and name N) (residue 139 and name O) 2.90 0.20 0.40
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assign (residue 136 and name N) (residue 100 and name O) 2.90 0.20 0.40
assign (residue 1913 and name NE2) (residue 112 and name O) 2.90 0.20 0.40
assign (residue 114 and name N) (residue 1913 and name OE1) 2.90 0.20 0.40

assign (residue 1908 and name HN) (residue 1904 and name O) 1.80 0.00 0.60
 assign (residue 1909 and name HN) (residue 1905 and name O) 1.80 0.00 0.60
 assign (residue 1910 and name HN) (residue 1906 and name O) 1.80 0.00 0.60
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 assign (residue 1909 and name N) (residue 1905 and name O) 2.90 0.20 0.40
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 assign (residue 1914 and name N) (residue 1910 and name O) 2.90 0.20 0.40
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 assign (residue 1918 and name N) (residue 1914 and name O) 2.90 0.20 0.40
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 assign (residue 1921 and name N) (residue 1917 and name O) 2.90 0.20 0.40

Nav1.2_{IQD} Intramolecular Restraint File

assign (residue 1904 and name HA) (residue 1904 and name HN) 5.00 3.20 0.50
 assign (residue 1904 and name HA) (residue 1905 and name HN) 5.00 3.20 0.50
 assign (residue 1904 and name HA) (residue 1907 and name HN) 4.00 2.20 0.40
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assign (residue 1908 and name HN) (residue 1909 and name HN) 3.00 1.20 0.30
assign (residue 1909 and name HN) (residue 1909 and name HA) 4.00 2.20 0.40
assign (residue 1909 and name HN) (residue 1909 and name HB#) 3.00 1.20 0.30
assign (residue 1909 and name HN) (residue 1910 and name HN) 4.00 2.20 0.40
assign (residue 1910 and name HB) (residue 1910 and name HA) 4.00 2.20 0.40
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assign (residue 1910 and name HG1#) (residue 1910 and name HB) 3.00 1.20 0.30
assign (residue 1909 and name HB#) (residue 1910 and name HN) 4.00 2.20 0.40
assign (residue 1910 and name HN) (residue 1910 and name HA) 5.00 3.20 0.50
assign (residue 1910 and name HN) (residue 1910 and name HB) 5.00 3.20 0.50

assign (residue 1918 and name HN) (residue 1918 and name HD#) 5.00 3.20 0.50
assign (residue 1918 and name HN) (residue 1918 and name HG#) 5.00 3.20 0.50
assign (residue 1919 and name HA) (residue 1920 and name HN) 5.00 3.20 0.50
assign (residue 1916 and name HN) (residue 1919 and name HB#) 4.00 2.20 0.40
assign (residue 1919 and name HB#) (residue 1919 and name HA) 5.00 3.20 0.50
assign (residue 1919 and name HD#) (residue 1919 and name HA) 5.00 3.20 0.50
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assign (residue 1919 and name HD#) (residue 1919 and name HN) 5.00 3.20 0.50
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assign (residue 1919 and name HN) (residue 1919 and name HA) 5.00 3.20 0.50
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assign (residue 1919 and name HD#) (residue 1920 and name HA) 5.00 3.20 0.50
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assign (residue 1919 and name HD#) (residue 1920 and name HB#) 4.00 2.20 0.40
assign (residue 1919 and name HE#) (residue 1920 and name HB#) 5.00 3.20 0.50
assign (residue 1920 and name HB#) (residue 1920 and name HA) 4.00 2.20 0.40
assign (residue 1917 and name HA) (residue 1920 and name HD##) 5.00 3.20 0.50
assign (residue 1920 and name HD##) (residue 1920 and name HA) 3.00 1.20 0.30
assign (residue 1920 and name HD##) (residue 1920 and name HB#) 3.00 1.20 0.30
assign (residue 1917 and name HA) (residue 1920 and name HG) 6.00 4.20 0.50
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assign (residue 1920 and name HG) (residue 1920 and name HD##) 4.00 2.20 0.40
assign (residue 1919 and name HB#) (residue 1920 and name HN) 5.00 3.20 0.50
assign (residue 1919 and name HD#) (residue 1920 and name HN) 5.00 3.20 0.50
assign (residue 1919 and name HN) (residue 1920 and name HN) 4.00 2.20 0.40
assign (residue 1920 and name HN) (residue 1920 and name HA) 5.00 3.20 0.50
assign (residue 1920 and name HN) (residue 1920 and name HB#) 4.00 2.20 0.40
assign (residue 1920 and name HN) (residue 1920 and name HD##) 5.00 3.20 0.50
assign (residue 1920 and name HN) (residue 1920 and name HG) 4.00 2.20 0.40
assign (residue 1920 and name HN) (residue 1921 and name HA) 5.00 3.20 0.50
assign (residue 1921 and name HA) (residue 1921 and name HD##) 5.00 3.20 0.50
assign (residue 1921 and name HA) (residue 1921 and name HG) 5.00 3.20 0.50
assign (residue 1921 and name HB#) (residue 1921 and name HA) 4.00 2.20 0.40
assign (residue 1921 and name HD##) (residue 1921 and name HB#) 4.00 2.20 0.40
assign (residue 1917 and name HA) (residue 1921 and name HG) 5.00 3.20 0.50
assign (residue 1921 and name HG) (residue 1921 and name HB#) 3.00 1.20 0.30
assign (residue 1921 and name HG) (residue 1921 and name HD##) 3.00 1.20 0.30
assign (residue 1920 and name HN) (residue 1921 and name HN) 4.00 2.20 0.40
assign (residue 1921 and name HN) (residue 1921 and name HA) 5.00 3.20 0.50
assign (residue 1921 and name HN) (residue 1921 and name HB#) 3.00 1.20 0.30
assign (residue 1921 and name HN) (residue 1921 and name HD##) 5.00 3.20 0.50
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assign (residue 1925 and name HB) (residue 1925 and name HA) 4.00 2.20 0.40

TALOS Dihedral Angle Restraint File

Values derived using chemical shift index

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1.0 -59.000 14 2

assign (resid 82 and name N)
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 (resid 83 and name N)
1.0 -39.000 18 2

assign (resid 82 and name C)
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 (resid 83 and name CA)
 (resid 83 and name C)
1.0 -61.000 10 2

assign (resid 83 and name N)
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 (resid 83 and name C)
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1.0 -42.000 12 2

assign (resid 83 and name C)
 (resid 84 and name N)
 (resid 84 and name CA)
 (resid 84 and name C)
1.0 -64.000 12 2

assign (resid 84 and name N)
 (resid 84 and name CA)
 (resid 84 and name C)
 (resid 85 and name N)
1.0 -42.000 8 2

assign (resid 84 and name C)
 (resid 85 and name N)
 (resid 85 and name CA)
 (resid 85 and name C)
1.0 -63.000 12 2

assign (resid 85 and name N)

(resid 85 and name CA)
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(resid 86 and name N)
1.0 -45.000 12 2

assign (resid 85 and name C)
(resid 86 and name N)
(resid 86 and name CA)
(resid 86 and name C)
1.0 -61.000 8 2

assign (resid 86 and name N)
(resid 86 and name CA)
(resid 86 and name C)
(resid 87 and name N)
1.0 -44.000 14 2

assign (resid 86 and name C)
(resid 87 and name N)
(resid 87 and name CA)
(resid 87 and name C)
1.0 -61.000 12 2

assign (resid 87 and name N)
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(resid 87 and name C)
(resid 88 and name N)
1.0 -43.000 12 2

assign (resid 87 and name C)
(resid 88 and name N)
(resid 88 and name CA)
(resid 88 and name C)
1.0 -67.000 6 2

assign (resid 88 and name N)
(resid 88 and name CA)
(resid 88 and name C)
(resid 89 and name N)
1.0 -41.000 18 2

assign (resid 88 and name C)
(resid 89 and name N)
(resid 89 and name CA)
(resid 89 and name C)
1.0 -65.000 8 2

assign (resid 89 and name N)
 (resid 89 and name CA)
 (resid 89 and name C)
 (resid 90 and name N)
 1.0 -43.000 6 2

assign (resid 89 and name C)
 (resid 90 and name N)
 (resid 90 and name CA)
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(resid 106 and name N)
1.0 -40.000 16 2

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assign (resid 122 and name N)

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 1.0 -16.000 34 2

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 1.0 136.000 54 2

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1.0 -110.000 64 2

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(resid 136 and name C)
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1.0 127.000 42 2

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(resid 137 and name C)
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1.0 110.000 36 2

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assign (resid 140 and name C)

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1.00 -65.00 20.00 2

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1.00 -65.00 20.00 2

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1.00 -65.00 20.00 2

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1.00 -65.00 20.00 2

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1.00 -65.00 20.00 2

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1.00 -65.00 20.00 2

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1.00 -65.00 20.00 2

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1.00 -65.00 20.00 2

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(residue 1917 and name CA)
(residue 1917 and name C)
1.00 -65.00 20.00 2

assign
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(residue 1918 and name CA)
(residue 1918 and name C)
1.00 -65.00 20.00 2

assign
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(residue 1919 and name CA)
(residue 1919 and name C)
1.00 -65.00 20.00 2

assign

(residue 1919 and name C)
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(residue 1920 and name CA)
(residue 1920 and name C)
1.00 -65.00 20.00 2

assign

(residue 1920 and name C)
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(residue 1921 and name CA)
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1.00 -65.00 20.00 2

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